# PROGRAMME PROJECT REPORT Bachelor of Science Programme (3 Year Programme in accordance with NEP-2020)





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#### 1. Bachelor's Degree Programme

The National Education Policy (NEP) 2020 envisions a new vision that enable an individual to study one or more specialized areas of interest at a deep level, and also develop capabilities across a range of disciplines including sciences, social sciences, arts, humanities, languages, as well as professional, technical, and vocational subjects. The NEP 2020 focuses on the formulation of expected learning outcomes for all higher education programmes. It states that "National Higher Education Qualifications Framework (NHEQF)" shall be align with the National Skills Qualifications Framework (NSQF) to ease the integration of vocational education into higher education. It also points out that higher educationqualifications leading to a degree/diploma/certificate shall be described by the NHEQF in terms of Outcome Based Education (OBE).

The design of B.Sc. under UGC Choice Based Credit System (CBCS) programme in line with NHEQF offers opportunities and avenues to learn core subjects but also to explore additional avenues of learning beyond the core subjects for holistic development of a learner.

Programme:	Bachelor of Science [B.Sc.]				
Year	First Introduction year: 2002				
	Revision of Programme in accordance with NEP-2020				
Init	Initiation year of revision 2022				
Completion year of revision		2023			

The salient advantages of the choice-based credit system are as follows:

- CBCS allows learner to choose inter-disciplinary, intra-disciplinary courses, skill-oriented courses (even from other disciplines according to their learning needs, interests and aptitude) and have more flexibility.
- CBCS offers flexibility for learner to study at different times and at different institutions to complete one course (ease of mobility of learner). Credits earned at one institution can be easily transferred to other universities.
- Learner may undertake as many credits as they can cope with without repeating all the courses in a given semester if they fail in one/more courses.
- Shift in focus from the teacher-centric to learner-centric education.

The uniform grading system will also enable potential employers in assessing the performance of the learner. In order to bring uniformity in evaluation system and computation of the Cumulative Grade Point Average (CGPA) based on learner's performance in examinations, guidelines framed by the UGC are followed. Hence, adoption of NHEQF helps to overcome the gap between university degree and employability by introducing skills and competencies in the graduates.

### 2. B.Sc. Programme

The structure and duration of undergraduate programme of Bachelor of Science in accordance with NEP 2020 includes multiple exit options within this period, with appropriate certifications:

- Level 5: a **certificate** after completing 1 year (2 semesters) of study in the chosen discipline or field, including vocational and professional areas;
- Level 6: a **diploma** after 2 years (4 semesters) of study;
- Level 7: a **Bachelor's** degree after a 3-year (6 semesters) programme.

### 2.1 Programme Mission & Objectives

In line with the mission of the University to provide flexible learning opportunities to all, particularly to those who could not join regular colleges or universities owing to social, economic and other constraints, the 3-year Undergraduate Programme in Science, B.Sc. aims at providing holistic and value based knowledge and guidance to promote scientific temper in everyday life. The program offers a platform to the learners to fulfill the eligible criteria in various scientific jobs in government and private sector.

The Programme aims at the following objectives:

- 1. To provide a sound academic base from which an advanced career in science can be developed.
- 2. To provide basic understanding about science among learners.
- **3**. To develop academically competent and professionally motivated personnel, equipped with objective, critical thinking, right moral and ethical values that compassionately foster scientific temper with a sense of social responsibility.
- 4. To enable learner to become globally competent.
- 5. To inculcate entrepreneurial skills among learners.

#### 2.2 Relevance of the Programme with Mission and Goals

The 3-year Undergraduate Programme in Science, B.Sc. is designed with the objective of equipping learners to cope with the emerging trends and challenges in the scientific domain. In congruence with goals of the University the Programme also focuses to provide skilled manpower to the society to meet global demands. The Programme is designed with three major subjects so that a successful learner can go for higher studies in any one of the major subjects of his/ her choice. The Programme also aims at making the learners fit for taking up various jobs.

#### 2.3 Nature of Prospective Target Group of Learners

The Program is targeted to all individuals looking to earn a graduation degree for employment, further higher education, promotion in career and professional development.

# 2.4 Appropriateness of Programme to be conducted in ODL mode to acquire specific skills & competence

	Learning outcomes after Level 5					
Learning	Elements of the	Level 5 (Undergraduate Certificate)				
Outcomes	descriptor					
LO 1	Knowledge and	• knowledge of facts, concepts, principles, theories, and processes				
	understanding	in multidisciplinary areas in science.				
		• understanding of the linkages between various disciplines.				
LO 2	Skills required to	• acquire cognitive and technical skills for selecting and using				
	perform and	relevant methods, tools, and materials to assess the				
	accomplish tasks	appropriateness of approaches to solving problems associated with				
		the science.				
LO 3	Application of	• apply the acquired knowledge, and a range of cognitive and				
	knowledge and	practical skills to select and use basic methods, tools, materials, and				
	skills	information to generate solution s to specific problems relating to				
		the science.				

LO 4	Generic learning	• listen carefully, read texts related to the science analytically and
	outcomes	present information in a clear and concise manner.
		• express thoughts and ideas effectively in writing and orally and
		present the results/findings of the experiments carried out in a
		clear and concise manner to different groups.
		<ul> <li>meet own learning needs relating to the science.</li> </ul>
		• pursue self-directed and self-managed learning to upgrade
		knowledge and skills required to pursue higher level of education
		and training.
		• gather and interpret relevant quantitative and qualitative data to
		identify problems,
		• critically evaluate the essential theories, policies, and practices by
		following scientific approach to knowledge development and take
		actions to generate solutions to specific problems associated with
		the science.
		• make judgement and take decision, based on analysis of data and
		evidence, for formulating responses to issues/problems associated
		with the science.
LO 5	Constitutional,	• embrace constitutional, humanistic, ethical, and moral values and
	humanistic,	practice these values in real-life situations.
	ethical and moral	
	values	
LO 6	Employment	<ul> <li>perform effectively in a defined job relating to the science.</li> </ul>
	ready skills, and	• ability to exercise responsibility for the completion of assigned
	entrepreneurship	tasks.
	skills and mindset	

	Learning outcomes after Level 6					
Learning Outcomes	Elements of the descriptor	Level 6 (Undergraduate Diploma)				
LO 1	Knowledge and understanding	<ul> <li>theoretic al and technical knowledge in multidisciplinary contexts,</li> <li>deeper knowledge and understanding of one of the learning areas and its underlying principles and theories,</li> <li>procedural knowledge required for performing skilled or paraprofessional tasks associated with the chosen fields of learning.</li> </ul>				
LO 2	Skills required to perform and accomplish tasks	<ul> <li>cognitive and technical skills required for performing and accomplishing complex tasks relating to the chosen fields of learning.</li> <li>cognitive and technical skills required to analyse and synthesize ideas and information from a range of sources.</li> <li>act on information to generate solutions to specific problem s associated with the chosen fields of learning.</li> </ul>				
LO 3	Application of	• apply the acquired specialized or theoretical knowledge, and a				

	knowledge and skills	range of cognitive and practical skills to gather quantitative and qualitative data,
LO 4	Generic learning outcomes	• listen carefully, read texts analytically and present complex information in a clear and concise manner,
		• communicate in writing and orally the information, arguments,
		and results of the experiments and studies conducted accurately and effectively.
		• critically evaluate the essential theories, policies, and practices by following scientific approach to knowledge development.
		• make judgement and take decision, based on the analysis and
		evaluation of information, for determining solutions to a variety of
		unpredictable problems.
LO 5	Constitutional,	• embrace constitutional, humanistic, ethical, and moral values,
	humanistic,	and practice these values in life,
	ethical and moral	
	values	
LO 6	Employment	• take up job/employment or professional practice requiring the
	ready skills, and	exercise of full personal responsibility for the completion of tasks
	entrepreneurship	and for the outputs of own work.
	skills and mindset	• exercise self- management within the guidelines of study and work contexts.
		• supervise the routine work of others, taking some responsibility
		for the evaluation and improvement of work or study activities.

Learning outcomes after Level 7					
Learning	Level 7 (Bachelor in Science)				
Outcomes	descriptor				
LO 1	Knowledge and understanding	<ul> <li>comprehensive, factual, theoretical, and specialized knowledge in multidisciplinary contexts with depth in the underlying principles and theories.</li> <li>knowledge of the current and emerging issues and developments.</li> </ul>			
LO 2	Skills required to perform and accomplish tasks	<ul> <li>cognitive and technical skills required for performing and accomplishing complex tasks to evaluate and analyse complex ideas.</li> <li>cognitive and technical skills required to generate solutions to specific problems.</li> </ul>			
LO 3	Application of knowledge and skills	• apply the acquired specialized technical or theoretic alknowledge, and cognitive and practical skills to gather and analyse quantitative/ qualitative data to assess the appropriateness of different approaches to solving problems,			
LO 4	Generic learning outcomes	<ul> <li>listen carefully, to read text related to the chosen fields of learning analytically and present complex information in a clear and concise manner to different groups/audiences.</li> <li>communicate in writing and orally the constructs and methodologies adopted for the studies undertaken relating to the chosen fields of learning.</li> </ul>			

		<ul> <li>critically evaluate evidence for taking actions to generate solutions to specific problems based on empirical evidence.</li> <li>make judgement and take decisions based on the analysis and evaluation of information for formulating responses to problems, including real-life problems,</li> </ul>
LO 5	Constitutional, humanistic, ethical and moral values	<ul> <li>embrace the constitutional, humanistic, ethical, and moral values, and practice these values in life.</li> <li>identify ethical issues in science,</li> <li>formulate coherent arguments about ethical and moral issues, including environmental and sustainable development issues.</li> <li>follow ethical practices in all aspects of research and development</li> </ul>
LO 6	Employment ready skills, and entrepreneurship skills and mindset	<ul> <li>knowledge and essential skills set and competence that are necessary to: take up a professional job and professional practice,</li> <li>entrepreneurship skills and mindset required for setting up and running an economic enterprise or pursuing self-employment</li> <li>exercise management and supervision in the contexts of work or study activities involving unpredictable work processes and working environment</li> </ul>

### **2.5 Instructional Design**

#### 2.5.1 3-year B.Sc. Programme Structure

The University follows the credit system in all its programmes. One credit is equal to 30 hours of learner's study time which is equivalent to 15 lectures in conventional system. To earn a Bachelor's Degree, a learner has to earn 120 credits in minimum six semesters (three years) with 20 credits per semester. For earning 120 credits, a learner has to opt from the following categories of courses:

- (a) Discipline Specific Core Courses
- (b) Discipline Specific Electives Courses (DEC)
- (c) Ability Enhancement Compulsory Courses (AECC)
- (d) Skill Enhancement Courses (SEC)

#### Programme Structure of B.Sc. Programme under NHEQF

Level	Year	Sem	First Selected Subject Discipline SpecificCore papers with credit	Second Selected Subject Discipline SpecificCore papers with credit	Third Selected Subject Discipline SpecificCore papers with credit	Ability Enhancement Compulsory Course (AECC)	Skill Enhancement Course (SEC)	Discipline Specific Elective Course (DEC)	Literature Survey/ Research Project	Total credit
5	1	1 <sup>st</sup>	4	4	4	4	4	-	-	20
		2 <sup>nd</sup>	4	4	4	4	4	-	-	20
6	2	3 <sup>rd</sup>	4	4	4	4	4	-	-	20
		4 <sup>th</sup>	4	4	4	4	4	-	-	20
7	3	5 <sup>th</sup>	-	-	-	-	4	12	4	20
		6 <sup>th</sup>	-	-	-	-	4	12	4	20
Total o	redit		16	16	16	16	24	24	8	120

#### Explanation of terms used for categorization of courses:

- A. **Discipline Specific Core Courses:** A course, which should compulsorily be studied by a learner as a core requirement is termed as a Core course.
- B. Elective Course (DE): Generally, a course which can be chosen from a pool of courses and which may be very specific or specialized or advanced or supportive to the discipline/ subject of study or which provides an extended scope or which enables an exposure to some other discipline/subject/domain or nurtures the candidate's proficiency/skill is called an Elective Course. The Elective course may be offered in following types:
  - a) **Discipline Specific Elective Course (DCE):** Elective courses may be offered by the main discipline/subject of study is referred to as Discipline Specific Elective.
  - b) Industrial Training/ Survey/ Research Project/ Field Work/Apprenticeship/ Dissertation/Internship: An elective course designed to acquire special/advanced knowledge, such as supplement study/support study to a project work, and a learner studies such a course on his own with an advisory support by a counsellor/faculty member. Currently, Literature survey and Research Project in 5<sup>th</sup> amd 6<sup>th</sup> semester respectively is offered under code; LS101N and RP102N.
  - c) **Generic Elective (GE) Course:** An elective course chosen generally from an unrelated discipline/subject, with an intention to seek exposure is called a Generic Elective. In B.Sc. programme presently we are not offering any such course.

P.S.: A core course offered in a discipline/subject may be treated as an elective by other discipline/subject and vice versa and such electives may also be referred to as Generic Elective.

- C. Ability Enhancement Compulsory Courses (AECC): AECC may be of two kinds: Ability Enhancement Courses (AEC) and Skill Enhancement Courses (SEC). "AECC" courses are the courses based upon the content that leads to knowledge enhancement. SEC courses are value-based and/or skillbased and are aimed at providing hands-on-training, competencies and skills.
  - (a) Ability Enhancement Courses (AEC): English Communication/Hindi Communication, Human Rights and Duties/Health & Hygiene, Environmental Science/Solid Waste Management, Disaster Management/Nutrition for Community.

(U)	
Semester	Ability Enhancement Courses (AECC)
1	Ability Enhancement Course in English [AECEG]
	OR
	Ability Enhancement Course in Hindi [AECHD]
2	Ability Enhancement Course in Human Rights and Duties [AECHRD]
	OR
	Ability Enhancement Course in Health & Hygiene [AECHH]
3	Ability Enhancement Course in Environment Awareness [AECEA]
	OR
	Ability Enhancement Course in Solid Waste Management [AESWM]
4	Ability Enhancement Course in Nutrition for Community [AECNC]
	OR
	Ability Enhancement Course in Disaster Management [AECDM]

(c) Skill Enhancement Courses (SEC): These courses may be chosen from a pool of courses designed to provide value-based and/or skill-based knowledge. In B.Sc. programme. Presently we are not offering options to choose from pool, however courses are fixed for respective semesters.

S.	Combinati	Semester, course code and credits					
No.	on of three	1 <sup>st</sup> sem	2 <sup>nd</sup> sem	3 <sup>rd</sup> sem	4 <sup>th</sup> sem	5 <sup>th</sup> sem	6 <sup>th</sup> sem
	courses	Course code/ Credit	Course code/ Credit	Course code/ Credit	Course code/ Credit	Course code/ Credit	Course code/ Credit
1.	PCM	SBSEVS-01N/(4)	SBSCHE-02N/(04)	SBSCHE-01N /(04)	SBSPHS-02N/(04)	SBSMM-03N/(04)	SBSSTAT-04N/(04)
2.	PCS	SBSEVS-01N/(4)	SBSCHE-02N/(04)	SBSCHE-01N /(04)	SBSPHS-02N/(04)	SBSMM-03N/(04)	SBSSTAT-04N/(04)
3.	PMCs	SBSEVS-01N/(4)	SBSCS-02N/(4)	SBSCS-01 N/(04)	SBSPHS-02N/(04)	SBSMM-03N/(04)	SBSSTAT-04N /(04)
4.	PMS	SBSEVS-01N/(4)	SBSCS-02N/(4)	SBSCS-01 N /(04)	SBSPHS-02N/(04)	SBSMM-03N/(04)	SBSSTAT-04N/(04)
5.	PSCs	SBSEVS-01N/(4)	SBSCS-02N/(4)	SBSCS-01 N /(04)	SBSPHS-02N/(04)	SBSMM-03N/(04)	SBSSTAT-04N/(04)
6.	MSCs	SBSEVS-01N/(4)	SBSCS-02N/(4)	SBSCS-01N /(04)	SBSPHS-02N/(04)	SBSMM-03N/(04)	SBSSTAT-04N/(04)
7.	BZC	SBSEVS-01N/(4)	SBSCHE-02N/(04)	SBSCHE-01N /(04)	SBSBY-02N/(04)	SBSZY-03N/(04)	SBSBCH-04N/(04)
8.	BZBch	SBSEVS-01N/(4)	SBSCHE-02N/(04)	SBSBCH-01N/(04)	SBSBY-02N/(04)	SBSZY-03N/(04)	SBSBCH-04N/(04)
9.	PCsEVS	SBSEVS-01N/(4)	SBSEVS-02N/(4)	SBSCS-01 N /(04)	SBSPHS-02N/(04)	SBSCS-02N/(4)	SBSSTAT-04N/(04)
10.	PCEVS	SBSEVS-01N/(4)	SBSEVS-02N/(4)	SBSCHE-01N /(04)	SBSPHS-02N/(04)	SBSCHE-01N/(4)	SBSSTAT-04N/(04)
11.	PSEVS	SBSEVS-01N/(4)	SBSEVS-02N/(4)	SBSCHE-01 N/(04)	SBSPHS-02N/(04)	SBSCHE-02N/(4)	SBSSTAT-04N/(04)
12.	BCEVS	SBSEVS-01N/(4)	SBSEVS-02N/(4)	SBSCHE-01N /(04)	SBSBY-02N/(04)	SBSZY-03N/(04)	SBSBCH-04N/(04)
13.	BZEVS	SBSEVS-01N/(4)	SBSEVS-02N/(4)	SBSBCH-01N/(04)	SBSZY-02N/(04)	SBSZY-03N/(04)	SBSSTAT-04N/(04)
14.	ZCEVS	SBSEVS-01N/(4)	SBSEVS-02N/(4)	SBSCHE-01N/(04)	SBSZY-02N/(04)	SBSZY-03N/(04)	SBSSTAT-04N/(04)
15.	BBchEVS	SBSEVS-01N/(4)	SBSEVS-02N/(4)	SBSBCH-01N/(04)	SBSBY-02N/(04)	SBSZY-03N/(4)	SBSBCH-04N/(04)
16.	ZBchEVS	SBSEVS-01N/(4)	SBSEVS-02N/(4)	SBSBCH-01N/(04)	SBSBY-02N/(04)	SBSZY-03N/(04)	SBSBCH-04N/(04)
17.	PMEVS	SBSEVS-01N/(4)	SBSEVS-02N/(4)	SBSCHE-02N/(04)	SBSPHS-02N/(04)	SBSMM-03N/(04)	SBSSTAT-04N/(04)

The format of **Skill Enhancement Courses** for different combination is given below:

It is mandatory for every learner to offer any combination of subjects listed below to complete his/her program for the degree. Total credits allotted against each course of all three subjects together with AECC and SEC will be 120 distributed in 06 semesters (three years) separately.

The List of Skill E	Enhancement courses	offered in B.Sc.	programme is give	en below:
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Course Code	Skill Enhancement Courses	Credit
SBSEVS-01N	Energy Resources and Green Technology	4
SBSEVS-02N	Environmental Impact Assessment and Legislation	4
SBSCHE-01N	Organic Chemistry II (Advance Organic Chemistry)	4
SBSCHE-02N	Advance Analytical Chemistry	4
SBSCS-01N	Discrete Mathematics	4
SBSCS-02N	Python Programming Programming	4
SBSBCH-01N	Bio-analytical techniques	4
SBSBCH-04N	Clinical biochemistry	4
SBSZY-02N	Fundamental of Animal Behavior	4
SBSZY-03N	Economic zoology and environmental biology	4
SBSPHS-02N	Modern physics	4
SBSBY-02N	Ecology	4
SBSMM-03N	Elementary Analysis	4
SBSSTAT-04N	Numerical Methods & Basic Computer Knowledge	4

#### **Combinations of undergraduate science programs:**

The learner has to offer any one of the following combinations:

Life Science Group	Physical Science Group			
BZC:Botany, Zoology, Chemistry	PCM:Physics, Chemistry, Mathematics			
BZBch:Botany, Zoology, Biochemistry	PCS: Physics, Chemistry, Statistics			
BCEVS: Botany, Chemistry, Environmental Science	PMCs: Physics, Mathematics, Computer Science			
BZEVS: Botany, Zoology, Environmental Science	PMS: Physics, Mathematics, Statistics			
<b>ZCEVS:</b> Zoology, Chemistry, Environmental Science	PSCs:Physics, Statistics, Computer Science			
BBchEVS: Botany, Biochemistry, Environmental Science	MSCs:Mathematics, Statistics, Computer Science			
ZBchEVS: Zoology, Biochemistry, Environmental Science	PCEVS: Physics, Chemistry, Environmental Science			
	PCsEVS: Physics, Computer Science, Environmental Science			
	PSEVS: Physics, Statistics, Environmental Science			
	PMEVS: Physics, Mathematics, Environmental Science			

- A learner has to study and pass the 20 credits each from the three selected **Core Course** (subjects) in the first and second year.
- The learner has to choose and study one paper out of AECEG or ACEHD in first semester, AECHRD or AECHH in second semester, AECEA or AECSWM in third semester and AECNC or AECDM in fourth semester, each of 4 credits under Ability Enhancement Compulsory Courses.
- Under **Skill Enhancement Courses**, it is compulsory to study 24 credit papers from 1<sup>st</sup> to 6<sup>th</sup> semester.
- The learner has to choose any two groups among A, B, C, D, E, F, G, H and I from **Discipline Specific Elective Course** in each selected subject in fifth and sixth semester. The Selection criterion for **Discipline Specific Elective Course** (**DEC**) for **5**<sup>th</sup> and **6**<sup>th</sup> Semester is given below:

Subjec ts/ semest	Physics (Group A	r)	Chemistr (Group B	y 5)	Mathema (Group C	ntics C)	Computer (Group D)	Science	Statistics (Group E	)	Biochemistr (Group F)	ŗy	Botany (Group G	<b>;</b> )	Zoology (Group H	D	Environm Science (Group I)	ental
er	Paper code	credit	Paper code	credit	Paper code	credit	Paper code	credit	Paper code	Paper code	Paper code	credit	Paper code	credit	Paper code	cre dit	Paper code	cred it
5 <sup>th</sup> Semest er	DECPHS -105N DECPHS -106N DECPHS -107N(P)	2 2 2	DECCH E -105N DECCH E -106N DECCH E- 107N(P)	2 2 2	DECMM -109N DECMM -110N DECMM -111N(P)	2 2 2	DECCS - 105N DECCS - 106N DECCS- 107N(P)	2 2 2	DECSTA T -105N DECSTA T -106N DECSTA T- 107N(P)	2 2 2	DECBCH - 105N DECBCH - 106N DECBCH 107N(P)	2 2 2	DECBY -105N DECBY -106N DECBY 107N(P)	2 2 2	DECZY- -105N DECZY -106N DECZY 107N(P)	2 2 2	DECEV S-105N DECEV S -106N DECEV S- 107N(P)	2 2 2
6 <sup>th</sup> Semest er	DECPHS -108N DECPHS -109N DECPHS -110N(P	2 2 2	DECCH E -108N DECCH E -109N DECCH E- 110N(P)	2 2 2	DECMM -112N DECMM -113N DECMM -114N(P)	2 2 2	DECCS - 108N DECCS - 109N DECCS- 110N(P)	2 2 2	DECSTA T-108N DECSTA T -109N DECSTA T- 110N(P)	2 2 2	DECBCH - 108N DECBCH - 109N DECBCH 110N(P)	2 2 2	DECBY -108N DECBY -109N DECBY 110N(P)	2 2 2	DECZY -108N DECZY -109N DECZY 110N(P)	2 2 2	DECEV S-108N DECZY -109N DECEV S- 110N(P)	2 2 2

In this way, the learner must complete his 40 credit in the first year, 40 credit in the second year and 40 credits in the third year totaling of 120 credits.

- 2.5.2 Course curriculum: The details of syllabus is given in Appendix-I
- **2.5.3 Language of Instruction:** SLM will be provided in English. However, learner can write assignment and give Term End Examination (TEE) either in Hindi or English.
- **2.5.4 Duration of the Programme**

Minimum duration in years: 03

Maximum duration in years: 06

#### 2.5.5 Faculty & Support Staff requirement

Professor (3), Associate Professor (1), Assistant Professor (9) and support staff (3)

#### 2.6 Instructional Delivery Mechanisms

The Open University system is more learner-oriented, and the student is an active participant in the teaching-learning process. Most of the instructions are imparted through distance rather than face-to-face communication.

The University follows a multi-media approach for instruction. It comprises of:

- self-instructional printed material (Self Learning Material)
- audio and video lectures
- face-to-face counselling
- assignments

- laboratory work
- Project work in some courses
- teleconference/web conference
- Web Enabled Academic Support Portal

#### 2.6.1 Self-Learning Material

The Self Learning Material (SLMs) are prepared in line with the UGC guidelines on preparation of SLMs. The prepared study materials are self-instructional in nature.

The course material is divided into blocks. Each block contains a few units. Lessons, which are called Units, are structured to facilitate self-study. The units of a block have similar nature of contents. The first page of each block indicates the numbers and titles of the units comprising the block. In the first block of each course, we start with course introduction. This is followed by a brief introduction to the block. After the block introduction, emphasis is given on contribution of ancient Indian knowledge into that specific course. Next, each unit begins with an introduction totalk about the contents of the unit. The list of objectives are outlined to expect the learning based outcome after working through the unit. This is followed by the main body of the unit, which is divided into various sections and subsections. Each unit is summarized with the main highlights of the contents.

Each unit have several "Check Your Progress" Questions and Terminal Questions /exercises. These questions help the learner to assess his/her understanding of the subject contents. At the end of units,additional references/books/suggested online weblink for MOOCs/Open Educational Resources for additional reading are suggested.

#### 2.6.2 Audio and Video lectures

Apart from SLM, audio and video lectures have been prepared for some courses. The audiovideo material is supplementary to print material. The video lectures are available at YouTube channel of university

#### 2.6.3 Counselling Classes

The face to face (F2F) counselling classes are conducted at head quarter and study centers. The purpose of such a contact class is to answer some of questions and clarify the doubts of learner which may not be possible through any other means of communication. Well experienced counsellors at study centers provide counselling and guidance to the learner in the courses that (s)he has chosen for study. The counselling sessions for each of the courses will be held at suitable intervals throughout the whole academic session. The time table for counselling classes are displayed at head quarter as well as by the coordinator of study center, however, attending counselling sessions is not compulsory. It is noted that to attend the counselling sessions, learner has to go through the course materials and note down the points to be discussed as it is not a regular class or lectures.

#### 2.6.4 Assignments

The purpose of assignments is to test the comprehension of the learning material that learner receives and also help to get through the courses by providing self-feedback to the learner. The course content given in the SLM will be sufficient for answering the assignments.

Assignments constitute the continuous evaluation component of a course. The assignments are available at the SLM section of the home page of university website. In any case, learner has to submit assignment before appearing in the examination for any course. The

assignments of a course carry 30% weightage while 70% weightage is given to the term-end examination (TEE). The marks obtained by learner in the assignments will be counted in the final result. Therefore, It is advised to take assignments seriously. However, there will be no written assignments for Lab courses.

#### 2.6.5 Laboratory Work

Laboratory courses are an integral component of the B.Sc. programme. While designing the curricula for laboratory courses, particular care has been taken to weed out experiments not significant to the present-day state of the discipline. Importance has been given to the utility of an experiment with respect to real life experience, development of experimental skills, and industrial applications. It is planned to phase the laboratory courses during suitable periods (such as summer or autumn vacations) so that in-service persons can take them without difficulty. Laboratory courses worth 2 credits will require full-time presence of the student at the Study Centre for one week continuously. During this time a student has to work for around 60 hours. Around 40 hours would be spent on experimental work and the remaining time will be used for doing calculations, preparations of records, viewing or listening to the video/audio programmes.

#### 2.6.6 Teleconference/Web conference

Teleconference/web conference, using done through ZOOM/webex in form of online special counselling sessions is another medium to impart instruction to and facilitate learning for a distance learner. The students concerned would be informed about the teleconferencing schedule and the place where it is to be conducted by sending bulk SMS.

#### 2.6.7 Web Enabled Academic Support Portal

The University also provide Web Enabled Academic Support Portal to access the course materials, assignments, and other learning resources.

#### 2.6.8 Learner Support Service Systems

#### (a) Study Centre

A Study Centre has following major functions:

- (i) **Counselling:** Counselling is an important aspect of Open University System. Face to face contact-cum-counselling classes for the courses will be provided at the Study Centre. The detailed programme of the contact-cum-counselling sessions will be sent to the learner by the Coordinator of the Study Centre. In these sessions learner will get an opportunity to discuss with the Counsellors his/her problems pertaining to the courses of study.
- (ii) Evaluation of Assignments: The evaluation of Tutor Marked Assignments (TMA) will be done by the Counsellors at the Study Centre. The evaluated assignments will be returned to the learner by the Coordinator of Study Centre with tutor comments and marks obtained in TMAs. These comments will help the learner in his/her studies.
- (iii) **Library:** Every Study Centre will have a library having relevant course materials, reference books suggested for supplementary reading prepared for the course(s).
- (iv) **Information and Advice:**The learner will be given relevant information about the courses offered by the University. Facilities are also provided to give him/her guidance in choosing courses.
- (v) **Interaction with fellow-students:** In the Study Centre learner will have an opportunity to interact with fellow students. This may lead to the formation of self-help groups.

#### (b) Learner Support Services (LSS)

The University has formed an LSS cell at the head quarter. The LSS cell coordinate with the Study Centre to get rid ofany problem faced by the learner.

#### 2.7 Procedure for admissions, curriculum transaction and evaluation

#### **Admission Procedure** 2.7.1

- (a) The detailed information regarding admission will be given on the UPRTOU website and on the admission portal. Learners seeking admission shall apply online.
- (b) Direct admission to 3-year B.Sc. program is offered to the interested candidates.
- (c) Eligibility: The candidate should pass the 10+2 level with science group. To opt B.Sc. life science combination, candidate should pass 10+2 with Biology and to opt for B.Sc. Physical Science combination, candidate should pass 10+2 with Mathematics.
- 2.7.2 **Programme Fee:** Rs. 8000 / year. The fee is deposited through online admission portal only.

#### 2.7.3 **Evaluation**

The evaluation consists of two components: (1) continuous evaluation through assignments, and (2) term-end examination. Learner must pass both in continuous evaluation as well as in the term-end examination of a course to earn the credits assigned to that course.For each course there shall be one written Terminal Examination. The evaluation of every course shall be in two parts that is 30% internal weightage through assignments and 70% external weightage through terminal exams.

(a) Theory course	Max. Marks	
Terminal Examination	70	
Assignment	30	
Total	100	
(b) Practical course:	Max. Marks	
Terminal Practical Examination	on 100	
Marks of Terminal Practical Examinations	shall be awarded as per following sch	eme:
i. Write up /theory work	30	
ii. Viva-voce	30	

ii.	Viva-voce

- Execution/Performance/Demonstration 20 iii.
- iv. Lab Record

The following 10-Point Grading System for evaluating learners' achievement is used for CBCS programmes:

20

#### **10-Point Grading System in the light of UGC-CBCS Guidelines**

Letter Grade	Grade Point	% Range		
O (Outstanding)	10	91-100		
A+ (Excellent)	9	81-90		
A (Very Good)	8	71-80		
B+ (Good)	7	61-70		
B (Above Average)	6	51-60		
C (Average)	5	41-50		
P (Pass)	4	36-40		
NC (Not Completed)	0	0-35		
Ab (Absent)	0			
Q	Qualified	Applicable only for Non-Credit		
NQ	Not Qualified	courses		

Learner is required to score at least a 'P' grade (36% marks) in both the continuous evaluation (assignments) as well as the term-end examination. In the overall computation also, learner must get at least a 'P' grade in each course to be eligible for the B. Sc. degree.

#### **Computation of CGPA and SGPA**

(a)	Following	formula	shall be	used for	calculation	of CGPA	and SGPA
(a)	Following	ioimuia	shan be	useu ioi	calculation	UI CUFF	and SOFA

For jth semester	where,
SGPA (Sj) = $\Sigma$ (Ci *Gi)/ $\Sigma$ Ci	Ci = number of credits of the ith course in jth semester Gi= grade point scored by the learner in the ith course in jth semester.
$CGPA = \Sigma (Cj *Sj) / \Sigma Cj$	where, Sj = SGPA of the jth semester Cj = total number of credits in the jth semester

The CGPA and CGPA shall be rounded off up to the two decimal points. (For e.g., if a learner obtained 7.2345, then it will be written as 7.23 or if s(he) obtained 7.23675 then it be will written as 7.24)

CGPA will be converted into percentage according to the following formula:

Equivalent Percentage = CGPA \* 9.5

(b) Award of Division

The learner will be awarded division according to the following table:

Division Classification			
1 <sup>st</sup> Division	6.31 or more and less than 10 CGPA		
2 <sup>nd</sup> Division	4.73 or more and less than 6.31 CGPA		
3 <sup>rd</sup> Division	3.78 or more and less than 4.73 CGPA		

#### 2.7.4 Multiple Entry and Multiple Exit options

The 3-year B.Sc. programme is an Outcome-Based Education (OBE) for qualifications of different types. The qualification types and examples of title/nomenclature for qualifications within each type are indicated in Table 1.

Table 1									
Level Qualification		Programme duration	Entry Option	Exit option					
	title	_		_					
	Undergraduate	Programme duration: First	10+2 level with science group	Exit followed by an exit 10- credit bridge					
5	Certificate in	year (first two semesters) of		course(s) lasting two months, including at					
	Science	the B.Sc. programme		least 6- credit job-specific					
				internship/apprenticeship					
	Undergraduate	Programme duration: First	Undergraduate Certificate	Exit followed by an exit 10- credit bridge					
6	Diploma in	two years (first four	obtained after completing the	course(s) lasting two months, including at					
	Science	semesters) of the of the	first year (two semesters) of the	least 6- credit job-specific					
		B.Sc. programme	B.Sc. programme	internship/apprenticeship					
	Bachelor in	Programme duration: First	Undergraduate diploma	Exit followed by an exit 10- credit bridge					
7	Science	three years (first six	obtained after completing two	course(s) lasting two months, including at					
		semesters) of the of the	years (four semesters) of the	least 6- credit job-specific					
		B.Sc. programme	B.Sc. programme	internship/apprenticeship					

Level	Yea r	Credits	Required Bridge Course of 10 cre	Award of Certificate/ Diploma/Degree		
			courses	Credits	Durati	Undergraduate
Level 5					on	Certificate in
	1	40	job-specific skill course	4	02 - 03	Science
			job-specific	6	months	
			internship/apprenticeship			
				Total: 10		
			courses	Credits	Durati	Undergraduate
Level 6					on	Diploma in
	2	40	job-specific skill course	4	02 - 03	Science
			job-specific	6	months	
			internship/apprenticeship			
				Total: 10		
			courses	Credits	Durati	Bachelor in
Level 7					on	Science
	3	40	job-specific skill course	4	02 – 03	
			job-specific	6	months	
			internship/apprenticeship			
				Total: 10		

#### Exit requirements from Level 5 to Level 7

#### Norms for 10- credit bridge course(s):

- 1. The job-specific skill course is of 4 credits. Only assignment has to be submitted by learner with 100% evaluation weightage.
- 2. The job-specific internship/apprenticeship of 02-03 months or more of 6 credits, after 2nd or 4th semester, will be mandatory for the learners desirous of exiting with a certificate or Diploma, respectively. The continuing learners may, however, undergo optional research internships after 2nd / 4th semester, to enhance their research capabilities, by engagement as interns in HEI/Research Institute/Industrial R&D labs/any other organization.
- 3. Under exit option from Level 5 to 7, the learner can choose HEI/Research Institute/Industrial R&D labs/any organization (Private/State Govt/Central Govt.) for internship/apprenticeship for job-specific bridge course by own or choose job-specific bridge course from the list provided by the University. After successful completion, he/she submit the certificate obtained from organization to the Training & Placement (T & P) Office of the University to get Undergraduate certificate/diploma for successful completion. The monitoring of such learners shall be done by T & P Cell.

#### 4. Evaluation of Bridge Course of 10 credit to exit from each Level

Bridge Course components	Credit	Mode of Evaluation
(a) job-specific skill course	4	Assignment
(b) job-specific internship/apprenticeship	6	Test/Viva voce/Practical
		conducted at organization level
Total credits	10	

5. Following is the list of courses under 10 credit bridge course.

Level		<b>Concerned Person</b>		
	Course Code / 4 credit	Job specific Course Title	Internship Domain Area/ 6 credit	to contact
5	BCOT-01	Office Tools	DTP Publishing	In-charge, Training & Placement Cell
6	BCCPLT-02	CPLT	Laboratory Technique	

#### 2.8 Requirement of the laboratory support and Library Resources

The practical sessions are held in the science laboratories of the Study Centre. In these labs, the learner will have the facility to use the equipment and consumables relevant to the syllabus. The SLM, supplementary text audio and video material of the various courses of the program is available through the online study portal of the University. The University also have a subscription of National Digital Library to provide the learners with the ability to enhance access to information and knowledge of various courses of the programme.

#### 2.9 Cost estimate of the programme and the provisions

3-year B.Sc. programme consists of 111 courses and 20 laboratory courses. One course is of 2 credits which consists of approx. 10 units. The total approximated expenditure on the development of 111 courses is:

S. No.	Item	Cost per Unit (writing	Total cost (Rs.)
		& editing)	
1	Total no. of units in 111 courses = $1110$	4500	495000
2	BOS Meetings, etc.	300000	300000
		Total	525000

#### 2.10 Quality assurance mechanism and expected programme outcomes

(a) **Quality assurance mechanism:** The program structure is developed under the guidance of the Board of studies comprising external expert members of the concerned subjects followed by the School board. The program structure and syllabus is approved by the Academic Council of the University. The course structure and syllabus is reviewed time to time according to the feedback received from the stakeholders and societal needs.

The Centre for Internal Quality Assurance will monitor, improve and enhance effectiveness of the program through the following:

- ✓ Annual academic audit
- ✓ Feedback analysis for quality improvement
- ✓ Regular faculty development programs
- ✓ Standardization of learning resources
- ✓ Periodic revision of program depending upon the changing trends by communicating to the concerned school

## (b) Expected programme outcomes (POs)

Knowledge	PO1	Demonstrate a fundamental/coherent understanding of the
and		academic field of science, its different learning areas and
understanding		applications, and its linkages with related disciplinary
_		areas/subjects
Skills related	<b>PO 2</b>	Demonstrate skills involving the constructive use of
to		knowledge in the subfields of physical and life science, and
specialization		other related fields of science in a range of settings, including
-		for pursuing higher studies related to the science.
Application	<b>PO 3</b>	Identify and apply appropriate principles and methodologies to
of knowledge		solve different types of problems with well-defined solutions.
and skills	<b>PO 4</b>	Apply knowledge of typical and atypical development across
		the lifespan of an individual
Generic	PO 5	Communicate accurately the findings of the experiments/
learning		investigations while relating the conclusions/findings to
outcomes		relevant theories of science.
	PO 6	Read texts and research papers analytically and present
		complex information and the findings of the
		experiments/investigations while relating the conclusions to
		relevant courses in science.

# **Combinations, Detailed Programme Structure & Syllabus**

S. No.	Groups	Programme structure for different combinations	Page No
1.		<b>BZC:</b> Botany, Zoology, Chemistry	110
2.		<b>BZBch:</b> Botany, Zoology, Biochemistry	
3.	Life Sciences	BCEVS: Botany, Chemistry, Environmental Science	
4.		BZEVS: Botany, Zoology, Environmental Science	
5.		<b>ZCEVS:</b> Zoology, Chemistry, Environmental Science	
6.		BBchEVS: Botany, Biochemistry, Environmental Science	
7.		ZBchEVS: Zoology, Biochemistry, Environmental Science	
8.		PCM: Physics, Chemistry, Mathematics	
9.		PCS: Physics, Chemistry, Statistics	
10.		PMCs: Physics, Mathematics, Computer Science	
11.		<b>PMS:</b> Physics, Mathematics, Statistics	
12.	Physical Sciences	PSCs: Physics, Statistics, Computer Science	
13.		MSCs: Mathematics, Statistics, Computer Science	
14.		PCEVS: Physics, Chemistry, Environmental Science	
15.		PCsEVS: Physics, Computer Science, Environmental	
		Science	
16.		<b>PSEVS:</b> Physics, Statistics, Environmental Science	
17.		<b>PMEVS:</b> Physics, Mathematics, Environmental Science	
S. No.		Syllabus	
1.	Physics		
2.	Chemistry		
3.	Mathematics		
4.	Biochemistry		
5.	Botany		
6.	Zoology		
7.	Computer Science		
8.	Statistics		
9.	Environmental Scie	ence	

Semester	Course Code	Title of course	Credits
		Compulsory Core Course	
	UGBY -101N	Cytology and Genetic	2
	UGCHE-101N	Inorganic Chemistry I (Basic Inorganic Chemistry)	2
	UGZY-101N	Animal Physiology	2
	UGBY -101(P)N	Practical Work based on UGBY -101N	2
I	UGCHE-101(P)N	Practical Work based on UGCHE-101N	2
	UGZY-101(P)N	Practical Work Based on UGZY -101N	2
		Skill Enhancement Course	
	SBSEVS-01N	Energy Resources and Green Technology	4
		Ability Enhancement Course	
	AECEG	Ability Enhancement Course in English	
	OR	OR	OR
	AECHD	Ability Enhancement Course in Hindi	4
		Total Credit (1 <sup>st</sup> Semester)	20
		Compulsory Core Course	
	UGBY -102N	Plant Physiology	2
	UGCHE-102N	Organic Chemistry I (Basic Organic Chemistry)	2
	UGZY-102N	Diversity of Animal life	2
	UGBY -102(P)N	Practical Work based on UGBY -102N	2
	UGCHE-102(P)N	Practical Work based on UGCHE-102N	2
	UGZY-102(P)N	Practical Work Based on UGZY -102N	2
ш		Skill Enhancement Course	
	SBSCHE-02N	Advance Analytical Chemistry	4
		Ability Enhancement Course	
	AECUDD	Ability Enhancement Course in Human Dialta and Dation	4
	OR	Ability Enhancement Course in Human Rights and Duties OR	4
	AECHH	Ability Enhancement Course in Health & Hygiene	4
		Total Credit (2 <sup>nd</sup> Semester)	20
		Compulsory Core Course	
	UGBY -103N	Plant Diversity-I	2
ш	UGCHE-103N UGZV 103N	Physical Chemistry I (Basic Physical Chemistry)	2
	UGBY -103(P)N	Practical Work based on UGBY -103N	2
	UGCHE-103(P)N	Practical Work based on UGCHE-103N	2
	UGZY-103(P)N	Practical Work Based on UGZY -103N	2
		Skill Enhancement Course	
	SBSCHE-01N	Organic Chemistry II (Advance Organic Chemistry)	4
		Ability Enhancement Course	
	AECEA	Ability Enhancement Course in Environment Awareness	4
	AECSWM	Ability Enhancement Course in Solid Waste Management	4
		Total Credit (3 <sup>rd</sup> Semester)	20
		Compulsory Core Course	
	UGBY -104N	Plant Diversity-II	2
	UGCHE-104N	Inorganic Chemistry II (Advance Inorganic Chemistry)	2
	UGZY-104N	Hemichordata and Chordata	2
	UGCHE-104(P)N	Practical Work based on UGCHE-104N	2
<b>TX</b> 7	UGZY-104(P)N	Practical Work Based on UGZY -104N	2
IV		Skill Enhancement Course	
	SBSBY-02N	Ecology	4
	ADONO	Ability Enhancement Course	
	OR	Adding Enhancement Course in Nutrition for Community [AEUNC] OR	4
	AECDM	Ability Enhancement Course in Disaster Management [AECDM]	4
		Total Credit (4 <sup>th</sup> Semester)	20

#### C-1: Combination (Botany, Zoology, Chemistry)

V     SBSZY03N     Economic zoology and environmental biology     4       Discipline Centric Elective Course (Chose any two group from G, B and H)       OCEBY -105N DCEBY -106N DCEBY -106N DCEBY -107P)N     Embryology and Morphogenesis Practical Work based on DCEBY -105N and DCEBY -106N     2       OCEBY -106N DCECHE -106N DCECHE -106N DCECHE -106N DCECHE -106N DCECHE -106N DCECHE -106N DCECY -106N DCECY -106N     DOECHE -106N DCECY - 106N DCECY - 106N     2       OCEEY -106N DCECY -106N DCECY -106N DCECY - 106N DCECY - 106N     Inorganic Chemistry III (Selected Topics In Inorganic Chemistry) Practical Work based on DCECHE -106N 2     2       UCEZY -106N DCEZY - 106N DCEZY - 106N     DCECY - 105N and DCECHE - 106N 2       UCEZY -107(P)N     Taxonomy and Evolution Practical Work based on DCEZY - 105N and DCEZY - 106N 2       VI       UGLS-101N     Literature Survey of Elected Subject-1     4       VI       Skill Enhancement Course       OCEBY -108N DCEBY -109N DCEBY -109N DCEBY -109N DCEBY -109N DCECHY -109N DCECHY -109N DCECHY -109N DCECHY -109N DCECHY -109N DCECHY -109N DCECHE -108N DCECHE -108N DCECHY -109N DCECHE -108N DCECHY -100PN DCECHE -108N DCECHY -100PN DCECHY -108N DCECHY -100PN DCECHY -100PN DCECHY -108N DCECHY -100PN DCECHY -100PN DCECHY -100PN D				Skill Enhancement Course		
V         Discipline Centric Elective Course (Chose any two group from G, B and H)           Group-G         DCEBY -105N DCEBY -106N DCEBY -107(P)N         Embryology and Microbiology Practical Work based on DCEBY -105N and DCEBY -106N         2           Group-B         DCECHE -108N DCECHE -108N DCECHE -108N DCECHE -107(P)N         Physical Chemistry III (Selected Topics In Inorganic Chemistry)         2           Group-H         DCECY -106N DCEZY -106N DCEZY -106N         Animal distribution and ecology Practical Work based on DCEZY -105N and DCEZY -106N         2           UGLS-101N         Literature Survey of Elected Subject-1         4           Vitarature Survey           UGLS-101N         Clinical biochemistry         4           DCECHE -108N DCEZY -107(P)N           SBSBCH-04N         Clinical biochemistry         4           Total Credit (5 <sup>th</sup> Semester)           20         SBSBCH-04N         Clinical biochemistry         4           DCECHE -108N DCEBY -108N DCEBY -109N DCEBY -109N DCEBY -109N DCEBY -109N DCEBY -109N         DCEBY -108N DCECHE -108N DCECH		SBSZY03N	Economic	zoology and environmental biology	4	
V     Group-G     DCEBY -105N DCEBY -107(P)N DCEBY -107(P)N DCECHE -105N DCECHE -105N DCECHE -105N DCECHE -105N DCECHE -106N DCECHE -106N DCECHE -106N DCECHE -106N DCECHE -106N DCECHE -107(P)N DCECHE -106N DCECHE -106N DCECHE -106N DCECHE -106N DCECHE -107(P)N DCECHE -106N DCECHE -108N DCECHE -	*7		Discipline Ce	ntric Elective Course (Chose any two group from G, B and H)		
Group-G         DCEBY 106N DCEBY 107(P)N         Plant Pathology and Microbiology Practical Work based on DCEBY -105N and DCEBY -106N         2           Group-B         DCECHE -105N DCECHE -107(P)N         Physical Chemistry II (Advance Physical Chemistry)         2           Group-H         DCECHE -106N DCEZY -105N DCEZY -107(P)N         Animal distribution and ecology         2           Group-H         DCEZY -105N DCEZY -107(P)N         Animal distribution and ecology         2           Taxonomy and Evolution DCEZY -107(P)N         Practical Work based on DCECTY -105N and DCEZY -106N         2           UGLS-101N         Literature Survey of Elected Subject-1         4           Total Credit (5 <sup>th</sup> Semester)           SBSBCH-04N         Clinical biochemistry         4           Discipline Centric Elective Course (Chose any two group as per Sen. V)           DCEBY -108N DCEBY -108N         Molecular Genetics and Biotechnology Practical Work based on DCEBY -108N and DCEBY -109N         2           Group-G         DCEBY -108N DCEBY -108N DCEBY -108N         Organic Chemistry III (Selected Topics In Organic Chemistry)         2           VI         Group-B         DCEEH -108N DCEEH -109N DCEEH -109N         Organic Chemistry III (Selected Topics In Physical Chemistry)         2           VI         DCEEH -108N DCEEH -108N DCECH -109N         Dreactical Work based on DCECHE -108N and DCEBY -109N	v	DCEBY -105N		Embryology and Morphogenesis	2	
VI         DCEBY -107(P)N DCECHE -105N DCECHE -105N DCECHE -106N         Practical Work based on DCEBY -105N and DCEBY -106N         2           Group-B         DCECHE -105N DCECHE -107(P)N         Physical Chemistry III (Selected Topics In Inorganic Chemistry) Practical Work based on DCECHE -105N and DCECHE -106N         2           Group-H         DCEZY -105N DCEZY -106N DCEZY -106N DCEZY -106N         Animal distribution and ecology Taxonomy and Evolution         2           UGLS-101N         Literature Survey of Elected Subject-1         4           VIETER SUPPORT           SBSBCH-04N         Clinical biochemistry         4           DCEEY -108N DCEBY -108N DCEBY -108N DCEBY -108N DCEBY -108N         Molecular Genetics and Biotechnology Paleobotany, Palynology and Economic DCEBY -108N         2           Skill Enhancement Course           VI           Group-G         DCEEY -108N DCEBY -108N DCEBY -108N DCEBY -108N DCEBY -108N DCEBY -108N DCEBY -108N DCEBY -108N DCEBY -108N         Molecular Genetics and Biotechnology Paleobotany, Palynology and Economic Practical Work based on DCEBY -108N and DCEBY -109N         2           OCEBY -108N DCEEH -109N DCECHE -		Group-G	DCEBY -106N	Plant Pathology and Microbiology	2	
Group-B         DCECHE -108N DCECHE -107(P)N DCECHE -107(P)N         Physical Chemistry II (Advance Physical Chemistry)         2           Group-H         DCEZY-105N DCEZY -105N DCEZY -106N         Animal distribution and ecology         2           Taxonomy and Evolution DCEZY - 107(P)N         Animal distribution and ecology         2           UGLS-101N         Literature Survey of Elected Subject-1         4           VI           Skill Enhancement Course           SBSBCH-04N           Clerize Course (Chose any two group as per Sem. V)           DCECHE -108N DCEBY -109N           DCEEBY -108N DCEBY -109N           DCECHE -108N DCECHE -109N DCEBY -100P)N           DCECHE -108N DCECHE -109N DCECHE -109N           DCECHE -109N         Organic Chemist			DCEBY -107(P)N	Practical Work based on DCEBY -105N and DCEBY -106N	2	
Group-B         DCECHE - 100N DCECHE - 107(P)N         Inorganic Chemistry III (Selected Topics In Inorganic Chemistry)         2           Group-H         DCEZY - 106N DCEZY - 106N DCEZY - 107(P)N         Inorganic Chemistry III (Selected Topics In Inorganic Chemistry)         2           Group-H         DCEZY - 106N DCEZY - 107(P)N         Animal distribution and ecology         2           Taxonomy and Evolution Practical Work based on DCEZY - 105N and DCEZY - 106N         2           UGLS-101N         Literature Survey of Elected Subject-1         4           Total Credit (5 <sup>th</sup> Semester)         20           Skill Enhancement Course           SBSBCH-04N         Clinical biochemistry         4           DEERY - 108N DCEBY - 108N DCEBY - 110(P)N         Molecular Genetics and Biotechnology Practical Work based on DCEBY - 108N and DCEBY - 109N         2           Group-G         DCECHE - 108N DCECHE - 108N DCECHE - 108N DCECHE - 109N         Organic Chemistry III (Selected Topics In Organic Chemistry)         2           Group-H         DCECT+ 108N DCECHE - 108N DCECHE - 100PN         Developmental Biology Practical Work based on DCECHE - 108N and DCECHE - 109N         2           Practical Work based on DCECHE - 108N DCECHE - 109N         2           DCECHE - 108N DCECHE - 100PN         Developmental Biology DCECHE - 100PN         2           DCECHE - 108N DCECH			DCECHE -105N	Physical Chemistry II (Advance Physical Chemistry)	2	
VI     DEEHE-IN(P)N DCEZY -105N DCEZY -107(P)N     Practical Work based on DCECHE -105N and DCECHE -106N     2       Group-H     DCEZY -105N DCEZY -107(P)N     Animal distribution and ecology Taxonomy and Evolution Practical Work based on DCEZY -105N and DCEZY -106N     2       UGLS-101N     Literature Survey of Elected Subject-1     4       Total Credit (5 <sup>th</sup> Semester)     20       Skill Enhancement Course       Skill Enhancement Course       Skill Enhancement Course       OCEBY -108N DCEBY -109N       Discipline Centric Elective Course (Chose any two group as per Sem. V)       DEEBY -108N DCEBY -109N DCEBY -109N DCEBY -109N DCEBY -109N       DCECHE -108N DCEBY -109N DCEEH - 108N DCECHE -108N DCECHE -108N DCECHE -108N DCECHE -108N DCECHE -108N DCECHE -109N     Molecular Genetics and Biotechnology Practical Work based on DCEBY -108N and DCEBY -109N     2       Practical Work based on DCEBY -108N and DCEBY -109N       OCECHE -108N DCECHE -108N DCECHE -109N       DCECHE -108N DCECHE -109N       DCECHE -108N DCECHE -109N       DCECHE -108N DCECHY -109N       DCECH -109N DCECHY -109N       DCECH -109N DCECHY -109N       DCECH -109N DCECHY -109N       DCECH -109N DCECHY -109N       DCECY -108N DCECY -109N       DCECY -108N D		Group-B	DCECHE -106N	Inorganic Chemistry III (Selected Topics In Inorganic Chemistry)	2	
VI     DCEZY - 105N DCEZY - 107(P)N DCEZY - 107(P)N     Animal distribution and ecology Taxonomy and Evolution Practical Work based on DCEZY - 105N and DCEZY - 106N     2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2			DCECHE-107(P)N	Practical Work based on DCECHE -105N and DCECHE -106N	2	
Group-H     DCEZY - 106N DCEZY - 107(P)N     Taxonomy and Evolution Practical Work based on DCEZY -105N and DCEZY -106N     2       Literature Survey     Literature Survey       UGLS-101N     Literature Survey of Elected Subject-1     4       Total Credit (5 <sup>th</sup> Semester)     20       SBSBCH-04N     Clinical biochemistry     4       Discipline Centric Elective Course (Chose any two group as per Sem. V)     Molecular Genetics and Biotechnology Paleobotany, Palynology and Economic     2       Group-G     DCEBY -109N DCEBY -109N     Molecular Genetics and DCEBY -108N and DCEBY -109N     2       DCECHE -108N DCECHE -108N     DCECHE -108N DCECHE -109N     Molecular Genetics and Biotechnology Paleobotany, Palynology and Economic     2       Group-B     DCECHE -108N DCECHE -109N     DCECHE -108N and DCEBY -109N     2       DCECHE -109N DCECHE -109N     DCECHE -108N DCECHE -109N     DCECHE -108N and DCECHE -108N and DCECHE -109N     2       Group-H     DCEZY -108N DCECY + 100PN     Developmental Biology DCEZY -100N     Developmental Biology Practical Work based on DCEZY -108N & DCEZY -109N     2       UGRP-102N     Research Project on Elected Subject-2     4			DCEZY-105N	Animal distribution and ecology	2	
VI     DCE2T - 107(P)N     Practical Work based on DCEZY -105N and DCEZY -106N     2       Literature Survey       UGLS-101N     Literature Survey of Elected Subject-1     4       Total Credit (5 <sup>th</sup> Semester)     20       Skill Enhancement Course       SBSBCH-04N     Clinical biochemistry     4       Discipline Centric Elective Course (Chose any two group as per Sem. V)       DCEBY -108N       DCEBY -108N       DCEBY -108N       DCEBY -108N       DCEBY -108N       DCEBY -108N       DCEEBY -109N       DCEEBY -109N       DCEEBY -108N       DCECHE -109N		Group-H	DCEZY -106N	Taxonomy and Evolution	2	
Literature Survey           UGLS-101N         Literature Survey of Elected Subject-1         4           Total Credit (5th Semester)         20           Skill Enhancement Course         20           Skill Enhancement Course         20           SBSBCH-04N         Clinical biochemistry         4           Discipline Centric Elective Course (Chose any two group as per Sem. V)           Discipline Centric Elective Course (Chose any two group as per Sem. V)           DCEBY -108N         Molecular Genetics and Biotechnology         2           DCEBY -108N         Molecular Genetics and Biotechnology         2           DCEBY -108N         DCEBY -108N         Practical Work based on DCEBY -108N and DCEBY -109N         2           DECECHE -108N         Organic Chemistry III (Selected Topics In Organic Chemistry)         2           DECEV -108N         DCECHE -108N         DCECHE -108N and DCECHE -108N and DCECHE -109N         2           DCEEY -108N         DCECHE -108N         DCECHE -108N DCEZY -108N DCEZY -108N and DCECHE -108N and DCECHE -109N         2           DCEEY -108N         DCEEY -108N DCEZY -108N DCEZY -109N         Developmental Biology and Genetic Engineering DCEZY -108N DCEZY -109N DCEZY -108N DCEZY -108N DCE			DCEZ I -10/(P)N	Practical Work based on DCEZY -105N and DCEZY -106N	2	
UGLS-101N     Literature Survey of Elected Subject-1     4       Total Credit (5 <sup>th</sup> Semester)     20       Skill Enhancement Course       SBSBCH-04N     Clinical biochemistry     4       Discipline Centric Elective Course (Chose any two group as per Sem. V)       OCEBY -108N       DCEBY -108N     DCEBY -108N     Molecular Genetics and Biotechnology     2       Paleobotany, Palynology and Economic     2     Practical Work based on DCEBY -108N and DCEBY -109N     2       DCEBY -110(P)N     DCECHE - 108N     Organic Chemistry III (Selected Topics In Organic Chemistry)     2       Practical Work based on DCEHE - 108N DCECHE - 109N     Organic Chemistry III (Selected Topics In Physical Chemistry)     2       Practical Work based on DCECHE - 108N DCECHE - 109N     DCECHE - 108N DCECHE - 109N     2       DCEEY -102N     DCEZY -108N DCECHE - 109N DCECHE - 109N     2       DCECHZ - 108N DCEZY - 109N DCEZY - 109N     DCECHE - 109N DCEZY - 108N BOCECHE - 109N     2       DCEZY - 108N DCEZY - 109N DCEZY - 109N     DCECHE - 108N DCEZY - 109N DCEZY - 109N DCEZY - 100P)     2       Broup-H     DCEZY - 108N DCEZY - 100P)N     Practical Work based on DCEZY - 108N & DCEZY - 109N DCEZY - 100P)N     2       Research Project       UGRP-102N     Research Project on Elected Subject-2     4				Literature Survey		
VI         Total Credit (5th Semester)         20           Skill Enhancement Course           SBSBCH-04N         Clinical biochemistry         4           Discipline Centric Elective Course (Chose any two group as per Sem. V)           Molecular Genetics and Biotechnology         2           Broup-G         DCEBY -108N DCEBY -109N DCEBY -110(P)N         Molecular Genetics and Biotechnology         2           Paleobotany, Palynology and Economic         2         Practical Work based on DCEBY -108N and DCEBY -109N         2           Broup-B         DCECHE -108N DCECHE -109N DCECHE -109N DCECHE -110(P)N         Organic Chemistry III (Selected Topics In Organic Chemistry)         2           Physical Chemistry III (Selected Topics In Physical Chemistry)         2         Practical Work based on DCECHE -108N and DCECHE -109N         2           Group-H         DCEZY -108N DCEZY -109N DCEZY -109N DCEZY -109N         Developmental Biology         2         2           Molecular Biology and Genetic Engineering Practical Work Based on DCEZY -108N & DCEZY -109N         2         2           Research Project           UGRP-102N         Research Project on Elected Subject-2         4		UGLS-101N	UGLS-101N Literature Survey of Elected Subject-1			
VI         Sill Enhancement Course         4           SBSBCH-04N         Clinical biochemistry         4           Discipline Centric Elective Course (Chose any two group as per Sem. V)           Group-G         DCEBY -108N DCEBY -109N DCEBY -110(P)N         Molecular Genetics and Biotechnology Practical Work based on DCEBY -108N and DCEBY -109N         2           Group-B         DCECHE -108N DCECHE -109N DCECHE -109N DCECHE -109N         Organic Chemistry III (Selected Topics In Organic Chemistry)         2           Physical Chemistry III (Selected Topics In Physical Chemistry)         2         2           Practical Work based on DCECHE -108N DCECHE -109N         Organic Chemistry III (Selected Topics In Physical Chemistry)         2           Physical Chemistry III (Selected Topics In Physical Chemistry)         2         2           Practical Work based on DCECHE -108N and DCECHE -109N         2         2           Physical Chemistry III (Selected Topics In Physical Chemistry)         2         2           Practical Work based on DCECHE -108N and DCECHE -109N         2         2           Practical Work based on DCEZY -108N DCECY -109N         2         2           Practical Work Based on DCEZY -108N & DCEZY -109N         2         2           Practical Work Based on DCEZY -108N & DCEZY -109N         2         2           Practical Work Based on DCEZY -1				Total Credit (5 <sup>th</sup> Semester)	20	
SBSBCH-04N       Clinical biochemistry       4         Discipline Centric Elective Course (Chose any two group as per Sem. V)         Group-G       DCEBY -108N DCEBY -109N DCEBY -110(P)N       Molecular Genetics and Biotechnology Paleobotany, Palynology and Economic       2         Paleobotany, Palynology and Economic       2         Practical Work based on DCEBY -108N DCECHE -108N       Organic Chemistry III (Selected Topics In Organic Chemistry)       2         Physical Chemistry III (Selected Topics In Organic Chemistry)       2       Physical Chemistry III (Selected Topics In Physical Chemistry)       2         Group-B       DCECY - 108N DCECHE -110(P)N       Developmental Biology       2       2         Physical Chemistry III (Selected Topics In Organic Chemistry)       2       2         Physical Chemistry III (Selected Topics In Physical Chemistry)       2       2         Practical Work based on DCECHE -108N DCECHE -109N       2       2         Practical Work based on DCECHE -108N and DCECHE -109N       2       2         Practical Work based on DCECHE -108N DCECHE -109N       2       2         Practical Work based on DCECHE -108N DCECHE -109N       2       2         Practical Work based on DCECHE -108N and DCECHE -109N       2       2         DCECTY -109N DCEZY -109N       Developmental Biology and Genetic Engineering Practical Work Based on				Skill Enhancement Course		
Discipline Centric Elective Course (Chose any two group as per Sem. V)         Group-G       DCEBY -108N DCEBY -109N DCEBY -110(P)N       Molecular Genetics and Biotechnology Paleobotany, Palynology and Economic       2         VI       Group-B       DCECHE -108N DCECHE -109N DCECHE -109N       Organic Chemistry III (Selected Topics In Organic Chemistry)       2         VI       DCECHE -109N DCECHE -109N       Organic Chemistry III (Selected Topics In Organic Chemistry)       2         Group-B       DCECHE -109N DCECHE -110(P)N       Organic Chemistry III (Selected Topics In Physical Chemistry)       2         B       DCECHE -109N DCECHE -109N       Developmental Biology       2         B       DCECHE -109N DCEZY -109N       Developmental Biology       2         Physical Chemistry III (Selected Topics In Physical Chemistry)       2         DCEZY -109N DCEZY -109N       Developmental Biology       2         Practical Work based on DCEZY -108N & DCEZY -109N       2         DCEZY -100N       Developmental Biology and Genetic Engineering DCEZY -110(P)N       2         Practical Work Based on DCEZY -108N & DCEZY -109N       2         Practical Work Based on DCEZY -108N & DCEZY -109N       2         Practical Work Based on DCEZY -108N & DCEZY -109N       2         Practical Work Based on DCEZY -108N & DCEZY -109N       2         Protect		SBSBCH-04N	Clinical bi	ochemistry	4	
VI     DCEBY -108N DCEBY -109N DCEBY -110(P)N     Molecular Genetics and Biotechnology Paleobotany, Palynology and Economic     2       Paleobotany, Palynology and Economic     2       Practical Work based on DCEBY -108N DCECHE -109N     0       DCECHE -108N DCECHE -109N     Organic Chemistry III (Selected Topics In Organic Chemistry)     2       Physical Chemistry III (Selected Topics In Physical Chemistry)     2       Practical Work based on DCECHE -108N DCECHE -109N     2       DCECHE -110(P)N     Physical Chemistry III (Selected Topics In Physical Chemistry)     2       Physical Chemistry III (Selected Topics In Physical Chemistry)     2       DCEZY -108N DCEZY -109N     Developmental Biology     2       Molecular Biology and Genetic Engineering     2       Practical Work Based on DCEZY -108N & DCEZY -109N     2       DCEZY -109N     Developmental Biology and Genetic Engineering     2       Practical Work Based on DCEZY -108N & DCEZY -109N     2       VI     Research Project on Elected Subject-2     4		Discipline Centric Elective Course (Chose any two group as per Sem. V)				
Group-G       DCEBY -109N DCEBY -110(P)N       Paleobotany, Palynology and Economic Practical Work based on DCEBY -108N and DCEBY -109N       2         Group-B       DCECHE -108N DCECHE -109N DCECHE -109N DCECHE -109N       Organic Chemistry III (Selected Topics In Organic Chemistry)       2         Hysical Chemistry III (Selected Topics In Physical Chemistry)       2       Physical Chemistry III (Selected Topics In Physical Chemistry)       2         Broup-H       DCEZY -108N DCEZY -109N DCEZY -109N       Developmental Biology Molecular Biology and Genetic Engineering Practical Work Based on DCEZY -108N & DCEZY -109N       2         VI       Research Project on Elected Subject-2       4			DCEBY -108N	Molecular Genetics and Biotechnology	2	
VI       DCEBY -110(P)N       Practical Work based on DCEBY -108N and DCEBY -109N       2         Group-B       DCECHE -108N DCECHE -109N DCECHE -109N DCECHE -110(P)N       Organic Chemistry III (Selected Topics In Organic Chemistry)       2         Physical Chemistry III (Selected Topics In Physical Chemistry)       2         DCEZY -109N DCEZY -109N       Developmental Biology       2         Molecular Biology and Genetic Engineering DCEZY -110(P)N       Developmental Biology and Genetic Engineering Practical Work Based on DCEZY -108N & DCEZY -109N       2         Research Project         UGRP-102N       Research Project on Elected Subject-2       4		Group-G	DCEBY -109N	Paleobotany, Palynology and Economic	2	
VI       DCECHE -108N DCECHE -109N DCECHE -109N DCECHE -109N       Organic Chemistry III (Selected Topics In Organic Chemistry)       2 Physical Chemistry         WI       Dread Control Contrection			DCEBY -110(P)N	Practical Work based on DCEBY -108N and DCEBY -109N	2	
VI     Group-B     DCECHE - 109N DCECHE - 110(P)N     Physical Chemistry III (Selected Topics In Physical Chemistry)     2       Practical Work based on DCECHE - 108N and DCECHE - 109N     2       Practical Work based on DCEZY - 108N & DCEZY - 109N     2       Practical Work Based on DCEZY - 108N & DCEZY - 109N     2       Practical Work Based on DCEZY - 108N & DCEZY - 109N     2       Practical Work Based on DCEZY - 108N & DCEZY - 109N     2       Practical Work Based on DCEZY - 108N & DCEZY - 109N     2       Practical Work Based on DCEZY - 108N & DCEZY - 109N     2			DCECHE -108N	Organic Chemistry III (Selected Topics In Organic Chemistry)	2	
Group-H     DCEZY - 108N DCEZY - 109N     Developmental Biology Molecular Biology and Genetic Engineering Practical Work Based on DCEZY - 108N & DCEZY - 109N     2       Research Project       UGRP-102N     Research Project on Elected Subject-2     4	VI	Group-B	DCECHE -109N	Physical Chemistry III (Selected Topics In Physical Chemistry)	2	
Group-H     DCEZY -108N DCEZY -109N DCEZY -110(P)N     Developmental Biology Molecular Biology and Genetic Engineering Practical Work Based on DCEZY -108N & DCEZY -109N     2       Research Project       UGRP-102N       Research Project on Elected Subject-2     4	V1		DCECHE-110(F)N	Practical Work based on DCECHE -108N and DCECHE -109N	2	
Group-H     DCE21-109N DCEZY-110(P)N     Molecular Biology and Genetic Engineering Practical Work Based on DCEZY-108N & DCEZY-109N     2       Research Project     Research Project     4		~	DCEZY -108N	Developmental Biology	2	
DELET FIGURA     Practical Work Based on DCEZY -108N & DCEZY -109N     2       Research Project       UGRP-102N     Research Project on Elected Subject-2     4		Group-H	DCEZY -109N	Molecular Biology and Genetic Engineering	2	
Research Project       UGRP-102N     Research Project on Elected Subject-2     4			DODET HIGH/I	Practical Work Based on DCEZY -108N & DCEZY -109N	2	
UGRP-102N Research Project on Elected Subject-2 4		Research Project				
		UGRP-102N	Research Project or	n Elected Subject-2	4	
Total Credit (6 <sup>th</sup> Semester)     20				Total Credit (6 <sup>th</sup> Semester)	20	

C-2: (	: Combination (Botany, Zoology, Biochemistry)				
Semester	Course Code	Title of course	Credits		
		Compulsory Core Course			
	UGBY -101N	Cytology and Genetic	2		
	UGBCH-101N	Introduction to biocnemistry	2		
	UGZY-101N	Animal Physiology	2		
т	UGBY -101(P)N	Practical Work based on UGBY -101N	2		
1	UGBCH-101(P)N	Practical Work Based on UGBCH -101N	2		
	UGZY-101(P)N	Practical Work Based on UGZY -101N	2		
		Skill Enhancement Course			
	SBSEVS-01N	Energy Resources and Green Technology	4		
		Ability Enhancement Course			
	AECEG	Ability Enhancement Course in English	4		
	OR	OR	OR		
	AECHD	Ability Enhancement Course in Hindi	4		
		Total Credit (1 <sup>st</sup> Semester)	20		
	LICEN LOON	Compulsory Core Course			
	UGBY -102N	Plant Physiology	2		
	UGBCH-102N	Nutritional blochemistry	2		
	UGZY-102N	Diversity of Animal life	2		
	UGBY -102(P)N	Practical Work based on UGBY -102N	2		
	UGBCH-102(P)N	Practical Work Based on UGBCH -102N	2		
п	UGZY-102(P)N	Practical Work Based on UGZY -102N	2		
11		Skill Enhancement Course			
	SBSCHE-02N	Advance Analytical Chemistry	4		
	Ability Enhancement Course				
	AECUDD	Ability Enhancement Course in Human Dishts and Dation	4		
	AECHKD	Ability Enhancement Course in Human Rights and Duties OR	4		
	AECHH	Ability Enhancement Course in Health & Hygiene	4		
		Total Credit (2 <sup>nd</sup> Semester)	20		
		Compulsory Core Course			
	UGBY -103N	Plant Diversity-I	2		
ш	UGBCH-103N UGZV 103N	Intermediary metabolism	2		
	UGBY -103(P)N	Practical Work based on UGBY -103N	2		
	UGBCH-103(P)N	Practical Work Based on UGBCH -103N	2		
	UGZY-103(P)N	Practical Work Based on UGZY -103N	2		
	SDSDCU 01N	Skill Enhancement Course	4		
	SDSDCH-01N	Bio-analytical techniques	4		
		Ability Enhancement Course			
	AECEA	Ability Enhancement Course in Environment Awareness	4		
	AECSWM	Or Ability Enhancement Course in Solid Weste Management	4		
	ALCSWM	Total Credit (3 <sup>rd</sup> Semester)	20		
		Compulsory Core Course			
	UGBY -104N	Plant Diversity-II	2		
	UGBCH-104N	Enzymology	2		
	UGZY-104N	Hemichordata and Chordata	2		
	UGBY -104(P)N UGBCH-104(P)N	Practical Work Based on UGBY -104N Practical Work Based on UGBCH -104N	2		
***	UGZY-104(P)N	Practical Work Based on UGZY -104N	2		
1V		Skill Enhancement Course			
	SBSBY-02N	Ecology	4		
	AECNC	Ability Enhancement Course Ability Enhancement Course in Nutrition for Community [AECNC]	Λ		
	OR	OR	+		
	AECDM	Ability Enhancement Course in Disaster Management [AECDM]	4		
		Total Credit (4 <sup>th</sup> Semester)	20		

			Skill Enhancement Course		
	SBSZY-03N	Econom	ic zoology and environmental biology	4	
		Discipline C	entric Elective Course (Chose any two group from G, F and H)		
v		DCEBY -105N	Embryology and Morphogenesis	2	
	Group-G	DCEBY -106N	Plant Pathology and Microbiology	2	
		DCEBY -107(P)N	Practical Work based on DCEBY -105N and DCEBY -106N	2	
		DCEBCH -105N	Microbiology	2	
	Group-F	DCEBCH -106N	Spectroscopy	2	
		DCEBCH -10/(P)N	Practical Work Based on DCEBCH -105N & DCEBCH -106N	2	
		DCEZY-105N	Animal distribution and ecology	2	
	Group-H	DCEZY -106N DCEZY 107(P)N	Taxonomy and Evolution	2	
		DCEZI -10/(F)N	Practical Work based on DCEZY -105N and DCEZY -106N	2	
			Literature Survey		
	UGLS-101N Literature Survey of Elected Subject-1		4		
			Total Credit (5 <sup>th</sup> Semester)	20	
	Skill Enhancement Course				
	SBSBCH-04	Clinical	biochemistry	4	
	Discipline Centric Elective Course (Chose any two group as per Sem. V)				
		DCEBY -108N	Molecular Genetics and Biotechnology	2	
	Group-G	DCEBY -109N	Paleobotany, Palynology and Economic	2	
		DCEBY -110(P)N	Practical Work based on DCEBY -108N and DCEBY -109N	2	
		DCEBCH -108N	Plant biochemistry	2	
VI	Group-F	DCEBCH -109N DCEBCH -110(P)N	Immunology	2	
		DCEDCH -110(1)N	Practical Work Based on DCEBCH -108N & DCEBCH -109N	2	
	<i>a</i> <b>v</b>	DCEZY -108N	Developmental Biology	2	
	Group-H	DCEZY -110(P)N	Molecular Biology and Genetic Engineering	2	
	Practical Work Based on DCEZY -108N & DCEZY -109N 2				
	Research Project				
	UGRP-102N	Research Project o	n Elected Subject-2	4	
			Total Credit (6 <sup>th</sup> Semester)	20	

Semester	Course Code	Title of course	Credits
		Compulsory Core Course	
	UGBY -101N	Cytology and Genetic	2
	UGCHE-101N	Inorganic Chemistry I (Basic Inorganic Chemistry)	2
	UGEVS-101N	Fundaments of Environmental Sciences	2
_	UGBY -101(P)N	Practical Work based on UGBY -101N	2
	UGCHE-101(P)N	Practical Work based on UGCHE-101N	2
	UGEVS-101(P)N	Practical Work based on UGEVS-101N	2
		Skill Enhancement Course	
	SBSEVS-01N	Energy Resources and Green Technology	4
		Ability Enhancement Course	
	AECEG	Ability Enhancement Course in English	4
	OR	OR	OR
	AECHD	Ability Enhancement Course in Hindi	4
		I otal Credit (1 <sup>st</sup> Semester)	20
	LICEN 102N	Compulsory Core Course	2
	UGDT -102N	Plain Physiology	2
	UGCHE-102N		2
	UGEVS-102N	Ecology and Biodiversity Conservation	2
	UGBY -102(P)N	Practical Work based on UGBY -102N	2
	UGCHE-102(P)N	Practical Work based on UGCHE-102N	2
т	UGEVS-102(P)N	Practical Work based on UGEVS-102N	2
11		Skill Enhancement Course	
	SBSEVS-02N	Environmental Impact Assessment and Legislation	4
		Ability Enhancement Course	•
	A E CUID D	Ability Enhancement Course	
	AECHKD	Ability Enhancement Course in Human Rights and Duties	4
	AECHH	Ability Enhancement Course in Health & Hygiene	4
		Total Credit (2 <sup>nd</sup> Semester)	20
		Compulsory Core Course	
	UGBY -103N	Plant Diversity-I	2
TT	UGCHE-103N	Physical Chemistry I (Basic Physical Chemistry)	2
111	UGEVS-103N	Environmental Microbiology and Biotechnology	2
	$\frac{\text{UGBY} - 103(\text{P})\text{N}}{\text{UGCUE} - 102(\text{P})\text{N}}$	Practical Work based on UCCHE 102N	2
	UGCHE-103(P)N	Practical Work based on LIGEVS 102N	2
	UUE V 5-105(F)IV	Skill Enhancement Course	2
	SBSCHE-01N	Organic Chemistry II (Advance Organic Chemistry)	4
		Ability Enhancement Course	
		Ability Enhancement Course in Environment Amount	4
	AECEA	Or	4
	AECSWM	Ability Enhancement Course in Solid Waste Management	4
		Total Credit (3 <sup>rd</sup> Semester)	20
		Compulsory Core Course	
	UGBY -104N	Plant Diversity-II	2
	UGCHE-104N	Inorganic Chemistry II (Advance Inorganic Chemistry)	2
	UGEVS-104N	Plant Physiology and Biochemistry	2
	UGBY -104(P)N	Practical Work based on UGBY -104	2
<b>TX</b> 7	UGCHE-104(P)N	Practical Work based on UGCHE-104	2
IV	UGEVS-104(P)N	Practical Work based on UGEVS-104N	2
	CDCDV 02N	SKIII Enhancement Course	A
	5B2B1-05N	Ecology	4
	AFONO	Ability Enhancement Course	A
	AECNC	Aduity Ennancement Course in Nutrition for Community [AECINC]	4
	AECDM	Ability Enhancement Course in Disaster Management [AECDM]	4

### C-3: Combination (Botany, Chemistry, Environmental Science)

			Total Credit (4 <sup>th</sup> Semester)	20	
	Skill Enhancement Course				
	SBSZY-03N Economic zoology and environmental biology				
<b>X</b> 7		Discipline C	entric Elective Course (Chose any two group from G, F and H)		
v		DCEBY -105N	Embryology and Morphogenesis	2	
	Group-G	DCEBY -106N	Plant Pathology and Microbiology	2	
		DCEBY -10/(P)N	Practical Work based on DCEBY -105N and DCEBY -106N	2	
		DCECHE -105N	Physical Chemistry II (Advance Physical Chemistry)	2	
	Group-F	DCECHE - 106N DCECHE - 107(P)N	Inorganic Chemistry III (Selected Topics In Inorganic Chemistry)	2	
		DCEVE 105N	Practical Work based on DCECHE -105N and DCECHE -106N	2	
		DCEVS-105N	Environmental Pollutions Demote Sensing, CIS and Hydrology	2	
	Group-H	DCEVS-100N DCEVS 107(D)N	Remote Sensing, GIS and Hydrology	2	
		DCEVS-107(F)IN	Practical Work based on DCEVS-105N & DCEVS-106N	2	
			Literature Survey		
		Literature Survey	of Elected Subject-1	4	
	UGLS-101N				
			Total Credit (5 <sup>th</sup> Semester)	20	
	Skill Enhancement Course				
	SBSBCH-04 Clinical biochemistry				
	Discipline Centric Elective Course (Chose any two group as per Sem. V)				
		DCEBY -108N	Molecular Genetics and Biotechnology	2	
	Group-G DCEB	DCEBY -109N DCEBY -110(P)N	Paleobotany, Palynology and Economic	2	
		DCECHE 100N	Practical Work based on DCEBY -108 and DCEBY -109N	2	
		DCECHE - 108N DCECHE - 109N	Molecular Genetics and Biotechnology	2	
<b>X</b> / <b>T</b>	Group-F	DCECHE-110(P)N	Practical Work based on DCERY -108 and DCERY -109	2	
V I		DCEVS 108	Statistics and Environmental Quality Assassment	- 2	
		DCEV5-100	Environmental geology and earth resources	$\frac{2}{2}$	
	Group-H DCE		Environmental geology and cartin resources	2	
	Group-H	DCEVS-110P	Practical Work based on DCEVS-108N & DCEVS-109N	2	
	Group-H	DCEVS-110P	Practical Work based on DCEVS-108N & DCEVS-109N	2	
	Group-H	DCEVS-110P	Practical Work based on DCEVS-108N & DCEVS-109N Research Project	2	
	Group-H UGRP-102N	DCEVS-110P Research Project o	Practical Work based on DCEVS-108N & DCEVS-109N Research Project n Elected Subject-2	2	

Semester	Course Code	Title of course	Credits
		Compulsory Core Course	
	UGBY -101N	Cytology and Genetic	2
	UGZY-101N	Animal Physiology	2
	UGEVS-101N	Fundaments of Environmental Sciences	2
_	UGBCH-101(P)N	Practical Work Based on UGBCH -101N	2
1	UGZY-101(P)N	Practical Work Based on UGZY -101N	2
	UGBY -101(P)N	Practical Work based on UGBY -101N	2
		Skill Enhancement Course	
	SBSEVS-01N	Energy Resources and Green Technology	4
		Ability Enhancement Course	
	AECEG	Ability Enhancement Course in English	4
	AECHD	Ability Enhancement Course in Hindi	OR 4
		Total Credit (1 <sup>st</sup> Semester)	16
		Compulsory Core Course	
	UGBY -102N	Plant Physiology	2
	UGZY-102N	Diversity of Animal life	2
	UGEVS-102	Ecology and Biodiversity Conservation	2
	UGBCH-102(P)N	Practical Work Based on UGBCH -102N	2
	UGZY-102(P)N	Practical Work Based on UGZY -102N	2
	UGBY -102(P)N	Practical Work based on UGBY -102N	2
II		Skill Enhancement Course	
	SBSEVS-02N	Environmental Impact Assessment and Legislation	4
		Ability Enhancement Course	7
		Ability Enhancement Course	
	AECHRD	Ability Enhancement Course in Human Rights and Duties	4
	AECHH	Ability Enhancement Course in Health & Hygiene	4
		Total Credit (2 <sup>nd</sup> Semester)	20
		Compulsory Core Course	
	UGBY -103N	Plant Diversity-I	2
	UGZY-103N	Genetic and Cell Biology	2
111	UGEVS-103	Environmental Microbiology and Biotechnology	2
	UGBCH-103(P)N	Practical Work Based on UGBCH -103N Practical Work Based on UGBCH 102N	2
	UGBY -103(P)N	Practical Work based on UGBY -103N	2
		Skill Enhancement Course	
	SBSBCH-01N	Bio-analytical techniques	4
		Ability Enhancement Course	
	AECEA	Ability Enhancement Course in Environment Awareness	4
	AECSWM	Or Ability Enhancement Course in Solid Wests Management	4
	AECSWM	Addity Enhancement Course in Solid waste Management Total Credit (3 <sup>rd</sup> Semester)	20
		Compulsory Core Course	-0
	UGBY -104N	Plant Diversity-II	2
	UGZY-104N	Hemichordata and Chordata	2
	UGEVS-104N	Plant Physiology and Biochemistry	2
	UGBCH-104(P)N	Practical Work Based on UGBCH -104N	2
	UGZ 1-104(P)N	Practical Work Based on UG2Y -104N Practical Work based on UGBY -104N	2
IV	0001-104(1)1	Skill Enhancement Course	2
	SBSZY-02N	Fundamental of Animal Behavior	4
		Ability Enhancement Course	
	AECNC	Ability Enhancement Course in Nutrition for Community [AECNC]	4
	AECDM	OK Ability Enhancement Course in Disaster Management [AECDM]	4
		Total Credit (4 <sup>th</sup> Semester)	20

#### C-4: Combination (Botany, Zoology, Environmental Science)

	Skill Enhancement Course				
	SBSZY-03N	Economi	c zoology and environmental biology	4	
*7		Discipline Co	entric Elective Course (Chose any two group from G, F and H)		
v		DCEBY -105N	Embryology and Morphogenesis	2	
	Group-G	DCEBY -106N	Plant Pathology and Microbiology	2	
		DCEBY -107(P)N	Practical Work based on DCEBY -105N and DCEBY -106N	2	
		DCEZY-105N	Animal distribution and ecology	2	
	Group-F	DCEZY -106N DCEZY 107(P)N	Taxonomy and Evolution	2	
		DCEZT -107(F)N	Practical Work based on DCEZY -105N and DCEZY -106N	2	
		DCEVS-105N	Environmental Pollutions	2	
	Group-H	DCEVS-106N	Remote Sensing, GIS and Hydrology	2	
	01000-11	DCEVS-107(P)N	Practical Work based on DCEVS-105N & DCEVS-106N	2	
	Literature Survey				
	UGLS-101N Literature Survey of Elected Subject-1			4	
			Total Credit (5 <sup>th</sup> Semester)	20	
			Skill Enhancement Course		
	SBSSTAT-04N	Numerica	al Methods & Basic Computer Knowledge	4	
	Discipline Centric Elective Course (Chose any two group as per Sem. V)				
		DCEBY -108N	Molecular Genetics and Biotechnology	2	
	Group-G	DCEBY -109N	Paleobotany, Palynology and Economic	2	
		DCEBY -110(P)N	Practical Work based on DCEBY -108N and DCEBY -109N	2	
		DCEZY -108N	Developmental Biology	2	
VI	Group-F	DCEZY -109N DCEZY 110(P)N	Molecular Biology and Genetic Engineering	2	
VI		DCEZT -ITO(T)N	Practical Work Based on DCEZY -108 & DCEZY -109	2	
		DCEVS-108N	Statistics and Environmental Quality Assessment	2	
	Group-H	DCEVS-109N	Environmental geology and earth resources	$\frac{2}{2}$	
	<b>P</b>	DCEVS-110P	Practical work based on DCEVS-108N & DCEVS-109N	2	
		1	Research Project		
	UGRP-102N	Research Project or	n Elected Subject-2	4	
			Total Credit (6 <sup>th</sup> Semester)	20	

Semester	Course Code	Title of course	Credits
		Compulsory Core Course	
	UGZY-101N	Animal Physiology	2
	UGCHE-101N	Inorganic Chemistry I (Basic Inorganic Chemistry)	2
	UGEVS-101N	Fundaments of Environmental Sciences	2
			2
I	UGZY-101(P)N	Practical Work Based on UGZY -101N	2
	UGCHE-101(P)N	Practical Work based on UGCHE-101N	2
			2
		Skill Enhancement Course	
	SBSEVS-01N	Energy Resources and Green Technology	4
		Ability Enhancement Course	
	AECEC	Ability Enhancement Course in English	4
	OR	OR	OR
	AECHD	Ability Enhancement Course in Hindi	4
		Total Credit (1 <sup>st</sup> Semester)	20
		Compulsory Core Course	
	UGZY-102N	Diversity of Animal life	2
	UGCHE-102N	Organic Chemistry I (Basic Organic Chemistry)	2
	UGEVS-102	Ecology and Biodiversity Conservation	2
	UG7Y-102(P)N	Practical Work Based on LIGZY -102N	2
			2
	UGCHE-102(P)N	Practical Work based on UGCHE-102N	2
п			2
		Skill Enhancement Course	
	SBSEVS-02N	Environmental Impact Assessment and Legislation	4
		Ability Enhancement Course	
	AECUDD	Ability Enhancement Course in Human Bichts and Duties	4
	ALCHIND	OR	4
	AECHH	Ability Enhancement Course in Health & Hygiene	4
		Total Credit (2 <sup>nd</sup> Semester)	20
		Compulsory Core Course	
	UGZY-103N	Genetic and Cell Biology	2
	UGCHE-103N	Physical Chemistry I (Basic Physical Chemistry)	2
111	UGEVS-103	Environmental Microbiology and Biotechnology	2
	UGZY-103(P)N	Practical Work Based on UGZY -103N	2
	UGCHE-103(P)N	Practical Work based on UGCHE-103N	2
		Skill Enhancement Course	2
	SBSCHE-01N	Organic Chemistry II (Advance Organic Chemistry)	4
			•
		Ability Ennancement Course	
	AECEA	Ability Enhancement Course in Environment Awareness	4
	AFCSWM	Or Ability Enhancement Course in Solid Waste Management	4
	THEO WIN	Total Credit (3 <sup>rd</sup> Semester)	20
		Compulsory Core Course	
	107X 104N		2
	UGZY-104N	Hemichordata and Chordata	2
	UGCHE-104N	Diant Diant Diant and Diant and Diant and Diant	2
	UGEVS-104N	Plant Physiology and Biochemistry Practical Work Based on LICZY 104N	2
	UGCHE 104(P)N	Practical Work based on UCCHE 104N	2
IV	00001112-104(1)14		2
		Skill Enhancement Course	4
	SBSZY-02N	Fundamental of Animal Behavior	
		Ability Enhancement Course	
	AECNC	Ability Enhancement Course in Nutrition for Community [AECNC]	4
	OR	OR	
	AECDM	Ability Enhancement Course in Disaster Management [AECDM]	4

C-5: Combination (Zoology, Chemistry, Environmental Science)

	Total Credit (4 <sup>th</sup> Semester)				20
	Skill Enhancement Course				
	SBSZY03N		Economic	zoology and environmental biology	4
*7		Dis	scipline Ce	ntric Elective Course (Chose any two group from G, B and H)	
v	DCEZY		-105N	Animal distribution and ecology	2
	Group-G	DCEZY	-106N	Taxonomy and Evolution	2
		DCEZY	-10/(P)N	Practical Work based on DCEZY -105N and DCEZY -106N	2
	DCEC		E -105N	Physical Chemistry II (Advance Physical Chemistry)	2
	Group-B	DCECHE -106N DCECHE 107(P)N		Inorganic Chemistry III (Selected Topics In Inorganic Chemistry)	2
		DODU	C 105N	Practical Work based on DCECHE -105N and DCECHE -106N	2
		DCEV	S-105N	Environmental Pollutions	2
	Group-H	DCEV	S-100IN S-107(D)N	Remote Sensing, GIS and Hydrology	$\frac{2}{2}$
		DCEV	3-10/(F)IN	Practical Work based on DCEVS-105N & DCEVS-106N	2
					L
				Literature Survey	
	UGLS-101N	Literatu	ure Survey	of Elected Subject-1	4
	Total Credit (5 <sup>th</sup> Semester)				20
				Skill Enhancement Course	
	SBSSTAT-04N		Numerica	l Methods & Basic Computer Knowledge	4
		D	Discipline C	entric Elective Course (Chose any two group as per Sem. V)	
		DCEZY	-108N	Developmental Biology	2
	Group-G	DCEZY	-109N	Molecular Biology and Genetic Engineering	2
		DCEZI	-110(P)IN	Practical Work Based on DCEZY -108N & DCEZY -109N	2
	~ ~	Group-B DCECHE -108N DCECHE -109N DCECHE-110(P)N		Organic Chemistry III (Selected Topics In Organic Chemistry)	2
	Group-B			Physical Chemistry III (Selected Topics In Physical Chemistry)	2
VI		DOEVO	1 100	Practical work based on DCECHE -108N and DCECHE -109N	2
		DCEVS	-108	Statistics and Environmental Quality Assessment	2
	Group-H	DCEVS	-109 110P	Practical Work based on DCEVS 108N & DCEVS 100N	2
	-	DCEVS	-1101	Tractical work based on DCEVS-1001 & DCEVS-1091	2
	Research Project				
	UGRP-102N	Researc	ch Project o	n Elected Subject-2	4
	Total Credit (6 <sup>th</sup> Semester)				20

Semester	Course Code	Title of course	Credits		
		Compulsory Core Course			
	UGBY -101N	Cytology and Genetic	2		
	UGBCH-101N	Introduction to biochemistry	2		
	UGEVS-101N	Fundaments of Environmental Sciences	2		
	UGBY -101(P)N	Practical Work based on UGBY -101N	2		
I	UGBCH-101(P)N	Practical Work Based on UGBCH -101N	2		
	UGEVS-101(P)N	Practical Work Based on UGEVS-101N	2		
		Skill Enhancement Course			
	SBSEVS-01N	Energy Resources and Green Technology	4		
		Ability Enhancement Course			
	AECEG	Ability Enhancement Course in English	4		
	OR	OR	OR		
	AECHD	Ability Enhancement Course in Hindi Total Credit (18 Semester)	4		
		Total Credit (1" Semester)	20		
		Compulsory Core Course			
	UGBY -102N	Plant Physiology	2		
	UGBCH-102N	Nutritional biochemistry	2		
	UGEVS-102	Ecology and Biodiversity Conservation	2		
	UGBY -102(P)N	Practical Work based on UGBY -102N	2		
	UGBCH-102(P)N	Practical Work Based on UGBCH -102N	2		
п	UGEVS-102(P)N	Practical Work Based on UGEVS-102N	2		
11		Skill Enhancement Course			
	SBSEVS-02N	Environmental Impact Assessment and Legislation	4		
		Ability Enhancement Course			
	AECHRD	Ability Enhancement Course in Human Rights and Duties	4		
	· Libolindo	OR			
	AECHH	Ability Enhancement Course in Health & Hygiene	4		
		Total Credit (2 <sup>nd</sup> Semester)	20		
		Compulsory Core Course			
	UGBY -103N	Plant Diversity-I	2		
TTT	UGBCH-103N	Intermediary metabolism	2		
111	UGEVS-103	Environmental Microbiology and Biotechnology	2		
	UGBY -103(P)N	Practical Work based on UGBY -103N	2		
	UGEVS 103(P)N	Practical Work Based on LIGEVS 103N	2		
	002101000(1)11	Skill Enhancement Course	2		
	SBSBCH-01N	Bio-analytical techniques	4		
		Ability Enhancement Course			
	AECEA	Ability Enhancement Course in Environment Awareness	4		
	1 DOGUD (	Or			
	AECSWM	Ability Enhancement Course in Solid Waste Management	4		
		Compulsory Core Course	20		
	UGBY -104N	Plant Diversity-II	2		
	UGBCH-104N	Enzymology	2		
	UGEVS-104N	Plant Physiology and Biochemistry	2		
	UGBY -104(P)N	Practical Work based on UGBY -104N	2		
	UGBCH-104(P)N	Practical Work Based on UGBCH -104N	2		
IV	UGEVS-104(P)N	Practical Work Based on UGEVS-104N	2		
	Skill Enhancement Course				
	SBSBY-02N Ecology				
	AECNC	Ability Enhancement Course in Nutrition for Community [AECNC]	4		
	OR	OR			
	AECDM	Ability Enhancement Course in Disaster Management [AECDM]	4		
		Total Credit (4 <sup>th</sup> Semester)	20		

#### C-6 Combination (Botany, Biochemistry, Environmental Science)

	Skill Enhancement Course					
	SBSZY-03N Economic zoology and environmental biology			4		
*7	Discipline Centric Elective Course (Chose any two group from G, F and H)					
v		DCEBY -105N	Embryology and Morphogenesis	2		
	Group-G	DCEBY -106N	Plant Pathology and Microbiology	2		
		DCEBY -107(P)N	Practical Work based on DCEBY -105N and DCEBY -106N	2		
		DCEBCH -105N	Microbiology	2		
	Group-F	DCEBCH -106N DCEBCH -107(P)N	Spectroscopy	2		
		DCEDCH 107(1)11	Practical Work Based on DCEBCH -105N & DCEBCH -106N	2		
		DCEVS-105N	Environmental Pollutions	2		
	Group-H	DCEVS-106N DCEVS 107(D)N	Remote Sensing, GIS and Hydrology	2		
	5 <b>F</b>	DCEVS-10/(P)N	Practical Work based on DCEVS-105N & DCEVS-106N	2		
			Literature Survey			
	UGLS-101N Literature Survey of		f Elected Subject-1	4		
			Total Credit (5 <sup>th</sup> Semester)	20		
			Skill Enhancement Course			
	SBSBCH-04	Clinical b	biochemistry	4		
		Discipline (	Centric Elective Course (Chose any two group as per Sem. V)			
		DCEBY -108N	Molecular Genetics and Biotechnology	2		
	Group-G	DCEBY -109N DCEBY -110(P)N	Paleobotany, Palynology and Economic	2		
		DCEDT -110(T)IN	Practical Work based on DCEBY -108N and DCEBY -109N	2		
	Crown E	DCEBCH -108N DCEBCH -109N	Plant biochemistry	2		
VI	Group-r	DCEBCH -110(P)N	Practical Work Based on DCEBCH -108N & DCEBCH -109N	$\frac{2}{2}$		
V1		DCEVS-108N	Statistics and Environmental Quality Assessment	2		
		DCEVS-109N	Environmental geology and earth resources	2		
	Group-H	DCEVS-110(P)N	Practical Work based on DCEVS-108N & DCEVS-109N	2		
	Research Project					
	UGRP-102N	Research Project or	n Elected Subject-2	4		
	Total Credit (6 <sup>th</sup> Semester)			20		

Semester	Course Code	Title of course	Credits
		Compulsory Core Course	
	UGZY-101N	Animal Physiology	2
	UGBCH-101N	Introduction to biochemistry	2
	UGEVS-101N	Fundaments of Environmental Sciences	2
	UGZY-101(P)N	Practical Work Based on UGZY -101N	2
1	UGBCH-101(P)N	Practical Work Based on UGBCH -101N	2
	UGEVS-101(P)N	Practical Work Based on UGEVS-101N	2
		Skill Enhancement Course	
	SBSEVS-01N	Energy Resources and Green Technology	4
		Ability Enhancement Course	
	AECEG	Ability Enhancement Course in English	4
	OR	OR	OR
	AECHD	Ability Enhancement Course in Hindi Total Cradit (1 <sup>st</sup> Samastar)	4
			20
	UGZY-102N	Diversity of Animal life	2
	UGBCH-102N	Nutritional biochemistry	2
	LIGEVS-102N	Ecology and Biodiversity Conservation	2
	UCZV 102(D)N	Denotical Work Decod on LICTV 102N	2
	UGZ 1-102(P)N		2
	UGBCH-102(P)N	Practical Work Based on UGBCH -102N	2
п	UGEVS-102(P)N	Practical Work Based on UGEVS-102N	2
		Skill Enhancement Course	
	SBSEVS-02N	Environmental Impact Assessment and Legislation	4
		Ability Enhancement Course	
	AECHRD	Ability Enhancement Course in Human Rights and Duties	4
		OR	
	AECHH	Ability Enhancement Course in Health & Hygiene Total Credit (2 <sup>nd</sup> Somester)	4
			20
	UGZY-103N	Genetic and Cell Biology	2
	UGBCH-103N	Intermediary metabolism	2
III	UGEVS-103N	Environmental Microbiology and Biotechnology	2
	UGZY-103(P)N	Practical Work Based on UGZY -103N	2
	UGBCH-103(P)N UGEVS-103(P)N	Practical Work Based on UGBCH -103N Practical Work Based on UGEVS-103N	2
		Skill Enhancement Course	2
	SBSBCH-01N	Bio-analytical techniques	4
		Ability Enhancement Course	
	AECEA	Ability Enhancement Course in Environment Awareness	4
		Or	
	AECSWM	Ability Enhancement Course in Solid Waste Management	4
		Total Credit (5 Semester)	20
	UGZY-104N	Hemichordata and Chordata	2
	UGBCH-104N	Enzymology	2
	UGEVS-104N	Plant Physiology and Biochemistry	2
	UGZY-104(P)N	Practical Work Based on UGZY -104N	2
	UGBCH-104(P)N UGEVS-104(P)N	Practical Work Based on UGBCH - 104N Practical Work Based on UGEVS-104N	2
IV	001/5-104(1)14	Skill Enhancement Course	2
	SBSBY-02N	Ecology	4
		Ability Enhancement Course	
	AECNC	Ability Ennancement Course in Nutrition for Community [AECNC] OR	4
	AECDM	Ability Enhancement Course in Disaster Management [AECDM]	4
		Total Credit (4 <sup>th</sup> Semester)	20

C-7: Combination (Zoology, Biochemistry, Environmental Science)

	Skill Enhancement Course				
	SBSZY-03N Economic zoology and environmental biology			4	
	Discipline Centric Elective Course (Chose any two group from G, F and H)				
v		DCEZY-105N	Animal distribution and ecology	2	
	Group-G	DCEZY -106N	Taxonomy and Evolution	2	
		DCEZY -107(P)N	Practical Work based on DCEZY -105N and DCEZY -106N	2	
		DCEBCH -105N	Microbiology	2	
	Group-F	DCEBCH -106N	Spectroscopy	2	
		DCEBCH -10/(P)N	Practical Work Based on DCEBCH -105N & DCEBCH -106N	2	
		DCEVS-105N	Environmental Pollutions	2	
	Carry II	DCEVS-106N	Remote Sensing, GIS and Hydrology	2	
	Group-н	DCEVS-107(P)N	Practical Work based on DCEVS-105N & DCEVS-106N	2	
	Literature Survey				
		T :	f Elected Secking 1	4	
	UGLS-101N		Elected Subject-1	4	
			Total Credit (5 <sup>th</sup> Semester)	20	
			Skill Enhancement Course		
	SBSBCH-04	Clinical t	piochemistry	4	
		Discipline (	Centric Elective Course (Chose any two group as per Sem. V)		
		DCEZY -108N	Developmental Biology	2	
	Group-G	DCEZY -109N	Molecular Biology and Genetic Engineering	2	
		DCE21 -110(P)N	Practical Work Based on DCEZY -108N & DCEZY -109N	2	
		DCEBCH -108N	Plant biochemistry	2	
	Group-F	DCEBCH -109N DCEBCH 110/DN	Immunology	2	
VI		DCEBCII -110(I)N	Practical Work Based on DCEBCH -108N & DCEBCH -109N	2	
		DCEVS-108N	Statistics and Environmental Quality Assessment	2	
	Group-H	DCEVS-109N	Environmental geology and earth resources	2	
	Group-II	DCEVS-110(P)N	Practical Work based on DCEVS-108N & DCEVS-109N	2	
	Research Project				
	UGRP-102N				
		•	Total Credit (6 <sup>th</sup> Semester)	20	

Semester	Course Code	Title of course	Credits
		Compulsory Core Course	
	UGPHS -101N	Vector, Mechanics and General Physics	2
	UGCHE-101N	Inorganic Chemistry I (Basic Inorganic Chemistry)	2
	UGMM-101N	Differential Calculus	2
т	UGMM-102N	Analytical Geometry	2
1	UGPHS -101(P)N	Practical Work based on UGPHS -101N	2
	UGCHE-101(P)N	Practical Work based on UGCHE-101N	2
		Skill Enhancement Course	
	SBSEVS-01N	Energy Resources and Green Technology	4
		Ability Enhancement Course	
	AECEG	Ability Enhancement Course in English	4 OP
	AECHD	Ability Enhancement Course in Hindi	4
		Total Credit (1 <sup>st</sup> Semester)	20
		Compulsory Core Course	
	UGPHS -102N	Oscillation, waves and electrical circuits	2
	UGCHE-102N	Organic Chemistry I (Basic Organic Chemistry)	2
	UGMM-103N	Integral Calculus	2
	UGMM-104N	Differential Equation	2
	UGPHS -102(P)N	Practical Work based on UGPHS -102N	2
п	UGCHE-102(P)N	Practical Work based on UGCHE-102N	2
11		Skill Enhancement Course	
	SBSCHE_02N	Advance Analytical Chemistry	4
	5650112-021	Advance Anarytical Chemistry	4
		Ability Ennancement Course	
	AECHRD	Ability Enhancement Course in Human Rights and Duties	4
	AECHH	Ability Enhancement Course in Health & Hygiene	4
		Total Credit (2 <sup>nd</sup> Semester)	20
		Compulsory Core Course	
	UGPHS -103N UGCHE 103N	Electromagnetism Physical Chemistry I (Basic Physical Chemistry)	2
III	UGMM-105N	Mechanics-I (Statics and Dynamics)	2
	UGMM-106N	Mechanics-II (Dynamics and Hydrodynamics)	2
	UGPHS -103(P)N	Practical Work based on UGPHS -103N	2
	UGCHE-103(P)N	Practical Work based on UGCHE-103N	2
	ODOCHE OIN	Skill Enhancement Course	4
	SBSCHE-01N	Ability Enhancement Course	4
	AECEA	Ability Enhancement Course in Environment Awareness	4
		Or	
	AECSWM	Ability Enhancement Course in Solid Waste Management	4
		Compulsory Core Course	20
	LICDUS 104N	Angles and Distal Electronics	2
	UGPHS -104N UGCHE-104N	Analog and Digital Electronics	2
	UGMM-107N	Linear Algebra	2
	UGMM-108N	Calculus of function of several variable and Vector Calculus	2
	UGPHS -104(P)N	Practical Work based on UGPHS -104N	2
IV	UGCHE-104(P)N	Practical Work based on UGCHE-104N	2
	SBSDHS 02N	Skill Enhancement Course	Λ
	5D51 FI3-02IN	Ability Enhancement Course	+
	AECNC	Ability Enhancement Course in Nutrition for Community	4
		or	
	AECDM	Ability Enhancement Course in Disaster Management	4
		Skill Enhancement Course	20

C-8: Combination (Physics, Chemistry, Mathematics)

	SBSMM-03N	I Elementary	Analysis	4	
	Discipline Centric Elective Course (Chose any two group from A, B and C)				
$\mathbf{V}$		DCEPHS -105N	Optics	2	
	Course A	DCEPHS -106N	Thermal Physics	2	
	Group-A	DCEPHS-	Practical Work based on DCEPHS -105N and DCEPHS -106N	2	
		107(P)N			
		DCECHE -105N	Physical Chemistry II (Advance Physical Chemistry)	2	
	Crown P	DCECHE -106N	Inorganic Chemistry III (Selected Topics In Inorganic Chemistry)	2	
	Group-D	DCECHE-	Practical Work based on DCECHE -105N and DCECHE -106N	2	
		107(P)N			
		DCEMM -109N	Abstract Algebra	2	
	Group-C	DCEMM -110N	Number Theory	2	
	Group-C	DCEMM-	Viva Voce	2	
		111(P)N			
			Literature Survey		
	UGLS-101N Literature Survey of Elected Subject-1				
			Total Credit (5 <sup>th</sup> Semester)	20	
			Skill Enhancement Course		
	SBSSTAT-04	Numerical	Methods & Basic Computer Knowledge	4	
		Discipline Ce	ntric Elective Course (Chose any two group as per Sem. V)		
		DCEPHS -108N	Quantum mechanics and spectroscopy	2	
	Course A	DCEPHS -109N	Solid State Physics and Advanced Electronics	2	
	Group-A	DCEPHS-	Practical Work based on DCEPHS-108N and DCEPHS-109N	2	
		110(P)N			
		DCECHE -108N	Organic Chemistry III (Selected Topics In Organic Chemistry)	2	
	Group-B	DCECHE -109N	Physical Chemistry III (Selected Topics In Physical Chemistry)	2	
VI	Group-D	DCECHE-	Practical Work based on DCECHE -108N and DCECHE -109N	2	
		110PN			
		DCEMM -112N	Advance Analysis	2	
	Group-C	DCEMM -113N	Function of Complex Variable	2	
	oroup o	DCEMM-	Viva Voce	2	
		114(P)N			
	Research Project				
	UGRP-102N Research Project on Elected Subject-2			4	
	Total Credit (6 <sup>th</sup> Semester)			20	
Grand Total Credit (All Semesters)				120	

Semester	Course Code	Title of course	Credits		
		Compulsory Core Course			
	UGPHS -101N	Vector, Mechanics and General Physics	2		
	UGCHE-101N	Inorganic Chemistry I (Basic Inorganic Chemistry)	2		
	UGSTAT-101N	Statistical Methods	2		
T	UGPHS -101(P)N	Practical Work based on UGPHS -101N	2		
1	UGCHE-101(P)N	Practical Work based on UGCHE-101N	2		
	UGSTAT-101(P)N	Practical Work based on UGSTAT -101N	2		
		Skill Enhancement Course			
	SBSEVS-01N	Energy Resources and Green Technology	4		
		Ability Enhancement Course			
	AECEG	Ability Enhancement Course in English	4		
	OR AFCHD	OR Ability Enhancement Course in Hindi	OR 4		
	ALCIID	Total Credit (1 <sup>st</sup> Semester)	20		
		Compulsory Core Course			
	UGPHS -102N	Oscillation, waves and electrical circuits	2		
	UGCHE-102N	Organic Chemistry I (Basic Organic Chemistry)	2		
	UGSTAT-102N	Probability, Distribution and Statistical Inference	2		
	UGPHS -102(P)N	Practical Work based on UGPHS -102N	2		
	UGCHE-102(P)N	Practical Work based on UGCHE-102N	2		
п	UGSTAT-102(P)N	Practical Work based on UGSTAT -102N	2		
11		Skill Enhancement Course			
	SBSCHE-02N	Advance Analytical Chemistry	4		
		Ability Enhancement Course			
	AECHRD	Ability Enhancement Course in Human Rights and Duties	4		
		OR			
	AECHH	Ability Enhancement Course in Health & Hygiene Total Credit (2 <sup>nd</sup> Semester)	4		
			20		
	UGPHS -103N	Electromagnetism	2		
	UGCHE-103N	Physical Chemistry I (Basic Physical Chemistry)	2		
III	UGSTAT-103N	Sampling Theory and Design of Experiments	2		
	UGPHS -103(P)N	Practical Work based on UGPHS -103N	2		
	UGCHE-103(P)N	Practical Work based on UGCHE-103N	2		
	0051A1-105(P)N	Skill Enhancement Course	2		
	SBSCHE-01N	Organic Chemistry II (Advance Organic Chemistry)	4		
		Ability Enhancement Course			
	AECEA	Ability Enhancement Course in Environment Awareness	4		
		Or			
	AECSWM	Ability Enhancement Course in Solid Waste Management	4		
			20		
		Compulsory Core Course			
	UGPHS -104N	Analog and Digital Electronics	2		
	UGCHE-104N UGSTAT 104N	Inorganic Chemistry II (Advance Inorganic Chemistry)	2		
	UGPHS -104(P)N	Practical Work based on UGPHS -104N	2		
	UGCHE-104(P)N	Practical Work based on UGCHE-104N	2		
IV	UGSTAT-104(P)N	Practical Work based on UGSTAT -104N	2		
	an antic and	Skill Enhancement Course			
	SBSPHS-02N	Modern physics	4		
	ADONO	Adding Enhancement Course			
	AECNC	Ability Enhancement Course in Nutrition for Community or	4		
	AECDM	Ability Enhancement Course in Disaster Management	4		

### C-9: Combination (Physics, Chemistry, Statistics)

		Total Credit (4 <sup>th</sup> Semester)					
	Skill Enhancement Course						
	SBSMM-03N	SBSMM-03N Elementary Analysis					
*7		Discipline Cent	tric Elective Course (Chose any two group from A, B and E)				
v	Group-A	DCEPHS -105N DCEPHS -106N DCEPHS-107(P)N	Optics Thermal Physics Practical Work based on DCEPHS -105N and DCEPHS -106N	2 2 2			
	Group-B	DCECHE -105N DCECHE -106N DCECHE-107(P)N	Physical Chemistry II (Advance Physical Chemistry) Inorganic Chemistry III (Selected Topics In Inorganic Chemistry) Practical Work based on DCECHE -105N and DCECHE -106N	2 2 2			
	Group-E	DCESTAT -105N DCESTAT -106N DCESTAT- 107(P)N	Advance Statistical Inference Basic Knowledge of Statistical Softwares Practical Work based on DCESTAT -105N and DCESTAT -106N	2 2 2			
			Literature Survey				
	UGLS-101N	Literature Survey of	of Elected Subject-1	4			
			Total Credit (5 <sup>th</sup> Semester)	20			
	Skill Enhancement Course						
	SBSSTAT-04N	Numerical	Methods & Basic Computer Knowledge	4			
		Discipline Cer	ntric Elective Course (Chose any two group as per Sem. V)				
VI	Group-A	DCEPHS -108N DCEPHS -109N DCEPHS-110(P)N	Quantum mechanics and spectroscopy Solid State Physics and Advanced Electronics Practical Work based on DCEPHS 108N and DCEPHS 109N	2 2 2			
	Group-B	DCECHE -108N DCECHE -109N DCECHE-110(P)N	Organic Chemistry III (Selected Topics In Organic Chemistry) Physical Chemistry III (Selected Topics In Physical Chemistry) Practical Work based on DCECHE -108N and DCECHE -109N	2 2 2 2			
	Group-E	DCESTAT-108N DCESTAT -109N DCESTAT- 110(P)N	Official Statistics Operation Research Practical Work based on DCESTAT -108N and DCESTAT -109N	2 2 2			
	Research Project						
	UGRP-102N	GRP-102N Research Project on Elected Subject-2					
	Total Credit (6 <sup>th</sup> Semester)						
Semester	Course Code Title of course						
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	Compulsory Core Course						
	UGPHS -101N	Vector, Mechanics and General Physics	2				
	UGCS-101N	Computer Fundamental & PC Software					
	UGMM-101N	Differential Calculus	2				
	UGMM-102N	Analytical Geometry	2				
I	UGPHS -101(P)N	Practical Work based on UGPHS -101N	2				
	UGCS-101(P)N	Practical Work based on UGCS -101N	2				
		Skill Enhancement Course					
	SBSEVS-01N Energy Resources and Green Technology						
		Ability Enhancement Course					
	AECEG	Ability Enhancement Course in English	4				
	OR	OR	OR				
	AECHD	Ability Enhancement Course in Hindi	4				
		Total Credit (1 <sup>st</sup> Semester)	20				
		Compulsory Core Course					
	UGPHS -102N	Oscillation, waves and electrical circuits	2				
	UGCS-102N	C Programming	2				
	UGMM-103N	Integral Calculus	2				
	UGMM-104N	Differential Equation	2				
	UGPHS -102(P)N	Practical Work based on UGPHS -102N	2				
	UGCS-102(P)N	Practical Work based on UGCS -102	2				
II		Skill Enhancement Course					
	SPSCS 02N	Buthon Drogramming	4				
	SDSCS-02N		4				
		Ability Enhancement Course					
	AECHRD	Ability Enhancement Course in Human Rights and Duties	4				
	AECHH	Ability Enhancement Course in Health & Hygiene	4				
		Total Credit (2 <sup>nd</sup> Semester)	20				
		Compulsory Core Course					
	UGPHS -103N	Electromagnetism	2				
TTT	UGCS-103N	Data Structures	2				
111	UGMM-105N UCMM-106N	Mechanics-I (Statics and Dynamics)	2				
	UGMM-100N UGPHS -103(P)N	Practical Work based on LIGPHS -103N	2				
	UGCS-103(P)N	Practical Work based on UGCS -103N	2				
		Skill Enhancement Course					
	SBSCS-01N	Discrete Mathematics	4				
	AECEA	Ability Enhancement Course	4				
	ALCEA	Or	4				
	AECSWM	Ability Enhancement Course in Solid Waste Management	4				
		Total Credit (3 <sup>rd</sup> Semester)	20				
		Compulsory Core Course					
	UGPHS -104N	Analog and Digital Electronics	2				
	UGCS-104N	Introduction to Database Management System	2				
	UGMM-10/N	Linear Algebra	2				
	UGMM-108N UGPHS -104(P)N	Practical Work based on UGPHS -104N	2				
IV	UGCS-104(P)N	Practical Work based on UGCS -104N	2				
11		Skill Enhancement Course					
	SBSPHS-02N	Modern physics	4				
	AFONO	Ability Enhancement Course	4				
	AECNU	Addity Ennancement Course in Nutrition for Community [AECNU]	4				
	AECDM	Ability Enhancement Course in Disaster Management [AECDM]	4				
		Total Credit (4 <sup>th</sup> Semester)	20				

#### C-10: Combination (Physics, Mathematics, Computer Science)

	Skill Enhancement Course					
	SBSMM-03N	Elementary Analy	sis	4		
<b>X</b> 7	]	Discipline Centric Ele	ctive Course (Chose any two group from A, D and C)			
v	Group-A	DCEPHS -105N DCEPHS -106N DCEPHS-107(P)N	Optics Thermal Physics Practical Work based on DCEPHS -105N and DCEPHS -106N	2 2 2		
	Group-D	DCECS -105N DCECS -106N DCECS-107(P)N	Computer Network Operating System Practical Work based on DCECS -106N	2 2 2 2		
	Group-C	DCEMM -109N DCEMM -110N DCEMM-111(P)N	Abstract Algebra Number Theory Viva Voce	2 2 2		
	Literature Survey					
	UGLS-101N Literature Survey of Elected Subject-1					
		•	Total Credit (5 <sup>th</sup> Semester)	20		
	Skill Enhancement Course					
	SBSSTAT-04N Numerical Methods & Basic Computer Knowledge					
	Discipline Centric Elective Course (Chose any two group as per Sem. V)					
	Group-A	DCEPHS -108N DCEPHS -109N DCEPHS-110(P)N	Quantum mechanics and spectroscopy Solid State Physics and Advanced Electronics Practical Work based on DCEPHS-108N and DCEPHS-109N	2 2 2		
VI	Group-D	DCECS -108N DCECS -109N DCECS-110(P)	C <sup>++</sup> and Object Oriented Programming Software Engineering Practical Work based on DCECS -108N	2 2 2		
	Group-C	DCEMM -112N DCEMM -113N DCEMM-114(P)N	Advance Analysis Function of Complex Variable Viva Voce	2 2 2		
			Research Project			
	UGRP-102N	Research Project of	n Elected Subject-2	4		
	Total Credit (6 <sup>th</sup> Semester)					

Semester	Course Code	Course Code Title of course			
		Compulsory Core Course			
	UGPHS -101N	Vector, Mechanics and General Physics	2		
	UGSTAT-101N	Statistical Methods	2		
	UGMM-101N	Differential Calculus	2		
Ŧ	UGMM-102N	Analytical Geometry			
1	UGPHS -101(P)N	Practical Work based on UGPHS -101N	2		
	UGSTAT-101(P)N	Practical Work based on UGSTAT -101N	2		
		Skill Enhancement Course			
	SBSEVS-01N	Energy Resources and Green Technology	4		
		Ability Enhancement Course			
	AECEG	Ability Enhancement Course in English	4		
	OR	OR	OR		
	AECHD	Ability Enhancement Course in Hindi	4		
		lotal Credit (1 <sup>th</sup> Semester)	20		
	LICENSE 100N	Compulsory Core Course	2		
	UGPHS -102N	Uscillation, waves and electrical circuits	2		
	UGSTAT -102N	Probability, Distribution and Statistical Inference	2		
	UGMM-103N	Integral Calculus	2		
	UGMM-104N	Differential Equation	2		
	UGPHS -102(P)N	Practical Work based on UGPHS -102N	2		
TT I	UGSTAT -102(P)N	Practical Work based on UGSTAT -102N	2		
ш		Skill Enhancement Course			
	SBSCS-02N	Python Programming	4		
		Ability Enhancement Course			
	AECHRD	Ability Enhancement Course in Human Rights and Duties	4		
		OR OR	4		
	AECHH	Total Credit (2 <sup>nd</sup> Semester)	<u>4</u> 20		
		Compulsory Core Course	-		
	UGPHS -103N	Electromagnetism	2		
	UGSTAT -103N	Sampling Theory and Design of Experiments	2		
III	UGMM-105N	Mechanics-I (Statics and Dynamics)	2		
	UGMM-106N	Mechanics-II (Dynamics and Hydrodynamics)	2		
	UGPHS -103(P)N	Practical Work based on UGPHS -103N	2		
	UGSTAT -103(P)N	Practical Work based on UGSTAT -103N	2		
	SBSCS 01N	Discrete Mathematics	1		
	50505-011	Ability Enhancement Course	-		
	AECEA	Ability Enhancement Course in Environment Awareness	4		
	AECSWM	Or Ability Enhancement Course in Solid Waste Management	4		
	ALCSWIN	Total Credit (3 <sup>rd</sup> Semester)	20		
		Compulsory Core Course			
	LIGPHS -104N	Analog and Digital Electronics	2		
	UGSTAT -104N	Applied Statistics	2		
	UGMM-107N	Linear Algebra	2		
	UGMM-108N	Calculus of function of several variable and Vector Calculus	2		
	UGPHS -104(P)N	Practical Work based on UGPHS -104N	2		
IV	UGSTAT -104(P)N	Practical Work based on UGSTAT -104N	2		
		Skill Enhancement Course			
	SBSPHS-02N	Modern physics	4		
	AECNC	Ability Enhancement Course	Λ		
	OR	OR	7		
	AECDM	Ability Enhancement Course in Disaster Management [AECDM]	4		
		Total Credit (4 <sup>th</sup> Semester)	20		

C-11: Combination (Physics, Mathematics, Statistics)

		Skill Enhancement Course				
	SBSMM-03N	I-03N Elementary Analysis				
		Discipline Centri	ic Elective Course (Chose any two group from A, E and C)			
v	Crown A	DCEPHS -105N DCEPHS -106N	Optics Thormal Dhypics	2		
	Group-A	DCEPHS-107(P)N	Practical Work based on DCEPHS -105N and DCEPHS -106N	$\frac{2}{2}$		
		DCESTAT -105N	Advance Statistical Inference	2		
	Group-E	DCESTAT- 107(P)N	Basic Knowledge of Statistical Softwares Practical Work based on DCESTAT -105N and DCESTAT -106N	2		
	Crown C	DCEMM -109N DCEMM -110N	Abstract Algebra	2		
	Group-C	DCEMM-111(P)N	Viva Voce	2		
		Literature Survey				
	UGLS-101N	Literature Survey o	4			
			Total Credit (5 <sup>th</sup> Semester)	20		
	Skill Enhancement Course					
	SBSSTAT-04N	SBSSTAT-04N Numerical Methods & Basic Computer Knowledge				
	Discipline Centric Elective Course (Chose any two group as per Sem. V)					
	Group-A	DCEPHS -108N DCEPHS -109N	Quantum mechanics and spectroscopy Solid State Physics and Advanced Electronics	2 2		
		DCEPHS-110(P)N	Practical Work based on DCEPHS-108N and DCEPHS-109N	2		
		DCESTAT -108N DCESTAT -109N	Official Statistics	2		
VI	Group-D	DCESTAT- 110(P)N	Practical Work based on DCESTAT -108N and DCESTAT -109N	2		
		DCEMM -112N	Advance Analysis	2		
	Group-C	DCEMM -113N DCEMM-114(P)N	Function of Complex Variable	2		
			Research Project	2		
	LICER 100N	Descent Duringt	on Elected Subject 2	1		

Semester	Course Code         Title of course				
		Compulsory Core Course			
	UGPHS -101N	Vector, Mechanics and General Physics	2		
	UGCS-101N	Computer Fundamental & PC Software	2		
	UGSTAT-101N	Statistical Methods	2		
	LICOHS 101/D/N	Dractical Work based on LICOHS 101N	2		
I		Destinal Work based on UCCC 101N	2		
	UGCS-101(P)N	Practical work based on UGCS -101N	2		
	UGSTAT-101(P)N	Practical Work based on UGSTAT -101N	2		
		Skill Enhancement Course			
	SBSEVS-01N	Energy Resources and Green Technology	4		
		Ability Enhancement Course			
	AECEG	Ability Enhancement Course in English	4		
	OR	OR	OR		
	AECHD	Ability Enhancement Course in Hindi Total Credit (1st Semester)	4		
		Total Credit (1 Semester)	20		
	LICELIS 102N	Compulsory Core Course	2		
	UGPHS -102N	C Programming	2		
	UGC3-1021V		2		
	UGSTAT-102N	Probability, Distribution and Statistical Inference	2		
	UGPHS -102(P)N	Practical Work based on UGPHS -102N	2		
	UGCS-102(P)N	Practical Work based on UGCS -102N	2		
п	UGSTAT-102(P)N	Practical Work based on UGSTAT -102N	2		
11		Skill Enhancement Course			
	SBSCS-02N Python Programming				
	55555 0211	Ability Enhancement Course			
	AECHPD	Ability Enhancement Course in Human Pichts and Duties	4		
	ALCIND	OR	+		
	AECHH	Ability Enhancement Course in Health & Hygiene	4		
		Total Credit (2 <sup>nd</sup> Semester)	20		
		Compulsory Core Course			
	UGPHS -103N	Electromagnetism	2		
TIT	UGCS-103N	Data Structures	2		
	UGSTAT-TUSN UGPHS -103(P)N	Practical Work based on LIGPHS -103N	2		
	UGCS-103(P)N	Practical Work based on UGCS -103N	2		
	UGSTAT-103(P)N	Practical Work based on UGSTAT -103N	2		
		Skill Enhancement Course			
	SBSCS-01N	Discrete Mathematics	4		
		Ability Enhancement Course			
	AECEA	Ability Enhancement Course in Environment Awareness	4		
		Or			
	AECSWM	Ability Enhancement Course in Solid Waste Management Total Credit (3 <sup>rd</sup> Semester)	4		
			20		
			2		
	UGPHS -104N	Analog and Digital Electronics	2		
	UGSTAT-104N	Applied Statistics	2		
	UGPHS -104(P)N	Practical Work based on UGPHS -104N	2		
<b>TX</b> 7	UGCS-104(P)N	Practical Work based on UGCS -104N	2		
10	UGSTAT-104(P)N	Practical Work based on UGSTAT -104N	2		
	SBSPHS-02N	SKIII Ennancement Course	4		
	5551115 0211	Ability Enhancement Course	-1		
	AECNC	Ability Enhancement Course in Nutrition for Community [AECNC]	4		
	OR	OR			
	AECDM	Ability Enhancement Course in Disaster Management [AECDM]	4		

C-12: Combination (Physics, Statistics, Computer Science)

		Total Credit (4 <sup>th</sup> Semester)				
		Skill Enhancement Course				
	SBSMM-03N	Elementary An	alysis	4		
<b>T</b> 7		Discipline Centri	c Elective Course (Chose any two group from A, B and E)			
v		DCEPHS -105N	Optics	2		
	Group-A	DCEPHS -106N	Thermal Physics	2		
		DCEPHS-107(P)N	Practical Work based on DCEPHS -105N and DCEPHS -106N	2		
	<i>a</i> <b>b</b>	DCECS -105N DCECS 106N	Computer Network	2		
	Group-D	DCECS-100N DCECS-107(P)N	Operating System	2		
		DCESTAT -105N	Advance Statistical Informac	2		
	Group-E	DCESTAT-106N	Basic Knowledge of Statistical Softwares	2		
	Group-L	DCESTAT-107(P)N	Practical Work based on DCESTAT -105N and DCESTAT -106N	2		
			Literature Survey			
	UGLS-101N	Literature Survey of	Elected Subject-1	4		
			Total Credit (5 <sup>th</sup> Semester)	20		
	Skill Enhancement Course					
	SBSSTAT-04N	Numerical Met	hods & Basic Computer Knowledge	4		
	Discipline Centric Elective Course (Chose any two group as per Sem. V)					
		DCEPHS -108N	Quantum mechanics and spectroscopy	2		
	Group-A DC	DCEPHS -109N	Solid State Physics and Advanced Electronics	2		
		DCEPHS-110(P)N	Practical Work based on DCEPHS-108 and DCEPHS-109	2		
	~ ~	DCECS -108N	C <sup>++</sup> and Object Oriented Programming	2		
	Group-D	DCECS-109N DCECS-110(P)N	Software Engineering	2		
VI		DCFSTAT -108N	Official Statistics	2		
	Group-E	DCESTAT -109N	Operation Research	2		
	Oloup-12	DCESTAT-110(P)N	Practical Work based on DCESTAT -108N and DCESTAT -109N	2		
	Research Project					
	UGRP-102N	Research Project on I	Elected Subject-2	4		
		Total Credit (6 <sup>th</sup> Semester)				

Semester	Course Code Title of course				
		Compulsory Core Course			
	UGSTAT -101N	Statistical Methods	2		
	UGCS-101N	Computer Fundamental & PC Software	2		
	UGMM-101N	Differential Calculus	4		
	UGMM-102N	Analytical Geometry			
1	UGSTAT -101(P)N	Practical Work based on UGSTAT -101N	2		
	UGCS-101(P)N	Practical Work based on UGCS -101N	2		
		Skill Enhancement Course			
	SBSEVS-01N	Energy Resources and Green Technology	4		
		Ability Enhancement Course			
	AECEG	Ability Enhancement Course in English	4		
	OR	OR Ability Enhancement Course in Uin di	OR		
	AECHD	Total Credit (1 <sup>st</sup> Semester)	<u>4</u> 20		
		Compulsory Core Course			
	UGSTAT -102N	Probability, Distribution and Statistical Inference	2		
	UGCS-102N	C Programming	2		
	UGMM-103N	Integral Calculus	2		
	UGMM-104N	Differential Equation	2		
	UGSTAT -102(P)N	Practical Work based on UGSTAT -102N	2		
п	UGCS-102(P)N	Practical Work based on UGCS -102N	2		
		Skill Enhancement Course			
	SBSCS-02N Python Programming				
		Ability Enhancement Course			
	AECHRD	Ability Enhancement Course in Human Rights and Duties	4		
	AECHH	Ability Enhancement Course in Health & Hygiene	4		
		Total Credit (2 <sup>nd</sup> Semester)	20		
	LICOTAT 102N	Compulsory Core Course	2		
	UGSTAT-103N UGCS-103N	Sampling Theory and Design of Experiments	2		
III	UGMM-105N	Mechanics-I (Statics and Dynamics)	2		
	UGMM-106N	Mechanics-II (Dynamics and Hydrodynamics)	2		
	UGSTAT -103(P)N	Practical Work based on UGSTAT -103N	2		
	UGCS-103(P)N	Practical Work based on UGCS -103N	2		
	SBSCS-01N	Discrete Mathematics	4		
	51565 011	Ability Enhancement Course	т		
	AECEA	Ability Enhancement Course in Environment Awareness	4		
	AECSWM	Ability Enhancement Course in Solid Waste Management	4		
		Total Credit (3 <sup>rd</sup> Semester)	20		
		Compulsory Core Course			
	UGSTAT -104N	Applied Statistics	2		
	UGCS-104N UGMM-107N	Introduction to Database Management System	2		
	UGMM-108N	Calculus of function of several variable and Vector Calculus	2		
	UGSTAT -104(P)N	Practical Work based on UGSTAT -104N	2		
IV	UGCS-104(P)N	Practical Work based on UGCS -104N	2		
	OD OD IG OON	Skill Enhancement Course	4		
	SBSPHS-02N	Modern physics Ability Enhancement Course	4		
	AECNC	Ability Enhancement Course in Nutrition for Community [AECNC]	4		
	OR	OR	·		
	AECDM	Ability Enhancement Course in Disaster Management [AECDM]	4		
		Total Credit (4 <sup>m</sup> Semester)   Skill Enhancement Course	20		
	SBSMM-03N	Elementary Analysis	4		

#### C-13: Combination (Mathematics, Statistics, Computer Science)

	Discipline Centric Elective Course (Chose any two group from E, D and C)				
V	Group-E	DCESTAT -105N DCESTAT -106N DCESTAT- 107(P)N	Advance Statistical Inference Basic Knowledge of Statistical Softwares Practical Work based on DCESTAT -105N and DCESTAT -106N	2 2 2	
	Group-D	DCECS -105N DCECS -106N DCECS -107(P)N	Computer Network Operating System Practical Work based on 106N	2 2 2	
	Group-C	DCEMM -109N DCEMM -110N DCEMM-111(P)N	Abstract Algebra Number Theory Viva Voce	2 2 2	
			Literature Survey		
	UGLS-101N	Literature Survey of	of Elected Subject-1	4	
		÷	Total Credit (5 <sup>th</sup> Semester)	20	
			Skill Enhancement Course		
	SBSSTAT-04N	Numerical	Methods & Basic Computer Knowledge	4	
	Discipline Centric Elective Course (Chose any two group as per Sem. V)				
	Group-E	DCESTAT -108N DCESTAT -109N	Official Statistics	2	
		DCESTAT- 110(P)N	Practical Work based on DCESTAT -108N and DCESTAT -109N	2	
VI	Group-D	DCECS-108N DCECS -109N DCECS -110(P)N	C <sup>++</sup> and Object Oriented Programming Software Engineering Practical Work based on 108N	2 2 2	
	Group-C	DCEMM -112N DCEMM -113N DCEMM-114(P)N	Advance Analysis Function of Complex Variable Viva Voce	2 2 2	
			Research Project		
	UGRP-102N	Research Project of	n Elected Subject-2	4	
			Total Credit (6 <sup>th</sup> Semester)	20	

Semester	Course Code Title of course				
		Compulsory Core Course			
	UGPHS -101N	Vector, Mechanics and General Physics	2		
	UGCHE-101N	Inorganic Chemistry I (Basic Inorganic Chemistry)	2		
	UGEVS-101N	Fundaments of Environmental Sciences	2		
	UGPHS -101(P)N	Practical Work based on UGPHS -101N	2		
1	UGCHE-101(P)N	Practical Work based on UGCHE-101N	2		
	UGEVS-101(P)N	Practical Work Based on UGEVS-101N	2		
		Skill Enhancement Course			
	SBSEVS-01N	Energy Resources and Green Technology	4		
		Ability Enhancement Course			
	AECEG	Ability Enhancement Course in English	4		
	OR	OR	OR		
	AECHD	Ability Enhancement Course in Hindi Total Credit (1 <sup>st</sup> Semester)	4		
		Compulsory Core Course	20		
	UGPHS -102N	Oscillation, waves and electrical circuits	2		
	UGCHE-102N	Organic Chemistry I (Basic Organic Chemistry)	2		
	UGEVS-102	Ecology and Biodiversity Conservation	2		
	UGPHS -102(P)N	Practical Work based on UGPHS -102N	2		
	UGCHE-102(P)N	Practical Work based on UGCHE-102N	2		
	UGEVS-102(P)N	Practical Work Based on UGEVS-102N	2		
ш		Skill Enhancement Course			
	SBSEVS-02N	Environmental Impact Assessment and Legislation	4		
		Ability Enhancement Course			
	AECHRD	Ability Enhancement Course in Human Rights and Duties	4		
		OR			
	AECHH	Ability Enhancement Course in Health & Hygiene Total Cradit (2 <sup>nd</sup> Somester)	4		
		Complementary Complementary	20		
	LIGPHS -103N	Flectromagnetism	2		
	UGCHE-103N	Physical Chemistry I (Basic Physical Chemistry)	2		
III	UGEVS-103	Environmental Microbiology and Biotechnology	2		
	UGPHS -103(P)N	Practical Work based on UGPHS -103N	2		
	UGCHE-103(P)N	Practical Work based on UGCHE-103N	2		
	UGEVS-105(P)N	Skill Enhancement Course	2		
	SBSCHE-01N	Organic Chemistry II (Advance Organic Chemistry)	4		
		Ability Enhancement Course			
	AECEA	Ability Enhancement Course in Environment Awareness	4		
		Or			
	AECSWM	Ability Enhancement Course in Solid Waste Management	4		
		Compulsory Core Course	20		
	UGPHS -104N	Analog and Digital Electronics	2		
	UGCHE-104N	Inorganic Chemistry II (Advance Inorganic Chemistry)	2		
	UGEVS-104N	Plant Physiology and Biochemistry	2		
	UGPHS -104(P)N	Practical Work based on UGPHS -104N	2		
	UGCHE-104(P)N UGEVS 104(P)N	Practical Work based on UGCHE-104N Practical Work Based on UGEVS 104N	2		
IV	00275-104(1)14	Skill Enhancement Course	2		
	SBSPHS-02N	Modern physics	4		
		Ability Enhancement Course			
	AECNC	Ability Enhancement Course in Nutrition for Community	4		
	AECDM	or Ability Enhancement Course in Disaster Management	4		
		Total Credit (4 <sup>th</sup> Semester)	20		

**C-14: Combination** (Physics, Chemistry, Environmental Science)

			Skill Enhancement Course				
	SBSCHE-02N	Advance A	analytical Chemistry	4			
		Discipline Cen	tric Elective Course (Chose any two group from A, B and E)				
v		DCEPHS -105N	Optics	2			
	Group-A	DCEPHS -106N	Thermal Physics	2			
		DCEPHS-107(P)N	Practical Work based on DCEPHS -105 and DCEPHS -106	2			
		DCECHE -105N	Physical Chemistry II (Advance Physical Chemistry)	2			
	Group-B	DCECHE -106N	Inorganic Chemistry III (Selected Topics In Inorganic Chemistry)	2			
		DCECHE-10/(P)N	Practical Work based on DCECHE -105 and DCECHE -106	2			
		DCEVS-105N	Environmental Pollutions	2			
	<i>a</i> <b>b</b>	DCEVS-106N	Remote Sensing, GIS and Hydrology	2			
	Group-E	DCEVS-107(P)N	Practical Work based on DCEVS-105N & DCEVS-106N	2			
			Literature Survey				
	UGLS-101N Literature Survey		of Elected Subject-1	4			
			Total Credit (5 <sup>th</sup> Semester)	20			
	Skill Enhancement Course						
	SBSSTAT-04N Numerical		Methods & Basic Computer Knowledge	4			
		Discipline Centric Elective Course (Chose any two group as per Sem. V)					
	DCEPHS -108N		Quantum mechanics and spectroscopy	2			
	Group-A DC	DCEPHS -109N	Solid State Physics and Advanced Electronics	2			
		DCEPHS-110(P)N	Practical Work based on DCEPHS-108 and DCEPHS-109	2			
	~ ~	DCECHE -108N	Organic Chemistry III (Selected Topics In Organic Chemistry)	2			
<b>X</b> 7 <b>X</b>	Group-B	DCECHE -109N DCECHE-110(P)N	Physical Chemistry III (Selected Topics In Physical Chemistry)	2			
V1		DEFENS 100	Practical Work based on DCECHE -108 and DCECHE -109	2			
		DCEVS-108	Statistics and Environmental Quality Assessment	2			
	Group-E	DCEVS-109	Environmental geology and earth resources	2			
		DCEVS-110P	Practical work based on DCEVS-108N & DCEVS-109N	2			
		1	Research Project				
	UGRP-102N	Research Project of	on Elected Subject-2	4			
	Total Credit (6 <sup>th</sup> Semester)						

Semester	Course Code Title of course			
		Compulsory Core Course		
	LIGPHS -101N	Vector Mechanics and General Physics	2	
	UGCS-101N	Computer Fundamental & PC Software	2	
	LICENG 101N		2	
	UGEVS-101N	Fundaments of Environmental Sciences	2	
т	UGPHS -101(P)N	Practical Work based on UGPHS -101N	2	
1	UGCS-101(P)N	Practical Work based on UGCS -101N	2	
	UGEVS-101(P)N	Practical Work Based on UGEVS-101N	2	
		Skill Enhancement Course		
	SBSEVS-01N	Energy Resources and Green Technology	4	
	5555115 0111	Ability Enhancement Course		
	AFOFO		4	
	AECEG	Ability Enhancement Course in English	4 OR	
	AECHD	Ability Enhancement Course in Hindi	4	
		Total Credit (1 <sup>st</sup> Semester)	20	
		Compulsory Core Course		
	UGPHS -102N	Oscillation, waves and electrical circuits	2	
	UGCS-102N	C Programming	2	
	LIGEVS-102	Ecology and Biodiversity Conservation	2	
	LICPUS 102/D/N	Proof of Work based on LICENS 102N	2	
	00FH3 -102(F)N		2	
	UGCS-102(P)N	Practical Work based on UGCS -102N	2	
п	UGEVS-102(P)N	Practical Work based on UGEVS-102N	2	
		Skill Enhancement Course		
	SBSEVS-02N	Environmental Impact Assessment and Legislation	4	
		Ability Enhancement Course		
	AECUDD	Ability Enhancement Course in Human Biohts and Dution	4	
	AECHKD	OR	4	
	AECHH	Ability Enhancement Course in Health & Hygiene	4	
		Total Credit (2 <sup>nd</sup> Semester)	20	
		Compulsory Core Course		
	UGPHS -103N	Electromagnetism	2	
	UGCS-103N	Data Structures	2	
111	UGEVS-103	Environmental Microbiology and Biotechnology	2	
	UGPHS -103(P)N	Practical Work based on UGPHS -103N	2	
	UGCS-103(P)N	Practical Work based on UGCS -103N	2	
	UGEVS-103(P)N	Practical Work Based on UGEVS-103N	2	
	SPSCS 01N	Skill Enhancement Course	4	
	3D3C3-01N		4	
		Ability Enhancement Course		
	AECEA	Ability Enhancement Course in Environment Awareness	4	
	AECSWM	Ability Enhancement Course in Solid Waste Management	4	
		Total Credit (3 <sup>rd</sup> Semester)	20	
		Compulsory Core Course		
	UGPHS -104N	Analog and Digital Electronics	2	
	UGCS-104N	Introduction to Database Management System	2	
	UGEVS-104N	Plant Physiology and Biochemistry	2	
	UGPHS -104(P)N	Practical Work based on UGPHS -104N	2	
	UGCS-104(P)N	Practical Work based on UGCS -104N	2	
IV	UGEVS-104(P)N	Practical Work Based on UGEVS-104N	2	
1		Skill Enhancement Course		
	SBSPHS-02N	Modern physics	4	
	AFONO	Ability Enhancement Course	4	
	AECNC	Ability Enhancement Course in Nutrition for Community [AECNC]	4	
	AECDM	Ability Enhancement Course in Disaster Management [AFCDM]	4	
		Total Credit (4 <sup>th</sup> Semester)	20	

**C-15: Combination** (Physics, Computer Science, Environmental Science)

		Skill Enhancement Course					
	SBSCS-02N		Python Programming	4			
		Discipline Centr	ic Elective Course (Chose any two group from A, B and E)				
v	Group-A	DCEPHS -105N DCEPHS -106N DCEPHS-107(P)N	Optics Thermal Physics	2 2 2			
	Group-D	DCECS -105N DCECS -106N DCECS-107(P)N	Practical Work based on DCEPHS -105N and DCEPHS -106N Computer Network Operating System Practical Work based on 106N	2 2 2 2			
	Group-E	DCEVS-105N DCEVS-106N DCEVS-107P	Environmental Pollutions Remote Sensing, GIS and Hydrology Practical Work based on DCEVS-105N & DCEVS-106N	2 2 2 2			
		·	Literature Survey				
	UGLS-101N						
			Total Credit (5 <sup>th</sup> Semester)	20			
		S	kill Enhancement Course				
	SBSSTAT-04N	Numerical M	lethods & Basic Computer Knowledge	4			
	Discipline Centric Elective Course (Chose any two group as per Sem. V)						
	Group-A	DCEPHS -108N DCEPHS -109N DCEPHS- 110(P)N	Quantum mechanics and spectroscopy Solid State Physics and Advanced Electronics Practical Work based on DCEPHS-108 and DCEPHS-109	2 2 2			
	Group-D	DCECS -108N DCECS -109N DCECS-110(P)N	C <sup>++</sup> and Object Oriented Programming Software Engineering Practical Work based on 108	2 2 2			
VI	Group-E	DCEVS-108N DCEVS-109N DCEVS-110(P)N	Statistics and Environmental Quality Assessment Environmental geology and earth resources Practical Work based on DCEVS-108 & DCEVS-109	2 2 2			
		Research Project					
	UGRP-102N	Research Project of	n Elected Subject-2	4			
		1	Total Credit (6 <sup>th</sup> Semester)	20			

Semester	Course Code	Title of course	Credits
		Compulsory Core Course	
	UGPHS -101N	Vector, Mechanics and General Physics	2
	UGSTAT-101N	Statistical Methods	2
	UGEVS-101N	Fundaments of Environmental Sciences	2
	UGPHS -101(P)N	Practical Work based on UGPHS -101N	2
I	UGSTAT-101(P)N	Practical Work based on UGSTAT -101N	2
			2
		Skill Enhancement Course	2
	SDSEVS 01N	Energy Pasoureas and Green Technology	1
	SDSEVS-01IN	Energy Resources and Green Technology	4
		Ability Enhancement Course	
	AECEG	Ability Enhancement Course in English	4
	AECHD	Ability Enhancement Course in Hindi	
	Theorem	Total Credit (1 <sup>st</sup> Semester)	20
		Compulsory Core Course	-
	UGPHS -102N	Oscillation, waves and electrical circuits	2
	UGSTAT-102N	Probability, Distribution and Statistical Inference	2
	LIGEVS-102	Ecology and Biodiversity Concervation	2
	UCDUS-102		2
	UGPHS -102(P)N	Practical Work based on UGPHS -102N	2
	UGSTAT-102(P)N	Practical Work based on UGSTAT -102N	2
п			2
		Skill Enhancement Course	
	SBSEVS-02N	Environmental Impact Assessment and Legislation	4
		Ability Enhancement Course	
	A E CUID D		
	AECHRD	Ability Enhancement Course in Human Rights and Duties	4
	AECHH	Ability Enhancement Course in Health & Hygiene	4
		Total Credit (2 <sup>nd</sup> Semester)	20
		Compulsory Core Course	
	UGPHS -103N	Electromagnetism	2
TT	UGSTAT-103N	Sampling Theory and Design of Experiments	2
111	UGEVS-103	Environmental Microbiology and Biotechnology	2
	UGPHS -103(P)N UGSTAT-103(P)N	Practical Work based on UGPHS -103N Practical Work based on UGSTAT -103N	2
	UGEVS-103(P)N	Practical Work Based on UGEVS-103N	2
		Skill Enhancement Course	
	SBSCHE-02N	Advance Analytical Chemistry	4
		Ability Enhancement Course	
	AECEA	Ability Enhancement Course in Environment Awareness	4
	ALCEN	Or	·
	AECSWM	Ability Enhancement Course in Solid Waste Management	4
		Total Credit (3 <sup>rd</sup> Semester)	20
		Compulsory Core Course	
	UGPHS -104N	Analog and Digital Electronics	2
	UGSTAT-104N	Applied Statistics	2
	UGEVS-104N	Plant Physiology and Biochemistry	2
	UGPHS -104(P)N	Practical Work based on UGPHS -104N Practical Work based on UGSTAT 104N	2
IV	UGEVS-104(P)N	Practical Work Based on UGEVS-104N	2
		Skill Enhancement Course	-
	SBSPHS-02N	Modern physics	4
		Ability Enhancement Course	
	AECNC	Ability Enhancement Course in Nutrition for Community [AECNC]	4
	AECDM	Ability Enhancement Course in Disaster Management [AECDM]	4

C-16: Combination (Physics, Statistics, Environmental Science)

			Total Credit (4 <sup>th</sup> Semester)	20
			Skill Enhancement Course	
	SBSMM-03N	Elementary A	Analysis	4
		Discipline Centr	ic Elective Course (Chose any two group from A, B and E)	
v		DCEPHS -105N	Optics	2
	Group-A	DCEPHS -106N	Thermal Physics	2
	-	DCEPHS-107(P)N	Practical Work based on DCEPHS -105N and DCEPHS -106N	2
		DCESTAT -105N	Advance Statistical Inference	2
	Group-D	DCESTAT-106N	Basic Knowledge of Statistical Softwares	2
	<b>F</b> -	DCESTAT- 107(P)N	Practical Work based on DCESTAT -105N and DCESTAT -106N	2
		DCEVS-105N	Environmental Pollutions	2
	<b>a b</b>	DCEVS-106N	Remote Sensing, GIS and Hydrology	2
	Group-E	DCEVS-107(P)N	Practical Work based on DCEVS-105N & DCEVS-106N	2
			Literature Survey	
	UGLS-101N	Literature Survey of	of Elected Subject-1	4
			Total Credit (5 <sup>th</sup> Semester)	20
		S	kill Enhancement Course	
	SBSSTAT-04N	Numerical M	lethods & Basic Computer Knowledge	4
		Discipline Cent	ric Elective Course (Chose any two group as per Sem. V)	
		DCEPHS -108N	Quantum mechanics and spectroscopy	2
	Group-A	DCEPHS -109N	Solid State Physics and Advanced Electronics	2
	_	DCEPHS-110(P)N	Practical Work based on DCEPHS-108N and DCEPHS-109N	2
		DCESTAT -108N	Official Statistics	2
	Group-D	DCESTAT -109N	Operation Research	2
	•	DCESTAT- 110(P)N	Practical Work based on DCESTAT -108N and DCESTAT -109N	2
VI		DCEVS-108	Statistics and Environmental Quality Assessment	2
	<b>a b</b>	DCEVS-109	Environmental geology and earth resources	2
	Group-E	DCEVS-110P	Practical Work based on DCEVS-108N & DCEVS-109N	2
			Research Project	
		Research Project of	n Elected Subject-2	4
	UGRP-102N			
			Total Credit (6 <sup>th</sup> Semester)	20

Semester	Course Code	Title of course	Credits
		Compulsory Core Course	
	UGPHS -101N	Vector, Mechanics and General Physics	2
	UGMM-101N	Differential Calculus	2
	UGEVS-101N	Fundaments of Environmental Sciences	2
-	UGPHS -101(P)N	Practical Work based on UGPHS -101	2
1	UGMM-102N	Analytical Geometry	2
	UGEVS-101(P)N	Practical Work Based on UGEVS-101N	2
		Skill Enhancement Course	
	SBSEVS-01N	Energy Resources and Green Technology	4
		Ability Enhancement Course	
	AECEG	Ability Enhancement Course in English	4
	OR	OR	OR
	AECHD	Ability Enhancement Course in Hindi	4
		Total Credit (1 <sup>st</sup> Semester)	20
		Compulsory Core Course	
	UGPHS -102N	Oscillation, waves and electrical circuits	2
	UGMM-103N	Integral Calculus	2
	UGEVS-102N	Ecology and Biodiversity Conservation	2
	UGPHS -102(P)N	Practical Work based on UGPHS -102	2
	UGMM-104N	Differential Equation	2
	UGEVS-102(P)N	Practical Work Based on UGEVS-102N	2
11	. ,	Skill Enhancement Course	
	SBSEVS-02N	Environmental Impact Assessment and Legislation	4
		Ability Enhancement Course	
	AFCHRD	Ability Enhancement Course in Human Rights and Duties	4
	ALCIND	OR	-
	AECHH	Ability Enhancement Course in Health & Hygiene	4
		Total Credit (2 <sup>na</sup> Semester)	20
		Compulsory Core Course	
	UGPHS -103N	Electromagnetism	2
TIT	UGMM-105N	Mechanics-I (Statics and Dynamics)	2
111	UGEVS-105N	Environmental Microbiology and Biotechnology	2
	UGPHS -105(P)N	Practical work based on UGPHS -105	2
	UGMINI-100N	Dreasting Work Deced on LICEVS 102N	2
	UGE V 5-105(P)IN	Practical work based on OGEVS-105N	2
	SBSCHE-02N	Advance Analytical Chemistry	4
	SESCIE 0211	Advance Anarytical Chemistry Ability Enhancement Course	4
	AECEA	Ability Enhancement Course in Environment Awareness	4
	AECSWM	Or Ability Enhancement Course in Solid Words Management	4
	ALCSWW	Total Credit (3 <sup>rd</sup> Semester)	20
		Compulsory Core Course	
	LICPHS 104N	Analog and Digital Electronics	2
	UGI IIS -104IN	Linear Algebra	2
	UCEVS 104N	Diant Divisiology and Dischamistry	2
	UGEVS-104N	Prant Physiology and Diochemistry	2
	UGPHS -104(P)N	Coloring of function of conversional and Martin Coloring	2
	UGIMINI-108IN	Calculus of function of several variable and vector Calculus	2
IV	UGE V 5-104(P)IN	Skill Enhoncoment Course	2
	SBSDHS OOM	SKIII EIIIantement Course	Λ
	SDSFID-02IN	Ability Enhancement Course	4
	AECNC	Ability Enhancement Course in Nutrition for Community [AECNC]	4
	OR	OR	
	AECDM	Ability Enhancement Course in Disaster Management [AECDM]	4
		Total Credit (4 <sup>th</sup> Semester)	20

#### C-17: Combination (Physics, Mathematics, Environmental Science)

			Skill Enhancement Course	
	SBSMM-03N	Elementary Analysi	S	4
		Discipline Centr	ic Elective Course (Chose any two group from A, E and C)	
v	Group-A	DCEPHS -105N DCEPHS -106N DCEPHS-107(P)N	Optics Thermal Physics Practical Work based on DCEPHS -105N and DCEPHS -106N	2 2 2
	Group-E	DCEMM -109N DCEMM -110N DCEMM-111(P)N	Abstract Algebra Number Theory Viva Voce	2 2 2
	Group-C	DCEVS-105N DCEVS-106N DCEVS-107(P)N	Environmental Pollutions Remote Sensing, GIS and Hydrology Practical Work based on DCEVS-105N & DCEVS-106N	2 2 2
			Literature Survey	
	UGLS-101N	Literature Survey of	f Elected Subject-1	4
		4	Total Credit (5 <sup>th</sup> Semester)	20
		S	Skill Enhancement Course	
	SBSSTAT-04N	Numerical Methods	& Basic Computer Knowledge	4
		Discipline Cen	tric Elective Course (Chose any two group as per Sem. V)	
	Group-A	DCEPHS -108N DCEPHS -109N DCEPHS-110(P)N	Quantum mechanics and spectroscopy Solid State Physics and Advanced Electronics Practical Work based on DCEPHS-108N and DCEPHS-1098	2 2 2
VI	Group-D	DCEMM -112N DCEMM -113N DCEMM-114(P)N	Advance Analysis Function of Complex Variable Viva Voce	2 2 2
	Group-C	DCEVS-108N DCEVS-109N DCEVS-110(P)N	Statistics and Environmental Quality Assessment Environmental geology and earth resources Practical Work based on DCEVS-108N & DCEVS-109N	2 2 2
			Research Project	
	UGRP-102N	Research Project of	n Elected Subject-2	4
			Total Credit (6 <sup>th</sup> Semester)	20

1.	APPENDIX-I	Detailed Programme Structure & Syllabus
2.	APPENDIX-II	Guidelines for Research Project (UGRP-102N) is available at link:
		http://14.139.237.190/upload pdf/01 02 2023 Guidelines fo Project Lit Survey Dissertation.pdf
3.	APPENDIX-	Guidelines for Preparing report on Literature Survey is available at link:
	III	http://14.139.237.190/upload pdf/01 02 2023 Common Guidelines for Literature Review.pdf
4.	APPENDIX-IV	Internship Policy: Guidelines and Procedures
		(With Effect From Academic Year 2023-24)
		is available at link: <u>http://14.139.237.190/upload_pdf/01_02_2023_Guidelines_for_Internship.pdf</u>

# Year-2023-2024

## Syllabus of B.Sc./B.A. Programme: [Subject Name: Statistics] In accordance with NEP-2020

Year	Semester	Course Code	Title of Paper	Theory/ Practical	Credit	Min. Marks	Max. Marks
		UGSTAT -101 (N)	Statistical Methods	Theory	2	36	100
	I	UGSTAT -101 (NP)	Practical and Viva voce	Practical	2	36	100
I <sup>st</sup> Yeaı	II	UGSTAT -102 (N)	Probability, Distribution and Statistical Inference	Theory	2	36	100
		UGSTAT -102 (NP)	Practical and Viva voce	Practical	2	36	100
-		UGSTAT -103 (N)	Sampling Theory and Design of Experiments	Theory	2	36	100
<sup>nd</sup> Yea		UGSTAT -103 (NP)	Practical and Viva voce	Practical	2	36	100
7		UGSTAT -104(N)	Applied Statistics	Theory	2	36	100
	IV	UGSTAT -104 (NP)	Practical and Viva voce	Practical	2	36	100
			<b>Discipline Centric Elec</b>	ctive Course			
		DECSTAT -105 (N)	Advance Statistical Inference	Theory	2	36	100
	V	DECSTAT -106 (N)	Basic Knowledge of Statistical Softwares	Theory	2	36	100
<u>ب</u>		DECSTAT -107 (NP)	Practical and Viva voce	Practical	2	36	100
Yea			<b>Discipline Centric Elec</b>	ctive Course			
p		DECSTAT -108 (N)	Official Statistics	Theory	2	36	100
m		DECSTAT -109 (N)	Operation Research	Theory	2	36	100
	VI	DECSTAT -110 (NP)	Practical and Viva voce	Practical	2	36	100
			Skill Enhancement	Course			
		SBSSTAT-04 (N)	Numerical Methods & Basic Computer	Theory	4	36	100
			Total Credit		32	540	1500

# Syllabus

# of

# B. Sc. (Statistics) / B. A. (Statistics)

Course pro	erequisites: For the study of the sai	d cours	e, the learner mus	st fulfill all the eligibility criteria
prescribed	by the university for the concerned	course.		
Programme	B.Sc./B.A.	Year:	-	Semester:I
Subject: Stat	istics	-		
<b>Course Code</b>	e: UGSTAT -101 N	Cours	e Title:Statistical	Methods
CO: To sumr	narize the data and to obtain its salient	features	from the vast mass	of original data. To understand the
concept of at	tributes. To understand and analyze and	d also ir	terpret the data the	rough graphical and diagrammatical
representation	of the data. Acquainting the Learner wi	th variou	is statistical method	ls. To introduce students to different
measurement	scales, qualitative and quantitative and d	iscrete a	nd continuous data.	To help learner to organize data into
frequency dis	tribution graphs, including bar graphs,	histogra	ns, polygons, and	Ogives. Students should be able to
understand the	e purpose for measuring central tendency	, variatio	on, skewness and ku	irtosis and should be able to compute
them as well.	Learners should be able to understand a	nd comp	oute various statistic	cal measures of correlation, fitting of
curve and reg	ression, theory of Attributes.			
Course Outc	comes:			
CO1:Learner	r will be able to understand about the	e concep	ot of data collection	on, tabulation and also about its
graphical and	d diagrammatical representation.			
CO 2:This c	ourse provides the knowledge about al	ll measu	res of central tende	ency and measures of dispersion
with its merit	s, demerits and further applications			
CO 3:Learne	r will gain sufficient knowledge about	moment	s, skewness and ku	rtosis.
<b>CO 4:</b> Lerner will able to handle data with use of raw moments for ungrouped data, raw moments for grouped				
data, Central	moments, Interrelationship between	various	moments, effect of	f change of origin and scale on
moments, Ch	arlier's checks, Sheppard''s correction	for mom	ents Factorial mon	nents.
CO 5: Learn	er should persist knowledge of correla	ation, in	ter and intra class	correlation and regression
<b>CO 6:</b> For 1	the qualitative analysis, learner will	able to	understand theory	of attributes and dichotomous
classification	s and measures of association.			
Credits: 2			Type of Course: C	Core
-		1		
Max. Marks	:100	Min. F	assing Marks:36	
Block 1	<b>Data Collection and Its Repre</b>	esentat	tion	
	Data Collection and Tabulation :	Meani	ngs, Definitions	and Applications of Statistics,
Unit I	Measurements and Scale, Measurer	nents of	qualitative data,	Methods of data collection, Types
	of data.		1	
Unit II	<b>Representation of Data- I</b> (Dias	gramm	atical representa	tion): Frequency distribution.
	Tabulation of data. Diagrammati	cal Rei	presentation of d	ata. Bar diagram. Multiple bar
	diagram Divided bar diagram Perc	entage	bar diagram. Pie c	hart, Pictogram, leaf chart
Unit III	Representation of Data- I (Gra	nhical	renresentation).	Graphical representation of
Oline III	frequency distribution Histogram	<b>Pincai</b> Frequen	cy polygon Frequ	lency curve Ogive
Block 2	Massures of Control Tondone	word	Disporsion	
DIOCK 2	Measures of Central Tendend	<i>y anu</i> T		· 1 · 1 · A · · 1 · ·
Unit IV	Measures of Central Tendency:	Types of	of measures of ce	intral tendency, Arithmetic mean,
	Fundamental Theorems on Arithr	netic m	iean, Geometric i	mean, Harmonic mean, Median,
	Mode, Percentiles, Deciles, and Qu	artiles.		
Unit V	Measures of Dispersion : Types of	measur	es of Dispersion, I	Range, Mean Deviation, Variance
	and Standard deviation, Effect of c	change o	of origin and scale	e, Relationship between measures
	of central tendency and measures of	f dispers	sion, Coefficient o	f variation.
Block 3	Moments, Skewness and Kur	tosis		

Unit VI	Moments, Rav ungrouped dat Interrelationshi Charlier's chec	<b>v Moments and Central Moments :</b> Det a, raw moments for grouped data, C p between various moments, effect of ch ks, Sheppard''s correction for moments.	finition of moments, raw moments for entral moments, Factorial moments, hange of origin and scale on moments,
Unit VII	Skewness and coefficient, Boy and scale.	<b>I Kurtosis:</b> Definition of skewness, wley's coefficients, Kurtosis, Measures of	Measures of skewness, Pearson's of Kurtosis, effect of change of origin
Block 4	Correlation a	nd Regression	
Unit VII	II <b>Bivariate Dat</b> correlation, Pro- change of origin	a and Correlation: Scatter Dia operties of correlation coefficient, limits n and scale on correlation coefficient.	gram, Karl Pearson's coefficient of s of correlation coefficient, Effect of
Unit IX	Regression: lines, Regressio	Regressions, linear regression model, on coefficient, Properties of Regression co	principal of least square, Regression pefficients.
Unit X	Correlation ar correlation coef remarks on Intr	<b>Intra Class Correlation:</b> Rank co ficients, rank correlation coefficient for t a-class correlation.	rrelation coefficient, Spearman's rank ied ranks, Intra-class correlation, some
Unit XI	Theory of AttDichotomous Ctables, IndependenceCorrection.	<b>Combinations, Classes</b> Classification, Consistency of data, joint indence and Association of Attributes	and Class frequencies of Attributes, distribution of attributes, Contingency s, Measures of Association, Yates
Sugges	ted Text Book R	eadings:	
• Bhat	B.R, Srivenkatramana	Г and Rao Madhava K.S.(1996): Statistics: А	Beginner's Text, Vol. I, New Age
Interr	national (P) Ltd.		
• Croxt	ton F.E, Cowden D.J an	d Kelin S (1973): Applied General Statistics,	Prentice Hall of India.
• Goon	A.M., Gupta M.K., Da	s Gupta. B. (1991): Fundamentals of Statistic	cs, Vol.I, World Press, Calcutta.
Ande     Coole	rson 1. w and Sciove S.	L (1978) An introduction to the Statistical A	halysis of Data, Houghton Millin Co.
<ul> <li>Cook</li> <li>Moor</li> </ul>	e, Clainer and Clarke () I A M Graybill F A and	Basic Statistical Computing, Chapman and Boes D C (1974): Introduction to the Theorem	nall. ry of Statistics McGraw Hill
Snede	ecor G W and Cochran	W G (1967): Statistical Methods Iowa State	e University Press
<ul> <li>Spieg</li> </ul>	gel, M. R. (1967): Theor	v & Problems of Statistics, Schaum's Publis	hing Series.
Gupta	a S. C. and Kapoor V. K	K. (1999): Fundamentals of Mathematical Sta	tistics, , S. Chand Publication, Delhi
This cou	urse can be opted as	s an elective by the students of follow	ing subjects:
U.G/P.C	b. in Mathematics, D	ata Science, Computer Science, Agricu	ltural Sciences, all Life Sciences and
Social S	ciences subjects stud	ents etc.	
Suggest	ed equivalent online	e courses (MOOCs) for credit transfe	er: NA
Learner Analysis	<b>c can join this for th</b> s, Prof. Soumen Mait	<b>teir own knowledge:</b> <u>https://onlinecour</u>	rses.nptel.ac.in/noc, Regression
Electro	nic media and other	digital components in the curriculur	n:
Choose	any one or more th	an one: (Electronic Media: Audio/Vide	o Lectures, Online
Counsel	ling/Virtual Classes/	E-Contents/e-SLM/OER/supplementar	y links for reference/Video
Confere	ncing/Radio broadca	st/Web Conferencing/ Other electronic	and digital contents)
<u>5. NO.</u>	<u>Course Code</u>	<u>Title</u>	Uploaded Date & link
1.	UGSTAT – 101	Correlation Analysis	uploaded on 25.09.2020, https://youtu.be/G4U98nAh78c

2.	UGSTAT - 101Block - IV, Unit - X(Module - 3)	Statistical Methods, Correlation and Regression (Two Variables and Association),Rank Correlation and Intra Class Correlation	uploaded on <b>05.09.2020,</b> <u>https://youtu.be/L1RLSOjjORs</u>
3.	UGSTAT - 101 Block - IV, Unit - IX (Module - 2)	Statistical Methods, Correlation and Regression (Two Variables and Association), Regression Analysis	uploaded on 27.08.2020, https://youtu.be/s9z9yHhITVE
4.	UGSTAT - 101 Block - IV, Unit - VIII (Module - 1)	Statistical Methods, Correlation and Regression (Two Variables and Association), Bivariate Data and Correlation	uploaded on <b>26.08.2020,</b> <u>https://youtu.be/1Bsi3YW3ySo</u>
5.	UGSTAT – 01 Block – III	Statistical Methods, Moments, Skewness and Kurtosis	uploaded on 18.06.2019, https://youtu.be/SNAZ8U2Mhjl
6.	UGSTAT – 01 Block – II, Unit – II	Statistical Methods, Measures of Central Tendency and Dispersion,Measures of Dispersion	uploaded on <b>11.06.2019</b> , <u>https://youtu.be/E1c9t8kXJOQ</u>
7.	UGSTAT – 01 Block – II, Unit – I	Statistical Methods, Measures of Central Tendency and Dispersion, Measures of Central Tendency	uploaded on <b>05.01.2019</b> , <u>https://youtu.be/JIPZjEyiLA</u>
8.	UGSTAT – 01 Block – I	Statistical Methods, Data Collection and Its Representation	uploaded on 05.01.2019, https://youtu.be/JwKFj74qICQ

Course prerequisites: For the study of the	e said course, the	learner must fulfill all the
eligibility criteria prescribed by the universi	ty for the concerne	ed course.
Programme: B.Sc./B.A.	Year: I	Semester: I
Subject: Statistics		
Course Code: UGSTAT-101NP	Cours	e Title: Practical and Viva voce
Course Objectives: The main objective of this c	ourse is to develop as	skill to: understand the practical
methods and tests related to estimation of real-life	data.	
<ul> <li>Course Outcomes:</li> <li>CO1: Learner should able to solve the numeric tendency.</li> <li>CO2: Learner should able to solve the numeric CO3:Learner should able to solve the numeric CO4:Learner should able to solve the numeric Credits: 2</li> </ul>	nerical problems r ical problems relate cal problems relate cal problems related <b>Type of Course:</b>	elated with measures of central ed with measures of dispersion. d with skewness and kurtosis. I with correlation and regression. Core
Max. Marks: 100	Min. Passing Ma	<b>rks:</b> 36
Practical based on UGSTAT-101N		

Course pre	requisites:For the study	v of the said course, the	earner must fulfill all the eligibility criteria
prescribed b	by the university for the	concerned course.	
<b>Programme:</b>	B.Sc./B.A.	Year:I	Semester:II
Subject: Stati	stics		
<b>Course Code</b>	:UGSTAT -102 (N)	Course Title: Probabilit	y, Distribution and Statistical Inference
Course Obje	ctives: To understand the	concept of random variab	les, probability distributions and expectation.
Understanding	probability theory at basic	and advance level, random	variables and also their convergences at weak
and strong leve	els. Different probability di	stribution (discrete and con	tinuous). Learner will also gain the knowledge
about small sai	nple test, large sample test	as well as non parametric te	sts.
<b>Course Outc</b>	omes:		
CO1:Learner	will able to understand	d the concept of random	n experiment, random variables, probability,
condi	tional probability and also	o Baye's Theorem.	
CO2:Under	this course, learner gain	knowledge about the bas	sics of probability distributions, expectations,
inequ	alities for moments, morr	ent generating functions	etc.
CO3: This co	ourse gives the complete	knowledge about discrete	distributions and their properties and also the
limiti	ng case, relation between	the discrete distributions	and also fitting of distribution.
CO4:This co	urse gives the complete	knowledge about continu	ous distributions and their properties and also
the lir	niting case relation betw	een them and also the lac	k of memory property area property
<b>COE</b> . In this of	ourse learner will have th	a knowledge of the infere	ntial statistics in which they able to understand
about	estimation, procedure o	f estimation, properties of	of estimators and also the properties of good
estima	ators and also knowledge	about the Fisher's transfo	rmations.
CO6:This cou	rse also provide the kno	wledge of testing of hypo	othesis, critical region, types of errors, test of
signifi	cance: which helps for ma	king the scientific and stat	istical decisions.
<b>CO7</b> . This cour	rse gives knowledge about	large sample test naram	etric and nonnarametric tests
Condition 2	se gives knowledge abou		
Creans: 2		Type of C	ourse: core
Max. Marks:	100	Min. Passing Ma	rks:36
Block 1	<b>Probability Theory</b>		
	Random experiment	and Probability: Dete	erministic and random experiments. Sample
Unit I	space, Events, Algebra	of Events, Axiomatic de	finition of Probability, Classical definition of
	Probability, Statistical	definition of probability.	Addition Theorem of Probability
Unit II	Conditional Probabi	<b>ity:</b> Conditional probab	ility. Multiplicative theorem of Probability.
	Independent events. Pa	rtition of sample space. F	ave's Theorem.
Block 2	Probability Distributi	ons and Expectations	
Unit III	Random Variables ar	nd Probability Distributi	ons. Definition and types of random variable
	Cumulative distributio	n function and its prope	rties Probability Mass Function Probability
	Density Function	in runction and its prope	rues, 1100aomry Mass 1 diedon, 1100aomry
Unit IV	Expectation: Defin	ition and types of Math	ematical Expectation Moments in terms of
Chit I V	expectation Mathemat	tical and Multiplication t	becrems of Expectation other theorems on
	expectation, Mathema	itedi didi Multiplication	incorems of Expectation, other medicins on
Unit V	Inequalities for Mom	ents. Cauchy-Schwartz	Inequality Markov's inequality Chebyshev's
	inequality	chus. Cauchy-Schwaltz	inequality, markov s inequality, Chebysilev s
Block 3	Concent of Drobability	Distributions	
Linit VI	Universite Distributi	DISUIDUUUUIS	on Binomial Distribution mean and variance
	of hinomial distribution	ution Momenta Marri	on, Dimonital Distribution, mean and variance
	Multiplicative recent	$\frac{1}{1000}$	momenta Eitting of Discription Distribution
	Numpricative property	, Recurrence relation to	r moments, Fluing of Binomial Distribution,
	POISSON DISTRIBUTION.	oisson distribution as a	minung case of Binomial Distribution, mean

	and variance of Poisson distribution, Moments, Moment Generating Function, Additive and
	Reproductive property, Recurrence relation for moments, fitting of Poisson Distribution.
Unit VII	<b>Discreet Distribution:</b> Geometric Distribution, mean and variance, moment generating
	function of geometric distribution, Negative Binomial Distribution, Moment Generating
	Function, Mean and Variance, Recurrence formulae for negative Binomial Distribution,
	Poisson Distribution as a limiting case of Negative Binomial Distribution, Hyper Geometric
	Distribution, Mean and Variance, Recurrence relation for Hyper Geometric distribution
Unit VIII	<b>Normal Distribution:</b> Normal Distribution and its parameters, Standard Normal Distribution Momente Concreting Function Area Property property and a formal
	Distribution, Moments, Moments Generating Function, Area Property, properties of normal ourse. Standard Scores, Advantages and Characteristics of Z Scores
Unit IX	Continuous Distribution Uniform Distribution Moment Generating Function Distribution
	Function Moments of Uniform Distribution Exponential Distribution Moments Moment
	Generating Function, Lack of Memory Property
Unit V	Sompling Distribution: Sompling distribution of a statistic Darameter Derivation of $x^2$ t. E.
Om A	z distributions. Beta Gamma Chauchy densities
Block 4	Basic Principles of Statistical Inference
Unit XI	<b>Estimation:</b> Point Estimation properties of a good estimator Consistency Unbiasedness
	Efficiency, Sufficiency, Confidence Interval Estimation
Unit XII	<b>Method of Estimation:</b> Procedures of Estimation. Method of Moments, method of
0	Maximum Likelihood, Method of Scoring, Properties of Estimators.
Unit XIII	<b>Testing of Hypothesis:</b> Statistical Hypothesis. Simple and Composite Hypothesis.
	Critical Region, Two kinds of Error, One-tailed and Two-tailed tests, Test of Significance,
	Most Powerful Test, Uniformly Most Powerful Test.
Block 5	Test of Significance
Unit XIV	<b>Exact Tests and Fisher's transformations:</b> Tests of Significance based on Chi-Square
	Linet Tests and Tisher's transformations. Tests of Significance Subed on Significance
	Distribution, Tests of Significance based on t – Distribution, Tests of Significance based on F
	Distribution, Tests of Significance based on Fisher's Z - Distribution.
Unit XV	<ul> <li>Distribution, Tests of Significance based on t – Distribution, Tests of Significance based on F</li> <li>– Distribution, Tests of Significance based on Fisher's Z - Distribution.</li> <li>Large Sample Tests: Testing Significance of Mean, Testing Equality of Means, Testing</li> </ul>
Unit XV	<ul> <li>Distribution, Tests of Significance based on t – Distribution, Tests of Significance based on F</li> <li>– Distribution, Tests of Significance based on Fisher's Z - Distribution.</li> <li>Large Sample Tests: Testing Significance of Mean, Testing Equality of Means, Testing Significance of Proportion, Testing Equality of Proportions, Testing Significance of Standard</li> </ul>
Unit XV	<ul> <li>Distribution, Tests of Significance based on t – Distribution, Tests of Significance based on F – Distribution, Tests of Significance based on Fisher's Z - Distribution.</li> <li>Large Sample Tests: Testing Significance of Mean, Testing Equality of Means, Testing Significance of Proportion, Testing Equality of Proportions, Testing Significance of Standard Deviation.</li> </ul>
Unit XV Unit XVI	<ul> <li>Distribution, Tests of Significance based on t – Distribution, Tests of Significance based on F</li> <li>– Distribution, Tests of Significance based on Fisher's Z - Distribution.</li> <li>Large Sample Tests: Testing Significance of Mean, Testing Equality of Means, Testing Significance of Proportion, Testing Equality of Proportions, Testing Significance of Standard Deviation, Testing Equality of Standard Deviation</li> <li>Non-Parametric Tests: Non Parametric Tests, Sign Test, Wilcoxon Signed- Rank Test,</li> </ul>
Unit XV Unit XVI	<ul> <li>Distribution, Tests of Significance based on t – Distribution, Tests of Significance based on F</li> <li>– Distribution, Tests of Significance based on Fisher's Z - Distribution.</li> <li>Large Sample Tests: Testing Significance of Mean, Testing Equality of Means, Testing Significance of Proportion, Testing Equality of Proportions, Testing Significance of Standard Deviation, Testing Equality of Standard Deviation</li> <li>Non-Parametric Tests: Non Parametric Tests, Sign Test, Wilcoxon Signed- Rank Test, Mann- Whitney U-Test, Run Test.</li> </ul>
Unit XV Unit XVI Suggested	<ul> <li>Distribution, Tests of Significance based on t – Distribution, Tests of Significance based on F</li> <li>– Distribution, Tests of Significance based on Fisher's Z - Distribution.</li> <li>Large Sample Tests: Testing Significance of Mean, Testing Equality of Means, Testing Significance of Proportion, Testing Equality of Proportions, Testing Significance of Standard Deviation, Testing Equality of Standard Deviation</li> <li>Non-Parametric Tests: Non Parametric Tests, Sign Test, Wilcoxon Signed- Rank Test, Mann- Whitney U-Test, Run Test.</li> <li>Text Book Readings:</li> </ul>
Unit XV Unit XVI Suggested • Bhat B.J	<ul> <li>Distribution, Tests of Significance based on t – Distribution, Tests of Significance based on F – Distribution, Tests of Significance based on Fisher's Z - Distribution.</li> <li>Large Sample Tests: Testing Significance of Mean, Testing Equality of Means, Testing Significance of Proportion, Testing Equality of Proportions, Testing Significance of Standard Deviation, Testing Equality of Standard Deviation</li> <li>Non-Parametric Tests: Non Parametric Tests, Sign Test, Wilcoxon Signed- Rank Test, Mann- Whitney U-Test, Run Test.</li> <li>Text Book Readings:</li> <li>R, Srivenkatramana T and Rao Madhava K.S. (1997):Statistics: A Beginner's Text, Vol. II, New</li> </ul>
Unit XV Unit XVI Suggested • Bhat B.J Age Inte	<ul> <li>Distribution, Tests of Significance based on t – Distribution, Tests of Significance based on F – Distribution, Tests of Significance based on Fisher's Z - Distribution.</li> <li>Large Sample Tests: Testing Significance of Mean, Testing Equality of Means, Testing Significance of Proportion, Testing Equality of Proportions, Testing Significance of Standard Deviation, Testing Equality of Standard Deviation</li> <li>Non-Parametric Tests: Non Parametric Tests, Sign Test, Wilcoxon Signed- Rank Test, Mann- Whitney U-Test, Run Test.</li> <li>Text Book Readings:</li> <li>R, Srivenkatramana T and Rao Madhava K.S. (1997):Statistics: A Beginner's Text, Vol. II, New ernational (P) Ltd.</li> </ul>
Unit XV Unit XVI Suggested • Bhat B.J Age Inte • Edward	<ul> <li>Distribution, Tests of Significance based on t – Distribution, Tests of Significance based on F</li> <li>– Distribution, Tests of Significance based on Fisher's Z - Distribution.</li> <li>Large Sample Tests: Testing Significance of Mean, Testing Equality of Means, Testing Significance of Proportion, Testing Equality of Proportions, Testing Significance of Standard Deviation, Testing Equality of Standard Deviation</li> <li>Non-Parametric Tests: Non Parametric Tests, Sign Test, Wilcoxon Signed- Rank Test, Mann- Whitney U-Test, Run Test.</li> <li>Text Book Readings:</li> <li>R, Srivenkatramana T and Rao Madhava K.S. (1997):Statistics: A Beginner's Text, Vol. II, New ernational (P) Ltd.</li> <li>P.J., Ford J.S.and Lin (1974): Probability for Statistical Decision-Making, Prentice Hall.</li> </ul>
Unit XV Unit XVI Suggested • Bhat B.I Age Inte • Edward • Goon A	<ul> <li>Distribution, Tests of Significance based on t – Distribution, Tests of Significance based on F – Distribution, Tests of Significance based on Fisher's Z - Distribution.</li> <li>Large Sample Tests: Testing Significance of Mean, Testing Equality of Means, Testing Significance of Proportion, Testing Equality of Proportions, Testing Significance of Standard Deviation, Testing Equality of Standard Deviation</li> <li>Non-Parametric Tests: Non Parametric Tests, Sign Test, Wilcoxon Signed- Rank Test, Mann- Whitney U-Test, Run Test.</li> <li>Text Book Readings:</li> <li>R, Srivenkatramana T and Rao Madhava K.S. (1997):Statistics: A Beginner's Text, Vol. II, New ernational (P) Ltd.</li> <li>P.J., Ford J.S. and Lin (1974): Probability for Statistical Decision-Making, Prentice Hall.</li> <li>.M., Gupta M.K., Das Gupta.B. (1999): Fundamentals of Statistics, Vol.II, World Press,</li> </ul>
Unit XV Unit XVI Suggested • Bhat B.I Age Inte • Edward • Goon A Calcutta	<ul> <li>Distribution, Tests of Significance based on t – Distribution, Tests of Significance based on F – Distribution, Tests of Significance based on Fisher's Z - Distribution.</li> <li>Large Sample Tests: Testing Significance of Mean, Testing Equality of Means, Testing Significance of Proportion, Testing Equality of Proportions, Testing Significance of Standard Deviation, Testing Equality of Standard Deviation</li> <li>Non-Parametric Tests: Non Parametric Tests, Sign Test, Wilcoxon Signed- Rank Test, Mann- Whitney U-Test, Run Test.</li> <li>Text Book Readings:</li> <li>R, Srivenkatramana T and Rao Madhava K.S. (1997):Statistics: A Beginner's Text, Vol. II, New ernational (P) Ltd.</li> <li>P.J., Ford J.S. and Lin (1974): Probability for Statistical Decision-Making, Prentice Hall.</li> <li>.M., Gupta M.K., Das Gupta.B. (1999): Fundamentals of Statistics, Vol.II, World Press, A.</li> </ul>
Unit XV Unit XVI Suggested • Bhat B.I Age Inte • Edward • Goon A Calcutta • Mood A	<ul> <li>Distribution, Tests of Significance based on t – Distribution, Tests of Significance based on F – Distribution, Tests of Significance based on Fisher's Z - Distribution.</li> <li>Large Sample Tests: Testing Significance of Mean, Testing Equality of Means, Testing Significance of Proportion, Testing Equality of Proportions, Testing Significance of Standard Deviation, Testing Equality of Standard Deviation</li> <li>Non-Parametric Tests: Non Parametric Tests, Sign Test, Wilcoxon Signed- Rank Test, Mann- Whitney U-Test, Run Test.</li> <li>Text Book Readings:</li> <li>R, Srivenkatramana T and Rao Madhava K.S. (1997):Statistics: A Beginner's Text, Vol. II, New ernational (P) Ltd.</li> <li>P.J., Ford J.S. and Lin (1974): Probability for Statistical Decision-Making, Prentice Hall.</li> <li>M., Gupta M.K., Das Gupta.B. (1999): Fundamentals of Statistics, Vol.II, World Press, M.</li> <li>M, Graybill F.A and Boes D.C. (1974): Introduction to the Theory of Statistics, McGraw Hill.</li> </ul>
Unit XV Unit XVI Suggested • Bhat B.J Age Inte • Edward • Goon A Calcutta • Mood A • Cooke, •	<ul> <li>Distribution, Tests of Significance based on t – Distribution, Tests of Significance based on F – Distribution, Tests of Significance based on Fisher's Z - Distribution.</li> <li>Large Sample Tests: Testing Significance of Mean, Testing Equality of Means, Testing Significance of Proportion, Testing Equality of Proportions, Testing Significance of Standard Deviation, Testing Equality of Standard Deviation</li> <li>Non-Parametric Tests: Non Parametric Tests, Sign Test, Wilcoxon Signed- Rank Test, Mann- Whitney U-Test, Run Test.</li> <li>Text Book Readings:</li> <li>R, Srivenkatramana T and Rao Madhava K.S. (1997):Statistics: A Beginner's Text, Vol. II, New ernational (P) Ltd.</li> <li>P.J., Ford J.S. and Lin (1974): Probability for Statistical Decision-Making, Prentice Hall.</li> <li>.M., Gupta M.K., Das Gupta.B. (1999): Fundamentals of Statistics, Vol.II, World Press,</li> <li>M, Graybill F.A and Boes D.C. (1974): Introduction to the Theory of Statistics, McGraw Hill.</li> <li>Cramer and Clarke (): Basic Statistical Computing, Chapman and Hall.</li> </ul>
Unit XV Unit XVI Suggested • Bhat B.I Age Inte • Edward • Goon A Calcutta • Mood A • Cooke, 0 • David S	<ul> <li>Distribution, Tests of Significance based on t – Distribution, Tests of Significance based on F – Distribution, Tests of Significance based on Fisher's Z - Distribution.</li> <li>Large Sample Tests: Testing Significance of Mean, Testing Equality of Means, Testing Significance of Proportion, Testing Equality of Proportions, Testing Significance of Standard Deviation, Testing Equality of Standard Deviation</li> <li>Non-Parametric Tests: Non Parametric Tests, Sign Test, Wilcoxon Signed- Rank Test, Mann- Whitney U-Test, Run Test.</li> <li>Text Book Readings:</li> <li>R, Srivenkatramana T and Rao Madhava K.S. (1997):Statistics: A Beginner's Text, Vol. II, New ernational (P) Ltd.</li> <li>P.J., Ford J.S.and Lin (1974): Probability for Statistical Decision-Making, Prentice Hall.</li> <li>.M., Gupta M.K., Das Gupta.B. (1999): Fundamentals of Statistics, Vol.II, World Press,</li> <li></li></ul>
Unit XV Unit XVI Suggested • Bhat B.I Age Inte • Edward • Goon A Calcutta • Mood A • Cooke, 0 • David S • Hoel P.C	<ul> <li>Distribution, Tests of Significance based on t – Distribution, Tests of Significance based on F – Distribution, Tests of Significance based on F – Distribution, Tests of Significance based on F – Distribution.</li> <li>Large Sample Tests: Testing Significance of Mean, Testing Equality of Means, Testing Significance of Proportion, Testing Equality of Proportions, Testing Significance of Standard Deviation.</li> <li>Non-Parametric Tests: Non Parametric Tests, Sign Test, Wilcoxon Signed- Rank Test, Mann- Whitney U-Test, Run Test.</li> <li>Text Book Readings:</li> <li>R, Srivenkatramana T and Rao Madhava K.S. (1997):Statistics: A Beginner's Text, Vol. II, New ernational (P) Ltd.</li> <li>P.J., Ford J.S.and Lin (1974): Probability for Statistical Decision-Making, Prentice Hall.</li> <li>.M., Gupta M.K., Das Gupta.B. (1999): Fundamentals of Statistics, Vol.II, World Press,</li> <li>M, Graybill F.A and Boes D.C. (1974): Introduction to the Theory of Statistics, McGraw Hill.</li> <li>(1996): Elementary Probability, Oxford Press.</li> <li>G (1971): Introduction to Mathematical Statistics, Asia Publishing House.</li> </ul>
Unit XV Unit XVI Suggested • Bhat B.I Age Inte • Edward • Goon A Calcutta • Mood A • Cooke, 0 • David S • Hoel P.0 • Meyer F	<ul> <li>Distribution, Tests of Significance based on t – Distribution, Tests of Significance based on F – Distribution, Tests of Significance based on Fisher's Z - Distribution.</li> <li>Large Sample Tests: Testing Significance of Mean, Testing Equality of Means, Testing Significance of Proportion, Testing Equality of Proportions, Testing Significance of Standard Deviation, Testing Equality of Standard Deviation</li> <li>Non-Parametric Tests: Non Parametric Tests, Sign Test, Wilcoxon Signed- Rank Test, Mann- Whitney U-Test, Run Test.</li> <li>Text Book Readings:</li> <li>R, Srivenkatramana T and Rao Madhava K.S. (1997):Statistics: A Beginner's Text, Vol. II, New ernational (P) Ltd.</li> <li>P.J., Ford J.S.and Lin (1974): Probability for Statistical Decision-Making, Prentice Hall.</li> <li>.M., Gupta M.K., Das Gupta.B. (1999): Fundamentals of Statistics, Vol.II, World Press, J.</li> <li>.M., Graybill F.A and Boes D.C. (1974): Introduction to the Theory of Statistics, McGraw Hill.</li> <li>Cramer and Clarke (): Basic Statistical Computing, Chapman and Hall.</li> <li>(1996): Elementary Probability, Oxford Press.</li> <li>G (1971): Introduction to Mathematical Statistics, Asia Publishing House.</li> <li>P.L (1970): Introductory Probability and Statistical applications. Addision Wesley.</li> </ul>
Unit XV Unit XVI Suggested • Bhat B.I Age Inte • Edward • Goon A Calcutta • Mood A • Cooke, 0 • David S • Hoel P.C • Meyer F • Freund J	<ul> <li>Distribution, Tests of Significance based on t – Distribution, Tests of Significance based on F</li> <li>– Distribution, Tests of Significance based on Fisher's Z - Distribution.</li> <li>Large Sample Tests: Testing Significance of Mean, Testing Equality of Means, Testing Significance of Proportion, Testing Equality of Proportions, Testing Significance of Standard Deviation.</li> <li>Non-Parametric Tests: Non Parametric Tests, Sign Test, Wilcoxon Signed- Rank Test, Mann- Whitney U-Test, Run Test.</li> <li>Text Book Readings:</li> <li>R, Srivenkatramana T and Rao Madhava K.S. (1997):Statistics: A Beginner's Text, Vol. II, New ernational (P) Ltd.</li> <li>P.J., Ford J.S. and Lin (1974): Probability for Statistical Decision-Making, Prentice Hall.</li> <li>M., Gupta M.K., Das Gupta.B. (1999): Fundamentals of Statistics, Vol.II, World Press,</li> <li></li></ul>
Unit XV Unit XVI Suggested • Bhat B.I Age Inte • Edward • Goon A Calcutta • Mood A • Cooke, 0 • David S • Hoel P.0 • Meyer F • Freund J • Hodges	<ul> <li>Distribution, Tests of Significance based on t – Distribution, Tests of Significance based on F – Distribution, Tests of Significance based on Fisher's Z - Distribution.</li> <li>Large Sample Tests: Testing Significance of Mean, Testing Equality of Means, Testing Significance of Proportion, Testing Equality of Proportions, Testing Significance of Standard Deviation, Testing Equality of Standard Deviation</li> <li>Non-Parametric Tests: Non Parametric Tests, Sign Test, Wilcoxon Signed- Rank Test, Mann- Whitney U-Test, Run Test.</li> <li>Text Book Readings:</li> <li>R, Srivenkatramana T and Rao Madhava K.S. (1997):Statistics: A Beginner's Text, Vol. II, New ernational (P) Ltd.</li> <li>P.J., Ford J.S. and Lin (1974): Probability for Statistical Decision-Making, Prentice Hall.</li> <li>M., Gupta M.K., Das Gupta.B. (1999): Fundamentals of Statistics, Vol.II, World Press,</li> <li></li></ul>

- Rohatgi V.K (1967): An Introduction to Probability Theory and Mathematical Statistics, John Wiley & Sons.
- Snedecor G.W and Cochran W. G. (1967): Statistical Methods. Iowa State University Press.

#### This course can be opted as an elective by the students of following subjects:

U.G/P.G. in Mathematics, Data Science, Computer Science, Agricultural Sciences, all Life Sciences and Social Sciences subjects students etc

Suggested equivalent online courses (MOOCs) for credit transfer: NA

#### Learner can join this for their own knowledge:

1. <u>https://onlinecourses.nptel.ac.in/noc</u>, Probability and RandomProcesses, Prof Rohit Sinha and Prof. Ribhu

2. https://onlinecourses.nptel.ac.in/noc, Non-parametric Statistical Inference, Prof. Niladri Chatterjee

#### Electronic media and other digital components in the curriculum:

**Choose any one or more than one:**(Electronic Media: Audio/Video Lectures, Online Counselling/Virtual Classes/E-Contents/e-SLM/OER/supplementary links for reference/Video Conferencing/Radio broadcast/Web Conferencing/ Other electronic and digital contents)

<u>S. No.</u>	<u>Course Code</u>	<u>Title</u>	Uploaded Date & link
1.	UGSTAT – 102	Probability Distribution and Statistical Inference.	uploaded on <b>20.09.2021</b> .
	Block – I, Unit – II	Probability Theory,Conditional Probability	https://youtu.be/7P-dvJIEs1g
2.	UGSTAT – 102 Block – I, Unit – I	Probability Distribution and Statistical Inference, Probability Theory,Random Experiments and Probability	uploaded on <b>23.09.2021</b> , <u>https://youtu.be/hr0frb7K-9c</u>

Course prerequisites: For the study of the said course, the learner must fulfill all the				
eligibility criteria prescribed by the university for the concerned course.				
Programme: B.Sc./B.A.	Year	: I Semester: II		
Subject: Statistics				
Course Code: UGSTAT-102NP		Course Title: Practical and Viva voce		
Course Objectives: The main objective of this co	ourse is to de	evelop a skil	l to: understand the practical	
methods and tests related to estimation of real-life	data.			
Course Outcomes:				
<b>CO1:</b> Learner should able to solve the numerical problems related with Probability theory.				
<b>CO2:</b> Learner should able to solve the numerical problems related with distributions.				
<b>CO3:</b> Learner should able to solve the numerical problems related with sampling distributions.				
<b>CO4:</b> Learner should able to solve the numerical problems related with Non parametric tests.				
Credits: 2 Type of Course: Core				
Max. Marks: 100	Marks: 100Min. Passing Marks: 36			
Practical based on UGSTAT-102N				

**Course prerequisites:**For the study of the said course, the learner must fulfill all the eligibility criteria prescribed by the university for the concerned course.

Semester:III

Programme: B.Sc./B.A.

Year: II

Subject: StatisticsCourse Code:UGSTAT -103 (N)Course Title:Sampling Theory and Design of Experiments

**Course Objectives:** To understand the concept of sampling distributions and their applications in statistical inference. To understand the process of hypothesis testing. To have a clear understanding of when to apply various tests of hypothesis about population parameters using sample statistics and draw appropriate conclusions from the analysis. To learn how the mathematical ideas of Statistics carry over into the world of applications. Drawing inference about the unknown population parameters based on random samples. Validating our estimation/ inference about the population using hypothesis testing. To provide tools and techniques for selecting a sample of elements from a target population keeping in mined the objectives to be fulfilled and nature of population. To obtain estimator of the population parameter on the basis of selected sample and study its properties. To understand the knowledge about the principles of design of experiments, linear models and also CRD, RBD and LSD.

#### **Course Outcomes:**

**CO1:** This course gives the concept of population, census and statistic, types of survey, sampling and Sampling over complete enumeration .Simple Random Sampling with and without replacement, Stratified sampling, Systematic Sampling.

**CO2:** Under this course learner will able to understand the knowledge of auxiliary variable, Ratio and Regression Method of estimation, Cluster sampling, Two Stage Sampling, Two Phase Sampling and also Multi Stage Sampling, Non Sampling errors: Response Errors and Non Response Errors.

CO3: Learner will able to understand the Analysis of Variance (ANOVA), and linear model.

**CO4:** Learner will able to understand concept about the Design of Experiments, CRD, RBD and LSD.

Credits: 2	Type of Course: Core		
Max. Marks:	100 Min. Passing Marks: 36		
Block 1	Samplings Theory - I		
	Simple Random Sampling: Advantages of Sampling over Complete Enumeration Sampling		
Unit I	and Non Sampling Errors. Probability or Random Sampling. Bias of an Estimator. Measures of		
	Sampling Error, Simple Random Sampling Without Replacement (SRSWOR).		
Unit II	Stratified Random Sampling:		
	Introduction, Reasons & Advantages of Stratification, Some theorems.		
Unit III	Allocation of Sample Size and Systematic Sampling: Introduction, Equal Allocation,		
	Proportional Allocation, Variance of Stratified Mean under Proportional Allocation, Optimum		
	Allocation, Variance of Stratified Mean under Neyman Allocation, Relationship Among Three		
	Variances, Impact of Arbitrary Allocation, Practical difficulties in Implementing		
	NeymanAlloation, Systematic Random Sampling.		
Block 2	Sampling Theory - II		
Unit IV	Ratio and Regression Methods of Estimation: Introduction, Ratio and Regression Estimators,		
	Approximate Variances of the Ratio Estimators		
Unit V	Cluster and Two Stage Sampling: Cluster Sampling (Equal Cluster- Size), Estimation of		
	Mean with SRS at both Stages, Relative Efficiency of Cluster Sampling, Two Stage Sampling,		
	Estimation of Mean, Optimum Allocation when Cost Fixed and when Variance Fixed, Two-		
	Phase (Double) Sampling for Stratification, Estimation of Mean, Difference between Multistage		
	Sampling and Two Phase Sampling.		
Unit VI	Non- Sampling Errors: Response Error and Non Response Errors: Introduction, Errors		
	in Sampling, Sampling Errors, Non Sampling Errors, Response Errors, Sources of Non Sampling		
	Errors, Method of Minimizing Non- Response Errors.		
Block 3	Design and Analysis of Experiments		
Unit VII	Analysis of Variance, Design of Experiment and Completely Block Design: Analysis of		
	Variance, Linear Models and Analysis of Variance, Design of Experiment, Basic Principles of		

	Design of Experiments, Completely Randomized Design.	
Unit VIII	Randomized Block Design and Latin square Design: Randomized Block Design, Efficiency	
	of RBD, Missing Plot Technique, Latin Square Design, and Efficiency of LSD	
Unit IX	Factorial Experiment:	
	Definition, $2^2$ and $2^3$ factorial experiments with its ANOVA table	
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## **Suggested Text Book Readings:**

- Cochran W.G and Cox G.M (1957): Experimental Designs, John Wiley and Sons.
- Das M.N and Giri (1986): Design and Analysis of Experiments, Springer Verlag
- Murthy M. N (1967): Sampling Theory and Methods, Statistical Publishing Society, Calcutta.
- Sampath S. (2000): Sampling Theory and Methods, Narosa Publishing House.
- Sukhatme B. V (1984): Sample Survey methods and Its Applications, Indian Society of Agricultural Statistics.
- Des Raj (2000): Sample Survey Theory, Narosa Publishing House.
- Goon A. M., Gupta M. K., Das Gupta. B. (1986): Fundamentals of Statistics, Vol. II, World Press, Calcutta.
- Kempthorne O. (1965): The Design and Analysis of Experiments, Wiley Eastern.

This course can be opted as an elective by the students of following subjects:

U.G/P.G. in Agricultural Sciences, all Life Sciences and Social Sciences subjects students etc

Suggested equivalent online courses (MOOCs) for credit transfer:NA

Course prerequisites: For the study of the said course, the learner must fulfill all the				
eligibility criteria prescribed by the university for the concerned course.				
Programme: B.Sc./B.A. Year: II Semester: III				
Subject: Statistics				
Course Code: UGSTAT-103 (N)(P)Course Title: Practical and Viva vol				
Course Objectives: The main objective of this course	e is to develop a	skill to: understand the practical		
methods and tests related to estimation of real-life data				
Course Outcomes:				
<b>CO1:</b> Learner should able to solve the numerical problems related with Simple random sampling theory.				
<b>CO2:</b> Learner should able to solve the numerical	problems relat	ed with stratified sampling.		
<b>CO3:</b> Learner should able to solve the numerical problems related with systematic sampling.				
<b>CO4:</b> Learner should able to solve the numerical problems related with cluster sampling.etc				
<b>CO5:</b> Learner should able to solve the numerical	problems relat	ed with CRD, RBD and LSD.		
<b>CO6:</b> Learner should able to solve the numerical problems related with one way and two way analysis etc				
Credits: 2 Ty	pe of Course:	Core		
Max. Marks: 100 Min. Passing Marks: 36				
Practical based on UGSTAT-103(N)				

Course pre	requisites: For the study of the said course, the learner must fulfill all the eligibility criteria				
prescribed b	by the university for the concerned course.				
Programme:	B.Sc./B.A. Year: II Semester:IV				
Subject: Stati					
Course Code:	<i>Course Title: Applied Statistics</i>				
Course Objec	cuves:				
Course Outco	omes:				
COI: This C	burse gives the complete knowledge about the index number, Price index number, Cost index				
CO2. Under	this course, learner will able to understand the knowledge about the Time Series Analysis, its				
utility	component mathematical models, determination of trends and seasonal indices				
CO3. Learne	r will able to understand about the concept of Demography vital statistics concept of fertility				
morte	ity with their measurements and also knowledge of life table and measures of reproductively				
	may with their measurements and also knowledge of the table and measures of reproductively.				
CO4: Learne	r will also persist the knowledge about Statistical Quality Control, control charts for variables,				
contro	of charts for attributes and also gain the knowledge of principles of acceptance sampling.etc				
Credits: 2	Type of Course: Core				
Max. Marks:	100 Min. Passing Marks:36				
Block 1	Index Numbers				
	Index Number: General Theory: Definition & Construction of an Index number, Price				
Unit I	Relatives, Quantity or Volume Relatives, Value Relatives, Link & Chain Relatives, Problem				
	involved in computation of an Index Number .				
Unit II	Index Numbers: Important Formulae: Introduction, Calculation of Index Number,				
	Laspeyre's, Paasche's, Marshall- Edgeworth's, fisher's formulae, other indices, Quantity Index,				
	Criteria of good Index Number				
Unit III	Price Index Number (CPI) Steps in construction of CPI Use & Limitations of CPI Base Shifting				
	of Index Numbers, Splicing of Index Number Series, Deflating the Index Number, Index of				
	Industrial Production				
Block 2	Time Series Analysis				
Unit IV	<b>Time Series:</b> Introduction, Utility of Time Series Analysis, Component of Time Series,				
	Mathematical Models for Time Series Analysis.				
Unit V	<b>Determination of Trends:</b> Introduction, Graphic Method, Method of Semi Averages, Method				
	of Curve Fitting by the Principle of Least Squares, Method of Moving Averages (when Period is				
	Even & Odd)				
Unit VI	<b>Determination of Seasonal Indices:</b> Introduction, Measurement of Seasonal Indices,				
	Method of Simple Averages, Ratio to Trend Method, Ratio to Moving Average Method, Method				
	of Link Relatives				
Block 3	Demography				
Unit VII	Sources of Demographic Data : Introduction, Demography & Vital Statistics, Sources of Demographic Data, Errors in Data Collection, Evaluation & its Adjustments, Rates & Ratios				
Unit VIII	Measures of Mortality:				
	Introduction, Measures of Mortality, CDR, SDR, StDR, MMR, IMR.				
Unit IX	Measures of Fertility :				
	Introduction, Measures of Fertility, CBR, GFR, ASFR, TFR.				
Unit X	Life Tables:				
11	Introduction, Description & Construction of Complete Life Table, Uses of a Life Table.				
Unit XI	Introduction GDD NDD				
Block 3	Statistical Quality Control				

Unit XII	Introduction of Statistical Quality Control : Introduction, Advantages of Quality Control,				
	Quality Characteristics, Basic Principles & Operating Characteristics of Control Charts, Choice of				
	Control Limits, Sample Size & Sample Frequency, Rational Subgroups, Analysis of Pattern on				
	Control Charts, Rate of Detection of Change in Average Level				
Unit XIII	Control Charts for Variables: Introduction, Control Charts for Mean, Control Charts FDor				
	Range, Control Charts for Standard Deviation				
Unit XIV	Control Charts for Attributes: Introduction, Control Charts for Fraction Defectives, Control				
	Charts for Number of Defectives, Control Charts for Number of Defects				
Unit XV	Principles of Acceptance Sampling:				
	Introduction, AQL, LTPD, Producer's Risk, Consumer's Risk, OC Function, AOQ,				
	Average Total Inspection, Average Sample Number, Single Sampling Plan, Double Sampling				
	Plan, Sampling Inspection by Variables				
Suggested T	ext Book Readings:				

- Croxton F.E and Cowden D.J. (1969) : Applied General Statistics, Prentice Hall of India.
- Goon A.M., Gupta M. K., Das Gupta. B. (1986): Fundamentals of Statistics, Vol. II, World Press, Calcutta.
- Guide to Current Indian Official Statistics: Central Statistical Organization, Govt. ofIndia, New Delhi.
- Saluja M. P () Indian Official statistical Systems, Statistical Publishing Society, Calcutta.
- Srivatava O.S (1983): A Textbook of Demography, Vikas Publishing.
- Gupta and Mukhopadhyay P.P () Applied Statistics, Central Book Agency.
- Pressat R (1978) : STATISTICAL Demography, Methuen and Co. Ltd.

This course can be opted as an elective by the students of following subjects: U.G/P.G. inCommerce, Business studies, population studies, economics, all Life Sciences and Social Sciences subjects students etc

Suggested equivalent online courses (MOOCs) for credit transfer:NA

Course prerequisites: For the study of the said course, the learner must fulfill all the				
eligibility criteria prescribed by the university for the Concerned Course.				
Programme: B.Sc./B.A.Year: IISemester: IV			Semester: IV	
Subject: Statistics				
Course Code: UGSTAT-104NPCourse Title: Practical and Viva v			itle: Practical and Viva voce	
Course Objectives: The main objective of this c	course is to de	evelop a ski	ll to: understand the practical	
methods and tests related to estimation of real-life	data.			
Course Outcomes:				
<b>CO1:</b> Learner should able to solve the numerical problems related with Index number.				
<b>CO2:</b> Learner should able to solve the numerical problems related with Time series analysis.				
<b>CO3:</b> Learner should able to solve the numerical problems related with Demography.				
<b>CO4:</b> Learner should able to solve the numerical problems related with Statistical Quality				
Control.				
Credits: 2 Type of Course: Core				
Max. Marks: 100 Min. Passing Marks: 36				
Practical based on UGSTAT-104N				

Course prerequisites: For the study of the said course, the learner must fulfill all the eligibility criteria				
prescribed by the university for the concerned course.				
Programme:	B.Sc./B.A. Year: III Semester: V			
Subject: Stati	UCSTAT 105 (N) Course Title: A duance Statistical Informace			
Course Code	<b>Course The:</b> Advance Statistical Inference			
Factorization 7	Theorem Complete Sufficient Statistics & Rao Blackwell Theorem Complete Sufficient Statistics			
MP & UMP T	Cests, Nevman- Pearson Lemma, Likelihood Ratio Test & Their Uses, Shortest Unbiased Confidence			
Intervals.				
<b>Course Outc</b>	comes:			
CO1: Under	this course learner will able to understand about the concept of statistical inference, point			
estima	ation, cramer rao inequality and MVUE.			
CO2: Learn	er will able to understand about the Sufficiency and factorization theorem, rao Blackwell			
theore	em and invariance property.			
CO3: This co	ourse also discuss about the minimum variance unbiased estimation, completeness, lehmann Scheffe			
theore	em and Neyman Pearson Lemma.			
CO4: Learne	er should also able to understand about the Neyman- Pearson Lemma, Likelihood Ratio Test & Their			
Uses,	Testing of Means of Normal Population, Confidence Interval & Confidence Coefficient, Neyman's			
Princi	ple of Shortest Confidence Interval, Unbiased Confidence Interval and Shortest Unbiased Confidence			
Interv	al.			
Credits: 2	<b>Type of Course:</b> Discipline Centric Elective			
Max. Marks:	100Min. Passing Marks:36			
(S	yllabi should be framed block wise/unit wise; No of blocks and units may change)			
Block 1	Point Estimation			
	Introduction to Statistical Inference:			
Unit I	Introduction, Parameter & Statistic, Parametric & Non-Parametric Methods, Likelihood			
	Function of Sample Values, Sampling Distribution, Standard Error of the Statistic			
Unit II	Point Estimation & Cramer Rao Inequality:			
	Introduction, Point Estimation, Properties of Estimators, Unbiasedness, Consistency,			
	Efficiency, MVUE, C-R Inequality			
Unit III	Sufficiency & Factorization Theorem:			
	Introduction, Sufficiency, Neymam- Fisher Factorization Theorem, Koopmam's form			
	of the Distribution, Invariance Property of Sufficient Statistics.			
Unit IV	Complete Sufficient Statistics & Rao Blackwell Theorem:			
Dia als 2	Introduction, Complete Family of Distributions, Rao-Blackwell Theorem			
BIOCK 2	MIN Estimatora			
Unit v	NUV Estimators:			
Unit VI	Complete Sufficient Statistics:			
	Introduction Sufficient Statistic & Completeness Lehmann-Scheffe Theorem Construction			
	of LIMVLIE			
Block 3	Testing of Hynothesis - I			
Unit VII	Preliminary Concents in Testing.			
	Introduction Types of Hypothesis Types of Error Critical Region Power Function			
Unit VIII	MP & UMP Tests :			
	Introduction, Most Powerful Test, Uniformly Most Powerful Test			
Block 4	Testing of Hypothesis -II			
Unit IX	Nevman- Pearson Lemma, Likelihood Ratio Test & Their Uses :			

	Introduction, Neyman-Pearson Lemma, Likelihood Ratio Test.		
Unit X Testing of Means of Normal Population :			
Introduction, One Sample Problem, Two Sample Problem			
Unit XI	Interval Estimation :		
	Introduction, Confidence Interval & Confidence Coefficient, C.I. For Sample Mean		
	from a Normal Population, C.I. for differences of Means From Two Normal Population.		
Unit XII	Shortest & Shortest Unbiased Confidence Intervals :		
	Introduction, Intervals of Shortest Length, Neyman's Principle of Shortest Confidence		
	Interval, Unbiased Confidence Interval, Shortest Unbiased Confidence Interval, Case of		
Suggested T	Discreet Random Variables.		
Suggested 1	ext Book Readings:		
• Bhat B.F	rnotional (P) I td		
Age fille	Induotial (r) Lu. D. L. Ford I. S. and I. in (1074): Probability for Statistical Decision Making, Prophico Hall		
Edward	M. Gunta M.K. Das Gunta B. (1900): Fundamentals of Statistics. Vol II. World Press		
<ul> <li>Goon A.</li> <li>Calcutta</li> </ul>	W., Oupla W.K., Das Oupla.B. (1999). Fundamentais of Statistics, Vol.11, World Hess,		
Mood A	M. Gravbill F A and Boes D C. (1974). Introduction to the Theory of Statistics. McGraw Hill		
Cooke (	Cramer and Clarke (): Basic Statistical Computing, Chapman and Hall		
<ul> <li>David S (1996): Elementary Probability Oxford Press</li> </ul>			
• Hoel P.G (1971): Introduction to Mathematical Statistics, Asia Publishing House.			
<ul> <li>Mever P L (1970): Introductory Probability and Statistical applications. Addision Wesley</li> </ul>			
Freund J	<ul> <li>Freund J E (2001): Mathematical Statistics. Prentice Hall of India</li> </ul>		
• Hodges.	• Hodges IL and Lehman EL (1964): Basic Concepts of Probability and Statistics Holden Day		
<ul> <li>Mood A.M. Gravbill F.A and Boes D.C. (1974): Introduction to the Theory of Statistics. McGraw Hil</li> </ul>			
Rohatgi V.K (1967): An Introduction to Probability Theory and Mathematical Statistics. John Wiley &			
Sons.	Sons.		
• Snedecor G.W and Cochran W. G. (1967): Statistical Methods. Iowa State University Press.			
This course	can be opted as an elective by the students of following subjects:		
U.G/P.G. in I	Mathematics, Data Science, Computer Science subjects students etc		

Suggested equivalent online courses (MOOCs) for credit transfer: NA

**Course prerequisites:** For the study of the said course, the learner must fulfill all the eligibility criteria prescribed by the university for the concerned course.

Semester: V

**Programme:** B.Sc./B.A.

Year: III

Subject: Statistics

Course Code: DECSTAT -106 (N) Course Title: Basic Knowledge of Statistical Software

**Course Objectives:** To introduce some advanced statistical computing techniques to extract information, visualization and knowledge about various industries. To learn the principles and methods of data analysis. To provide a basic understanding of methods of analyzing data from different fields. To learn R software. The main objective of this course is to allow the learners to learn the advanced techniques of modeling real data from diverse discipline

### Course Outcomes:

**CO1:** Learner should able to understand about the concept and practical hands on about statistical software. **CO2:** Understand about SPSS, Stata with statistical applications.

CO3: Understand about R and R commander with statistical applications..

CO4: Understand about Matlab and Latex etc..

Credits: 2	]	<b>Type of Course:</b> Discipline Centric Elective		
Max. Marks: 100 Min. P		sing Marks: 36		
Block 1	Statistics with MS Office			
Unit I	<b>MS Office and its components:</b> About Statistical Softwares, its features and the steps for data analysis with related softwares Introduction to system software and application software, word processing software – Microsoft office Word, spread sheet software – Microsoft office excel, presentation software – Microsoft office Power Point (Interface of all the three application software, file handling, editing, formatting and final output). Excel as data base software: cell referencing, concept of list, data sorting and filtering, manipulation of data, naming of cells			
Unit II	Computation with MS Excel: Functions specifically Numeric/Mathematical functions, Statistical			
	Functions, Logical Functions, lookup functions, Statistical Analysis using Excel – Descriptive Statistics, Curve fitting, correlation and regression analysis, graphs			
Block 2	Statistical Computation with R			
Unit III	<b>Basics of R:</b> Basics of R, R Studio and R files, Transformation of Data.	-Commander, creation of data files. Import Export of Data		
Unit IV	Statistical Analysis with R: Statistica	al Analysis using R – Descriptive Statistics, Curve fitting,		
	correlation and regression analysis, graphs.	-		
Unit V	<b>Testing of Hypothesis with R:</b> Testing o	f hypothesis using R.		
Suggested Text Book Readings:				

• Bhat B.R, Srivenkatramana T and Rao Madhava K.S. (1997): Statistics: A Beginner's Text, Vol. II, New Age International (P) Ltd.

- Edward P.J., Ford J.S.and Lin (1974): Probability for Statistical Decision-Making, Prentice Hall.
- Goon A.M., Gupta M.K., Das Gupta.B. (1999): Fundamentals of Statistics, Vol.II, World Press, Calcutta.
- Mood A.M, Graybill F.A and Boes D.C. (1974): Introduction to the Theory of Statistics, McGraw Hill.
- Cooke, Cramer and Clarke (): Basic Statistical Computing, Chapman and Hall.
- David S (1996): Elementary Probability, Oxford Press.
- Hoel P.G (1971): Introduction to Mathematical Statistics, Asia Publishing House.
- Meyer P.L (1970): Introductory Probability and Statistical applications. Addision Wesley.
- Freund J.E (2001): Mathematical Statistics, Prentice Hall of India.
- Hodges J.L and Lehman E.L (1964): Basic Concepts of Probability and Statistics, Holden Day.
- Mood A.M, Graybill F.A and Boes D.C. (1974): Introduction to the Theory of Statistics, McGraw Hill.
- Rohatgi V.K (1967): An Introduction to Probability Theory and Mathematical Statistics, John Wiley & Sons.

• Snedecor G.W and Cochran W. G. (1967): Statistical Methods. Iowa State University Press.

This course can be opted as an elective by the students of following subjects:

U.G/P.G. in Mathematics, Data Science, Computer Science, Agricultural Sciences, all Life Sciences and Social Sciences subjects students etc

Suggested equivalent online courses (MOOCs) for credit transfer: NA

### Learner can join this for their own knowledge:

- 1. <u>https://onlinecourses.nptel.ac.in/noc</u>, Descriptive Statistics With R Software, Prof. Prashant Jha
- 2. <u>https://onlinecourses.nptel.ac.in/noc</u>, Foundations of R Software, Prof. Shalabh
- **3.** <u>https://onlinecourses.nptel.ac.in/noc</u>, Probability –Ii with Examples Using R, Prof Siva Athreya.

Course prerequisites: For the study of the said course, the learner must fulfill all the				
eligibility criteria prescribed by the university for the concerned course.				
Programme: B.Sc./B.A.	Yea	r: III	Semester: V	
Subject: Statistics				
Course Code: DECSTAT-107(N)(P)       Course Title: Practical and Viva voce				
Course Objectives: The main objective of this c	ourse is to	develop a ski	ll to: understand the practical	
methods and tests related to estimation of real-life	data.			
Course Outcomes:				
CO1: Learner should able to solve the numeri	cal proble	ms related w	vith Statistical Inference.	
<b>CO2:</b> Learner should able to solve the numerical problems related based on descriptive statistics				
with R.				
<b>CO3:</b> Learner should able to solve the numerical problems related with Applied Statistics with R.				
<b>CO4:</b> Learner should able to solve the numerical problems related with official statistics with R.				
Credits: 2     Type of Course: Discipline Centric Elective				
Max. Marks: 100 Min. Passing Marks: 36				
Practical work based on DCESTAT-105 (N) and DCESTAT-106 (N)				
**Course prerequisites:**For the study of the said course, the learner must fulfill all the eligibility criteria prescribed by the university for the concerned course.

Programme: B.Sc./B.A.

Year: III

Semester: VI

Subject: Statistics

Course Code: *DECSTAT -108* (*N*) Course Title: *Official Statistics* 

**Course Objectives:** To understand the concept of all theories and their practical knowledge. This program will gives a complete knowledge about the official statistics. To understand the concept of all theories and their practical knowledge. This program will give a complete knowledge about the audit sampling, audit risk, official and Bio statistics.

### **Course Outcomes:**

- **CO1:** Learner will able to understand the knowledge about official statistics, Statistical Methods for Total Quality Management, Quality Systems, ISO 9000 standards, QS 9000 standards. Concept of six-sigma.
- **CO2:** Learner will understand concepts of population and sample need for sampling, census and sample surveys, basic concepts in sampling and designing of large scale surveys, non-sampling errors, randomized response technique (Warner's model only).
- **CO3:** Learner have a concept of need for design of experiments, fundamental principles of design of experiments., Factorial Experiments, 2n, 32 factorial experiments, illustrations, main effects and interactions, confounding and illustrations.
- **CO4:** This course persist the knowledge about statistical audit sampling, its advantage in audit, risk of statistical sampling, attributes vs variable sampling, audit hypothesis, testing of hypothesis, monetary unit sampling, risk based analysis, concept of alpha and beta risk, concept of tolerable misstatement.

Credits: 2	Type of Course: Discipline Centric Elective			
Max. Marks: 100 Min. Passing Marks: 36				
Block 1	Official Statistics			
Unit I	<b>Basics of Official Statistics:</b> About the official Statistics, Use of Statistics in different offices, Census, National Sample Survey Office, Sample Survey Organization, Birth and Death Registration. etc Introduction to Indian and International statistical systems. Role, function and activities of Central and State statistical organizations. Organization of large scale sample surveys. Role of National Sample Survey Organization. General and special data dissemination systems.			
Unit II	Application of Areas and Tools: Population growth in developed and developing			
	countries, evaluation of performance of family welfare programmes, projections of labour			
	force and manpower. Scope and content of population census of India. System of collection			
	of Agricultural Statistics. Crop forecasting and estimation, productivity, fragmentation of			
	notatings, support prices, buffer stocks, impact of irrigation projects. Statistics related to			
	industries, foreign trade and balance of payment, cost of living, inflation, educational and			
I Init III	other social statistics.			
Unit III	in India Methods of collection of official statistics, their reliability and limitations. Principal			
	number in the provided of the provided statistics, then remainly and minimum structure industry trade			
	prices labour and employment transport and communications banking and finance Various			
	official agencies responsible for data collection and their main functions.			
Block 2	Statistical Methods for Total Quality Management			
Unit IV	Objectives and Basics of TQM: Quality Systems, ISO 9000 standards, QS 9000			
	standards. Concept of six-sigma and the Define-Measure-Analyse-Improve-Control			
	Approach. Precision and accuracy in measurement systems. Estimation of Measurement			
	Uncertainty. Total Quality Management			
Unit V	Methodologies of TQM: Process Analysis and Optimization. Quality at Design stage,			

	Quality Function Deployment, Failure Mode and Effect Analysis, Conjoint Analysis. System, parameter and tolerance designs. Planning and analysis of fractional factorial experiments. Basic ideas of response surface methodology and contour plots
Unit VI	Process Quality and Capability Analysis: Quality in manufacturing, control charts for
	attribute and variable characteristics, process adjustments based on control chart evidences.
	Process capability and performance indices. Evolutionary operations. Measuring customer
	satisfaction, American Customer Satisfaction Index Model.
Suggested	Text Book Readings:
• Guide	to current Indian Official Statistics. Central Statistical Organisation, Govt. of India, New Delhi.

• Saluja, M.P. (): Indian official statistical systems. Statistical Publishing Society, Calcutta.

This course can be opted as an elective by the students of following subjects: U.G/P.G. in Business & Commerce, all Life Sciences and Social Sciences subjects students etc Suggested equivalent online courses (MOOCs) for credit transfer: NA

	Introduction, Basic Feasible Solution of a Transportation Problem, Mod		
	Distribution Method (MODI), Vogel's Approximation Method (VAM), Maximiz		
	in a Transportation Problem		
Unit VII	Solution of Assignment Problem With using Hungarian Method : Introduc		
	Solution of an Assignment Problem, Hungarian Method, Maximization in		
	Assignment Problem.		
Block 4	Theory of Games		
Unit	Basic Concepts of Game Theory : Introduction, A Game, Pure & M		
VIII	Strategies, Two- Person Zero- Sum Game, Pay-Off Matrix, Games without Sa		
	Point and Mixed Strategies, Methods of Solving Game Problems.		
Unit IX	Dominance Rule, Equivalence of Rectangular Games with Linear Programm		
	Introduction, Rectangular Games without Saddle Point, Dominance Property		
	reducing the Size of the Game, Solution Methods of Games without Saddle P		
	Equivalence of Rectangular Games with Linear Programming		
Suggested	l Text Book Readings:		
• Taha l	H.A. (1982) Operational Research: An Introduction; Macmillan.		
Hillier	<sup>•</sup> F.S. and Leiberman G.J. (1962) Introduction to Operations Research; Holden Day.		
<ul> <li>Kanti</li> </ul>	Swarup, Gupta, P.K. and Singh, M.M. (1985) Operations Research; Sultan Chand & Sons.		
Philip:	s D.T., Ravindran A. and Solberg J.() Operations Research, Principles and Practice.		
<ul> <li>Churc Wiley</li> </ul>	hman C.W., Ackoff R.L. and Arnoff E.L. (1957) Introduction to Operations Research; John		
<ul> <li>Hadle</li> </ul>	y G. (1964) Non-linear and Dynamic programming;		
Addise	on Wesley Murthy K.G. (1976) Linear and Combinatorial Programming;		
• John \	Wiley Kleinrock L. (1975) Queueing Systems, vol. 1, Theory;		
• John V	Wiley Saaty T.L. (1961) Elements of Queueing Theory with Applications; McGraw Hill		
<ul> <li>Hadle</li> </ul>	y G. and Whitin T.M. (1963) Analysis of Inventory Systems; Prentice Hall		
Starr I	M.K. and Miller D.W. (1962) Inventory Control-Theory and Practice; Prentice Hall		
Mckin	isey J.C.C. (1952) Introduction to the Theory of Games; McGraw Hill		
• Wagn	• Wagner H.M. (1973) Principles of O.R. with Applications to Managerial Decisions; Prentice Hall		
• Gross	, D. Harris, C.M. (1974) Fundamentals of Queueing Theory; John Wiley		
P G in cor	nputer science. Data science Mathematics MBA and engineering students etc.		
	aparter sectore, Data sectore, manematics, mist and engineering students etc.		
Suggested	equivalent online courses (MOOCs) for credit transfer: NA		

Course prerequisites: For the study of the said course, the learner must fulfill all the				
eligibility criteria prescribed by the universi	ty for th	e concerned	course.	
Programme: B.Sc./B.A.	Y	ear: III	Semester: VI	
Subject: Statistics				
Course Code: DECSTAT-110(N)(P)	Course Code: DECSTAT-110(N)(P)Course Title: Practical and Viva voce			
Course Objectives: The main objective of this of	course is t	o develop a sk	ill to: understand the practical	
methods and tests related to estimation of real-life	data.			
Course Outcomes:	Course Outcomes:			
<b>CO1:</b> Learner should able to solve the numerical problems related with Operation research.				
<b>CO2:</b> Learner should able to solve the numerical problems related based on descriptive statistics.				
<b>CO3:</b> Learner should able to solve the numerical problems related with Applied Statistics.				
<b>CO4:</b> Learner should able to solve the numerical problems related with official statistics.				
Credits: 2     Type of Course: Discipline Centric Elective				
Max. Marks: 100 Min. Passing Marks: 36				
Practical work based on DCESTAT-108 (N) and DCESTAT-109 (N)				

Course pre	prerequisites: For the study of the said course, the learner must fulfil	l all the eligibility criteria		
prescribed by the university for the concerned course.				
<b>Programme:</b>	e: B.Sc./B.A. Year: III Semester	er: VI		
Subject: Stati	tatistics			
Course Code	de: SBSSTAT -04 (N) Course Title: Numerical Methods & Basic	c Computer Knowledge		
Course Obje	bjectives: To study the Numerical Analysis, this is the study of algorithm	ithms that use numerical		
approximation	ion for the problems of mathematical analysis and also the basic knowledge of	the computers. To define,		
design and m	model; To analyze; To identify the real life applications of stochastic pro	ocesses. To study various		
Operational Re	Research Techniques and Models.			
Course Outc	itcomes:			
COI: This	s course provides the knowledge of finite differences, interpolation	h with equal and unequal		
interv	ervals, Lagrange's Interpolation.	noog Internet Internetica		
CO2: Under	marical Differentiation and also Numerical Integration	nces, inverse interpolation,		
	merical Differentiation and also Numerical integration.	tions of computers. It gives		
	uners also learn about the infroduction and instory of computers, general	tions of computers. It gives		
the Kr	knowledge about the hardware and system software			
CO4: This	is course also provides the knowledge about the basic computer	programming, concept of		
algori	orithm, flow charts and also programming languages			
Credits: 4	Type of Course: Skill Develo	pment		
Max. Marks:	ks: 100 Min. Passing Marks: 36			
Block 1	Solutions of Non-Linear Equations in one Variable			
	<b>Basic Properties of Equations:</b> Review of Calculus, Round of	ff Error, Truncation Error,		
Unit I	Some properties of equations, Iteration Methods for finding the roc	ots (zero's) of an equation.		
	Convergence Criterion, Initial Approximation to a Root, Bisection N	Aethod		
Unit II	Solutions of Non-Linear Equations: Fixed Point Iteration Mo	ethod, Chord Methods for		
	Finding Roots- Regula Falsi Method, Newton Raphson Method. O	rder of convergence		
Block 2	Finite Differences			
Unit III	Finite Differences: Forward Difference Operator, Difference 1	able, The Operator E, The		
	Operator D, Backward Differences, Factorial Polynomial, Co	entral Differences, Mean		
	Uperator.	Values Newton Cresser		
Unit Iv	Econyard & Deckward Intervals: Introduction, Missing	values, Newton- Gregory		
Unit V	Interpolation with Un Equal Intervals: Introduction Miss	ing Values Properties of		
Unit v	Divided Differences Newton's Divided Differences Intermelating Delynomial Error of the			
	interpolation Polynomial Divided Differences and Derivatives	i orynolinal, Error or the		
Unit VI	Lagrange's Internalation: Introduction Lagrange's Internal	ating Polynomial General		
	Error term or Reminder Term Linear Interpolation error in Linear	Interpolation		
Block 3	Central Differences			
Unit V	Central Difference Internolation Formulae: Introduction G	ouss Forward & Backward		
ome v	Formulae. Stirling's Formula, Bessel's Formula, Bessel's Formula	for halves		
Unit VI	<b>Inverse Interpolation:</b> Inverse Interpolation by Lagrange's met	hod, method of Successive		
	Approximation. Method of Reversion of Series			
Unit VII	<b>Numerical Differentiation:</b> Introduction, Numerical Differentiation	ntial for Equal Intervals.		
	Numerical Differential for Un-Equal Intervals, Approximation For	mulae for the Derivative of		
	a Function.			
Unit VIII	Numerical Integration: Introduction. Tranezoidal Rule. Sir	npson's One-Third Rule.		
	Simpson's Three-Eighth Rule, Waddle's Rule, Eular-Maculerian Fo	ormula.		
Block 4	Solution of Differential Equations			
DIUCK 4				

Unit IX	Numerical Solution of Ordinary Differential Equations-I:			
	(first order) by Picard's Iteration Method Fuler's Method Runge- Kutta Methods- 4 <sup>th</sup> Order			
Unit X	Numerical Solution of Ordinary Differential Equations-II:			
	(second order and simultaneous) by Dicard's Iteration Method Fuler's Method Runge Kutta			
	Mothoda 4 <sup>th</sup> Order			
Dlools 5	Computer			
BIOCK 5	Computer			
Unit XI	Introduction to Computer : Introduction, Characteristics of computer, Historical			
	Evaluation of Computer, Generation of Computers, Classification of computers.			
Unit XII	Hardware: Introduction, CPU, Memory Organization, Input-Output Devices			
Unit XIII	System Software: Introduction, System Software, File Commands, Editing, Commands,			
	Disk Management Commands, Number System			
Block 3	Basics of Computer Programming			
Unit XIV	Algorithm & Flow Charts :			
	Introduction, Algorithm, Flow Charts			
Unit XV	Programming Language:			
	Introduction, Machine Language, Assembly Language & Assembler, High Level Language,			
	Object Oriented Programming, Programming Language Generation			
Suggested Text Book Readings:				
• Guide	to current Indian Official Statistics. Central Statistical Organisation, Govt. of India, New Delhi.			
<ul> <li>Saluia</li> </ul>	M.P. (): Indian official statistical systems. Statistical Publishing Society. Calcutta			
This cours	e can be onted as an elective by the students of following subjects.			
	e can be opted as an elective by the students of fonowing subjects.			
U.G/P.G. in ]	Business & Commerce, all Life Sciences and Social Sciences subjects students etc			
Suggested eq	uivalent online courses (MOOCs) for credit transfer: NA			

## Year-2023-2024

## Syllabus of B.Sc. Programme: [Subject Name: Zoology] In accordance with NEP-2020

Year	Sem.	Course Code	Paper Title	Theory/Practical	Max.	Credits
					Marks.	
	Ι	UGZY-101 (N)	Animal Physilogy	Theory	100	2
		UGZY-101 (N)(P)	Practical Work	Practical Work	100	2
		UGZY-102 (N)	Diversity of Animal Life	Theory	100	2
1	Π	UGZY-102 (N)(P)	Practical Work	Practical Work	100	2
		Skill Enhancement Co	ourse			
		SBSZY-02(N)	Fundamental of Animal Behavior			4
	III	UGZY-103 (N)	Genetic and Cell Biology	Theory	100	2
		UGZY-103(N) (P)	Practical Work	Practical Work	100	2
2	IV	UGZY-104 (N)	Hemichordata and Chordata	Theory	100	2
		UGZY-104 (N)(P)	Practical Work	Practical Work	100	2
		Discipline Centric Elective Course				
		DCEZY-105(N)	Animal Distribution and ecology	Theory	100	2
	V	DCEZY-106 (N)	Taxonomy and Evolution	Theory	100	2
		DCEZY-107 (N)(P)	Practical Work	Practical Work	100	2
		Skill Enhancement Co	ourse			
3		SBSZY-03(N)	Economic Zoology and	Theory	100	4
			environmental biology			
		Discipline Centric Ele	ctive Course			
		DCEZY-108 (N)	Developmental Biology	Theory	100	2
	VI	DCEZY-109 (N)	Molecular Biology and Genetic	Theory	100	2
			Engineering			
		DCEZY-110 (N) (P)	Practical Work	Practical Work	100	2
			Te	otal Marks/Credit	1500	36

# **B.Sc.: Subject: Zoology**

Course prerequ	uisites: To study this course, a student m	nust have qual	ified 10+2 with Biology	
Programme: B.	Sc.	Year: I	Semester: I	
Subject: Zoology				
Course Code: U	Course Code: UGZY-101 (N) Course Title: Animal Physiology			
Course Objectiv	ves- The course aim to develop, to knowle	dge about the	structure and functions of various organs	
in our body.				
Course Outcom	nes: (CO):			
<ul> <li>Anima</li> </ul>	al physiology is comprehensive subject that	t gives in dept	n knowledge of various physiological	
proces	ses of the animal kingdom.			
• Studer	its gain knowledge about the comparative	physiological of	concepts of nutrition, digestion,	
• Studer	the feel confident in teaching physiology as	s well as execu-	tive research projects	
Credits: 02	its reel confident in teaching physiology as		rse: Core	
credits. 02		Type of Cou	<b>132.</b> Core	
Max. Marks: 10	00	Min. Passin	g Marks: 36	
Block 1	Physiology I	•		
	Physiology of Digestion			
	• Nutrition – Carbohydrates, Lip	oids, Proteins, Y	Vitamins and Minerals	
	Feeding Mechanism			
Unit I	Digestive Tract and process of	Digestion		
	Digestive Enzymes, its Regula	tion and Contr	ol	
	• GIT System			
	Absorption of products of Dige	estion		
Unit II	Physiology of Respiration			
Modes of Respiration     Structural Operation of Longer and other Description				
	<ul> <li>Structural Organization of Lungs and other Respiratory Structures</li> <li>Process of Gaseous Exchange</li> <li>Hemoglobin,</li> <li>Basinington Caseo and its Transport</li> </ul>			
	Respiratory Gases and its Transport			
Regulation of Respiration				
Unit III	II Circulatory System			
	General plan of Circulatory S	Systems (Circu	lation)	
	Structure of Mammalian Hea	art		
	Excitation of Heart			
	Cardiac Output			
	Blood Vessels, Arteries, Vei	ns and Capilla	ries	
<ul><li>Blood Flow</li><li>Lymphatic System</li></ul>				
	Haemostatic Mechanisms			
Unit IV	Excretory System			
	Nitrogen Excretion with Form	mation of Amr	nonia	
	Ammonotelic, Ureotelic ,Ure	ecotelic Anima	ls	
	Glomerular Filtration			
	Reabsorption and Secretion i	n Renal Tubul	es	
	Function and Regulation of V	Vertebrate Kid	ney	
Block 2	Physiology II			
Unit V	Osmoregulation			
	Functional Principles of Osm	noregulation ar	d membrane permeability	

	Problems of Osmoregulation			
	Osmoregulation in Aqueous (Fresh, Marine) and Terrestrial Environment			
Unit VI	Nervous System			
	<ul> <li>Nervous System and Nerve Cells</li> </ul>			
	Nerve Impulse, Action Potential			
	Conduction of Nerve Impulse			
	Synaptic Transmission Chemical Synaptic Transmission Pos	t Synantic Potential		
	Nourotransmittars	t Synaptic I otential		
	Neurol Circuits			
	• Neural Circuits			
Unit VII	Muscular System			
	Structure of Vertebrate Skeletal Muscle			
	Mecahnism and Control of Muscle Contraction			
	Initiation of Muscle Contraction			
	Cardiac and Smooth Muscle			
Unit VIII	Endocrine system			
	Hormonal Control Mechanism			
	Chemical Nature, Synthesis and Storage of Hormones			
	• Secretion of Hormones			
	• Steroid			
	Thyroid and Peptide Hormones			
	Neuroendocrine Connection			
	Hypothalamus and Pituitary			
	Regulation of Hormones			
	Pheromones			
Suggested Text	t Book Readings:			
1. Knut S	chmidt-Nielsen: Animal physiology			
2. Philip	C. Withers: Comparative Animal Physiology			
3. Christo	opher D. Moyes and Patricia M. Schulte: Principles of Animal Physiology	ву		
4. Ian Ka	y: Introduction to Animal Physiology			
5. Thoma	Is Mills: A text book of animal physiology			
This course can	be opted as an elective by the students of following subjects: INA			
Electronic med	lia and other digital components in the curriculum:			
Choose any on	e or more than one: (Electronic Media: Audio/Video Lectures Online	e Counselling/Virtual		
Classes/E-Cont	ents/e-SLM/OER/supplementary links for reference/Video Conferencir	g/Radio broadcast/Web		
Conferencing/ C	Other electronic and digital contents)	6		
Name of electro	onic media			
https://youtu.	<u>be/d2ab1v7yIBU</u>			
https://youtu.be/h1PcmyJusQw https://youtu.be/iVV1SViv7nE				
https://youtu	https://youtu.be/IIAu36gcSNtO			
https://voutu.be/E8ns1b0o1s				
e-SLM				

Course prerequi	sites: To study this c	ourse, a student must l	nave qualified 10+2 with B	iology	
Programme: B.S	c.	Year: I	Semest	er: II	
Subject: Zoology	ý				
Course Code: UGZY-102 (N) Course Title: Diversity of Animal Life					
Course Objectiv	es- Students will be ab	le to identify and under	stand the basics of animal bi	ology with a	
comparative kno	wledge on the organiz	ation of various animals	s group.		
Course Outcome	es:(CO):				
Provide	s students with an in-c	lepth knwoldege of dive	ersity of animal life and their	systematic position.	
• To mak	e them aware of the ec	the diversity in non she	some classes.		
Credits: 02	e the students observe		nuales.	Type of Course: Core	
creans. 02				Type of course. core	
Max. Marks: 10	)	Min. Passing Marks:	36	I	
Block 1	<b>Comparative Form</b>	s and Functions-I			
	General characters &	c Classification of Prote	ozoa		
	<ul> <li>Locomotory</li> </ul>	y Organelles			
	Locomotion	1 in Protozoa.			
Unit I	• Viruses- a	Border Line Case betwe	en Living and Non Living t	hings.	
Unit I	Acellular ar	nd Cellular Organisms			
	<ul> <li>Prokaryotes</li> </ul>	and Eukaryotes			
	Biology of	Flagellated Protozoans,	Amoeboid Protozoans, Spor	e Forming Protozoans,	
	Ciliated Pro	tozoans and Parasitic P	rotozoans		
Unit II	Body Organization &	ganization & Characteristic of Metazoa			
	• Symmetry: Asymmetrical, Spherical, Radial, Biradial, Bilateral				
	Developme	Development patterns – Cleavage, Fate of Blastopore and Germ Layers			
	Body Cavit	y – Pseudocoelom and (	Coelom		
	• Origin and	Evolution of Metazoa			
	<ul> <li>Syncytial th</li> </ul>	eory, Colonial Theory,	Polyphyletic Theory		
Unit III	General characters an Nematoda	nd classification of Pori	fera, Cnidaria, Ctenophora,	Platyhelminthes,	
Unit IV	General characters and classification of Phylum Annelida, Arthropoda, Mollusca				
	Torsion and Detosion in molnsca.				
	• Echinoderm	nata- Laval forms in Ech	inodermata		
Block 2	Comparative Form	s and Functions-II			
Unit V	Comparative form an	nd Functions :			
	Locomotion	i : Significance of Hydr	aulic Pressure in Locomotio	n, Locomotion in	
	Coelenterat	es, Flatworms, Nemator	la, Annelida & Arthropoda		
	• Mollusca –	Foot in mollusa as a Cr	eeping and Crawling organ,	burrowing Organ ,	
	Leaping org	gan and Swimming orga	n		
	Ambulance	system in Echinoderma	ıta		
	Feeding and	1 Digestion in Sponges,	Coelenterates		
	• Structure and function of Protonephridia, Metanephridia, MalpighianTtubules and				
	Coelomodu	cts of Molluscs			
Init VI	Respiratory Circula	tory and Nervous system	n		
	Respiratory, Circula     Respiratory	System – Respiratory of	rgans, Process of Respiration	on, Respiratory Pigments	
	Circulatory	System – Open and clo	sed type of Circulatory Syst	em	
	Organosatic	on of Nervous System	Nerve Cell Neuroglia Ga	nolia	
	Norwous See	estem in Platyhalminthe	Annalida Arthronada an	d Molluson	
	• Inervous Sy	atem in riatyneminnes	s, Annenda, Artifopoda an	u monusoa	
Block 3	Adaption and Beha	vioral Pattern			

Unit VII	Reproductive system				
	• Reproductive Sytsem – Formation of Special Reproductive Unit				
	• Asexual Reproduction – The Gemmules, Regeneration, Autolomy and Regeneratio				
	• Epitoky, Polarity and Regeneration				
	Prevalence and its Significance				
	Sexual Reproduction and its Patterns. Sexual Dimorphism, The Reproductive Organs				
	Mating and Fertilization, Ovipary, Vivipary, Ovovivipary, Hermaphroditism				
	Parthenogenesis and Metagenesis				
Unit VIII	Adaptive Radition				
	<ul> <li>Colonial forms among Protozoans and Metozoans</li> </ul>				
	Adaptive Radiations in Annelida, Arthropoda and Mollusca				
	• Flight in Insects, Migration in Insects				
Unit IX	Behavioural patterns				
	<ul> <li>Social organization in insects – Advantage and disadvantage of</li> </ul>	Social Behavior			
	Kinds of Honey Bees, Production of Honey, Composition of H	oney. Honey Production			
	in India				
	<ul> <li>Industrial Products – Silk Lac Bees Way Pearl Sponge Dye</li> </ul>	s and Pigments			
	industrial Floducts Slik, Eac, Dees Wax, Fearl, Sponge, Dye	s and Tigments			
Unit X	Harmful and beneficial Non-Chordates				
	Parasitic PLatyhelminthes – Nematoda				
	Parasitic Nematoda				
	• Economic importance of Arthropods : in agriculture, soil facility, pollination, post				
	management, food chain, scavenger				
~					
Suggested Text	Book Readings:				
1. Darnes e	t al (2009), The Invertebrates: A synthesis, whey backwell 17				
3 Marshall	· Parker & Haswell Text Book of Zoology Vol 1 (7 <sup>th</sup> ed 1972, Macmi	llan)			
4. Moore:	An Introduction to the Invertebrates (2001, Cambridge University Press	s)			
5. Jan Pech	enik (2014) Biology of the invertebrates. McGraw Hill	,			
6. Thomos	C. Chung. General Parasitology. Hardcourt Brace and Co. Ltd., Asia, I	New Delhi.			
7. Bisht. D	S. Apiculture, ICAR Publication.				
This serves ser	be optid as an elective by the students of following subjects: NA				
This course can	be opted as an elective by the students of following subjects: INA				
Suggested equiv	valent online courses (MOOCs) for credit transfer: NA				
Electronic med	ia and other digital components in the curriculum:				
Choose any on	e or more than one: (Electronic Media: Audio/Video Lectures, Online	e Counselling/Virtual			
Classes/E-Cont	ents/e-SLM/OER/supplementary links for reference/Video Conferencin	ng/Radio broadcast/Web			
Conterencing/ C	Other electronic and digital contents)	<b>TX C1</b>			
Name of electro	nic media	Year of incorporation:			
6-SLIVI		2021-22			

Course prerequi	sites: To study this course, a student must have qualified 10+2	with Biology			
Programme: B.Sc. Year: I Semester: III					
Subject: Zoolog	Subject: Zoology				
Course Code: S	BSZY-02 (N) Course Title: Fundamental of Animal	Behavior			
Course Objectiv	es- To understand the natural behavioral of various animals.				
Knowledge the	difference between innate and learned behavioural.				
Course Outcom	es (CO):				
• By the	completion of this course, students will be expected to gain a co	mprehensive understanding of			
the beh	avior of animals.				
To desc	ribe innate Taxes, Reflexes, Instincts and Motivation, Kinesis.				
<ul> <li>To desc</li> </ul>	ribe the social behavior and parental care in fish and amphibia.				
<ul> <li>Underst</li> </ul>	and types of animal behavior and there importance to the organism	IS.			
Credits: 02	Type of Course: Core				
Max. Marks: 10	0 Min. Passing Marks: 36				
Block 1	Fundamental of Animal Behavior – I				
	General Survey of Various Kinds of Animal Behavior				
Unit I	General Survey of Various Kinds of Animal Behavior 7	Types of Innate Behavior - Taxes			
0	Reflexes Instincts and Motivation Kinesis	Types of finate Denavior Taxes,			
Unit II	Types of Rehavior				
olint II	Types of Denavior	ditioned Defleyes, Triel and Error			
	• Types of Learned Benavior Habituation, Imprinting, Con	unioned Renexes, Iriai and Erior,			
Unit III	Latent Learning, Reasoning.				
Unit III	Introduction and basic Mechanism of Benavior				
	• Introduction and Basic Mechanism of Behavior- Role of	Nervous System, Hormones,			
	Pheromones and Genetics in Behavior, Difficulties in St	udying Behavior Study of Human			
	Benavior Se siel Belessier in Lessate				
Unit IV	Social Benavior in Insects				
	<ul> <li>Social Behavior in Insects, Social structure and function</li> </ul>	ing of Bees and Termites.			
Block 2	Fundamental of Animal Behavior - II				
Unit V	Parental Care in Fishes and Amphibia				
	<ul> <li>Parental Care in Fishes and Amphibia, Maternal Vs Pate</li> </ul>	rnal care			
Unit VI	Nest Building				
	Nest Building, Nesting and Brooding Behavior in Birds				
Unit VII	Migration in Fishes and Birds				
	<ul> <li>Migration in Fishes and Birds- physiological and behavi</li> </ul>	oral changes. Cost and benefits of			
	migration.				
Unit VIII	Biological Clock				
	<ul> <li>Biological Clock, Colouration, Mimicry, Adaptation and</li> </ul>	l anti-predator behavior.			
Suggested Text	Book Readings:				
1. Animal beha	avior by Reena Mathur				
2. The marvels	of Animal Behaviour, A publication of National Geographic S	ociety, Washington, DC, USA.			
3. Wildlife We	alth of India (Resources and Management), Edited By T.C. Mo	Jupuria. Published and			
Distributed	By: Lecpress Service, Bangkok, Thailand.				
4. Wildine in I	ndia, By V.B. Sanaria, Natraj Publishers, Denradun.	Designed Dy Hone Johannes			
5. Indian Wild	inite, Edited By Samuel Israel and Toby Sincialr. Directed and I	Designed By Hans Jonannes			
6 Animal beh	apore.				
7 Animal Soc	al Behaviour, By James F Wittenberger Duybury Press, Boston	USΔ			
8 Animal Beh	aviour: An Evolutionary Approach By John Alcock Sinaver A	sociates Inc USA			
9. Sociology, By Edward O. Wilson. The Bellknan Press, USA					
This course can be onted as an elective by the students of following subjects: NA					
Suggested equiv	Suggested equivalent online courses (MOOCs) for credit transfer: $N\Delta$				
Electronic med	ia and other digital components in the curriculum.				
Choose any on	e or more than one: (Electronic Media: Audio/Video Lectures	Online Counselling/Virtual			
Classes/E-Conte	ents/e-SLM/OER/supplementary links for reference/Video Conf	erencing/Radio broadcast/Web			
Conferencing/ (	Other electronic and digital contents)				
Name of electro	Name of electronic media				
e-SLM		2021-22			
- ~					

Course prerequi	sites: To study this co	urse, a student m	ust have qualified 10+	2 with Biology	
Programme: B.	Sc.	Year: I		Semester: III	
Subject: Zoolog	SV				
Course Code: U	GZY-103 (N)	<b>Course Title: G</b>	enetics and Cell Biol	ogy	
Course Objectiv	ves- Students can under	stands the struct	ure and functions of ce	ell organelles.	
To understand t	he cellular components	under lying pro	cess of cell division in	both somatic and germ cell.	
Course Outcomes (CO):				_	
• Structu	ral and functional aspects	der ge-nemuatio	on me i.e. cell concepts	s. ociated with inhorn errors of	
metabo	lism.	der, ge-nemuatic	ins, various causes ass	belated with moorn errors of	
Credits: 02			Type of Course: Co	re	
Max Marke 10	0	Min Passing M	arke: 36		
Block 1	Genetics	will. Fassing wi	uks. 50		
DIOCK I	Molecular basis of ge	netic information	1		
	Genetic Vari	ation. Molecular	basis of genetic inform	nation	
	Human Chro	mosomes and H	uman Chromosomal A	bnormalities	
Unit I	Say Linkaga	and Dotorminati	on in Drosonhila and I	Man	
	Sex Ellikage			vian	
	• Sex Chronian				
Unit II	Dosage Com     Dosage Com	pensation and L	yon's hypothesis		
Unit II	Blood gloup, DNA a				
	Blood group	and haemoglobi	n, Genetics in Man Int	born Errors of Metabolism in Man	
	• DNA and RNA structure				
	Harchey chas	se experiment			
	Replication of	of DNA – Messe	lson and Stahl's Exper	riment	
Unit III	DNA Polymerase and Experiment	l In Vitro DNA S	Synthesis, Transcriptio	n, Genetic Code, Gene Cloning	
	DNA Polymerase and in Vitro DNA synthesis				
	• Transcription				
	Genetic Code				
	Gene Cloning Experiment				
Block 2	Cell Biology				
Unit IV	Cell Biology & Micr	oscopy			
	Definition and history of Cell Biology				
	<ul> <li>Microscopy – Light Microscopy and Electron Microscopy (Fundamental of TEM and SEM)</li> </ul>				
	• Principle of I	Fixation, Staining	g and Autoradiography	<i>i</i>	
Unit V	Plasma Membrane, N	Nucleus and Cell	cycle		
	Cell Cycle – Mitosis and Meiosis, Nucleus, Nuclear Membrane and Nucleolus				
	<ul> <li>Structure and</li> </ul>	l Function of Pla	sma Membrane (Passi	ve Transport and Active	
	Transport)		X	1	
	ENDOPLASMIC RE	ETICULUM, RI	BOSOMES		
	• Endoplasmic	Reticulum – Mo	orphology, Ultrastructu	ıre	
Unit VI	• Types of End	loplasmic Reticu	lum		
		Smooth ER and F	Rough ER		
	• (	Drigin of ER	-		

	• Function of ER			
	• Ribosomes – Occurrence and Distribution			
	• Types Of Ribosomes			
	70s Ribosomes			
	80s Ribosomes			
	Structure of Ribosomes			
	Dissociation and Reconstitution of Ribosomes			
Unit VII	Golgi body & Lysosomes			
	• Golgi Body – Occurrence, Distribution, Morphology, Chemical Composition,			
	Origin and Function			
Lysosomes – Chemical Composition, Lysosomal Enzymes, Lysosomal Membrane				
	<ul> <li>Kinds of Lysosomes – Primary and Secondary Lysosomes</li> </ul>			
	Origin and Function of Lysosomes			
	• Lysosomes and Disease			
Unit VIII	Mitochondria			
	Origin of Mitochondria			
	Mitochondria – Morphology, Chemical Composition			
	Function of Mitochondria			
	Mitochondria as Semi Autonomous Organelles			
Suggested Text 1. Johen Morra 2. Gunter Ed C 3. Frederic He 4. H C MacGra 5. Barbara Han 6. Cell Biology 7. Cytogenetics K Kukda, Aj 8. Genetics and 9. PadapPrajan Jaipur	Book Readings: ow: Eukaryotic Cell Genetics Obe: Cytogenetics: Basic & Applied Aspects cht: Textbook of cytogenetics egor: Introduction to Animal Cytogenetics mkalo : Molecular Cytogenetics. • And Genetics (Hindi) 2/e PBGupta P K (Hindi) rastogi Publications s, Plant Breeding, Evolution and Biostatistics ISBN #: 978-81-301-0066-1Sunil D Purohit &Gotam pex Publishing House d Biotechnology Sunil D Purohit, K. Ahmed &Gotam K Kukda Apex Publishing House an (Hindi) Hardcover – 1 January 2016 by Chandra Prakash Shukl (Author) Pointer Publishers,			
This course can	be opted as an elective by the students of following subjects: NA			
Suggested equiv	valent online courses (MOOCs) for credit transfer: NA			
Electronic medi	a and other digital components in the curriculum:			
Classes/E-Contents/e-SLM/OER/supplementary links for reference/Video Conferencing/Radio broadcast/Web				
Conferencing/ C	Other electronic and digital contents)			
Name of electro	nic media: e-SLM Year of incorporation: 2021-22			

Course prerequ	isites: To study this c	ourse, a student must	have qualified 10+2 wit	h Biology	
Programme: B.	Sc.	Year: I	Sen	nester: IV	
Subject: Zoolog	v	1041.1	ben		
Course Code: U	JGZY-104 (N)	Course Title: HEM	IICHORDATES & CHO	RDATES	
Course Objectiv	Course Objectives- To understand different categories of chordates.				
To understand	general characters of c	chordates and affinitie	es of hemichordates and	chordates.	
To understand	the comparative anato	my of chordates.			
Course Outcom	nes: (CO):				
Impart	s conceptual knowled	ge of vertebrates			
Classif	y phylum protocharda	ata to mammalia.			
• Unders	standing of origin and	salient features of Os	stracodrms to Actinopter	ygii, adaptive radiation of	
amphil	pian, reptiles, birds an	d mammals.			
To mal	ke the student observe	the diversity in chore	dates and their systemati	c position.	
To ma	ke them aware the eco	nomic importance of	some classes.		
Credits: 02		Т	ype of Course: Core		
Max. Marks: 10	00	Min. Passing Marks	s: 36		
Block 1	HEMICHORDATES	5 & CHORDATES			
	Hemichordata and C	Cephalochordata			
TT. '4 T	General Characters	of Hemichordata and	Affinities of Balanoglos	sus	
Unit I	Classification and D	Detailed Study (Habits	s, Morphology, Anatomy	and Physiology ) of	
	Branchiostoma	-			
Unit II	UROCHORDATA				
	Classification and D	Detailed Study (Habits	s, Morphology, Anatomy	y, Physiology and Post	
-	Embryonic Develop	oment) of Herdmania			
Unit III	FISH, Amphibia &	Reptilia			
	Classification and D	Detailed Study (Habits	s, Morphology, Anatomy	and Physiology) of	
	Scoliodon	1.01	A 1 1' 1 .''''		
	General Characters	and Classification of	Amphelia and reptilian u	ip to Order with examples	
Unit IV	General Characters and Classification of Aves Up To Order With Examples, Flying				
Block 2	Functional Anatomy of Chordates				
Unit V	Comparative Anatomy of vertebrates				
Oline v	Histology Compar	ative Study Of Integu	iment And Skeleton		
	Digestive system &	Respiratory system	ment 7 ma Skeleton		
Unit VI	Brief Account of Al	imentary Canal And	Digestive Glands in vert	ebrates	
	Brief Account of Gi	lls and Air Sacs, Swin	m Bladder		
Unit VII	Circulatory system	& Urinogenital syste	em		
	Evolution of Heart	And Aortic Arches in	vertebrates		
	Succession of Kidne	ey, Evolution Of Urin	nogenital Ducts		
Unit VIII	Nervous system &	Sense Organs			
	Comparative Accou	nt Of Brain			
	Types Of Receptors				
Suggested Tex	t Book Readings:				
1. Harvey	y et al: The Vertebrate	Life (2006)			
2. Kennel	th V. Kardong (2015)	vertebrates: Comarat	tive Anatomy, Function, $(1078, ELDS)$	Evolution McGraw Hill	
5. Parker	and Parsons: The Ver	ok of Zoology, vol. II rtabrata Rody (6th ad	1086 CBS Publishing I	anan)	
4. Nomer and raisons. The veneorate body (our ed 1980, CDS Publishing Japan) 5. Young: The life of vertebrates (3rd ed 2006 ELRS/Oxford)					
This course can be onted as an elective by the students of following subjects: $NA$					
Suggested equi	valent online courses	(MOOCs) for credit th	ransfer: NA		
Electronic med	Electronic media and other digital components in the curriculum:				
Choose any one or more than one: (Electronic Media: Audio/Video Lectures. Online Counselling/Virtual					
Classes/E-Cont	ents/e-SLM/OER/sup	plementary links for 1	reference/Video Confere	ncing/Radio broadcast/Web	
Conferencing/ Other electronic and digital contents)					
Name of electro	onic media: e-SLM			Year of incorporation: 2021-22	

Course prerequisites: To study this course, a student must have qualified 10+2 with Biology					
Programme: B.	Sc. Year: I Semester: V				
Subject: Zoolog	SA CONTRACTOR OF C				
Course Code: D	OCEZY-105 (N) Course Title: Animal Distribution and Ecology				
Course Objective ecology and var	ves- Students will be able to distribution of fauna in different realms interaction branches of rious kinds of animals adaptions.				
Course Outcom	es:(CO):				
Knowle	• Knowledge about branches of ecology and animal distribution.				
Knowle	edge of eras and evolution of species.				
Onders     Structu	re and organization of ecosystem with biotic and abiotic component				
Energy	flow and nutrient cycle in ecosystem and various kinds of animal adaptions.				
Comm	unity, population and role of ecology in human welfare.				
Credits: 02	Type of Course: Core $()$ /Elective				
Max. Marks: 10	00 Min. Passing Marks: 36				
Block 1	Animal distribution				
<b>TT</b> • . <b>T</b>	Geological and Geographical				
Unit I	• Animal Distribution – Geological and Geographical Distribution of Animals, with				
Unit II	Eassile Barriers and Dispersal				
Ollit II					
Block 2	Ecology – I				
	Branches and significance of Ecology				
	• Ecology – Definition, Branches of Ecology, Significance of Ecology For Man				
	Growth of Animal Ecology.				
Unit III	• Desert Ecology				
	Pollution Ecology				
	Pollution Ecology				
Unit IV	Atmosphere – Hydrosphere & Lithosphere				
	Various Zone of Atmosphere				
	• Hydrosphere (Water) – Physical and Chemical Properties Of Water				
	Effect of Factor of Aquatic Environment On Aquatic Organisms				
	• Lithosphere (Soil) – Process of Soil Formation				
	• Soil Types, Morphology of Soil				
	<ul> <li>Physical and chemical Properties of Soil</li> </ul>				
	Soil Fauna and Flora				
Unit V	Ecological Environmental Factors				
	<ul> <li>Ecological Environment, Factors (Biotic and abiote) and Limiting Factors</li> </ul>				
	• Component of Ecosystem , Tolerance Range And Limiting Factor , Tropic Level				
Block 3	Ecology – II				
Unit VI	Ecological Pyramids & Biogeochemical Cycle				
	Ecological Pyramids				
	Energy Flow				
	Food Chain and Food Web				
	Biogeochemical Cycle				
Unit VII	Population Ecology				

	• Population Dynamics – Density, Natality, Mortality, Age Distribution, Population Distribution				
	Population Growth – Factors Affecting Biotic Potential , Carrying Capacity				
	Population Regulation				
Unit VIII	Adaptation				
	• Adaptation of Animals In Deserts and Fresh Water				
Unit IX	Wildlife Conservation				
	• Wildlife Conservation – Defining Wildlife, Treats to Wildlife, Measures For Conservation of Wild Life				
Suggested Text	Book Readings:				
1. Clark	e: Elements of Ecology				
2. Euger	ne P. Odum: Ecology				
3. Edmo	ond Hillary: Ecology				
4. Allan	Frewin Jones: Environmental Biology				
5. F.S.	commental Rielegy and Divisionantential Biology (Finiciples of Ecology)				
6. Environmental Biology and Phylogeography ISBN #: 9/8-81-301-0004-7B. L. Chaudhary,					
Outaill K Kukuak Jilenara Kumar Joshi 7 Odum E B Fundamentals of Ecology Latest Ed Saundars					
7. Odum, F.F. Fundamentals of Ecology, Latest Ed., Saunders 8. Sharma, P.D. Elements of Ecology Latest Ed. Rastogi Publications					
9. Amba	isht, R.S. & Ambasht, N.K. A Text Book of Plant Ecology, Latest Ed., CBS Publication &				
Distributors					
10. Mani,	M.S. Bio-Geography of India, Latest Ed., Springer-Verlag.				
This course can	This course can be opted as an elective by the students of following subjects: NA				
Suggested equivalent online courses (MOOCs) for credit transfer: NA					
Electronic media and other digital components in the curriculum:					
Choose any one or more than one: (Electronic Media: Audio/Video Lectures, Online Counselling/Virtual					
Classes/E-Conte	Classes/E-Contents/e-SLM/OER/supplementary links for reference/Video Conferencing/Radio broadcast/Web				
Conferencing/ C	Other electronic and digital contents)				
Name of electro	nic media: e-SLM Year of incorporation: 2021-22				

Course prerequisites: To study this course, a student must have qualified 10+2 with Biology					
Programme: B.	Sc.	Year: I	Sen	nester: V	
Subject: Zoolog	Y				
Course Code: D	CEZY-106 (N)	Course Title:	<b>Taxonomy and Evolution</b>		
Course Objectiv	ves- This paper to intro	ducing the learne	er for the salient features of	Taxonomy and Evolution.	
Course Outcom	es:(CO):				
• To give a through understanding in the functional principles of systematic in which the animals are, to classify according to their characters and international rules of nomenclature.					
Introdu	ction of Elementary sta	atistics			
Unders	tanding of origin of life	e.	1		
Credits: 02			Type of Course: Core		
Max. Marks: 10	0	Min. Passing	Marks: 36		
Block 1	Taxonomy and Evol	ution – I			
	Taxonomy and biolog	gical species con	cept		
IIn:t I	• Principle of S	Systematics and	Taxonomy		
Unit I	<ul> <li>Biological St</li> </ul>	pecies Concept			
	<ul> <li>Taxonomy p</li> </ul>	ractices			
Unit II	Evidences of evolution	on and comparat	ive anatomy		
	<ul> <li>Evidences of</li> </ul>	evolution from	classification (taxonomy).		
	<ul> <li>Comparative</li> </ul>	anatomy conne	cting link homology analo	gy and vestigial organ	
Unit III	Evidences of evolution	on from compara	tive embryology, physiolog	v and biochemistry	
	<ul> <li>Evidences of</li> </ul>	evolution from	comparative embryology	5	
		physiology and	biochemistry		
	Taxonomy and Evol	ution – II	biochemisu y		
Block 2					
Unit IV	Classification and po	pulation taxono	my		
	<ul> <li>Objectives of classification, Theories of classification, grouping and ranking, diversity of individuals, principle of hierachy, population taxonomy, information retrival</li> </ul>				
Unit V	Modern concept in taxonomy				
	• Taxonomic and non-taxonomic attributes, morden concepts in taxonomy.				
Unit VI	International code of Zoological nomenclature				
	• Definitions,	Uses and applica	tion of international code of	f zoological nomenclature	
Unit VII	Elementary Statistics	11			
	• Elementary s	statistics, Mean,	Median and Mode, Measure	s of dispersion variation,	
Unit VIII	Origin of Life, Mutat	ion, Migrations,	Isolation		
	• Origin of life	e, synthetic theor	y of evolution, selection, m	utation, migration, genetic	
Concepted Trant	drit, mimicry	v isolation and sp	beciation		
1 Verma A	BOOK Readings:	l taxonomy			
2. Futuyarr	a. D.J. Evolution	rtaxonomy			
3. Lull, R.S.	S. Organic Evolution.				
4. Organic evolution by A.K. Berry.					
5. Richard E. Blackwelder: Taxonomy: a Text and Reference book.					
This course can be opted as an elective by the students of following subjects: NA					
Suggested equiv	valent online courses (N	violution (in the second	ut transfer: NA		
Choose any one or more than one: (Electronic Media: Audio/Video Loctures, Online Counselling/Virtuel					
Classes/E-Contents/e-SLM/OER/supplementary links for reference/Video Conferencing/Radio broadcast/Web					
Conferencing/ Other electronic and digital contents)					
Name of electro	Name of electronic media: e-SLMYear of incorporation: 2021-22				

Course prerequisites: To study this course, a student must have qualified 10+2 with Biology					
Programme: B.Sc. Year: I Semester: V					
Subject: Zoology					
Course Code: SBSZY-03 Course Title: Economic Zoology and environmental biology					
Course Objectives- The main objectives of the course is to make learner aware of economic important of					
various inverteb	brates and scope and methodology of aquaculture.				
Course Outcom	es:(CO):				
• Econor	nic uses of various animal products.				
• Olders	tand morphology, me cycle and economic important protozoa, pratyneminutes ascheminutis				
Unders	tands concepts of fisheries, fishing tools and site selection.				
<ul> <li>Aquact</li> </ul>	Ilture system, induced breeding techniques, post harvesting techniques.				
Credits: 02	Type of Course: Core $()$ /Elective				
Max. Marks: 10	0 Min. Passing Marks: 36				
Block 1	Economic Zoology				
	Protozoa				
Unit I	• Protozoan Parasitic Diseases Of Man And Domestic Animals With Special Reference				
Omti	To Zoonotic Significance Of Entamoeba histoltica Plasmodium				
	Protozoa And Soil Fertility				
Unit II	PLATYHELMINTHES & ASCHELMINTHES				
	Life Cycle and Zoonotic Significance of Diphyllobothrium latum				
	Life Cycle and Zoonotic Significance of Dracunculus medinensis				
Unit III	Arthropoda				
	Life Cycle and Zoonotic Significance of Representation Tick And Mite				
	Beneficial and Harmful Insects				
	Plant and stored grain pest and role of insecticides in their control				
Unit IV	• Interrelationship of mosquito with Malaria, Yellow fever, Dengue, Encephalitis and				
Unit IV Dermatobia, their presentation and control					
	Biological control of insect pests				
Block 2	Environmental Biology				
Unit V	Aquaculture				
	Its Basic Concepts, Management and Economics(Including Pearl Fishery)				
Unit VI	Air Pollution				
	• Nature of Pollutants, Their Sources and Effects On Humans, Plants And Animals				
	And Their Control				
Unit VII	Water Pollution& Soil Pollution				
	Sources, Consequences And Control				
	Sources, Nature And Harmful Effects				
Unit VIII	Environmental Health				
	Animal In Relation To Human Health				
	Water In Relation To Human Disease				
	Urbanisation Stress And Health				
	Behaviour Patterns Of Health And Disease				
Suggested Text	Book Readings:				
1. Harvey e	et al: The Vertebrate Life (2006)				
2. Parker and 3 Romer a	<ol> <li>raiker and Haswell: Text BOOK OF ZOOLOgy, Vol. II (1978, ELBS)</li> <li>Romer and Parsons: The Vertebrates Rody (6th ad 1086, CRS Publishing Japan)</li> </ol>				
4. Brusca a	nd Brusca (2016) Invertebrates. Sinauer.				
5. Bisht. D	.S. Apiculture, Oxford and IBH, New Delhi.				
6. Boyd, C	6. Boyd, C.E. & Tucker. C.S., Pond aquaculture water quality management.				

- 7. Pedigo, L.P. (2002), Entomology and Pest Management, Prentice Hall.
- 8. Ranganathan L.S., Vermicomposting technology-soil health to human health.
- 9. Bookhout, T.A. (1996). Research and Management Techniques for Wildlife and Habitats, 5<sup>th</sup> edition. The wildlife Society, Allen Press.

This course can be opted as an elective by the students of following subjects: NA

Suggested equivalent online courses (MOOCs) for credit transfer: NA

Electronic media and other digital components in the curriculum:

Choose any one or more than one: (Electronic Media: Audio/Video Lectures, Online Counselling/Virtual Classes/E-Contents/e-SLM/OER/supplementary links for reference/Video Conferencing/Radio broadcast/Web Conferencing/ Other electronic and digital contents)

Name of electronic media: e-SLM

Year of incorporation: 2021-22

Course prerequi	sites: To study this con	urse, a student m	ust have qualified 10+2	with Biology	
Programme: B.	Sc.	Year: I		Semester: VI	
Subject: Zoology					
Course Code: DCEZY-108 (N) Course Title: Development Biology					
Course Objectiv	ves- The main objective	e of course, stude	ents are able to know var	rious stages involved in the	
embryonic deve	elopment.				
To study of pro	cess of fertilization and	development of	various organs.		
Course Outcom	es:(CO):	thiclosy and an			
Known     Goin known	age about developmen	and or	ganogenesis	ion and role of hormonos in	
• Galli Ki	orphosis and regeneration	on.	e mechanisms, gasuura	ion and role of normones m	
Unders	tanding of evolutionary	v significance of	internal fertilization, neo	oteny etc.	
Credits: 02	<u> </u>	8	Type of Course: Core	····	
Max. Marks: 10	0	Min. Passing	Marks: 36		
Block 1	Development Biolog	y – I			
	Asexual reproduction	n (fission, buddin	ng, gemmule formation)		
Unit I	• The Morpho	ogenetic Proces	ses And The Stages	(Blastema ,Blastogenesis ,And	
o int i	Blastozooide	s), The Kinds (F	ission, Budding, Gemn	nule Formation) And Comparion	
	Between Bla	stogegesis And l	Embryogenesis		
Unit II	Sexual reproduction (	spermetogenesis	, oogenesis and vitellog	enesis)	
	Gametogenesis (Sper	matogenesis And	l Oogenesis ) Maturation	n Of Gametes : Vitellogenesis	
Unit III	Parthenogenesis				
Unit IV	Metamorphosis				
Ontry	The Morphogenetic P	rocesses And Ca	auation In Amphibians A	And Insects, Tissue Reactivity	
Block 2	Development Biology – II				
Unit V	Induction process (factors controlling moultingin insects)				
Unit VI	Regeneration (Ability of regeneration, amphibian limb regeneration)				
	The Morphogenetic P	rocess In Regen	eration ,Ability Of Rege	neration In Different Group Of	
	Animal, Amphibian	Limb Regenerati	on		
Unit VII	Growth and Ageing				
	Concept Of Growth,	Degrowth And	Cell Death , Mechanism	Of Growth	
Unit VIII	Growth curve and its	interpretation (t	ypes of cell growth, ag	eing)	
Suggested Text	Book Readings:				
1. Essential	Development Biology,	Johnahan, M.W.	Slack (3 <sup>rd</sup> ed.), Welly B	Blackwell. (2012)	
2. Current To	opics in Development E	Biology: Roger A	A, Pedersen, Gerald P. So	chatten, Elsevier. (1998)	
3. Developm	ent Biology: Werner A	. Moller, Spring	er Science & Business N att E. Gilbert, Oxford Un	Aledia. (2012)	
4. Developin	ent blology. Michael J	. г., ballesi, sec	nt F. Gilbert, Oxford Ul	liversity Fless (2019)	
This course can	be opted as an elective	by the students	of following subjects: N	A	
Suggested equiv	valent online courses (N	AOOCs) for cred	lit transfer: NA		
Electronic medi	a and other digital com	ponents in the c	urriculum:		
Choose any one or more than one: (Electronic Media: Audio/Video Lectures, Online Counselling/Virtual					
Classes/E-Contents/e-SLM/OER/supplementary links for reference/Video Conferencing/Radio broadcast/Web					
Conferencing/ Other electronic and digital contents)					
Name of electro	onic media: e-SLM			Year of incorporation: 2021-22	

Course prerequi	isites: To study this course, a student must have	e qualified 10+2 with Biology				
Programme: B.S	Sc. Year: I	Semester: VI				
Subject: Zoology	y					
Course Code: Do	CEZY-109 (N) Course Title: Molecula	Course Title: Molecular Biology & Genetic Engineering				
Course Objective	ves- This paper to the aimed to introduce molecula	r biology & genetic engineering.				
Course Outcomes:(CO):						
Imparts	s the knowledge to culture animal cells in artificia	l media.				
• Use in recombinant DNA technology, genetic manipulations and in a variety of industrial processes.						
• Types of	of immunity, antigens-antibodies and their proper	ties.				
Credits: 02	ations of DNA technology and molecular biology	rse: Core				
creans. 62	Type of Cou					
Max. Marks: 100	0 Min. Passing Marks: 3	6				
Block 1	Molecular Biology & Genetic Engineering – I					
	Prokaryotic and Eukaryotic genome					
Unit I	Eukaryotic genome and its organizatio	n, unique and repetitive DNA, recombination and				
	chromosome mapping in bacteria and v	virus, Molecular basis of gene regulation in				
Unit II	Concept of immunology	111				
Chit II	Introduction to Basic Concepts In Immuno	blogy				
	Components of Immune System					
	Principles of Innate and Adaptive Immune	System				
	Haemopoeisis					
	Cells of Immune System and Organs(Prim	ary And Secondary Lymphoid Organs)of The				
<b>11</b> · · <b>11</b>	Immune System	1 1' 1				
Unit III	<ul> <li>Basic properties of Antigens, Immune System and disorders</li> <li>Basic Properties of Antigens</li> <li>B and T Cells</li> <li>The Immune Sector and discrete UW</li> </ul>					
	<ul> <li>Antigen Antibody Interactions as Tools for Research and Diagnosis</li> </ul>					
	Anugen Anubody Interactions as Tools for Research and Diagnosis     Gene regulation in somatic cells. Antibody structure and classes					
	Gene Regulation in Heterokaryons and Somatic Cells					
	Sometic Hybridization And Studies In Malingnancy					
Unit IV Structure Classes And Functions Of Antibodies						
	Monoclonal Antibodias	Antibodies				
	Structure And Eurotion Of MHC					
Dlook 2	Molecular Piology & Constin Engineering					
Unit V	Immune system and disease various types of y	accines				
Unit VI	Scope of genetic engineering and nucleotides					
	Scope of Genetic Engineering					
	Restriction Enzymes And Their Uses	In Gene Cloning				
	Nucleotide Sequencing Isolation And	Ananlysis Of mRNA and cDNA Probes and Their				
	Synthesis	5				
Unit VII	Recombinant DNA Technology					
	• In Vitro Synthesis of Recombinant DNA A	And Gene Cloning Techniques				
	Non Coding Intervening Sequences Within	n Eukaryoticgenes				
Application Of Recombinant DNA Technology						
Microinjecting Gene Into Animal Oocytes, Eggs And Embryos						
Suggested Text	t Book Readings:	•				
1. Lodish et	al: Molecular Cell Biology: Freeman & Co, US	A (2004),				
2. Albert et a	al: Molecular Biology of the cell: Garland (2002	2)				
3. Karp: Cell	and Molecular Biology: Willey (2002), Pierce	B. Genetics. Freeman (2004)				
4. Lewin B. 5 Waston et	<ol> <li>Lewin D. Genes VIII, FeatSon (2004).</li> <li>Waston et al. Molecular Biology of the Gana Dearson (2004).</li> </ol>					
5. Waston Ct	an more and bronegy of the Oche. I carsoli (2)					

- 6. Thomas J, Kindit, Richard A. Goldsby, Barbara A. Osborne, Janis Kubykuby Immunology, W H Freeman (2007).
- Delves Peter J., Martin Seamus J., Burton Dennis R., Roitt Ivan M. Roitt's Essential Immunology, 13<sup>th</sup> Edition, Wiley Blackwell (2017).
- 8. Primrose, SB. 1995. Principles of Genome Analysis. Blackwell Science Ltd.Oxford, UK.
- 9. E.J. Gardner and D.P. Snustad. PRINCIPAL OF GENETICS (1984), John Wiley & Sons, Ney York.
- 10. Watson, J. D., Baker T.A., Bell, S. P., Gann, A., Levine, M., and Losick, R., 2008 Molecular Biology of the Gene 6th edition. Cold Spring Harbour Lab. Press, Pearson Pub.
- 11. P.K. Gupta. BIOTECHNOLOGY AND GENOMICS. Rastogi Publications, 7th Reprint (1st Edition): 2016-2017.
- 12. A Textbook of Basic and Molecular Genetics (pb)ISBN : 9788188826193Edition : 01Year : 2018Author : Dr. Parihar P

This course can be opted as an elective by the students of following subjects: NA

Suggested equivalent online courses (MOOCs) for credit transfer: NA

#### Electronic media and other digital components in the curriculum:

**Choose any one or more than one:** (Electronic Media: Audio/Video Lectures, Online Counselling/Virtual Classes/E-Contents/e-SLM/OER/supplementary links for reference/Video Conferencing/Radio broadcast/Web Conferencing/ Other electronic and digital contents)

Nome of electronic medic: a SI M	Year of incorporation:
Name of electronic media. e-SLM	2021-22

### Year-2023-2024

## Syllabus of B.Sc. Programme: [Subject Name: Biochemistry] In accordance with NEP-2020

Year	Semester	Course Code	Title of Paper	Theory/ Practical	Credits	Max. Marks
	т	UGBCH -101N	Introduction to biochemistry	Theory	2	100
	1	UGBCH -101(P)N	Practical Work	Practical	2	100
1	UGBCH -102N Nutritional bioc		Nutritional biochemistry	Theory	2	100
	11	UGBCH -102(P)N	Practical Work	Practical	2	100
		UGBCH -103N	Intermediary metabolism	Theory	2	100
	***	UGBCH -103(P)	Practical Work	Practical	2	100
2	111	Skill Enhancement	Course			
-		SBSBCH- 01N	Bio-analytical techniques	Theory	2	100
	IV	UGBCH -104N	Enzymology	Theory	2	100
		UGBCH -104(P)N	Practical Work	Practical	2	100
		Discipline Centric E		•		
	V	DCEBCH -105N	Microbiology	Theory	2	100
		DCEBCH -106N	Spectroscopy	Theory	2	100
		DCEBCH -107(P)N	Practical Work Based on 105 & 106	Practical	2	100
3		Discipline Centric E				
5		DCEBCH -108N	Plant biochemistry	Theory	2	100
		DCEBCH -109N	Immunology	Theory	2	100
	VI	DCEBCH -109(P)N	Practical Work Based on	Practical	2	100
			108 & 109			
		Skill Enhancement	Course			
		SBSBCH-04N	Clinical biochemistry	Theory	4	100
			Total Credit/Max. Marks		34	1600

Programme: I	<b>3.Sc.</b> Year: <b>First</b> Semester: <b>I</b>			
	Subject: Biochemistry			
Course Code:	UGBCH-101N         Course Title: Introduction to Biochemistry			
Course Obje	ctives:			
• To dis	cuss the basics of outline of biochemistry			
• To dis	cuss the solvent properties of water of biochemical reactions			
• To dis	cuss the protein that is the building block of living being			
To dis	cuss the basic concept of cell organelles and its role in biochemical functions			
<b>Course Outc</b>	omes:			
CO 1: Abl	e to understanding of history and scope of Biochemistry in brief.			
CO 2: Lea	rn about water properties and role of it in living beings.			
CO 3: To l	know about cell structure and their functions			
CO 4: Abl	e to understanding the structure and functions of various cell organelles.			
CO 5: Also	b able to understand the amino acid structure and its classification, and also know			
how	amino acids play important role in proteins formation.			
Credits: 2	Type of Course: Core			
Max. Marks:	100 Min. Passing Marks: 36			
Block 1	Life History and Cell Structure			
	Introduction to biochemistry:			
I Init I	The origin of biochemistry and unity of life-History, scope and current prospective			
Unit I	of biochemistry, unique properties of water, weak interactions in aqueous systems,			
ionization of water, buffers				
Unit II	Cell structure and functions			
	Prokaryotic and eukaryotic cell, animal and plant cells, units of measurement, light			
	microscope and electron microscope, centrifugation for subcellular fractionation.			
Unit III	Cell organelles:			
	Structure and functions of cell nucleolus, Endoplasmic reticulum, Golgi complex,			
	Lysosome, mitochondria, chloroplasts and peroxisomes.			
Block 2	Amino Acids, Proteins and Carbohydrate			
Unit IV	Amino Acids: General introduction, classification, structure and functions of			
Unit I v	amino acids, basic properties of amino acids.			
Unit V	Proteins the basic molecules:			
	Nature, classification and types of protein structure, Peptides classification and			
	conformational structure			
Unit VI	Carbohydrates:			
	General introduction, classification and structure, monosaccharides - structure of			
	aldoses and ketoses, ring structure of sugars, formation of disaccharides,			
	polysaccharides.			
Block 3	Lipids, Nucleic Acids and Vitamins			
Unit VII	Lipids:			

General introduction, classification, building blocks of lipids - fatty acids, glycerol,			
ceramide, structure of fatty acids and their derivati	ceramide, structure of fatty acids and their derivatives.		
t VIII Nucleic acids: Nucleotides and nucleosides-structure and properties, nitroger			
bases: purines and pyrimidines, structure of DNA and RNA.			
Unit IX Vitamins: Types and functions, structure and activ	ve forms of water soluble and fat		
soluble vitamins, deficiency diseases and symptom	ns.		
Suggested Text Book Readings:			
1. David L. Nelson and Michael Cox, "Lehninger Principles	of Biochemistry" International		
Edition-2021.			
2. Dr J L Jain, "Fundamentals of Biochemistry" S. Chand an	nd Company-2020.		
3. P S Verma and V K Agarwal, "Cell Biology (Cytology, B	Biomolecule and Molecular		
Biology" S. Chand Publication-2016.			
4. Talwar and Srivastava, "Textbook of Biochemistry and H	luman Biology" Eastern		
Economy Edition, Prentice Hall, India-2002.			
5. Satyanarayana U., "Biochemistry" Elsevier India, 2021			
Suggested online link:			
1. Cell Organelles: <u>Cell Organelles Notes.pdf (gwisd.us)</u>			
2. Carbohydrates: <u>CARBOHYDRATES (dhingcollegeonline.co.in)</u>			
3. Amino Acids: <u>Microsoft Word - Amino Acids Peptides Protein</u>	ns_Notes (srmist.edu.in)		
4. Vitamins:			
https://www.pearsonhighered.com/assets/samplechapter/0/1/3/2/0132181630.pdf			
This course can be opted as an elective by the students of following subjects: NA			
Suggested equivalent online courses (MOOCs) for credit transfer: NO			
Electronic media and other digital components in the curriculum:			
Choose any one or more than one: (Electronic Media: Audio/Video Lectures, Online			
Counseling/VirtualClasses/E-Contents/e-SLM/OER/supplementary links for reference/Video			
Conferencing/Radio broadcast/Web Conferencing/ Other electronic and digital contents)			
Name of electronic media: YoutubeYear of incorporation: 2023-24			

Programme: I	Programme: <b>B.Sc.</b> Year: <b>First</b> Semester: <b>II</b>			
Subject: Biochemistry				
Course Code: UGBCH-102N Course Title: Nutritional Biochemistry			utritional Biochemistry	
Course Obje	ctives:			
To dis	cuss the basic elem	ents of nutrition.		
To dis	cuss the basal meta	bolic rate and measuremer	nt of fuel value of foods.	
To dis	cuss the biological	oxidation of foodstuff.		
• To dis	cuss the basic conc	ept of micro and macro nu	trition	
Course Outc	omes:			
CO 1: Abl	e to know the basic	concept of different nutrit	ion and energy in brief.	
CO 2: Abl	e to understand the	role of different nutrients i	in growth of living beings.	
CO 3: Lea	rn about cell structu	re and their functions.		
CO 4: Lea	rn about the concep	t of digestion of carbohydr	rates, proteins and fats.	
CO 5: Also	o know the structure	e of proteins, carbohydrate	s and lipids.	
Credits: 2		Type of C	Course: Core	
Max. Marks:	100	Min. Passing Marks: 36		
Block 1	Nutrition and ox	idation of foodstuff		
	Elements of Nutrition:			
Unit I	Dietary requirements of carbohydrates, lipids and proteins. Essential amino acids,			
	essential fatty acids, Malnutrition.			
Unit II	Basal Metabolic	Rate (BMR):		
	Concept of BMR, factors affecting BMR, measurement of fuel value of foods. basal			
	and resting metabolism, physical activity, energy balance.			
Unit III	Biological oxidation of foodstuff:			
	Measurement of energy content of food, physiological energy value of foods,			
	measurement of energy expenditure, factors affecting thermogenesis.			
Block 2	Nutrition of carb	ohydrates, Proteins and	Vitamins	
TT '/ TT/	Dietary carbohydrate:			
Unit IV	Functions, digestion, absorption, storage and utilization of carbohydrates, hormonal			
Linit V	regulation of blood glucose.			
Unit v	Proteins:			
	Sources, functions, digestions and absorptions, essential and nonessential amino			
Unit VI	Actus, antagonism, toxicity and imbalance, effects of deficiency and kwashiorkor.			
	Nutrition importa	annis. De of dietary calcium: pho	sphorus: magnesium: iron: jodine: zinc	
	and copper requ	irements and deficiency	diseases associated with vitamin B	
	Complex C A D E and K			
Block 3	Food and drugs integration			
Unit VII	Lipids and fats:			

	Role of lipid in dietary supplement. Dietary fibe	r, role of fibre in lipid metabolism,		
	blood glucose level and GI tract functions, role of saturated fat. Sources, functions,			
	digestions and absorptions of fats, types of fats dietary fats, role of omega-3 fatty			
	acids in living human body, essential and nonessential fatty acids.			
Unit VIII	Food and drug interactions:			
	Nutrient interactions affecting ADME of drugs	s, alcohol and nutrient deficiency,		
	antidepressants, psychoactive drugs and nutrient	interactions.		
Unit IX	Nutritional status:			
	Anthropometric measurements, biochemical as	sessment, reactive oxygen species		
	(ROS), glycosylated Hb, differential diagnosis of	of $B_{12}$ and foliate.		
Suggested T	ext Book Readings:			
1. David L	. Nelson and Michael Cox, "Lehninger Principles	of Biochemistry" International		
Edition-	-2021.			
2. Sharma	D C, Nutritional Biochemistry, CBS Publications	5		
3. P S Ver	rma and V K Agarwal, "Cell Biology (Cytology, I	Biomolecule and Molecular		
Biology	" S. Chand Publication-2016.			
4. Tom Brody, Nutritional Biochemistry, Second edition, University of California at Berkely				
5. Satyanarayana U., "Biochemistry" Elsevier India, 2021				
Suggested	online link:			
1. Meta	abolism: Instruction_Metabolism_1_Medicine_V1.Pd	f (Umed.Wroc.Pl)		
2. Bmr: <u>Basal Metabolic Rate (Upsmfac.Org)</u>				
3. Minerals and vitamins: <u>https://sightandlife.org/wp-</u>				
cont	content/uploads/2017/03/sal_mvlex_web.pdf			
4. Food and drug interactions:				
Http	Https://www.omjournal.org/images/75_m_deatials_pdfpdf			
5. Nutritional status: <u>PowerPoint Presentation (zmchdahod.org</u> )				
This course can be opted as an elective by the students of following subjects: NA				
Suggested eq	uivalent online courses (MOOCs) for credit tra	nsfer:		
1. Nutritional and Clinical Biochemistry - Course (swayam2.ac.in)				
Electronic n	nedia and other digital components in the cur	rriculum:		
Choose any one or more than: (Electronic Media: Audio/Video Lectures, Online				
Counseling/Virtual Classes/E-Contents/e-SLM/OER/supplementary links for reference/Video				
Conferencing	/Radio broadcast/Web Conferencing/ Other electr	onic and digital contents)		
Name of elec	etronic media: YouTube	Year of incorporation: 2023-24		

Programme: <b>B.Sc.</b> Year: <b>Second</b> Semester: <b>III</b>		Semester: III		
Subject: Biochemistry				
Course Code: UGBCH-103N       Course Title: Intermediary Metabolism			diary Metabolism	
Course Obje	ctives:			
• To dis	cuss the basic outlin	e of bioenergetics		
• To dis	cuss the concept of	metabolism.		
• To dis	cuss the concept of	thermodynamics		
To dis	cuss the basic conce	pt of metabolism of proteins, ca	rbohydrates and lipids	
Course Outc	omes:			
CO 1: Abl	e to understanding the	he concept of bioenergetics and	metabolism	
CO 2: Kno	ow to the concept of	phosphorylation, and ATP cycle		
CO 3: Abl	e to know Glycolysi	s and Kreb's cycle		
CO 4: Lea	rn about the structur	e and functions of chloroplast ar	id mitochondria.	
CO 5: Abl	e the understand the	concept of photosynthesis and p	bhotosystem I and II.	
Credits: 2		Type of Course	: Core	
Max. Marks:	100	Min. Passing Marks: 36		
Block 1	Bioenergetics and	l Thermodynamics		
	<b>Bioenergetics:</b>			
Unit I	Introduction to bioenergetics, photochemical reaction in plants, chemical energy			
	organic substance.			
Unit II	<b>Thermodynamics:</b> Notions and laws of thermodynamics, state functions,			
	equilibrium constant, coupled reactions, free energy charge, and application to			
	chemical reaction.			
Unit III	AIT: ATD evole and formation of ATD by phosphorylation importance of ATD and other			
	ATP cycle and formation of ATP by phosphorylation, importance of ATP and other			
Block 2	Compounds of high energy potential.			
DIOCK 2	Metabolism of carbohydrates.			
Unit IV	Glycolysis Kreb's	s cycle electron transport syst	em in mitochondria Oxidative	
Chit I V	phosphorylation at	ad mechanism of ATP synthesis		
Unit V	Jnit V       Metabolism of lipids;         Catabolism of triglycerides, biosynthesis of cholesterol, B-oxidation of fatty			
	acids.		ý <b>j</b>	
Unit VI	Nitrogen metabolism:			
	Nitrogen fixation	and assimilation, amino acid	metabolism, the urea cycle,	
	chlorophylls.		- -	
Block 3	lock 3 Phosphorylation and Photosynthesis			
Unit VII	Oxidative phosph	orylation:		

	Mitochondria, electron transport chain-its organization and function, regulation of				
	oxidative phosphorylation, alternative respiratory pathways in plants.				
Unit VIII	Photophosphorylation:				
	Photophosphorylation in plants - structure of chloroplast, molecular architecture of				
	Photosystem I and Photosystem II. Photo in	nhibition.			
Unit IX	Photosynthesis: Pigments of Photosy	nthesis, Oxygenic and anoxygenic			
	Photosynthesis, adsorption of light by chlor	rophyll, Calvin cycle.			
Suggested To	ext Book Readings:				
1. David	L. Nelson and Michael Cox, "Lehninger Pri	nciples of Biochemistry" International			
Edition	n-2021.				
2. Simmi	Kharb, Intermediary Metabolism				
<b>3.</b> P S Ve	erma and V K Agarwal, "Cell Biology (Cyto	logy, Biomolecule and Molecular			
Biolog	y" S. Chand Publication-2016.				
4. Talwa	r and Srivastava, "Textbook of Biochemistry	and Human Biology" Eastern			
Econo	my Edition, Prentice Hall, India-2002.				
Sugge	sted online link:				
1. Bioenergetics: Microsoft PowerPoint - 426L4Bioen.ppt [Compatibility Mode] (unm.edu)					
2. Metaboli	sm of Biomolecules: <u>76633_ch07_5589.pdf (jbp</u>	oub.com)			
3. Nitrogen	n metabolism: <u>Nitrogen Metabolism (wou.edu</u>	<u>)</u>			
4. Photoph	osphorylation:				
5. <u>http://pp</u>	up.ac.in/download/econtent/pdf/Photophosp	horylation.pdf			
6. Photosy	6. Photosynthesis: https://www.rsb.org.uk/images/15_Photosynthesis.pdf				
This course can be opted as an elective by the students of following subjects: NA					
Suggested equivalent online courses (MOOCs) for credit transfer: NO					
Electronic media and other digital components in the curriculum:					
Choose any one or more than one:(Electronic Media: Audio/Video Lectures, Online					
Counseling/Virtual Classes/E-Contents/e-SLM/OER/supplementary links for					
reference/Video Conferencing/Radio broadcast/Web Conferencing/ Other electronic and					
digital contents)					
Name of electronic media: YouTubeYear of incorporation: 2023-24					

Programme: <b>B.Sc.</b>		Year: Second	Semester: III	
Subject: Biochemistry				
Course Code:	Course Code: SBSBCH-01N Course Title: BIO ANALYTICAL TECHNIQUES			
Course Obje	ctives:			
• To dis	cuss the basics concepts	s of bioanalytical technique	s	
• To dis	cuss the pH, buffer and	biological importance		
• To dis	cuss the about concept	of chromatography and spe	ctroscopy.	
• To dis	cuss the basic concept of	of electrophoresis		
<b>Course Outc</b>	omes:			
CO 1: Abl	e to understanding the c	concept of normality, molar	ity and molality.	
CO 2: Kno	ow the properties of ligh	t, optical rotation and optic	al rotator.	
CO 3: Abl	e to know about visible	and UV spectroscopy.		
CO 4: Lea	rn the basic principle of	FT-IR and NMR spectrom	eter.	
CO 5: Disc	cuss the principle of cen	trifugation and its applicat	ons.	
Credits: 4		Type of Course: Skill En	hancement Course	
Max. Marks:	100	Min. Passing Marks: 36		
Block 1	Basics of bio-analytical techniques			
	<b>Basic introductions:</b> Basic concept of bio analytical techniques, normality, morality and molality, brief			
Unit I				
	about purification, centrifugation, filtration, dialysis, homogenization			
Unit II	pH and buffer:			
	Hydrogen ion concentration, Buffer- definition, types and its preparation, buffers			
I In: 4 III	of biological importance such as carbonate bicarbonate, phosphate and acetate.			
Unit III	<b>Properties of Light:</b> light spectra, wave length, plane polarized light, optical			
Block 2	Plack II. Chromotography and anostroscopy			
DIUCK 2	2         Diock II- Unromatography and spectroscopy           Chromeomorphy         Dringingle of pertition shares to such as the sector of the state of the st			
Unit IV	chromatography. 11h	performance liquid chrom	tography, exchange, ger mutation	
Unit V	Snectroscony-I:			
Cint V	Concepts of spectroscopy Beer-Lambert's law Visible and UV Spectroscopy			
	applications of colorimetery.			
Unit VI	<b>Spectroscopy-II</b> Basic principle of FT-IR and NMR spectrometer and their role in			
	detection of organic molecules detection.			
Block 3	Electrophoresis and Centrifugation			
Unit VII	Electrophoresis: Prin	ciples of electrophoresis, se	paration of proteins by PAGE and	
	SDS-PAGE.			
Unit VIII	Centrifugation: Principles of centrifugation, differential centrifugation,			
	applications of centrifugation and density gradient.			

Unit IX Microbial techniques: Isolation of bacteria, antimicrobial activity by using DIS	Microbial techniques: Isolation of bacteria, antimicrobial activity by using DISC			
diffusion techniques, use of different solvent system for amino acid, carbohydra	diffusion techniques, use of different solvent system for amino acid, carbohydrate			
and lipid separation.				
Suggested Text Book Readings:				
1. David L. Nelson and Michael Cox, "Lehninger Principles of Biochemistry" International				
Edition-2021.				
2. Sabari Ghosal and Anupama Sharma Awasthi, Fundamentals of Bioanalytical Techniques				
and Instrumentation, Second edition.				
3. Abhilasha Shourie (Author), Shilpa S. Chapadgaonkar Bioanalytical Techniques				
4. Sabro Ghosal a. K. Srivastava, Fundamentals of Bioanalytical Techniques a	nd			
Instrumentation				
5. Talwar and Srivastava, "Textbook of Biochemistry and Human Biology" Eastern Economy	y			
Edition, Prentice Hall, India-2002.				
Suggested online link:				
1. pH and buffer: <u>pH and Buffers.ppt (csun.edu)</u>				
2. Properties of Light: <u>ACL7_light.ppt (umd.edu)</u>				
3. spectroscopy: <u>Spectroscopy.pdf (osti.gov)</u>				
<ol> <li>Chroniatographic Techniques. <u>222 Chapter 4.pdf (unipute.ac.m)</u></li> <li>Mienshielegiesl Leberstern: Techniques. Mienshielegiesl Leberstern: Techniques.</li> </ol>				
5. Microbiological Laboratory Techniques <u>Microbiological Laboratory Techniques</u>				
(mowr.gov.in)				
This course can be opted as an elective by the students of following subjects: NA				
Suggested equivalent online courses (MOOCs) for credit transfer:				
1. Analytical techniques: by Dr. Moganty r. Rajeswari Analytical Techniques - Course				
(swayam2.ac.in)				
Electronic media and other digital components in the curriculum:				
<b>Choose any one or more than one:</b> (Electronic Media: Audio/Video Lectures, Online				
Counseling/Virtual Classes/E-Contents/e-SLM/OER/supplementary links for				
reference/Video Conferencing/Radio broadcast/Web Conferencing/ Other electronic and				
digital contents)				
Name of electronic media: YouTubeYear of incorporation: 2023-24				

Programme: <b>B.Sc.</b>		Year: Second	Semester: IV		
		Subject: Biochemistry			
Course Code:	UGBCH-104N	Course Title: En	zymology		
Course Obje	Course Objectives:				
• To dis	cuss the basics of en	nzymology.			
• To dis	cuss the effects of v	arious factors on enzymatic-cataly	zed reactions		
• To dis	cuss the enzyme cot	factors and inhibition			
• To dis	cuss the regulation	of enzyme activity and its importa	nce.		
<b>Course Outc</b>	omes:				
CO 1: Lea	rn the enzyme classi	fication and kinetics			
CO 2: kno	wn the reversible an	d irreversible inhibition			
CO 3: Abl	e to discuss the mec	hanism of action of chymotrypsim	1		
CO 4: Kno	ow the enzymes used	l in clinical biochemistry as reage	nts,		
CO 5: Abl	e to discuss the prin	ciple of co-enzymes, prosthetic gro	oups and allosteric activators		
Credits: 3		Type of Course:	Core/Elective(Core)		
Max. Marks:	100	Min. Passing Marks: 21			
Block 1	Enzyme classifica	tion and kinetics			
	Introduction to en	nzymes:			
Unit I	Basic concept and	classification of enzymes, enzymes	nes as biocatalysts, effects of		
	various factors on enzymatic-catalyzed reactions.				
Unit II	Enzyme cofactors and inhibition: Role cofactors, mode of action of coenzymes,				
	principle co-enzymes, prosthetic groups, allosteric activators and inhibitors.				
Unit III	Enzymes kinetics: enzymes classification, concept of ES complex, Michaelis-				
	Menten equation, KM constant.				
Block 2	Enzyme action and mechanism				
Unit IV	Enzyme inhibition: Reversible and irreversible inhibition, competitive, non-				
	competitive and un-competitive inhibition.				
Unit V	Mechanism of en	<b>nzyme action:</b> Acid-base catalys	sis, chemical modification of		
	active site group; mechanism of action of chymotrypsim and lysozyme.				
Unit VI	Enzyme mechanism:				
	Mechanism of ac	ction of chymotrypsim, inhibitor	rs of enzymes - antibiotics,		
	regulation of enzyme activity and its importance.				
Block 3	Multienzyme System and its role in medicine				
Unit VII	Enzyme regulatio	n:			
	General mechanisms of enzyme regulation, inhibition, allosteric enzymes, positive				
	and negative coop	eratively with special reference to	aspartate, transcarbamoylase.		
Unit VIII	Multienzyme Sys	tem:			
	Mechanism of enzyme action and regulation of pyruvate dehydrogenase,				
	1soenzymes.				

Unit IX	Enzymes in medicine: Enzymes used in	clinical biochemistry as reagents,		
	diagnostics and therapy, role of immobilized enzymes in industry.			
Suggested To	ext Book Readings:			
1. David L	. Nelson and Michael Cox, "Lehninger Principl	les of Biochemistry" International		
Edition-	2021.			
2. Nicholas	s C. Price, Fundamentals Of Enzymology, 3rd I	Edition		
<b>3.</b> P S Veri	ma and V K Agarwal, "Cell Biology (Cytology	, Biomolecule and Molecular		
Biology	" S. Chand Publication-2016.			
<b>4.</b> Talwar a	and Srivastava, "Textbook of Biochemistry and	Human Biology" Eastern Economy		
Edition,	Prentice Hall, India-2002.			
5. Lewis S	Stevens and Nicholas Price, Fundamentals of Er	nzymology: Cell and Molecular		
Biology	of Catalytic Proteins			
Suggested of	online link:			
1. Introdu	uction to Enzymes: Microsoft Word - Introduct	ion (ufsc.br)		
2. Enzym	ne Kinetics: ENZYME KINETICS (columbia.ed	<u>du)</u>		
3. Regula	ation of Enzyme Activity: <u>Slide 1 (mgcub.ac.in</u> )	<u>)</u>		
4. Multie	nzyme Complexes: <u>Multienzyme Complexes (1</u>	<u>mlsu.ac.in)</u>		
This course c	an be opted as an elective by the students of f	following subjects: Any one		
Suggested equivalent online courses (MOOCs) for credit transfer:				
Enzymology: Enzymology - Course (swayam2.ac.in)				
Electronic media and other digital components in the curriculum: Choose any one or				
more than	one:(Electronic Media: Audio/Video Lec	tures, Online Counseling/Virtual		
Classes/E-Co	ntents/e-SLM/OER/supplementary lin	ks for reference/Video		
Conferencing/Radio broadcast/Web Conferencing/ Other electronic and digital contents)				
Name of elec	tronic media: Youtube	Year of incorporation: 2023-24		

Programme: <b>B.Sc.</b>		Year: Third	d	Semester: V
Subject: Biochemistry				
Course Code:	DCEBCH-105N		Course Title: Mi	crobiology
Course Obje	ctives:			
• To dis	cuss the diversity of n	nicrobial wor	·ld.	
• To dis	cuss the classification	of microbiol	logy, and their nom	enclature.
• To dis	cuss the genetic engin	eering and re	ecombination of bac	cteria.
• To dis	cuss the basic concept	t of bacterial	cell, gram positive	and negative bacteria.
Course Outco	mes:			
CO 1: Lea	rn the history of micro	obial world ar	nd development in l	biology.
CO 2: Able	e to know about bacte	ria, viruses a	nd algae.	
CO 3: Kno	w the biological nitro	gen fixation a	and biofertilizers	
CO 4: Lear	rn the structure and fu	nctions of fla	agella.	
CO 5: Able	e to know the role of b	bacteria in N,	P, S and C cycle.	
Credits: 2		•	Type of Course:	Elective
Max. Marks:	100	Min. Passin	ng Marks: 36	
Block 1	Microbial world the	eir developm	nent	
	Diversity of Microbial world:			
Unit I	Classification of microbiology, and their nomenclature. Whittaker's five kingdom			
	classification systems and their utility.			
Unit II	Development of microbiology:			
	Spontaneous generation vs. biogenesis, general characteristics of different groups,			
	acellular microorganisms (Viruses, Viroids, Prions).			
Unit III	Isolation and Cultivation:			
	Culture media, nutritional requirements and growth characteristics of bacteria,			
	development of various microbiological techniques.			
Block 2	Genetic engineering and Microorganism			
	Genetic:	1		
Unit IV	Recombination of	bacteria co	onjugation, transd	uction, and transformation,
Linit V	significance of genetic recombination in bacteria.			
Unit v	Bacteria: Structure of hostorial call grow positive and grow possitive hostoria mission			
	simple compound	applications of	of bacteria and arch	againe bacteria, incroscopy-
	and food			
Unit VI	Viruses:			
	General structure and	d classificatio	on properties of vir	uses structure and replication
	of poliovirus and HIV Protozoa. General characteristics with special reference to			
	Amoeba and Paramecium.			
Block 3	Microorganism and	d their agricu	ulture	
Unit VII	Algae:			
---	---	--	--	
	Types and occurrence, thallus organization, algae cell ultra structure, pigments,			
	flagella, eyespot food reserves.			
Unit VIII	Unit VIII Fungi:			
	General classification, occurrence, habitat of fungi, distribution, nutritional			
	requirements, fungal cell ultra- structure, role of fungi in agriculture, environme			
	Industry, medicine and food.			
Unit IX	Role of microorganism in Agriculture:			
	Biological nitrogen fixation, microbes as bio fe	ertilizers, role of bacteria in N, P, S,		
	C cycle, role of bacteria in nutrient cycle			
Suggested Te	ext Book Readings:			
1. Donald L	. Pavia (Author) Introduction to Spectroscopy			
2. Gauglitz,	John Wiley Handbook Of Spectroscopy 2Nd Ed	dition 4 Volume Set		
3. David L.	Nelson and Michael Cox, "Lehninger Principles	s of Biochemistry" International		
Edition-2	021.			
4. Banwell	(Author), Fundamentals of Molecular Spectrosco	opy   4th Edition		
5. P. S Kals	i, Spectroscopy of Organic Compounds			
Suggested on	line link:			
1. Microbia	al Diversity And Systematic: <u>1075x_Ch03_025.</u>	Qxd (Jblearning.Com)		
2. Microbi	ological Laboratory Techniques: Microbiologica	al Laboratory Techniques		
(Mowr.C	<u>Jov.ln)</u>			
3. Micro-O	organism: <u>Pdf (Usda.Gov)</u>	and Jacture 25 Die fartilieer de er		
4. Benefici	al microorganisms in Agriculture: Microsoft wa	Drd - Lecture 25 Bio fertiliser.docx		
This course c	an be opted as an elective by the students of fo	ollowing subjects: NA		
Suggested equ	uivalent online courses (MOOCs) for credit tr	ansfer: NO		
Electronic media and other digital components in the curriculum:				
Choose any	<b>Choose any one or more than one:</b> (Electronic Media: Audio/Video Lectures, Online			
Counseling/V	Counseling/Virtual Classes/E-Contents/e-SLM/OER/supplementary links for			
reference/Video Conferencing/Radio broadcast/Web Conferencing/ Other electronic and				
digital contents)				
Name of elec	tronic media: YouTube	Year of incorporation: 2023-24		

Programme: <b>B</b>	B.Sc. Year: Third Semester: V			
Subject: Biochemistry				
Course Code: DCEBCH-106N Course Title: Spectroscopy			se Title: Spectroscopy	
Course Object	tives:			
To disc	cuss the basics of UV-visit	ible and IR spectrosc	opy.	
To disc	cuss the fundamental law	of spectroscopy, elec	ctromagnetic radiation and atomic	
adsorpt	adsorption spectroscopy.			
To disc	cuss the NMR, atomic spe	ectroscopy and atomi	c emission spectroscopy	
To disc	cuss the basic concept of I	luminescence and ele	ectron spectroscopy.	
Course Outco	mes:			
CO 1: Able	to know the history spec	troscopy and electro	magnetic radiation in brief.	
CO 2: Able	to know the principle an	d instrumentation of	UV-Visible and applications	
CO 3: Able	to know the principle of	electron spectroscop	y and its applications.	
CO 4: Able	to know the principle of	flame photometry.		
CO 5: Learn	n the enzymatic kinetics i	reactions.		
Credits: 2		<b>Type of Course:</b> E	lective	
Max. Marks: 1	00	Min. Passing Marks	s: 36	
Block 1	UV-Visible and IR spec	troscopy		
	Spectroscopy:			
Unit I	Fundamental law of spectroscopy, electromagnetic radiation, origin of spectra,			
	application of spectrosco	py in biochemistry.		
Unit II	UV Visible spectroscop	y:		
]	Principle and instrume	ntation of UV-Visi	ble, Beer-Lambert law, qualitative and	
	quantitative analysis by	UV-Visible spectro	scopy. Origin of spectra and electronic	
1	transition, composition of	of color complex, a	oplication of UV-Visible spectrometer in	
	enzyme kinetics reaction			
Unit III	Spectroscopy:	C 1 /		
	I neory and principle of 1	nirared spectroscopy	, components of IR spectroscopy,	
Dlask 2	NMD and Atomic Speed	traggery A mine A a	ida Protoing and Carbohydrata	
DIOCK 2	NMR and Atomic Speci	troscopy Annio Ac	us, Proteins and Carbonyurate	
Unit IV	Drinciple of NMP spectr	oscony NMP spect	a measurement types of NMP chemical	
	Principle of NMR spectroscopy, NMR spectra measurement, types of NMR, chemical			
Unit V	Atomic adsorption spectroscopy:			
	Principle of adsorption	spectroscopy.	mentation and application of adsorption	
	spectroscopy			
Unit VI	Atomic Emission Spect	roscopy:		
	Principle of emission spe	ctroscopy. Instrumer	tation Emission spectroscopy, principle of	
	flame photometry	r,		
Block 3	Luminescence and Elec	tron spectroscopy		

Unit VII	ICP-atomic emission spectroscopy: Princip	le of plasma spectroscopy, application of	
	adsorption spectroscopy, comparison of ICP-AES with ASS		
Unit VIII	Luminescence spectroscopy:		
	Luminescence and chemiluminescence, principle of fluorescence, application of		
	Fluorimetry		
Unit IX	Electron spectroscopy:		
	Principle of electron spectroscopy, electron sp	pectroscopy for chemical analysis (ESCA),	
	chemical shift in ESCA.		
Suggested T	ext Book Readings:		
1. David L.	Nelson and Michael Cox, "Lehninger Principl	les of Biochemistry" International	
Edition-2	2021.		
2. Dr J L Ja	in, "Fundamentals of Biochemistry" S. Chand	and Company-2020.	
3. P S Verr	na and V K Agarwal, "Cell Biology (Cytology,	, Biomolecule and Molecular Biology" S.	
Chand P	ublication-2016.		
4. Talwar a	nd Srivastava, "Textbook of Biochemistry and	Human Biology" Eastern Economy	
Edition,	Prentice Hall, India-2002.		
Suggested o	nline link:		
1. Spectro	scopy: <u>Spectroscopy.pdf (osti.gov)</u>		
<b>2.</b> Nuclear	Magnetic Resonance: Nuclear Magnetic Reso	nance (NMR) (brown.edu)	
3. Electron	nic Spectroscopy: MSc Chemistry Paper-IX Ur	nit-4.pdf (nou.ac.in)	
This course of	can be opted as an elective by the students of	following subjects: NA	
Suggested equivalent online courses (MOOCs) for credit transfer: NO			
Electronic media and other digital components in the curriculum:			
Choose any one or more than one: (Electronic Media: Audio/Video Lectures, Online			
Counselling/Virtual Classes/E-Contents/e-SLM/OER/supplementary links for reference/Video			
Conferencing/Radio broadcast/Web Conferencing/ Other electronic and digital contents)			
Name of elec	Name of electronic media: YoutubeYear of incorporation: 2023-24		

Programme: I	ramme: <b>B.Sc.</b> Year: <b>Third</b> Semester: <b>VI</b>		Semester: VI
Subject: Biochemistry			
Course Code: DCEBCH -108N Course Title: PLANT BIOCHEMISTRY			BIOCHEMISTRY
Course Obje	ctives:		
• To dis	cuss the plant biocher	nistry and electron transport sys	stem in plants.
• To dis	cuss the nitrogen meta	abolism and nitrogen fixation a	nd assimilation.
• To une	derstand the carbon as	similation, respiration and plan	t growth regulator
• To une	derstand basic concept	t of hill reaction, photorespiration	on and photosynthesis
<b>Course Outc</b>	omes:		
CO 1: Abl	e to know the oxidativ	e phosphorylation	
CO 2: kno	w the concept of chlor	ophyll, pigments, and light har	vesting complexes.
CO 3: To l	know cyclic and non c	yclic photophosphorylation.	
CO 4: Abl	e to understanding the	regulation of plant glycolysis	
CO 5: Also	o know about abiotic a	and biotic stress, salinity, water	stress and pathogenesis.
Credits: 2		Type of Course:	Elective
Max. Marks:	100	Min. Passing Marks: 36	
Block 1	Electron Transport	System and nitrogen metabo	lism
Unit I	Electron Transport	System in Plants:	
	Oxidative phosphory	lation, mitochondrial respirator	ry complexes.
Unit II	Nitrogen Metabolism: Assimilation of nitrate, enzyme of nitrate reduction and		
	their regulation and assimilation of ammonia into organic compounds.		
Unit III	Nitrogen fixation and assimilation:		
	Biological nitrogen	ixation by free living and in sy	mbiotic association, structure
	and function of enzyme nitrogenase, nitrate assimilation.		
Block 2	Photosynthetic process and carbon assimilation		
	Photosynthetic pro		
	Chlorophylls, photo	pperiodism, photosynthetic m	embranes and organelles, z
Unit IV	involved in photosy	ndant reactions. Photosynthe	an of NADDH and ATD light
	hervesting complexe	indesis, Hill feaction, generation	on of NADER and ATE, light
Linit V	narvesting complexes.		
Unit v	Synthesis of photochemicals: Classification and biosynthesis of Terpenes,		
Unit VI	Carbon assimilation:		
	Cyclic and non cyclic photophosphorylation. Calvin cycle, and photorespiration		
	General introduction classification and structure monosaccharides - structure of		
	aldoses and ketoses ring structure of sugars formation of disaccharides		
	polysaccharides.		
Block 3	Plant stress growth	regulators	

Unit VII	Stress Metabolism in Plants: A biotic and biotic stress; salinity, water stress,		
	chilling, heat, pathogenesis, heavy metals and their impact on plant growth and		
	metabolism.		
Unit VIII	Respiration: Regulation of plant glycolysis, regulation of plant glycolysis,		
	translocation of metabolites across mitochondrial membrane, TCA cycle.		
Unit IX	Plant growth regulator: Phytohormones and its effect on plant growth and		
	development, regulation of plant morphogenetic processes by light.		
Suggested To	ext Book Readings:		
1. David I	L. Nelson and Michael Cox, "Lehninger Principles of Biochemistry" International		
Edition	-2021.		
2. Dr J L .	Jain, "Fundamentals of Biochemistry" S. Chand and Company-2020.		
3. PS Ver	rma and V K Agarwal, "Cell Biology (Cytology, Biomolecule and Molecular		
Biolog	y" S. Chand Publication-2016.		
4. Talwar	and Srivastava, "Textbook of Biochemistry and Human Biology" Eastern		
Econor	ny Edition, Prentice Hall, India-2002.		
Suggested	Online Link:		
<b>1.</b> Ele	ctron transport and oxidative phosphorylation: spring 2013 lecture 37 & 38		
<u>(pu</u>	rdue.edu)		
<b>2.</b> Niti	rogen metabolism: <u>lesson-10.pdf (nios.ac.in)</u>		
<b>3.</b> Car	bon dioxide assimilation and respiration: <u>chapter-2-carbon-dioxide-assimilation-</u>		
and	-respiration-for-pdf.pdf (asps.org.au)		
This course c	an be opted as an elective by the students of following subjects: Any one		
Suggested eq	uivalent online courses (MOOCs) for credit transfer: NO		
Electronic media and other digital components in the curriculum:			
Choose any o	e any one or more than one: (Electronic Media: Audio/Video Lectures, Online		
Counseling/V	rtual Classes/E-Contents/e-SLM/OER/supplementary links for		
reference/Vid	rence/Video Conferencing/Radio broadcast/Web Conferencing/ Other electronic and		
digital contents)			
Name of elec	tronic media: YouTube Year of incorporation: 2023-24		

Programme:	Programme: <b>B.Sc.</b> Year: <b>Third</b> Semester: <b>VI</b>		
	Subject: Biochemistry		
Course Code:	Course Code: DCEBCH-109N Course Title: Immunology		
Course Obje	ctives:		
• To dis	cuss the immunology and typ	es of immunity.	
• To dis	cuss the diversity in immune	system and types of imm	unoglobulin's.
• To dis	cuss the generation of antibod	ly diversity, B cell activa	ation and theory of clonal
selecti	on.		
• To dis	cuss the basic concept of disc	orderds of immune respon	nses and different
immu	noglobulin (IgG, IgM, IgA, Ig	gD and IgE)	
<b>Course Outc</b>	omes:		
CO 1: Abl	e to understanding of immune	e system and its types.	
CO 2: Abl	e to understanding about antig	gen, antibody and their ir	iteraction.
CO 3: kno	w about concept of autoimmu	nity, ELISA and T-cell r	receptor diversity.
CO 4: Also	o understanding about AIDS a	and active immunity & p	assive immunity.
CO 5: Brie	ef idea about SARS, hepatitis,	tolerance and hypersens	itivity.
Credits: 2		Type of Course: Elect	ive
Max. Marks:	100	Min. Passing Marks: 36	5
Block 1	Immune system		
Unit I	Immune system, innate and acquired immunity, definitions, non-immunological		
	barriers, cells and soluble mediators of innate immunity, cytokines.		
Unit II	Antigen and Antibody: Criteria of antigen city, haptens; classification, types and		
	functions of antibodies, anti-	genic determinants of im	munoglobulins
Unit III	Types of immunoglobulins	, generation of antibody	y diversity, B cell activation,
	theory of clonal selection, for	ormation of plasma and r	nemory cells.
Block 2	Diversity in Immune system	m	
Unit IV	Diversity in Immune system	n: Clonal selection theor	y, concept of antigen specific
	receptor, generation of antib	ody diversity,	
Unit V	Antigen-antibody: Measure	ment of antigen-antibod	ly interactions, agglutination,
	precipitations, opsonizatcon	, gel diffusion (ouchterlo	ony double immune diffusion).
Unit VI	Immune system and immunity: Enzyme linked immunosorbent assay (ELISA), T-		
	Cell receptor diversity, concept of autoimmunity.		
Block 3	Immune responses and Immunoglobulins		
Unit VII	Disorderds of immune responses: Autoimmunity, acquired immunodeficiency,		
	immune tolerance and hypersensitivity.		
Unit VIII	Cell mediated immunity: T-cell development, MHC locus, structure, function and		
	distribution of MHC glycop	roteins, cell mediated im	mune responses.
Unit IX	Immnogolobulins: IgG, IgN	M, IgA, IgD and IgE,	active immunity and passive
immunity. Brief idea of AIDS, SARS and hepatitis.			

#### **Suggested Text Book Readings:**

- 1. David L. Nelson and Michael Cox, "Lehninger Principles of Biochemistry" International Edition-2021.
- 2. Dr J L Jain, "Fundamentals of Biochemistry" S. Chand and Company-2020.
- 3. P S Verma and V K Agarwal, "Cell Biology (Cytology, Biomolecule and Molecular Biology" S. Chand Publication-2016.
- 4. Talwar and Srivastava, "Textbook of Biochemistry and Human Biology" Eastern Economy Edition, Prentice Hall, India-2002.

#### Suggested online link:

- 1. Introduction to Immunology: <u>Immunology.pdf (hmmcollege.ac.in)</u>
- 2. Components of Immune system: <u>components of immune system (dhingcollegeonline.co.in)</u>
- 3. Immune System: Immune System Handout (Soinc.Org)
- 4. Antigens: <u>Microsoft PowerPoint Chapter04-09 (nau.edu)</u>
- 5. Immunoglobulins: <u>Immunoglobulins.pdf (ndvsu.org)</u>

This course can be opted as an elective by the students of following subjects: Any one

Suggested equivalent online courses (MOOCs) for credit transfer: NO

Electronic media and other digital components in the curriculum:

Choose any one or more than one:(Electronic Media: Audio/Video Lectures, Online

Counseling/Virtual Classes/E-Contents/e-SLM/OER/supplementary links for

reference/Video Conferencing/Radio broadcast/Web Conferencing/ Other electronic and digital contents)

Name of electronic media: Youtube	Year of incorporation: 2023-24
Tume of electronic media. Toutabe	real of meorporation. 2023 21

Programme: 1	B.Sc.	Year: Third	Semester: VI
Subject: Biochemistry			
Course Code: SBSBCH-04N Course Title: Clinical biochemistry			ical biochemistry
Course Obje	Course Objectives:		
To dis	scuss the basics of clinic	cal biochemistry and clinical	enzymology.
• To dis	scuss the role and regula	ation of electrolyte content in	body fluids
• To dis	scuss the concept disord	lers of carbohydrate, lipids an	ıd nitrogen metabolism.
To dis	scuss the basic concept	of blood clotting, nutrition, d	rugs and cancer.
<b>Course Outc</b>	omes:		
CO 1: Abl	e to understanding of el	lectrolyte, enzymes, hormone	es and bone disorder.
CO 2: able	e to know the regulation	n of blood sugar, glycogen, an	nd diabetes mellitus.
CO 3: Abl	e to learn the density of	f lipoproteins, cholesterol, tri	glycerides and
pho	ospholipids in health and	d disease.	
CO 4: De	fine the concept of tube	feeding, parenteral nutrition	, drugs and alcohol
CO 5: Abl	e to known the types of	cancer, multiple steps of tur	nor development.
		1	
Credits: 4		Type of Course: Skill Enh	ancement Course
Max. Marks:	100	Min. Passing Marks: 36	
Block 1	ck 1 Basic in clinical chemistry and Clinical Enzymology		
	<b>Basic introduction:</b>		
Unit I	<b>Basic</b> concept of clinical biochemistry: A brief review of units and abbreviations		
	used in expressing concentrations and standard solutions.		
Unit II	Electrolytes and acid-base balance:		
	Role and regulation of electrolyte content in body fluids and maintenance of pH,		
	body fluids and fluid compartments		
Unit III	Clinical enzymology:		
	Enzymes and hormor	ies, plasma enzymes, isoenzy	mes with examples, liver
	damage, bone disorde	r.	
Block 2	Disorders of carboh	ydrate, lipids and Nitrogen	metabolism
TT ·/ TT7	Disorders of carbohydrate metabolism:		
Unit IV	Regulation of blood sugar, glycogen storage diseases, diabetes mellitus, glucose		
TT ·/ T7	and galactose tolerance tests, sugar levels in blood.		
Unit V	Disorders of lipids:		
	Low and high density lipoproteins, choiesterol, triglycerides and phospholipids in		
Linit VI	neath and disease, Gaucher's and Tay-Sach's disease		
Unit VI	Abnormalitian in nit	netabolism. Uramia	hyperuricensie permissioned
	factors affecting nitro	aen halance	nyperuncenna, porpnyna and
Block 3	Nutrition drugs and blood electring		
Unit VII	Nutrition and drugs		
	run non and drugs:	•	

	Routine hospital diets, special feeding methods, tube feeding, parenteral nutrition,			
	drugs, alcohol and toxicants			
Unit VIII	Diagnostic Enzymes:			
	Enzymes in health and diseases. I	Biochemical diagnosis of diseases by enzyme		
	assays- SGPT, CPK, LDH.			
Unit IX	X Blood Clotting:			
	Blood clotting mechanism-hem	orrhagic disorders-hemophilia, thrombotic		
	thrombocytopenic purpura, blood	groups, antigen and antibodies, circulating		
	anticoagulants.			
Unit X	Cancer:			
	Types of cancer, multiple steps of	tumor development, cell death and apoptosis,		
	carcinogens and cancer therapy.			
Suggested T	ext Book Readings:			
1. David L.	Nelson and Michael Cox, "Lehninge	er Principles of Biochemistry" International		
Edition-2	2021.			
<b>2.</b> Dr J L Ja	in, "Fundamentals of Biochemistry"	S. Chand and Company-2020.		
<b>3.</b> P S Vern	na and V K Agarwal, "Cell Biology (	Cytology, Biomolecule and Molecular		
Biology'	'S. Chand Publication-2016.			
<b>4.</b> Talwar a	nd Srivastava, "Textbook of Biochen	nistry and Human Biology" Eastern Economy		
Edition,	Prentice Hall, India-2002.			
5. Satyanar	5. Satyanarayana U., "Biochemistry" Elsevier India,2021			
Suggested online link:				
1. Pathop	1. Pathophysiology of Water and Electrolyte Metabolism: PowerPoint Presentation (bns-			
hungary.hu) 2 Netwient Dans Letomotions and East (002(1 add (calentate adv))				
<ol> <li>Nutrient-Drug Interactions and Food: <u>09361.pdf (colostate.edu)</u></li> <li>HANDROOK OF DRUG NUTRIENT INTERACTIONS: Handbook of Drug Nutrient</li> </ol>				
Interac	tions. 2nd Edition (Nutrition and Hea	lth) (usp.br)		
4. Enzym	es of diagnostic values: L12-Enzyme	es-of-diagnostic-values.pdf (ndvsu.org)		
5. Blood	Clotting Notes: <u>Blood Clotting Notes</u>	(murrieta.k12.ca.us)		
<b>6.</b> Cancer	: <u>book.pdf (tmc.gov.in)</u>			
This course can be opted as an elective by the students of following subjects: NA				
Suggested equivalent online courses (MOOCs) for credit transfer: NO				
Electronic n	Electronic media and other digital components in the curriculum:			
Choose any	one or more than one: (Electronic	Media: Audio/Video Lectures, Online		
Counseling/Virtual Classes/E-Contents/e-SLM/OER/supplementary links for				
disital contents)				
digital contents)				
Name of elec	ctronic media: Youtube	Year of incorporation: 2023-24		

#### Year-2023-2024 Syllabus of B.Sc. Programme<u>: [Subject Name: Botany]</u> In accordance with NEP-2020

Year	Sem.	Course Code	Paper Title	Theory/Practical	Max.	Credits
					Marks.	
	Ι	UGBY-101(N)	Cytology and Genetic	Theory	100	2
	Ι	UGBY-101(N) (P)	Practical Work	Practical Work	100	2
1	II	UGBY-102(N)	Plant Physiology	Theory	100	2
	II	UGBY-102(N) (P)	Practical Work	Practical Work	100	2
	III	UGBY-103(N)	Plant Diversity-I	Theory	100	2
	III	UGBY-103(N) (P)	Practical Work	Practical Work	100	2
	IV	UGBY-104(N)	Plant Diversity-II	Theory	100	2
2	IV	UGBY-104(N) (P)	Practical Work	Practical Work	100	2
	Skill I	Enhancement Course				
	IV	SBSBY-02(N)	Ecology	Theory	100	4
	Discip	oline Centric Elective C	ourse			
	V	DCEBY -105(N)	Embryology and	Theory	100	2
			Morphogenesis			
	V	DCEBY-106(N)	Plant Pathology and	Theory	100	2
			Microbiology			
3	V	DCEBY-107(N) (P)	Practical Work	Practical Work	100	2
	Discip	oline Centric Elective C	lourse			
	VI	DCEBY-108(N)	Molecular Genetics and	Theory	100	2
			biotechnology			
	VI	DCEBY-109(N)	Paleobotany,	Theory	100	2
			Palynology and			
			Economic Botany			
	VI	DCEBY-110(N) (P)	Practical Work	Practical Work	100	2
Total Marks/Credit   1500   32					32	

# **B.Sc.: Subject: Botany**

Course prereq	Course prerequisites: To study this course, a student must have qualified 10+2 with Biology			
Programme: B.	Programme: B.Sc. Year: I Semester: I			
Subject: Biolog	Subject: Biology			
Course Code: U	UGBY-101 (N) C	Course Title: Cytology and Gen	etic	
Course Objective	es: The main objective of	f the course is to make learners award	e of-	
	Cell science			
	• Eukaryotic plant ce	ell, various cellular organelles and ge	enetics	
Course Outcom	nes: (CO): Knowledge of	f plant cell and various cellular orgai	nelles.	
Salient	features of cell division in	in plants cell.		
Underst	tand the Mendel's laws of	f Heredity.		
Concep	t of linkage, crossing ove	er and chromosome mapping.		
Extranu	clear inheritance, structur	re, numerical abnormalities in chron	nosome and their effects.	
Knowle	edge of nature and structu	re of genetic material.		
Structure	re and function of gene.			
Credits: 02		Type of Course:	Core	
Max. Marks: 1	00 N	Min. Passing Marks: 36		
Block 1	Cytology			
	Cell structure and cell	llular organelles-I		
Unit I	<ul> <li>Chloroplast, I</li> </ul>	Mitochondria, Ribosome, Nucleo	lus and Nucleus, Plasma	
	membrane			
Unit II	Cell structure and cellular organelles-II			
	Endoplasmic	reticulum, Golgi-body, Lysoson	e and chromosome.	
Unit III	Cell cycle, Mitosis an	nd Meiosis		
Block 2	Genetics-I			
Unit IV	Pre-mendelian genetic	ics and Mendel's laws of inherita	nce	
Unit V	Linkage and crossing	gover		
Unit VI	Cytoplasmic inheritar	nce, sex linked		
Block 3	Genetics-II			
Unit VII	Pre-Chromosomal abo	perrations		
Unit VIII	Gene mutation and in	nduced mutation		
Unit IX	Genetics in Plant imp	provement		
Suggested Text Book Readings:				
Gardner, E.J., Simmons, M.J., Snustad, D.P. (1991). Principles of Genetics, John Wiley & sons, India. 8th				
Cell Biology And Genetics (Hindi) 2/e PBGupta P K (Hindi) rastogi Publications				
Cytogenetics, Plant Breeding, Evolution and Biostatistics ISBN #: 978-81-301-0066-1Sunil D Purohit &				
Gotam K Kukda, Apex Publishing House				
Genetics and Biotechnology Sunil D Purohit, K. Ahmed & Gotam K Kukda Apex Publishing House				
This course can be opted as an elective by the students of following subjects: NA				
Suggested equ	Suggested equivalent online courses (MOOCs) for credit transfer: NA			

Course prerequisites: To study this course, a student must have qualified 10+2 with Biology				
Programme: B.	Programme: B.Sc. Year: I Semester: II			
Subject: Biolog	Subject: Biology			
Course Code: U	Course Code: UGBY-102 (N) Course Title: Plant Physiology			
Course Objecti	ves: The main objective of the course is	to make learners aware	of-	
Plant w	ater relations, mineral nutrition, photosy	nthesis, Respiration and	growth harmones	
Course Outcom	nes (CO):			
Underst	and different process of plant water relation	tion.		
Underst	and process of photosynthesis.			
Process	of biological Nitrogen Fixation.			
Plant ho	ormones and their role in physiology of p	olant.		
Credits: 02	20	Type of Course: Cord	e	
Max. Marks: 10		Min. Passing Marks:	36	
BIOCK I	Plant Physiology-1	o and a second West	n Detential	
	Absorption of Water	on, Osmosis and wate	er Potential	
Unit II	Absorption of water	· ( 1 (		
	• water absorbing organ, act	ive water absorption t	neory and passive water	
Unit III	Accent of San			
Omt m	• Concepts theory of escent	of can with amphasia	on cohosion theory of ascent of	
	• Concepts, meory of ascent	of sap with emphasis	on conesion meory of ascent of	
Unit IV	Water Loss (Transpiration)			
Ollit I V	Stomatal structure mechan	ism of stomatal openi	ng and closing transpiration	
	guttation factors controllin	g transpiration	ing and closing, transpiration,	
Block 2	Plant Physiology-II			
Unit V	Mineral Nutrition			
	• Essential elements, macro and micro nutrient, role of essential elements.			
	transport of ion across cell membrane, active and passive transport.			
Unit VI	Photosynthesis			
	• Basic concept, equation of photosynthesis, evidence for the existence of light			
	and dark reaction, structure of chloroplast, discovery of two light reactions,			
	quantum yield, red drop, Emerson enhancement effect, photo system I & II, dark			
	reaction-the calvin cycle photorespiration and C <sub>4</sub> plants, CAM plant.			
Unit VII	Hormones			
	• Discovery and characteristic of plant hormones, role of auxins, giberellin,			
	cytokinin, ethylene, abscisi	c acid. Flowering har	mones, Phytochrome.	
Unit VIII	Respiration			
	• Aerobic and anaerobic	respiration, Glycoly	sis, TCA cycle, Oxidative	
	phosphorylation.			
Suggested Tex	t Book Readings:			
Jain, V.K. Fund	amental of Plant Physiology (7th ed	.) 2004. S. Chand and	Company.	
Salisbury, F.B.	& Ross, C.W. Plant Physiology (4th	ed.), 19992, Wadsow $d = 2006$ Wilson Dubli	orth Publishing Company.	
Panday, S.N. & Sinna, B.K. Plant Physiology (4th ed.), 2006, Vikas Publishing House Pvt. Ltd.				
Mukherjee, S. & Gilosh, A. Plant Physiology (2nd ed.), 2003, New Central Book Agency. Chaudhuri D. Kar, D.K. and Halder, S.A. Handbook of Plant Biogynthetic Pthysics 2008, New Central				
Chaudhuri, D., Kai, D.K., and Haider, S.A. Handbook of Flain Biosynthetic Pulways 2008, New Central Book Agencies				
Srivastava, HN, 2006, Pradeen's Botany Vol. V. Pradeen Publications, Jalandhar				
Verma, SK. Plant Physiology and Biochemistry, S. Chand & Sons, New Delhi.				
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This course can be opted as an elective by the students of following subjects: NA				
Suggested equ	Suggested equivalent online courses (MOOCs) for credit transfer: NA			

Electronic media and other digital components in the curriculum:				
Choose any one or more than one: (Electronic Media: Audio/Video Lectures, Online				
Counselling/Virtual Classes/E-Contents/e-SLM/OER/supplementary links f	for reference/Video			
Conferencing/Radio broadcast/Web Conferencing/ Other electronic and dig	ital contents)			
Name of electronic media Video Lectures				
https://youtu.be/AXMZ80EePQy				
https://youtu.be/2W5SKKFNdk	Year of incorporation:			
https://youtu.be/yEblrxy6mAU				
https://youtu.be/oElnm3y7Pzw 2021-22				
https://youtu.be/Vcs-4Ws/2Q				
e-SLM				

Course prereq	usites: To study this course, a student must have qualified 10+2 with Biology				
Programme: B	Sc. Year: I Semester: III				
Subject: Biolog	y .				
Course Code: I	JGBY-103 (N) Course Title: Plant Diversity-I				
Course Objecti	ves: The main objective of the course is to make learners aware of-				
Structure	e, reproduction and economic important of bacteria, virus and lichens.				
Morpho	blogy and life cycle of important groups of algae, fungi, bryophytes and pteridophytes.				
Course Outcon	nes:(CO):				
Knowle	dge of microbes and diversity of lower plants.				
Unders     Knowle	and the diversity of plant.				
Knowle     Habits	morphology, life cycle of fungi and their economic importance				
Credits: 02	Type of Course: Core				
Max. Marks: 1	0 Min. Passing Marks: 36				
Block 1	Plant Diversity-I				
	Bacteria, Virus and Lichen				
	Bacterial-Cell structure, Reproduction, Economic importance				
Unit I	• Virus- Biological status of virus, structure of bacteriophage & TMV,				
	replication.				
	• Lichen- Structure and economic importance of lichen.				
Unit II	Algae – I				
	• Morphology and life cycle of algae, unicellular form <i>chlamydomonas</i> , colonial				
	forms Volvox, Fliamentous form Nostoc, Heterotrichous forms Ectocarpus,				
	Thalloid form <i>Fucus</i> , Polysiphonoid form <i>Polysiphonia</i> .				
Unit III	I Algae – II				
	• Origin and evolution of sex, classification of Algae: Criteria for classification,				
	Economic importance of Alage, Habitats and distribution- Algae, Aquatic algae: Erash water, Marine babitate Special Habitate: Spill and sub-spice algae:				
Fresh water, Marine habitats Special Habitats; Soil and sub aerial algae.					
	Fungi				
Unit IV	• Introduction; Habitats, morphology, nutrition and reproduction, life cycle of				
Block 2	Plant Diversity I				
Unit V	Bryonhytes_I				
Unit v	• Introduction General characteristics adaptation to land habit morphology				
	anatomy and reproduction of Hepaticosida Riccia Marchantia Pellia				
	Anthoceropsida Anthoceros, Bryopsida-Sphagnum.				
	Bryophytes-II				
I Init VI	• Evolution of sporophytes in bryophytes. Importance and bio-functional uses of				
	bryophytes (Food, Medicine, Ecological Services, Industrial and Research				
	work)				
Unit VII	Pteridophytes-I				
	• General characteristics, and Life cycle of pteridophytes, Relationship with other				
	groups, stellar structure and evolution, Fern as a system for experimental				
<b>.</b>	studies, apogamy and apospory.				
Unit VIII Pteridophytes-II					
• Morphology, anatomy and life cycle <u>of Rhynia</u> , <u>Lycopodium, Selaginella</u> ,					
Suggested T	<u>Equisetum</u> , ana <u>Marsued</u> .				
Daribar NS (	1 DUUK NERUIII25. 1991) An introduction to Embryonhyta Vol I Bryonhyta Cantral Rook Danot Allahahad				
Rashid $\Delta$ (100	)) An Introduction to Pteridonhyta Vikas Publishing House Pyt I to New Delhi				
Vashishtha BR	Vashishtha BR, Sinha AK and Kumar A (2010) Botany for Degree Students – Pteridophyta, S. Chand and				
Company	Company				
Parihar NS (19	76) Biology and Morphology of Pteridophytes. Central Book Depot.				

Course prereq	uisites: To study thi	is course, a student must have qualifi	ed 10+2 with Biology		
Programme: B.	Programme: B.Sc. Year: I Semester: IV				
Subject: Biolog	<u>y</u>				
Course Code: U	JGBY-104 (N)	Course Title: Plant Diversity-II			
Course Objecti	ves: The main objectiv	e of the course is to make learners aware of	of-		
Gymno	sperms & life cycle of	Cycas & Punus.			
Various	aspects of anatomy of	vascular plants.			
System	of classification and de	etails of important dicot and monocot fami	lies.		
Course Outcon	nes:(CO):				
Underst	and morphology, anato	omy, life cycle and economic important ge	nera of gymnosperm.		
Anatom	y and secondary grow	th in some angiospermic plants.			
General	information of flower	ring plants.			
Underst	and aims, objective an	d importance of taxonomy.	a		
Various	system of plant classif	fication and description of some important	families.		
Credits: 02		Type of Course: Cor	re		
Max. Marks: 1	00	Min. Passing Marks: 36			
Block 1	Plant Diversity-II				
	Introduction of Gy	mnosperms			
Unit I	Introductio	n, characteristic, classification ar	nd economic importance of		
	gymnosper	m.			
Unit II	Cycas				
	Structure as	nd reproduction (life cycle)			
Unit III	Pinus				
	Structure as	nd reproduction (life cycle)			
Block 2	Plant Diversity-II	[			
Unit IV	Tissue system				
	<ul> <li>simple tissu</li> </ul>	ue, complex tissue.			
Unit V	Root				
	Primary an	d secondary structure of root			
Unit VI	Stem				
	<ul> <li>Primary an</li> </ul>	d secondary structure			
Unit VII	Anomalous Second	dary Growth			
	Anomalous	s secondary growth in <i>Bignonia</i> and <i>Ba</i>	<u>perhaavia</u> (dicot-stem),		
	<u>Dracaena</u>	(Monocot-stem)			
Block 3	Plant Taxonomy				
Unit VIII	Plant Taxonomy-I				
	History of	Economic botany with special reference	e to India, Bentham and		
	Hookers sy	stem of classification.			
Unit IX	Plant Taxonomy-II				
	<ul> <li>Details acc</li> </ul>	ount of following families: Dicot-Fan	nily- Asteraceae, Ranunculaceae		
	Brassicacea	ae, Solanaceae, Malvaceae, Mi	imosoideae, Caesalpinioideae,		
	Papilionoic	leae, Motocot-Family- Liliaceae, Orch	idaceae, Poaceae		
Suggested Text Book Readings:					
Bhatnagar, S.P. and Moitra, A. (1996). Gymnosperms. New Age International (P) Ltd Publishers, New					
Delhi, India.					
Sharma OP (1990) Textbook of Pteridophyta. MacMillan India Ltd. Delhi.					
Vashishtha BR	, Sinha AK and Kum	har A (2010) Botany for Degree Studen	nts – Gymnosperms, S. Chand		
Bhatnagar SP (	Brainagar SP (1990) Gymnosperms, New Age International Publisher.				
E.J.Eames . Mo	E.J.Eames . Worphology of vascular Flants, Standard University Press.				
Dickinson, W.	. (2000). Integrative	e Plant Anatomy. Harcourt Academic F	rtess, USA.		
Faini, A. (17/4). Flain Anatomy, reignon Fress, USA. Event D.E. (2006) Econy's Diant Anotomy, Mariatana, Calla, and Tianyan adding Diant Daday. Their					
Evert, R.F. (2006) Esau's Plant Anatomy: Meristems, Cells, and Tissues of the Plant Body: Their					

Structure, Function and Development. John Wiley and Sons, Inc.
Plant Systematics. Arun K. Pandey & Shruti Kansana. 2020. Jaya Publishing House
K. B. Anjaria, (2015)"Electronic Herbarium and Digital Database Preparation of Common Trees of
Anand District, Gujarat" MRP submitted to UGC, WRO, Pune 2015 (unpublished)
Pandey, B.P. 2007. Botany for Degree Students: Diversity of Seed Plants and their Systematics, Structure,
Development and Reproduction in Flowering Plants. S. Chand & Company Ltd,New Delhi.

Course prereq	uisites: To study this course, a student must have qualified 10+2 with Biology				
Programme: B	Programme: B.Sc. Year: I Semester: V				
Subject: Biolog	Subject: Biology				
Course Code:	SBSBY-02(N) Course Title: Ecology				
Course Objecti	ves: The main objective of the course is to make learners aware of-				
Ecosys	tem its various aspects which educate them about environment.				
Course Outcon	nes (CO):				
• Unders	tand the concept of environment, ecology and ecosystem.				
• Structu	re and organization of ecosystem with biotic and abiotic component.				
• Energy	flow and nutrient cycle in ecosystem.				
• Commu	unity, population and role of ecology in human welfare.				
Credits: 02	Type of Course: Core           00         Min. Descine Market 26				
Max. Marks: 1	00 Min. Passing Marks: 30				
BIOCK I	Ecology-1				
	Introduction To Ecology				
Unit II	Structure and function of Ecosystem				
	• Biolic and Adjoic components, Food chain, Food web, Pyramid, and Energy flow in accessistem. Biogeochemical cycle				
Unit III	Feelogical Succession				
	Basic concept succession in water and land (hydrosore and versare)				
	Dasic concept, succession in water and fand (flydrosere and xerosere)      Dellution				
Unit IV	Definition types of pollution: Air pollution water pollution Noise pollution				
Onitiv	control of pollution				
Block 2	Ecology-II				
Unit V	Ecological Adaptations in Plants				
	Hydrophytic and xerophytic adaptation.				
Unit VI	Edaphic Factors				
	• Definition and composition of soil, soil profile, soil erosion, soil conservation.				
Unit VII	Phytogeography				
	• Major plant community of world, soil, climate and vegetation of India.				
Unit VIII	Environmental Education				
Suggested Text Book Readings:					
Ecology And Environmental Biology by RBD Publisher Author: Bhatia - Jain - Kohli - Shrivastava -					
Singh – Verma					
Environmental Biology and Phytogeography ISBN #: 978-81-301-0064-7B. L. Chaudhary, Gotam K					
Kukda & Jitendra Kumar Joshi					
Odum, F.P. Fundamentals of Ecology, Latest Ed., Saunders					
Sharma, P.D. Elements of Ecology, Latest Ed., Rastogi Publications					
Singh, J. S., Singh, S.P. and Gupta, S. (2006). Ecology, Environment and Resource Conservation.					
Anamaya Publications, New Delhi.					

Course prerequ	uisites: To study this course, a student must have qualified 10+2 with Biology				
Programme: B.	Sc. Year: I Semester: V				
Subject: Biolog	<u>y</u>				
Course Code: I	DCEBY-105 (N) Course Title: Embryology and Morphogenesis				
Course Objecti	ves: The main objective of the course is to make learners aware of-				
Various	s aspects of angiosperm's embryology and phenomenon of morphogenesis in plant.				
Course Outcon	nes (CO):				
Knowle	edge about gametogenesis of anther and ovale.				
Pollinat	tion, fertilization along with development of embryo and endosperm.				
• Underst	tand Polyembryony, its application and morphogenesis.				
Credits: 02	I ype of Course: Core				
Max. Marks: 1	00 Min. Passing Marks: 36				
BIOCK I	Plant Embryology-1				
	Introduction to Embryology				
Unit II	Life Cycle of Angiosperm				
	Structure of flower; process of reproduction				
	Microsporogenesis				
Unit III	• Microsporogenesis and male gametophytes, microsporangium anther wan and sporogenesis tissue. Microsporogenesis Cytokinesis pollon tetrade Male				
	sporogenous issue. Microsporogenesis-Cytokinesis, polici terraus. Mate				
Unit IV	Megasporogenesis				
Oline I V	• Megasporagium and female gametophyte-1 Megasporangium - type of oyule				
	development of ovule parts of ovule 2 Megasporagenesis Female gametophyte				
	(Embryosac) structure of embryosac and types				
Block 2	Plant Embryology-II				
Block 2 Unit V	Plant Embryology-II Pollination				
Block 2 Unit V	<ul> <li>Plant Embryology-II</li> <li>Pollination</li> <li>Anther dehiscence, types of pollination, agents and types of cross pollination; artificial pollination.</li> </ul>				
Block 2 Unit V Unit VI	<ul> <li>Plant Embryology-II</li> <li>Pollination         <ul> <li>Anther dehiscence, types of pollination, agents and types of cross pollination; artificial pollination.</li> </ul> </li> <li>Fertilization</li> </ul>				
Block 2 Unit V Unit VI	Plant Embryology-II         Pollination       • Anther dehiscence, types of pollination, agents and types of cross pollination; artificial pollination.         Fertilization       • Germination of pollen grain, Growth of pollen tube, Entry of pollen tube into Ovule				
Block 2 Unit V Unit VI	<ul> <li>Plant Embryology-II</li> <li>Pollination         <ul> <li>Anther dehiscence, types of pollination, agents and types of cross pollination; artificial pollination.</li> </ul> </li> <li>Fertilization         <ul> <li>Germination of pollen grain, Growth of pollen tube, Entry of pollen tube into Ovule and Embryosac, Movement of sperm towards egg and polar nuclei. pollination and</li> </ul> </li> </ul>				
Block 2 Unit V Unit VI	Plant Embryology-II         Pollination         • Anther dehiscence, types of pollination, agents and types of cross pollination; artificial pollination.         Fertilization         • Germination of pollen grain, Growth of pollen tube, Entry of pollen tube into Ovule and Embryosac, Movement of sperm towards egg and polar nuclei. pollination and fertilization.				
Block 2 Unit V Unit VI Unit VI	<ul> <li>Plant Embryology-II</li> <li>Pollination <ul> <li>Anther dehiscence, types of pollination, agents and types of cross pollination; artificial pollination.</li> </ul> </li> <li>Fertilization <ul> <li>Germination of pollen grain, Growth of pollen tube, Entry of pollen tube into Ovule and Embryosac, Movement of sperm towards egg and polar nuclei. pollination and fertilization.</li> </ul> </li> <li>Post fertilization Development</li> </ul>				
Block 2 Unit V Unit VI Unit VII	<ul> <li>Plant Embryology-II</li> <li>Pollination <ul> <li>Anther dehiscence, types of pollination, agents and types of cross pollination; artificial pollination.</li> </ul> </li> <li>Fertilization <ul> <li>Germination of pollen grain, Growth of pollen tube, Entry of pollen tube into Ovule and Embryosac, Movement of sperm towards egg and polar nuclei. pollination and fertilization.</li> </ul> </li> <li>Post fertilization Development <ul> <li>Endosperm- Types of endosperm, Nuclear types, Cellular types, Helobial types,</li> </ul> </li> </ul>				
Block 2 Unit V Unit VI Unit VI	<ul> <li>Plant Embryology-II</li> <li>Pollination         <ul> <li>Anther dehiscence, types of pollination, agents and types of cross pollination; artificial pollination.</li> </ul> </li> <li>Fertilization         <ul> <li>Germination of pollen grain, Growth of pollen tube, Entry of pollen tube into Ovule and Embryosac, Movement of sperm towards egg and polar nuclei. pollination and fertilization.</li> </ul> </li> <li>Post fertilization Development         <ul> <li>Endosperm- Types of endosperm, Nuclear types, Cellular types, Helobial types, Function of endosperm, morphological nature of endosperm, embryogenesis –</li> </ul> </li> </ul>				
Block 2 Unit V Unit VI Unit VI	<ul> <li>Plant Embryology-II</li> <li>Pollination         <ul> <li>Anther dehiscence, types of pollination, agents and types of cross pollination; artificial pollination.</li> </ul> </li> <li>Fertilization         <ul> <li>Germination of pollen grain, Growth of pollen tube, Entry of pollen tube into Ovule and Embryosac, Movement of sperm towards egg and polar nuclei. pollination and fertilization.</li> </ul> </li> <li>Post fertilization Development         <ul> <li>Endosperm- Types of endosperm, Nuclear types, Cellular types, Helobial types, Function of endosperm, morphological nature of endosperm, embryogenesis – Development of dicot and mono-cot embryo, nutrition of embryo</li> </ul> </li> </ul>				
Block 2 Unit V Unit VI Unit VII	<ul> <li>Plant Embryology-II</li> <li>Pollination         <ul> <li>Anther dehiscence, types of pollination, agents and types of cross pollination; artificial pollination.</li> </ul> </li> <li>Fertilization         <ul> <li>Germination of pollen grain, Growth of pollen tube, Entry of pollen tube into Ovule and Embryosac, Movement of sperm towards egg and polar nuclei. pollination and fertilization.</li> </ul> </li> <li>Post fertilization Development         <ul> <li>Endosperm- Types of endosperm, Nuclear types, Cellular types, Helobial types, Function of endosperm, morphological nature of endosperm, embryogenesis – Development of dicot and mono-cot embryo, nutrition of embryo</li> </ul> </li> </ul>				
Block 2 Unit V Unit VI Unit VII Unit VIII	Plant Embryology-II         Pollination         • Anther dehiscence, types of pollination, agents and types of cross pollination; artificial pollination.         Fertilization         • Germination of pollen grain, Growth of pollen tube, Entry of pollen tube into Ovule and Embryosac, Movement of sperm towards egg and polar nuclei. pollination and fertilization.         Post fertilization Development         • Endosperm- Types of endosperm, Nuclear types, Cellular types, Helobial types, Function of endosperm, morphological nature of endosperm, embryogenesis – Development of dicot and mono-cot embryo, nutrition of embryo         Polyembryony and Apomixis         • Origin of polyembryony, causes of polyembryony and role of polyembryony in				
Block 2 Unit V Unit VI Unit VII Unit VIII	Plant Embryology-II         Pollination         • Anther dehiscence, types of pollination, agents and types of cross pollination; artificial pollination.         Fertilization         • Germination of pollen grain, Growth of pollen tube, Entry of pollen tube into Ovule and Embryosac, Movement of sperm towards egg and polar nuclei. pollination and fertilization.         Post fertilization Development         • Endosperm- Types of endosperm, Nuclear types, Cellular types, Helobial types, Function of endosperm, morphological nature of endosperm, embryogenesis – Development of dicot and mono-cot embryo, nutrition of embryo         Polyembryony and Apomixis         • Origin of polyembryony, causes of polyembryony and role of polyembryony in breeding. apomixis; Types of apomixes and it significance.				
Block 2 Unit V Unit VI Unit VII Unit VIII Block 2	<ul> <li>Plant Embryology-II</li> <li>Pollination         <ul> <li>Anther dehiscence, types of pollination, agents and types of cross pollination; artificial pollination.</li> </ul> </li> <li>Fertilization         <ul> <li>Germination of pollen grain, Growth of pollen tube, Entry of pollen tube into Ovule and Embryosac, Movement of sperm towards egg and polar nuclei. pollination and fertilization.</li> </ul> </li> <li>Post fertilization Development         <ul> <li>Endosperm- Types of endosperm, Nuclear types, Cellular types, Helobial types, Function of endosperm, morphological nature of endosperm, embryogenesis – Development of dicot and mono-cot embryo, nutrition of embryo</li> </ul> </li> <li>Polyembryony and Apomixis         <ul> <li>Origin of polyembryony, causes of polyembryony and role of polyembryony in breeding. apomixis; Types of apomixes and it significance.</li> </ul> </li> <li>Plant Morphogenesis and Polarity</li> </ul>				
Block 2 Unit V Unit VI Unit VII Unit VIII Block 2 Unit IX	<ul> <li>Plant Embryology-II</li> <li>Pollination         <ul> <li>Anther dehiscence, types of pollination, agents and types of cross pollination; artificial pollination.</li> </ul> </li> <li>Fertilization         <ul> <li>Germination of pollen grain, Growth of pollen tube, Entry of pollen tube into Ovule and Embryosac, Movement of sperm towards egg and polar nuclei. pollination and fertilization.</li> </ul> </li> <li>Post fertilization Development         <ul> <li>Endosperm- Types of endosperm, Nuclear types, Cellular types, Helobial types, Function of endosperm, morphological nature of endosperm, embryogenesis – Development of dicot and mono-cot embryo, nutrition of embryo</li> </ul> </li> <li>Polyembryony and Apomixis         <ul> <li>Origin of polyembryony, causes of polyembryony and role of polyembryony in breeding. apomixis; Types of apomixes and it significance.</li> </ul> </li> <li>Plant Morphogenesis and Polarity         <ul> <li>Morphogenesis</li> <li>Morphogenesis</li> </ul> </li> </ul>				
Block 2 Unit V Unit VI Unit VII Unit VIII Block 2 Unit IX	<ul> <li>Plant Embryology-II</li> <li>Pollination         <ul> <li>Anther dehiscence, types of pollination, agents and types of cross pollination; artificial pollination.</li> </ul> </li> <li>Fertilization         <ul> <li>Germination of pollen grain, Growth of pollen tube, Entry of pollen tube into Ovule and Embryosac, Movement of sperm towards egg and polar nuclei. pollination and fertilization.</li> </ul> </li> <li>Post fertilization Development         <ul> <li>Endosperm- Types of endosperm, Nuclear types, Cellular types, Helobial types, Function of endosperm, morphological nature of endosperm, embryogenesis – Development of dicot and mono-cot embryo, nutrition of embryo</li> </ul> </li> <li>Polyembryony and Apomixis         <ul> <li>Origin of polyembryony, causes of polyembryony and role of polyembryony in breeding. apomixis; Types of apomixes and it significance.</li> </ul> </li> <li>Plant Morphogenesis and Polarity         <ul> <li>Morphogenesis and Factors Affecting Morphogenesis</li> <li>Morphogenesis and Factors Affecting Morphogenesis</li> </ul> </li> </ul>				
Block 2 Unit V Unit VI Unit VII Unit VIII Block 2 Unit IX Unit-X	<ul> <li>Plant Embryology-II</li> <li>Pollination         <ul> <li>Anther dehiscence, types of pollination, agents and types of cross pollination; artificial pollination.</li> </ul> </li> <li>Fertilization         <ul> <li>Germination of pollen grain, Growth of pollen tube, Entry of pollen tube into Ovule and Embryosac, Movement of sperm towards egg and polar nuclei. pollination and fertilization.</li> </ul> </li> <li>Post fertilization Development         <ul> <li>Endosperm- Types of endosperm, Nuclear types, Cellular types, Helobial types, Function of endosperm, morphological nature of endosperm, embryogenesis – Development of dicot and mono-cot embryo, nutrition of embryo</li> </ul> </li> <li>Polyembryony and Apomixis         <ul> <li>Origin of polyembryony, causes of polyembryony and role of polyembryony in breeding. apomixis; Types of apomixes and it significance.</li> </ul> </li> <li>Plant Morphogenesis and Factors Affecting Morphogenesis</li> <li>Morphogenesis and Factors Affecting Morphogenesis</li> </ul>				
Block 2 Unit V Unit VI Unit VII Unit VIII Block 2 Unit IX Unit-X	<ul> <li>Plant Embryology-II</li> <li>Pollination         <ul> <li>Anther dehiscence, types of pollination, agents and types of cross pollination; artificial pollination.</li> </ul> </li> <li>Fertilization         <ul> <li>Germination of pollen grain, Growth of pollen tube, Entry of pollen tube into Ovule and Embryosac, Movement of sperm towards egg and polar nuclei. pollination and fertilization.</li> </ul> </li> <li>Post fertilization Development         <ul> <li>Endosperm- Types of endosperm, Nuclear types, Cellular types, Helobial types, Function of endosperm, morphological nature of endosperm, embryogenesis – Development of dicot and mono-cot embryo, nutrition of embryo</li> </ul> </li> <li>Polyembryony and Apomixis         <ul> <li>Origin of polyembryony, causes of polyembryony and role of polyembryony in breeding. apomixis; Types of apomixes and it significance.</li> </ul> </li> <li>Plant Morphogenesis and Polarity         <ul> <li>Morphogenesis and Factors Affecting Morphogenesis</li> <li>Morphogenesis and Factors Affecting Morphogenesis</li> </ul> </li> </ul>				
Block 2 Unit V Unit VI Unit VII Unit VIII Block 2 Unit IX Unit-X Suggested Tex Bhoiwani S S	<ul> <li>Plant Embryology-II</li> <li>Pollination         <ul> <li>Anther dehiscence, types of pollination, agents and types of cross pollination; artificial pollination.</li> </ul> </li> <li>Fertilization         <ul> <li>Germination of pollen grain, Growth of pollen tube, Entry of pollen tube into Ovule and Embryosac, Movement of sperm towards egg and polar nuclei. pollination and fertilization.</li> </ul> </li> <li>Post fertilization Development         <ul> <li>Endosperm- Types of endosperm, Nuclear types, Cellular types, Helobial types, Function of endosperm, morphological nature of endosperm, embryogenesis – Development of dicot and mono-cot embryo, nutrition of embryo</li> </ul> </li> <li>Polyembryony and Apomixis         <ul> <li>Origin of polyembryony, causes of polyembryony and role of polyembryony in breeding. apomixis; Types of apomixes and it significance.</li> </ul> </li> <li>Plant Morphogenesis and Polarity         <ul> <li>Morphogenesis and Factors Affecting Morphogenesis</li> <li>Polarity                 <ul> <li>Symmetry, Totipotency</li> <li>tt Book Readings:</li></ul></li></ul></li></ul>				
Block 2 Unit V Unit VI Unit VII Unit VIII Block 2 Unit IX Unit-X Suggested Tex Bhojwani, S.S. House	<ul> <li>Plant Embryology-II</li> <li>Pollination         <ul> <li>Anther dehiscence, types of pollination, agents and types of cross pollination; artificial pollination.</li> </ul> </li> <li>Fertilization         <ul> <li>Germination of pollen grain, Growth of pollen tube, Entry of pollen tube into Ovule and Embryosac, Movement of sperm towards egg and polar nuclei. pollination and fertilization.</li> </ul> </li> <li>Post fertilization Development         <ul> <li>Endosperm- Types of endosperm, Nuclear types, Cellular types, Helobial types, Function of endosperm, morphological nature of endosperm, embryogenesis – Development of dicot and mono-cot embryo, nutrition of embryo</li> </ul> </li> <li>Polyembryony and Apomixis         <ul> <li>Origin of polyembryony, causes of polyembryony and role of polyembryony in breeding. apomixis; Types of apomixes and it significance.</li> </ul> </li> <li>Plant Morphogenesis and Polarity         <ul> <li>Morphogenesis and Factors Affecting Morphogenesis</li> <li>Polarity                 <ul> <li>Symmetry, Totipotency</li> <li>tt Book Readings:</li></ul></li></ul></li></ul>				

Course prerequisites: To study this course, a student must have qualified 10+2 with Biology				
Programme: B.	Sc.	Year: I	Semester: V	
Subject: Biolog	<u>gy</u>			
Course Code: I	DCEBY-106 (N)	Course Title: Plant Path	ology and Microbiology	
Course Objectiv	ves: The main objective of	of the course is to make learn	ners aware of-	
• Scope a	nd importance of plant pa	athology.		
Know the second se	he prevention and control	measures of plant diseases.		
Life cyc	cle of some important plan	nt diseases.		
Soil, wa	ater and dairy microbiolog	gy.		
Course Outcom	nes (CO):			
Introduce	ction of plant pathology, s	symptions, dissemination and	d various control methods for disease.	
<ul> <li>Descrip</li> </ul>	tion of some important di	seases of plants.		
Knowle	dge of water, soil and dai	ry microbiology.		
Credits: 02		Type of C	ourse: Core	
Max. Marks: 10	00	Min. Passing Marks: 36		
Block 1	Plant Pathology			
Unit I	Introduction of plant	pathology		
Unit II	Symptoms of plant d	liseases caused by fungi,	bacteria and virus	
Unit III	Control of plant diseases : various methods used for disease control			
Unit IV	Dissemination of pathogen, epidemiology and disease forecasting			
Unit V	Diseases of plant.			
	<ul> <li>White Rust o</li> </ul>	f Crucifer, wilt of arhar,	damping off, Late Blight of potato, Early	
	Blight of Pota	ato, Black Rust of wheat, 7	Tikka Disease of Groundnut Loose smut of	
	wheat.			
Block 2	Microbiology			
Unit VI	Sewage microbiolog	У		
	<ul> <li>What is sewage</li> </ul>	ge, various process of trea	tment of sewage.	
Unit VII	Soil microbiology			
	Humus, Role	of microbes in various cyc	cles: Nitrogen, Carbon, Phosphorous and	
	sulpher in soil.			
Unit VIII	Jnit VIII Dairy microbiology			
Suggested Text Book Readings:				
Microbiology Fundamental and Applications (hindi) (pb)				
Modern Microbiology (hindi) (hb) ISBN: 9788177543599Edition : 1Year : 2018Author : Dr. Purohit SS,				
Dr. Singh T Publisher : Agrobios (India)				
"Plant patholog	gy by R.S. Mehrotra, Ta	ata McGraw-Hill Educatio	on" are included in reading resources list	
Sharma, P. D. 2	2012, Microbiology and	d Plant Pathology, Rastogi	Publication Pvt Ltd., Meerut, India.	
Singh, R. P. 20	07. Microbial Taxonom	ny and Culture Techniques	s, Kalyani Publication, New Delhi.	

Course prereq	uisites: To study this course, a student must have qualified 10+2 with Biology			
Programme: B.	.Sc. Year: I Semester: VI			
Subject: Biolog	 ۷			
Course Code: I	DCEBY-108 (N) Course Title: Molecular Genetics and Biotechnology			
Course Objecti	ves: The main objective of the course is to make learners aware of-			
• The ger	nomic organization.			
Genetic	c engineering.			
Concep	ot of operon its structure and regulation.			
Basic p	rotocols for plant tissue culture.			
Course Outcon	nes (CO):			
Underst	tand DNA, RNA, gene expression and regulation.			
Genetic	e engineering, biotechnology and its applications in human welfare with special reference to			
agricult	ture.			
Credits: 02	Type of Course: Core			
Max. Marks: 1	00 Min. Passing Marks: 36			
Block 1	Molecular genetics- I			
Unit I	Nucleic acids (DNA & RNA) and Genetic Materials			
Unit II	Structure of DNA, Replication and Types of RNA			
Unit III	Gene Expression, Transcription and Translation in Prokaryotes and Eukaryotes			
Block 2	Molecular genetics- II			
Unit IV	Gene Regulation, Operon Concept and Transposons			
Unit V	Recombinant DNA Technology and Transgenic Plant			
Unit VI	Monoclonal Antibodies, DNA Probe and DNA Fingerprinting			
Block 3	Biotechnology			
Unit VII	Tissue Culture and Its Techniques			
Unit VIII	Culture of Different Tissue, Meristem Culture, Anther Culture Pollen culture, Protoplast			
	culture and Embryo cultures.			
Unit IX	Applications of Biotechnology in human welfare with special reference to agriculture			
Suggested Tex	kt Book Readings:			
Karp, G. 2010.	Cell and Molecular Biology: Concepts and Experiments. 6th Edition. John Wiley & Sons.			
Inc.				
Advanced Met	hods In Physiology And Biochemistry (pb)ISBN : 9789381191132Edition : 01Year :			
2016Author : Padmanaban G , Chandrasekaran CN , Thangavelu AU , Dr. Sivakumar R , Kalimuthu N ,				
Dr. Boominathan P, Dr. Anbarasan P, Agrobios.				
Methods in Plant Biochemistry and Molecular Biology. 1997. Dashek, WV (ed.). CRC Press.				
Thimmaiah, SR. 2004. Standard Methods of Biochemical Analysis. Kalyani Publishers.				
Henry, RJ. 1997. Practical Application of Plant Molecular Biology. Chapman & Hall, London				
M K Raxdan An Introduction to Plant Tissue Culture -; Oxfird & IBH Publishing Co.Pvt. Ltd., New Delhi				
Veer Bala Rastogi (2008), Fundamentals of Molecular Biology Ane Books Pvt. Ltd				
J. Nair Introduc	ction to Genetic Engineering & Biotechnology.Jones & Bartlett Publishers, Boston,USA.			

Course prerequisites: To study this course, a student must have qualified 10+2 with Biology				
Programme: B.Sc.	ogramme: B.Sc. Year: I Semester: VI			
Subject: Biology		·		
Course Code: DCEH	3Y-109 (N)	Course Title: Pal	eobotany, Palynology and Economic	
		Botany		
Course Objectives: 1	The main objective of	of the course is to ma	ke learners aware of-	
Foss	sils, pollens and vari	ous plants of econon	nic use.	
Course Outcomes (C	CO):			
Understandin	g of fossils & Vario	us technology techn	que used for their study.	
Palynology a	nd its scope.	_		
• Economic us	es of various plant p	roducts.		
Credits: 02		Ту	pe of Course: Core	
Max. Marks: 100	1	Min. Passing Ma	rks: 36	
Block 1	Paleobotany ar	nd palynology		
Unit I	Introduction and	d techniques to stu	dy of fossils, Geological time scale	
Unit II	Kinds of fossils	s and reconstruction	on of fossil, form-genera, Organ-genera,	
	Reconstruction of	of fossil.		
Unit III	Concept and sc	ope of palynology	Y: Pollen units, Pollen preparation, acetolysis	
	method.			
Block 2	Economic bota	any is divided int	o five units as under	
Unit IV	Spices and flat	voring materials:	Ginger, Turmeric, Clove, Saffron, Coriander;	
Linit V	Botanical desci	and Coffeet Deten	and uses.	
	Etherer Jute El	and Conee; Botan	tear description, cultivation and uses.	
	Fibers: Jule, Fi	Waad Dubbara	tion; Bolanical description, cultivation and uses.	
Unit VII	Forest products:	wood, Rudders, w	Jum and Resines; botanical description,	
	Madiainal plant	ses.	dana Anirina Aniren Enhadrinas hataniaal	
Unit VIII	description oult	s: Kauwollia, Bell	idona, Quimine, Opium, Epnedrine; bolanicai	
Suggested Text De	description, cult	vation and uses.		
Suggested Text Book Readings:				
<b>P.K.K.</b> Naif- A lexibook of Palynology.				
Kocnnar, S.L. (2011). Economic Botany in the Tropics, MacMillan Publishers India Ltd., New Delhi. 4th				
Sambamurthy AVSS & Subrahmanyam NS (2000) Economic Botany of Cron Plants Asistech				
Publishers New Delhi				
Raychudhuri S.P. 1991 (Ed.) Recent advances in Medicinal aromatic and spice crops. Vol 1. Today&				
Tomorrow's printers and publishers. New Delhi				
Tomorrow sprincers and publishers, new Denn				

Raychudhuri, S.P., 1991. (Ed.) Recent advances in Medicinal aromatic and spice crops. Vol.1,Today& Tomorrow's printers and publishers, New Delhi.

#### Year-2023-2024 Syllabus of B.Sc. Programme<u>: [Subject Name: Chemistry</u>] In accordance with NEP-2020

Year	Sem.	Course Code	Paper Title Theory/Practical		Max.	Credits
					Marks.	
1	I	UGCHE -101N	INORGANIC CHEMISTRY I (BASIC INORGANIC CHEMISTRY)	VORGANIC CHEMISTRY I Theory ASIC INORGANIC HEMISTRY)		2
		UGCHE -101P(N)	Practical Work	Practical	100	2
		UGCHE -102N	ORGANIC CHEMISTRY I (BASIC ORGANIC CHEMISTRY)	Theory 100		2
1	П	UGCHE -102P(N)	Practical Work	Practical	100	2
		Skill Enhancen	ient Course			
		SBSCHE-02N	ADVANCED ANALYTICAL TECHNIQUES	Theory	100	4
		UGCHE -103N	PHYSICAL CHEMISTRY I (BASIC PHYSICAL CHEMISTRY)	Theory	100	2
2	Ш	UGCHE -103P(N)	Practical Work	Practical	100	2
		Skill Enhancen	ient Course			
	SBSCHE-01N ORGANIC CHEMISTRY II (ADVANCE ORGANIC CHEMISTRY)		Theory	100	4	
2	IV	UGCHE -104N	INORGANIC CHEMISTRY II (ADVANCE INORGANIC CHEMISTRY)	Theory	100	2
	UGCHE -104P(N) Practical Work Practical		100	2		
Discip	line Ce	ntric Elective Cours	Se			
2		DCECHE -105N	PHYSICAL CHEMISTRY II (ADVANCE PHYSICAL CHEMISTRY)	Theory	100	2
3	3 V DCECHE -106N INORGANIC CHEMISTRY III (SELECTED TOPICS IN INORGANIC CHEMISTRY)		Theory	100	2	
	DCECHE -107P(N) Practical Work Practical		Practical	100	2	
Discip	oline Ce	ntric Elective Cours	se			
2	. //	DCECHE -108N	ORGANIC CHEMISTRY III (SELECTED TOPICS IN ORGANIC CHEMISTRY)	Theory	100	2
3 VI		DCECHE -109N	PHYSICAL CHEMISTRY III (SELECTED TOPICS IN PHYSICAL CHEMISTRY)	Theory	100	2
		DCECHE -110P(N)	Practical Work	Practical	100	2
			۱	Fotal Marks/Credit	1600	32

	Syna	bus for [D.Sc.]: Subject				
Course prere	equisites: 10+2	with Chemistry				
Programme:B.Sc. Year: 1 Semester: I						
Subject: Chemistry						
Course Code: UGCHE -101N Course Title: INORGANIC CHEMISTRY I (BASIC INORGANIC CHEMISTRY)						
<b>Course Objec</b>	Course Objectives:					
To provide kn	owledge about s	ructure of atoms and	associated important rules, imp	portance of		
chemistry of	elements, bondi	ng and properties of a	ny compound/material. Several	parameters		
associated wit	th elements, Soli	d state chemistry and	chemistry of elements belonging	to s-block,		
noble gases ar	nd main group.					
<b>Course Outco</b>	omes:					
CO-1 Structur	re of atoms and a	ssociated important ru	es, importance of chemistry of e	lements.		
CO-2 Ionic, c	ovalent and non-	covalent bonding which	h always play pivotal role in dec	iding the		
chemistry and	properties of an	y compound/material.		-		
CO-3 Periodi	c properties of el	ements and several par	ameters associated with elements	8		
CO-4 Solid st	ate chemistry wl	nich forms the basis of	he development of targeted crys	talline		
solids inculcat	ting varied defec	ts which induces varie	y of materials properties viz.			
piezoelectricit	y.					
CO-5 Chemis	try of elements l	elonging to s-block, n	ble gases and main group.			
Credits: 2	•	Тур	e of Course: Core			
Max. Marks: 1	00	Min. Passing Marks: 3	5			
Block 1						
	Atomic Struct	ure				
	Historical con	cepts of atomic struc	ure. Idea of de-Broglie's ma	tter waves.		
	Heisenberg's u	ncertainly principle, s	gnificant of $\psi$ and $\psi^2$ . Schrodir	iger's wave		
Unit I	equation for H	atom: Radial and ang	ilar wave functions: quantum n	umbers and		
	shapes of s. p. o	and f orbitals: Aufba	and Pauli Exclusion Principle.	Variation of		
	orbital energie	s with atomic number	and energy level diagram: Lo	ng form of		
	periodic table h	ased on electronic con	iguration.	-8		
Unit II	Periodic prope	rties of elements				
e int fi	Types of radii	(Covalent Crystal an	l Van der Waal): Electron affi	nity and its		
	variation. Ionis	ation potential Factor	affecting the magnitude of I P	Concept of		
	effective nucle	ar charge and shieldin	effect (Calculation of Screeni	ng constant		
	with Slater's rul	es). Electronegativity	Pauling Mulliken and Allred Ro	chow scale)		
	and its variation		r aanne, mannen and rinea re-	enow seule)		
Unit III	Chemical Bon	ding				
Olint III	(i) Ionic Bondi	ng. Conditions favour	ng the ionic bond radius ratio an	d structure		
	of ionic solids	ing. Conditions favour	ig the forme bond, radius fatto an	a structure		
	Concept of latt	ice energy and Born-H	aber cycle Polarisation of ions	and Faian's		
	rules					
(ii) Covalent and brief idea of other bonds.						
	(II) <b>Covarent</b> a	cted valence bond the	ry (VBT) and hybrid orbital des	cription (sp		
	$sn^2 sn^3 sn^3 dn^3$	$d \sin^3 d^2$ ) using simple	illustrations determination of th	e shapes of		
	by, sp, sp u a	$ions viz NH_2 H_2O$	$H_2O^+$ SE4 CIE <sub>2</sub> ICl <sup>2-</sup> and I <sup>3-</sup>	by VCEDD		
	concept Conce	nt of maximum coval	$n_{3}$ , $n_{4}$ , $n_{3}$ , $n_{5}$ and $n_{5}$	entre bond		
	MO Theory ho	monuclear ( $H_2 H^{2+} R_2$	$N_2$ , $O_2$ , $Cl_2$ ) and heteronuclear ((	CN CO and		

from dipole moment and electronegativity. Multicenter bonding in electron deficient molecules.         (iii) Weak Interactions: Hydrogen bonding (Inter and Intra Molecular), Vander Waals forces.         (iv) Metallite Bond: Theories of bonding in metals; Free electron, VB and Band theories.         Block 2         Unit IV       General Studies of s block elements         Comparative study, diagonal relationships, salient features of hydrides, solvation and complexation tendencies including their function in biosystems, an introduction to alkyls and aryls; Chemical reactivity of alkali and earth alkaline metals; Uses of s-block elements and their compounds (Li, Na and K only), Organometallic compounds of Li, Na and K only), Organometallic compounds of Li, Na and K only), Organometallic softwares of diborane, borazine, hydrazine, interhalogens and polyhalides and fluorides of xenon. Structure and basicities of oxyacids of B, P and S. Structural features of hydrides, halides, oxides and oxyacids.         Unit VI       Oxidation and Reduction Electrochemical series and its applications. EMF diagrams and their utility. Principle involved in the extraction of the elements.         Suggested Text Book Readings:       Text Books (Cheory Course):         (a) Concise Inorganic Chemistry, J.D. Lee, Blackwell Science Ld.       (b) Inorganic Chemistry, J.D. Lee, Blackwell Science Ld.         (b) Inorganic Chemistry, Pure, Filen A. Keiter, Richard L. Keiter, Addison Wesley Longman (Singapore) Pvt. Ld.       (c) Basic Inorganic Chemistry, BL Huheey, Ellen A. Keiter, Richard L. Keiter, Addison Wesley Longman (Singapore) Pvt. Ld.         (b) Inorganic Chemistry, J.D. Lee, Blackwell Science Ld.		NO) diatomic molecules, bond strength, and bond energy, percent ionic character				
deficient molecules.         (iii) Weak Interactions: Hydrogen bonding (Inter and Intra Molecular), Vander Waals forces.         (iv) Metallic Bond: Theories of bonding in metals; Free electron, VB and Band theories.         Block 2         Unit IV       General Studies of s block elements         Comparative study, diagonal relationships, salient features of hydrides, solvation and complexation tendencies including their function in biosystems, an introduction to alkyls and aryls; Chemical reactivity of alkali and earth alkaline metals; Uses of s-block elements and their compounds (Li, Na and K only), Organometallic compounds of Li, Na, K, Be and Mg. Polyether complexes (Crown ether complexes) of alkali metals; Extraction and isolation of Li, Be and Ra from their minerals.         Unit V       General Studies of p- block elements         Group wise discussion with respect to electronic configuration, ionisation potential, electron affinity, electronegativity, atomic and ionic radii, oxidation states, catenation and inert pair effect (wherever applicable). Preparation, properties and structures of diborane, borazine, hydrazine, interhalogens and polyhalides and fluorides of xenon. Structure and basicities of oxyacids of B, P and S. Structural features of hydrides, halides, oxides and oxyacids.         Unit VI       Oxidation and Reduction         Electrode potential, electrochemical series and its applications. EMF diagrams and their utility. Principle involved in the extraction of the elements.         Suggested Text Book Readings:       Text Books (Theory Courses):         (a) Concise Inorganic Chemistry, T.D. Lee, Blackwell Science Ltd.		from dipole moment and electronegativity. Multicenter bonding in electron				
(iii) Weak Interactions: Hydrogen bonding (Inter and Intra Molecular), Vander Waals forces.         (iv) Metallic Bond: Theories of bonding in metals; Free electron, VB and Band theories.         Block 2         Unit IV       General Studies of s block elements         Comparative study, diagonal relationships, salient features of hydrides, solvation and complexation tendencies including their function in biosystems, an introduction to alkyls and aryls; Chemical reactivity of alkali and earth alkaline metals; Uses of s-block elements and their compounds (Li, Na and K Only), Organometallic compounds of Li, Na and K Only), Organometallic compounds of Li, Na and K Only), Organometallic scompounds of Li, Na and K Only), Organometallic compounds of Li, Na (K Be and Mg. Polyether complexes (Crown ether complexes) of alkali metals; Extraction and isolation of Li, Be and Ra from their minerals.         Unit V       General Studies of p-block elements         Group wise discussion with respect to electronic configuration, ionisation potential, electron affinity, electronegativity, atomic and ionic radii, oxidation states, catenation and inert pair effect (wherever applicable). Preparation, properties and structures of diborane, borazine, hydrazine, interhalogens and polyhalides and fluorides of xenon. Structure and basicities of oxyacids of B, P and S. Structural features of hydrides, halides, oxides and oxyacids.         Unit VI       Oxidation and Reduction         Eletrode potential, electrochemical series and its applications. EMF diagrams and their utility. Principle involved in the extraction of the elements.         Suggested Text Book Readings:       () Concist Ghorganic Chemistry, J.E. Lee, Blackwell Science Ltd. <td></td> <td colspan="5">deficient molecules.</td>		deficient molecules.				
Waals forces.       (iv) Metallic Bond: Theories of bonding in metals; Free electron, VB and Band theories.         Block 2       Unit IV       General Studies of s block elements         Comparative study, diagonal relationships, salient features of hydrides, solvation and complexation tendencies including their function in biosystems, an introduction to alkyls and aryls; Chemical reactivity of alkali and earth alkaline metals; Uses of s-block elements and their compounds (Li, Na and K only), Organometallic compounds of Li, Na, K, Be and Mg. Polyether complexes (Crown ether complexes) of alkali metals; Extraction and isolation of Li, Be and Ra from their minerals.         Unit V       General Studies of p- block elements         Group wise discussion with respect to electronic configuration, ionisation potential, electron affinity, electronegativity, atomic and ionic radii, oxidation states, catenation and inert pair effect (wherever applicable). Preparation, properties and structures of diborane, borazine, hydrazine, interhalogens and polyhalides and fluorides of xenon. Structure and basicities of oxyacids of B, P and S. Structural features of hydrides, halides, oxides and oxyacids.         Unit VI       Oxidation and Reduction         Electrode potential, electrochemical series and its applications. EMF diagrams and their utility. Principle involved in the extraction of the elements.         Suggested Text Book Readings:         Test Books (Theory Courses):         (a) Concise Inorganic Chemistry, J.D. Lee, Blackwell Science Ltd.         (b) Inorganic Chemistry, J.E. Huheey, Ellen A. Keiter, Richard L. Keiter, Addison Wesley Longman (Singapore) Pvt. Ltd.		(iii) Weak Interactions: Hydrogen bonding (Inter and Intra Molecular), Vander				
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Electrode potential, electrochemical series and its applications. EMF diagrams and their utility. Principle involved in the extraction of the elements.         Suggested Text Book Readings:         Text Books (Theory Courses):         (a) Concise Inorganic Chemistry, J.D. Lee, Blackwell Science Ltd.         (b) Inorganic Chemistry, Puri, Sharma, Kalia and Kaushal.         (c) Pradeep's Inorganic Chemistry, K.K. Bhasin, Pradeep Publication.         (d) Chemistry of degree students, R. L. Madan         Reference Books:         (a) Inorganic Chemistry, J.E.Huheey, Ellen A. Keiter, Richard L. Keiter, Addison Wesley Longman         (Singapore) Pvt. Ltd.         (b) Inorganic Chemistry, F.A. Cotton, G. Wilkinson and P.L. Gaus, Wiley.         (d) Concepts of Models of Inorganic Chemistry, B.Douglas,         D.McDaniel and J Alexander, John Wiley.         (e) Inorganic Chemistry, N.W. Porterfield, Addison - Wesley.         (f) Inorganic Chemistry, G.L. Meissler and D.A. Tarr, Prentice-Hall.         Suggested online links:         http://heecontent.upsdc.gov.in/Home.aspx         https://nptel.ac.in/courses/104/106/104106096/         https://nptel.ac.in/courses/104/103/11/#         Electronic media and other digital components in the curriculum:	Unit VI	Oxidation and Reduction				
Liter utility. Principle involved in the extraction of the elements.           Suggested Text Book Readings:           Text Books (Theory Courses):           (a) Concise Inorganic Chemistry, J.D. Lee, Blackwell Science Ltd.           (b) Inorganic Chemistry, Puri, Sharma, Kalia and Kaushal.           (c) Pradeep's Inorganic Chemistry, K.K. Bhasin, Pradeep Publication.           (d) Chemistry of degree students, R. L. Madan           Reference Books:           (a) Inorganic Chemistry, J.E.Huheey, Ellen A. Keiter, Richard L. Keiter, Addison Wesley Longman           (Singapore) Pvt. Ltd.           (b) Inorganic Chemistry, D.E.Shriver, P W. Atkins and C.H.L. Langford, Oxford.           (c) Casic Inorganic Chemistry, F.A. Cotton, G. Wilkinson and P.L. Gaus, Wiley.           (d) Concepts of Models of Inorganic Chemistry, B.Douglas,           D.McDaniel and J Alexander, John Wiley.           (e) Inorganic Chemistry, W.W. Porterfield, Addison - Wesley.           (f) Inorganic Chemistry, G.L. Meissler and D.A. Tarr, Prentice-Hall.           Suggested online links:           http://nptel.ac.in/courses/104/106/104106096/           https://nptel.ac.in/courses/104/103/104103071/#           Electronic media and other digital components in the curriculum:		Electrode potential electrochemical series and its applications EME diagrams and				
Suggested Text Book Readings:         Text Books (Theory Courses):         (a) Concise Inorganic Chemistry, J.D. Lee, Blackwell Science Ltd.         (b) Inorganic Chemistry, Puri, Sharma, Kalia and Kaushal.         (c) Pradeep's Inorganic Chemistry, K.K. Bhasin, Pradeep Publication.         (d) Chemistry for degree students, R. L. Madan         Reference Books:         (a) Inorganic Chemistry, J.E.Huheey, Ellen A. Keiter, Richard L. Keiter, Addison Wesley Longman         (Singapore) Pvt. Ltd.         (b) Inorganic Chemistry, F.A. Cotton, G. Wilkinson and P.L. Gaus, Wiley.         (d) Concepts of Models of Inorganic Chemistry, B.Douglas,         D.McDaniel and J Alexander, John Wiley.         (e) Inorganic Chemistry, G.L. Meissler and D.A. Tarr, Prentice-Hall.         Suggested online links:         http://heecontent.upsdc.gov.in/Home.aspx         https://nptel.ac.in/courses/104/106/104106096/         https://nptel.ac.in/courses/104/103/01/#         Electronic media and other digital components in the curriculum:		their utility. Principle involved in the extraction of the elements				
<ul> <li>Text Books (Theory Courses):</li> <li>(a) Concise Inorganic Chemistry, J.D. Lee, Blackwell Science Ltd.</li> <li>(b) Inorganic Chemistry, Puri, Sharma, Kalia and Kaushal.</li> <li>(c) Pradeep's Inorganic Chemistry, K.K. Bhasin, Pradeep Publication.</li> <li>(d) Chemistry for degree students, R. L. Madan</li> <li>Reference Books:</li> <li>(a) Inorganic Chemistry, J.E.Huheey, Ellen A. Keiter, Richard L. Keiter, Addison Wesley Longman</li> <li>(Singapore) Pvt. Ltd.</li> <li>(b) Inorganic Chemistry, D.E.Shriver, P W. Atkins and C.H.L. Langford, Oxford.</li> <li>(c) Basic Inorganic Chemistry, F.A. Cotton, G. Wilkinson and P.L. Gaus, Wiley.</li> <li>(d) Concepts of Models of Inorganic Chemistry, B.Douglas,</li> <li>D.McDaniel and J Alexander, John Wiley.</li> <li>(e) Inorganic Chemistry, A.G. Sharpe, ELBS</li> <li>(g) Inorganic Chemistry, G.L. Meissler and D.A. Tarr, Prentice-Hall.</li> <li>Suggested online links:</li> <li>http://heecontent.upsdc.gov.in/Home.aspx</li> <li>https://nptel.ac.in/courses/104/106/104106096/</li> <li>https://nptel.ac.in/courses/104/103/104103071/#</li> <li>Electronic media and other digital components in the curriculum:</li> </ul>	Suggested Text	Book Readings:				
<ul> <li>(a) Concise Inorganic Chemistry, J.D. Lee, Blackwell Science Ltd.</li> <li>(b) Inorganic Chemistry, Puri, Sharma, Kalia and Kaushal.</li> <li>(c) Pradeep's Inorganic Chemistry, K.K. Bhasin, Pradeep Publication.</li> <li>(d) Chemistry for degree students, R. L. Madan Reference Books: <ul> <li>(a) Inorganic Chemistry, J.E.Huheey, Ellen A. Keiter, Richard L. Keiter, Addison Wesley Longman</li> <li>(Singapore) Pvt. Ltd.</li> <li>(b) Inorganic Chemistry, D.E.Shriver, P W. Atkins and C.H.L. Langford, Oxford.</li> <li>(c) Basic Inorganic Chemistry, F.A. Cotton, G. Wilkinson and P.L. Gaus, Wiley.</li> <li>(d) Concepts of Models of Inorganic Chemistry, B.Douglas,</li> <li>D.McDaniel and J Alexander, John Wiley.</li> <li>(e) Inorganic Chemistry, A.G. Sharpe, ELBS</li> <li>(g) Inorganic Chemistry, G.L. Meissler and D.A. Tarr, Prentice-Hall.</li> </ul> </li> <li>Suggested online links: <ul> <li>http://heecontent.upsdc.gov.in/Home.aspx</li> <li>https://nptel.ac.in/courses/104/106/104106096/</li> <li>https://www2.chemistry.msu.edu/faculty/reusch/VirtTxtJml/intro1.htm</li> <li>https://nptel.ac.in/courses/104/103/104103071/#</li> </ul> </li> </ul>	Text Books (Th	eory Courses):				
<ul> <li>(b) Inorganic Chemistry, Puri, Sharma, Kalia and Kaushal.</li> <li>(c) Pradeep's Inorganic Chemistry, K.K. Bhasin, Pradeep Publication.</li> <li>(d) Chemistry for degree students, R. L. Madan Reference Books: <ul> <li>(a) Inorganic Chemistry, J.E.Huheey, Ellen A. Keiter, Richard L. Keiter, Addison Wesley Longman</li> <li>(Singapore) Pvt. Ltd.</li> <li>(b) Inorganic Chemistry, D.E.Shriver, P W. Atkins and C.H.L. Langford, Oxford.</li> <li>(c) Basic Inorganic Chemistry, F.A. Cotton, G. Wilkinson and P.L. Gaus, Wiley.</li> <li>(d) Concepts of Models of Inorganic Chemistry, B.Douglas, D.McDaniel and J Alexander, John Wiley.</li> <li>(e) Inorganic Chemistry, M.W. Porterfield, Addison - Wesley.</li> <li>(f) Inorganic Chemistry, G.L. Meissler and D.A. Tarr, Prentice-Hall.</li> </ul> </li> <li>Suggested online links: http://heecontent.upsdc.gov.in/Home.aspx https://nptel.ac.in/courses/104/106/104106096/</li> <li>https://nptel.ac.in/courses/104/103/104103071/#</li> <li>Electronic media and other digital components in the curriculum:</li> </ul>	(a) Concise Inor	ganic Chemistry, J.D. Lee, Blackwell Science Ltd.				
<ul> <li>(c) Pradeep's Inorganic Chemistry, K.K. Bhasin, Pradeep Publication.</li> <li>(d) Chemistry for degree students, R. L. Madan Reference Books: <ul> <li>(a) Inorganic Chemistry, J.E.Huheey, Ellen A. Keiter, Richard L. Keiter, Addison Wesley Longman</li> <li>(Singapore) Pvt. Ltd.</li> <li>(b) Inorganic Chemistry, D.E.Shriver, P W. Atkins and C.H.L. Langford, Oxford.</li> <li>(c) Basic Inorganic Chemistry, F.A. Cotton, G. Wilkinson and P.L. Gaus, Wiley.</li> <li>(d) Concepts of Models of Inorganic Chemistry, B.Douglas, D.McDaniel and J Alexander, John Wiley.</li> <li>(e) Inorganic Chemistry, W.W. Porterfield, Addison - Wesley.</li> <li>(f) Inorganic Chemistry, G.L. Meissler and D.A. Tarr, Prentice-Hall.</li> </ul> </li> <li>Suggested online links: http://heecontent.upsdc.gov.in/Home.aspx https://nptel.ac.in/courses/104/106/104106096/</li> <li>https://nptel.ac.in/courses/104/103/104103071/#</li> <li>Electronic media and other digital Components in the curriculum:</li> </ul>	(b) Inorganic Ch	nemistry, Puri, Sharma, Kalia and Kaushal.				
<ul> <li>(d) Chemistry for degree students, K. L. Madan</li> <li>Reference Books: <ul> <li>(a) Inorganic Chemistry, J.E.Huheey, Ellen A. Keiter, Richard L. Keiter, Addison Wesley Longman</li> <li>(Singapore) Pvt. Ltd.</li> </ul> </li> <li>(b) Inorganic Chemistry, D.E.Shriver, P W. Atkins and C.H.L. Langford, Oxford.</li> <li>(c) Basic Inorganic Chemistry, F.A. Cotton, G. Wilkinson and P.L. Gaus, Wiley.</li> <li>(d) Concepts of Models of Inorganic Chemistry, B.Douglas, D.McDaniel and J Alexander, John Wiley.</li> <li>(e) Inorganic Chemistry, A.G. Sharpe, ELBS</li> <li>(g) Inorganic Chemistry, G.L. Meissler and D.A. Tarr, Prentice-Hall.</li> </ul> <li>Suggested online links: <ul> <li>http://heecontent.upsdc.gov.in/Home.aspx</li> <li>https://nptel.ac.in/courses/104/106/104106096/</li> <li>https://nptel.ac.in/courses/104/103/104103071/#</li> </ul> </li> <li>Electronic media and other digital components in the curriculum:</li>	(c) Pradeep's Ind	organic Chemistry, K.K. Bhasin, Pradeep Publication.				
<ul> <li>(a) Inorganic Chemistry, J.E.Huheey, Ellen A. Keiter, Richard L. Keiter, Addison Wesley Longman (Singapore) Pvt. Ltd.</li> <li>(b) Inorganic Chemistry, D.E.Shriver, P W. Atkins and C.H.L. Langford, Oxford.</li> <li>(c) Basic Inorganic Chemistry, F.A. Cotton, G. Wilkinson and P.L. Gaus, Wiley.</li> <li>(d) Concepts of Models of Inorganic Chemistry, B.Douglas, D.McDaniel and J Alexander, John Wiley.</li> <li>(e) Inorganic Chemistry, W.W. Porterfield, Addison - Wesley.</li> <li>(f) Inorganic Chemistry, G.L. Meissler and D.A. Tarr, Prentice-Hall.</li> <li>Suggested online links: http://heecontent.upsdc.gov.in/Home.aspx https://nptel.ac.in/courses/104/106/104106096/ https://www2.chemistry.msu.edu/faculty/reusch/VirtTxtJml/intro1.htm https://nptel.ac.in/courses/104/103/104103071/#</li> </ul>	(d) Chemistry IC	or degree students, K. L. Madan				
<ul> <li>(Singapore) Pvt. Ltd.</li> <li>(b) Inorganic Chemistry, D.E.Shriver, P W. Atkins and C.H.L. Langford, Oxford.</li> <li>(c) Basic Inorganic Chemistry, F.A. Cotton, G. Wilkinson and P.L. Gaus, Wiley.</li> <li>(d) Concepts of Models of Inorganic Chemistry, B.Douglas,</li> <li>D.McDaniel and J Alexander, John Wiley.</li> <li>(e) Inorganic Chemistry, W.W. Porterfield, Addison - Wesley.</li> <li>(f) Inorganic Chemistry, G.L. Meissler and D.A. Tarr, Prentice-Hall.</li> <li>Suggested online links:</li> <li>http://heecontent.upsdc.gov.in/Home.aspx</li> <li>https://nptel.ac.in/courses/104/106/104106096/</li> <li>https://nptel.ac.in/courses/104/103/104103071/#</li> </ul>	(a) Inorganic Ch	emistry, J.E.Huheey, Ellen A. Keiter, Richard L. Keiter, Addison Wesley Longman				
<ul> <li>(b) Inorganic Chemistry, D.E.Shriver, P W. Atkins and C.H.L. Langford, Oxford.</li> <li>(c) Basic Inorganic Chemistry, F.A. Cotton, G. Wilkinson and P.L. Gaus, Wiley.</li> <li>(d) Concepts of Models of Inorganic Chemistry, B.Douglas, D.McDaniel and J Alexander, John Wiley.</li> <li>(e) Inorganic Chemistry, W.W. Porterfield, Addison - Wesley.</li> <li>(f) Inorganic Chemistry, A.G. Sharpe, ELBS</li> <li>(g) Inorganic Chemistry, G.L. Meissler and D.A. Tarr, Prentice-Hall.</li> <li>Suggested online links: http://heecontent.upsdc.gov.in/Home.aspx</li> <li>https://nptel.ac.in/courses/104/106/104106096/</li> <li>https://nptel.ac.in/courses/104/103/104103071/#</li> <li>Electronic media and other digital components in the curriculum:</li> </ul>	(Singapore) Pvt	. Ltd.				
<ul> <li>(c) Basic Inorganic Chemistry, F.A. Cotton, G. Wilkinson and P.L. Gaus, Wiley.</li> <li>(d) Concepts of Models of Inorganic Chemistry, B.Douglas,</li> <li>D.McDaniel and J Alexander, John Wiley.</li> <li>(e) Inorganic Chemistry, W.W. Porterfield, Addison - Wesley.</li> <li>(f) Inorganic Chemistry, A.G. Sharpe, ELBS</li> <li>(g) Inorganic Chemistry, G.L. Meissler and D.A. Tarr, Prentice-Hall.</li> <li>Suggested online links:</li> <li>http://heecontent.upsdc.gov.in/Home.aspx</li> <li>https://nptel.ac.in/courses/104/106/104106096/</li> <li>https://nptel.ac.in/courses/104/103/104103071/#</li> <li>Electronic media and other digital components in the curriculum:</li> </ul>	(b) Inorganic Cl	nemistry, D.E.Shriver, P W. Atkins and C.H.L. Langford, Oxford.				
<ul> <li>(d) Concepts of Models of Inorganic Chemistry, B.Douglas,</li> <li>D.McDaniel and J Alexander, John Wiley.</li> <li>(e) Inorganic Chemistry, W.W. Porterfield, Addison - Wesley.</li> <li>(f) Inorganic Chemistry, A.G. Sharpe, ELBS</li> <li>(g) Inorganic Chemistry, G.L. Meissler and D.A. Tarr, Prentice-Hall.</li> <li>Suggested online links:</li> <li>http://heecontent.upsdc.gov.in/Home.aspx</li> <li>https://nptel.ac.in/courses/104/106/104106096/</li> <li>https://nptel.ac.in/courses/104/103/104103071/#</li> <li>Electronic media and other digital components in the curriculum:</li> </ul>	(c) Basic Inorga	nic Chemistry, F.A. Cotton, G. Wilkinson and P.L. Gaus, Wiley.				
<ul> <li>D.McDaniel and J Alexander, John Wiley.</li> <li>(e) Inorganic Chemistry, W.W. Porterfield, Addison - Wesley.</li> <li>(f) Inorganic Chemistry, A.G. Sharpe, ELBS</li> <li>(g) Inorganic Chemistry, G.L. Meissler and D.A. Tarr, Prentice-Hall.</li> <li>Suggested online links:</li> <li>http://heecontent.upsdc.gov.in/Home.aspx</li> <li>https://nptel.ac.in/courses/104/106/104106096/</li> <li>https://nptel.ac.in/courses/104/103/104103071/#</li> <li>Electronic media and other digital components in the curriculum:</li> </ul>	(d) Concepts of Models of Inorganic Chemistry, B.Douglas,					
<ul> <li>(c) Inorganic Chemistry, W.W. Porterneud, Addison - Wesley.</li> <li>(f) Inorganic Chemistry, A.G. Sharpe, ELBS</li> <li>(g) Inorganic Chemistry, G.L. Meissler and D.A. Tarr, Prentice-Hall.</li> <li>Suggested online links:</li> <li>http://heecontent.upsdc.gov.in/Home.aspx</li> <li>https://nptel.ac.in/courses/104/106/104106096/</li> <li>https://nptel.ac.in/courses/104/103/104103071/#</li> <li>Electronic media and other digital components in the curriculum:</li> </ul>	D.McDaniel and	1 J Alexander, John Wiley.				
<ul> <li>(g) Inorganic Chemistry, G.L. Meissler and D.A. Tarr, Prentice-Hall.</li> <li>Suggested online links: http://heecontent.upsdc.gov.in/Home.aspx https://nptel.ac.in/courses/104/106/104106096/ https://www2.chemistry.msu.edu/faculty/reusch/VirtTxtJml/intro1.htm https://nptel.ac.in/courses/104/103/104103071/#</li> <li>Electronic media and other digital components in the curriculum:</li> </ul>	(f) Inorganic Ch	emistry A G Sharpe FLBS				
Suggested online links: http://heecontent.upsdc.gov.in/Home.aspx https://nptel.ac.in/courses/104/106/104106096/ https://www2.chemistry.msu.edu/faculty/reusch/VirtTxtJml/intro1.htm https://nptel.ac.in/courses/104/103/104103071/# Electronic media and other digital components in the curriculum: Chapter and an an an analysis of the second se	(g) Inorganic Ch	nemistry, G.L. Meissler and D.A. Tarr, Prentice-Hall.				
http://heecontent.upsdc.gov.in/Home.aspx https://nptel.ac.in/courses/104/106/104106096/ https://www2.chemistry.msu.edu/faculty/reusch/VirtTxtJml/intro1.htm https://nptel.ac.in/courses/104/103/104103071/# Electronic media and other digital components in the curriculum:	Suggested onli	ne links:				
https://nptel.ac.in/courses/104/106/104106096/ https://www2.chemistry.msu.edu/faculty/reusch/VirtTxtJml/intro1.htm https://nptel.ac.in/courses/104/103/104103071/# Electronic media and other digital components in the curriculum:	http://heeconter	nt.upsdc.gov.in/Home.aspx				
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https://nptel.ac.in/courses/104/103/104103071/# Electronic media and other digital components in the curriculum:	https://www2.chemistry.msu.edu/faculty/reusch/VirtTxtJml/intro1.htm					
Electronic media and other digital components in the curriculum:	https://nptel.ac.in/courses/104/103/104103071/#					
Change one on more than a CLM/ Other describes $1, 1', 1'$	Electronic media and other digital components in the curriculum:					
Choose any one or more than: e-SLM/ Other electronic and digital contents	Choose any	one or more than: e-SLM/ Other electronic and digital contents				
Name of electronic media: e-SLMYear of incorporation: 2020	Name of elec	tronic media: e-SLM Year of incorporation: 2020				

Course prerequisites: Chemistry in 10+2 Level					
Programme: B.Sc. Year: 1 Semester: 1					
		Subject:	Chemistry		
Course Code: U	GCHE 101P(N)	Course Title: ${f UG}$	CHE-LAB-WORK-	I	
Course Object	ctives:				
To understand	basic knowledge	and skills about	t laboratory methods ar	nd tests related to estimation	
of metals ions	and estimation of	f acids and alkali	l.		
Course Outco	omes:				
<b>CO-1</b> Upon cor	npletion of this co	urse the students	will have the knowledg	e and skills to: understand the	
laboratory met	hods and tests rela	ted to estimation	of metals ions and estin	nation of acids and alkali	
Contents in con	nmercial products.	•	Turne of Courses Co		
Credits: 2	100	M. D	Type of Course: Co	re	
Max. Marks:		Min. Passing N	larks: 36		
BIOCK I	Concerned Dut				
Unit I	General – Pri	nciple and worl	king of Chemical bala	ance. Calibration of fractional	
IL.: 4 II	Weights and the	ermometer.			
Unit II	Inorganic Cne	emistry		ining fine we light next of the	
	Qualitative and	alysis of an ino	rganic mixture conta	ling five radicals out of the	
	10110Wing prefe	$M_{2}^{++}$ C $^{++}$ S	$^{++}$ Do <sup>++</sup> $7n^{++}$ M $n^{++}$	$\begin{array}{c} \text{Ing insoluble substances}:\\ \text{Ni}^{++}  \text{Ca}^{++}  \text{A}^{+++}  \text{Ea}^{+++}  \text{Ca}^{+++} \end{array}$	
	$\mathbf{NH4}$ , $\mathbf{Na}$ , $\mathbf{K}$	, Mg , Ca , Sr + U <sub>2</sub> ++ Cd++ A	, Ba, Zn, Mn, J	$(N_1, C_0, A_1, F_2, F_2, C_1, F_2)$	
	$Cu^{-}, B1^{-}, Hg$	r, Hg <sup>-1</sup> , Cd <sup>-1</sup> $F$	$AS^{-}, SD^{-}, S\Pi^{-}, PD^{-}$	$P_{1}^{-}$ , $PO_{1}^{-}$ , $Ag_{1}^{-}$ , $CO_{3}^{-}$ , $NO_{1}^{-}$ , $S_{1}^{-}$ , $NO_{2}^{-}$ , $S_{2}^{-}$ , $NO_{2}^{-}$ , $NO_{2}^{-}$ , $S_{2}^{-}$ , $NO_{2}^{-}$ , $NO_{2}^{-}$ , $S_{2}^{-}$ , $NO_{2}^{-}$ , $S_{2}^{-}$ , $NO_{2}^{-}$ ,	
Suggested To	$503$ , $504$ , $\Gamma$	$, CI, DI, NO_3$	$, CH_3COO$ , Dolale, C	Dxalate, and Phosphate.	
1 Mondhom	Vogel's Quentity	igs. stive Chemical A	nalucia Deerson 2000		
1. Mendiani, J	. Vogel s Qualitita	anve Chemical A	ith Ed. Eraaman (2007)	Chapters 2 5	
2. Harris, D. C. Quantitative Chemical Analysis. 6th Ed., Freeman (2007) Chapters 3-5.					
3. Harris, D.C. Exploring Chemical Analysis, 9th Ed. New York, W.H. Freeman, 2016.					
4. Skoog, D.A. Holler F.J. and Nieman, T.A. Principles of Instrumental Analysis, Cengage Learning					
<b>Note</b> : For the r	promotion of Hind	i language cours	e books published in Hi	ndi may be prescribed by the	
University					
Suggestive digital platforms web links					
https://www.labster.com/chemistry-virtual-labs/ https://www.vlab.co.in/broad-area-chemical-sciences					
http://chemcoll	http://chemcollective.org/vlabs				
This course can be opted as an elective by the students of following subjects:					
Suggested eq	Suggested equivalent online courses (MOOCs) for credit transfer: NA				
Electronic n	nedia and other	digital compo	onents in the curricu	ılum:	
Choose any	one or more th	an: e-SLM/ Ot	her electronic and di	gital contents	
Name of elec	Name of electronic media: e-SLMYear of incorporation: 2021				

Course prerequisites: 10+2 Chemistry as subject					
Programme:B.	.Sc.	Year:1 Semester:2			
Subject: Chemistry					
Course Code:	UGCHE-102N	Course Title: ORGANIC CHEMISTRY I (BASIC ORGANIC CHEMISTRY)			
Course Objec	tives:	, ,			
• To uno basics	<ul> <li>To understand different organic compounds with respect to the functional groups and basics of chemical reactions.</li> </ul>				
• To und derive	lerstand differen mechanism of v	t principles of organic chemistry and various types of organic reactions.	predict outcomes and		
• To un prepar	derstand the contraction, reactivity	oncept of Aromaticity of benzeno and structure of aromatic compound	ids & nonbenzenoids. The s.		
To lea     compo	rn the prepration ounds.	s, reactivity & stereochemistry of SN	<sup>1</sup> &SN <sup>2</sup> reactions of Halogen		
Course Outco	omes:				
CO-1 Unders	tand different or	rganic compounds with respect to th	ne functional group and thus		
capable to nar	ne the organic c	ompounds as per IUPAC nomenclatu	ire.		
CO-2 Underst	and the basics of	f chemical reactions i.e. Substrate and	l Reagent, types of Reagents,		
Electrophilic	and Nucleophili	c Homolytic and heterolytic fission.	Electron mobility, Inductive		
effect etc.	effect etc.				
CO-3 Recogn	ize and draw c	onstitutional isomers, stereoisomers	, including enantiomers and		
diasteromers,	diasteromers, racemic mixture and meso compounds.				
CO-4. Unders	CO-4. Understand fundamental principles of organic chemistry and predict outcomes and derive				
mechanism of various types of organic reactions.					
CO-5 Underst	CO-5 Understand various types of reactive intermediates and factors affecting their stability				
CO-6 Underst	CO-6 Understand the nomenclature, synthesis, isomerism and physical properties of alkanes and				
cycloalkanes.					
CO-/ Understand the concept of Aromaticity of benzenoids & nonbenzenoids. The preparation,					
reactivity and structure of aromatic compounds.					
CO-8 Learn the preprations, reactivity & stereochemistry of $SN^{1}$ & $SN^{2}$ reactions of Halogen					
compounds.					
Credits:2	00	I ype of Course: Co	ore		
Max. Marks: 1	.00	Min. Passing Marks: 36			
BIOCK I	<u><u>G</u>(1) (1) (1) (1) (1) (1) (1) (1) (1) (1) </u>				
	Structure and	Bonding	an of mothana others others		
I India I	Atomic orbital	s, hydridization, orbital representati	on of methane, ethane, ethyne		
Unit I	Unit I and benzene. Polarity of bonds: Inductive, resonance and steric effe				
	nyperconjugation, and their influence on acidity and basicity of organic compound				
I Init II	Mochanism -	Concept of Carbocation	1, Cardanion and Free radicals.		
Unit II	Curved errors	Organic Reactions and Reaction I	to with allows half handed and		
	double based	arrows homolytic and hotorolytic he	us with anows, nan-neaded and		
	uouble-neaded arrows, nemolytic and neterolytic bond breaking. Types of reagent				
	-electrophiles	and nucleophnes, Types of organic re	actions, Energy considerations.		

	Reactive intermediates – Carbocations, carbanions, free radicals, carbenes, arynes
	and nitrenes (with examples). Assigning format charges on intermediates and other
	ionic species. Methods of determination of reaction mechanism (product analysis,
	intermediates, isotope effects, kinetic and stereochemical studies).
Unit III	Alkanes and Cycloalkanes
	IUPAC nomenclature of branched and unbranched alkanes, the alkyl group.
	classification of carbon atom in alkanes. Isomerism in alkanes, sources methods of
	formation (with special reference to Wurtz reaction Kolbe reaction Corey-House
	reaction and decarboxylation of carboxylic acids) physical properties and chemical
	reactions of alkanes. Mechanism of free radical halogenation of alkanes:
	orientation reactivity and selectivity
Dlook 2	
BIOCK 2	
Unit IV	Stereochemistry of Organic Compounds
	Concept of isomerism. Types of isomerism. Optical isomerism – elements of
	symmetry, molecular chirality, enantionmers, stereogenic center, optical activity,
	properties of enantiomers, chiral and achiral molecules with two stereogenic
	centers, diastereomers, threo and erythro diastereomers, meso compounds,
	resolution of enantiomers, inversion, retention and recemization. Relative and
	absolute configuration, sequence rules, D & L and R & S systems of nomenclaute.
	Geometric isomerism – determination of configuration of geometric isomers. E &
	Z system of nomenclature, geometric isomerism in oximes and alicyclic
	compounds. Conformational isomerism conformational analysis of ethane and n-
	butane: conformations of cyclohexane, axial and equatorial bonds, conformation of
	mono substituted cyclohexane derivatives. Newman projection and Sawhorse
	formulae Fischer and flying wedge formulae Difference between configuration
	and conformation
Unit V	Alkanes Cycloalkanes Dianes and Alkynes
Omt v	Nomanalatura of alkanas, mathads of formation, machanisms of dahudration of
	alcohols and dehydrohologenetion of alkul holidog, regiogalactivity in alcohol
	alcohols and denydronalogenation of arkyr handes, regioserectivity in alcohol
	denydration. The Saytzerr rule, Hormann elimination, physical properties and
	relative stabilities of alkenes. Chemical reactions of alkenes – mechanism involved
	in hydrogenation, electrophilic and free radical additions, Markownikoff's rule,
	hydroborationoxidation, oxymercuration-reduction. Epoxidation, ozonolysis,
	hydration, hydroxylation and oxidation with KMnO4. Polymerization of alkenes.
	Substitution at the allylic and vinylic positions of alkenes. Industrial applications of
	ethylene and propene. Methods of formation, conformation and chemical reactions
	of cycloalkenes. Nomenclature and classification of dienes: isolated, conjugated
	and cumulated dienes. Structure of allenes and butadiene, methods of formation,
	polymerization, Chemical reaction $-1,2$ and $1,4$ additions, Diets-Alder reaction.
	Nomenclature, structure and bonding in alkynes. Methods of formation. Chemical
	reactions of alkynes, acidity of alkynes. Mechanism of electrophilic and
	nucleophilic addition reactions, hydroboration-oxidation, metal-ammonia
	reductions, oxidation and polymerization
Unit VI	Arenes and Aromaticity
	Nomenclature of benzene derivatives. The arvl group Aromatic nucleus and side
	chain Structure of benzene, molecular formula and kakula structure. Stability and
1	I cham, Subclute of benzene, molecular formula and kekule subclute. Stability and

	carbon-carbon bond lengths of benzy	ene, resonance structure. MO picture.			
	Aromaticity: the Huckel rule, aromatic ions. Aromatic electrophilic substitution –				
	general pattern of the mechanism role	$\alpha$ of $\sigma$ and $\pi$ complexes Mechanism of			
	nitration halogenation sulphonation	mercuration and Friedel-Crafts reaction			
	Energy profile diagrams Activating and	deactivating substituents orientation and			
	ortho/para ratio Side chain reactions	of benzene derivatives. Birch reduction			
	Methods of formation and chemical rea	ctions of albylbanzanas, alkynylbanzanas			
	and biphonyl	etions of amytoenzenes, arkynytoenzenes			
IInit VII	Allwi and Awi Halidag				
	Nomeneletime and classes of allul h	lides methods of formation shamical			
	reactions Machanisms of nucleanhilis a	andes, methods of formation, chemical			
	and SN1 reactions with apargy prof	ile diagrams Delyhelegen compounds:			
	and SINT reactions with energy prof	le of formation of anyl halidae nyelean and			
	chloroform, carbon tetrachloride. Method	is of formation of aryl handes, nuclear and			
	side chain reactions. The addition-en	imination and the elimination-addition			
	mechanisms of nucleopning aromatic sur	Stitution reactions. Relative reactivities of			
<u> </u>	aikyi nandes vs ailyi, vinyi and aryi nand	les. Synthesis and uses of DD1 and BHC.			
Suggested Te	ext Book Readings:				
Text Books (	Theory Courses):				
(a) Organic (	chemistry, Vol. I, I.L. Finar, Pearson Edi	ucation.			
(b) Organic (	Chemistry, M.K. Jain, Shoban Lal& Co.				
(c) Pradeep's Organic Chemistry, S.N. Dhawan, Pradeep Publication.					
Reference Books:					
(a) Organic Chemistry, Morrison and Boyd, Prentice Hall.					
(b) Organic Unemistry, L.G. Wade Jr. Prentice Hall.					
(c) Fundame	(c) Fundamentals of Organic Chemistry Solomons, John Wiley.				
(d) Organic Chemistry, Vol. I, II, III S.M. Mukherji, S.P. Singh and R.P. Kapoor, Wiley					
Eastern Ltd. (New Age International)					
(e) Organic Chemistry, F.A. Carey, McGraw-Hill Inc.					
(f) Introducti	on to Organic Chemistry, Streitwiesser,	Hathcock and Kosover, Macmillan.			
Suggested only	ine links:				
http://heeconte	nt.upsdc.gov.in/Home.aspx				
https://nptel.ac.in/courses/104/105/104105124/					
nttps://nptel.ac.in/courses/103/106/105106204/					
https://https/					
https://nptel.ac.in/courses/104/102/104102016/					
https://nptel.ac.in/courses/104/106/104106106/					
https://nptel.ac.in/courses/104/105/104105120/					
This course of	an be opted as an elective by the student	ts of following subjects:			
Suggested eq	uivalent online courses (MOOCs) for cr	edit transfer:			
1. Mechani	sms in Organic Chemistry, Prof. Nandita Madhavan, NPTI	EL, https://onlinecourses.nptel.ac.in/noc22_cy42/preview			
Electronic media and other digital components in the curriculum:					
Choose any	one or more than: e-SLM/ Other electr	onic and digital contents			
Name of elec	etronic media: e-SLM	Year of incorporation: 2021			

Course prerequisites: 10+2 Chemistry as subject				
Programme: B	.Sc.	Year:1	Semester:2	
Subject: Chemistry				
Course Code: U	GCHE 102P (N)		Course Title: UGCHE-LAB-WORK-II	
Course Object	tives:			
This course wi	ll provide basic	qualitative and qu	antitative experimental knowledge of biomolecules	
such as carbo	hydrates, protei	ns, amino acids,	nucleic acids drug molecules. Upon successful	
completion of	this course stude	ents may get job	opportunities in food, beverage and pharmaceutical	
industries.				
Course Outcon	nes:	ana ana da		
CO1- Preparati	on of organic co	minotion of mole	ing points	
CO2- Crystalli	zation and deter	mination of men	True of Courses Core	
Credits:2	00	Min Dessine M	Type of Course: Core	
Max. Marks: 1	.00	Min. Passing M	arks: 30	
BIOCK I		· · · · · · · · · · · · · · · · · · ·		
	Organic Chen	nistry-1	1	
TT '/ T	(a) Preparation	of organic com	bounds:	
Unit I	1. Acetanilide	to mili da		
	2. p-bromoace	lannde		
TT '/ TT	3. picrates	· · · · · · · · · · · · · · · · · · ·		
Unit II	(h) Crustallize	nistry-11	notion of molting noint	
	(b) Crystallization and determination of melting point.			
	1. Phinalic acid from hot water (using fluted filter paper and stemless funnel)			
	2. Acetaminde from bolling water 3. Naphthalona from athanol			
	5. Naphthalene from ethanol 4. Panzoia agid from water			
4. Delizoic aciu itolii water				
Suggested Text Book Readings:				
Pearson (2012)				
2. Mann, F.G. & Saunders, B.C. <i>Practical Organic Chemistry</i> , Pearson Education.				
3. Vogel's Qualitative Inorganic Analysis, Revised by G. Svehla.				
4. Vogel, A.I. A Textbook of Quantitative Analysis, ELBS. 1986				
5. Furniss, B.S.; Hannaford, A.J.; Rogers, V.; Smith, P.W.G.; Tatchell, A.R. Vogel's Textbook of				
Practical Organic Chemistry, ELBS.				
6. Ahluwalia, V.K. & Aggarwal, R. Comprehensive Practical Organic Chemistry, Universities Pres				
7. Cooper, T.G. Tool of Biochemistry. Wiley-Blackwell (1977).				
8. Wilson, K. & Walker, J. Practical Biochemistry. Cambridge University Press (2009).				
<b>9.</b> variey, H., Gowenlock, A.H & Bell, M.: Practical Clinical Biochemistry, Heinemann, Suggestive digital platforms web links				
1 https://www.labster.com/chemistry-virtual-labs/				
2. https://www.vlab.co.in/broad-area-chemical-sciences				
3. http://chemcollective.org/vlabs				
This course can be opted as an elective by the students of following subjects:				
Suggested eq	uivalent online	courses (MOO	Cs) for credit transfer:	
Electronic media and other digital components in the curriculum.				
Choose any one or more than:e-SLM/ Other electronic and digital contents				
Name of elec	tronic media: e	-SLM	Year of incorporation: 2021	
	a one meata. e		1 cm 01 moorporation. 2021	

Course prerequisites: 10+2 Chemistry as subject					
Programme: E	B.Sc.	Year:1		Semester:2	
		Subject: C	hemistry		
Course Code: SI	BSCHE -02N	Course Title: ADVA	NCED ANALYTICA	L TECHNIQUES	
Course Object	Course Objectives:				
To gain basics	about analytical	chemistry plays an	enormous role in ou	ir society, such as in drug	
manufacturing	g, process control	in industry, environ	nmental monitoring,	medical diagnostics, food	
production, an	d forensic survey	vs. It is also of great	importance in diffe	erent research areas.	
Course Outco	omes:				
CO1- Students	will be able to ex	kplore Analytical che	emistry is a science t	that is directed towards creating	
new knowledg	e so that chemica	l analysis can be imp	proved to respond to	increasing or new demands.	
CO2- Students	will be able to ex	plore new areas of r	esearch in both chen	nistry and alled fields of science	
	y. will be able to fu	action as a mombor	of an interdisciplinar	warahlam salwing taam	
CO3- Students	will be skilled in	problem colving or	itical thinking and a	y problem solving team.	
scientific probl		problem solving, ci		harytical reasoning as applied to	
CO5- Students	will gain an unde	rstanding of how to	determine the struct	ure of organic molecules using	
IR and NMR sn	ectrosconic techr				
CO6- To develo	op basic skills requ	lired for purification	. solvent extraction.	TIC and column chromatography	
Credits: 4			Type of Course: Co	re	
Category of C	ourse		Value-added / em	plovability/	
Max. Marks: 1	100	Min. Passing Mar	ks: 36		
Block 1		88			
	Statistical Ana	alvsis			
	Definition of	terms mean and	median, precision	, standard deviation, relative	
<b>TT 1 T</b>	standard deviation, accuracy, absolute error, types of error in experimental data.				
Unit I determinate (systematic), indeterminate (or random) ar		) and gross, sources of errors			
	and effects up	on the analytical	results, methods	for reporting analytical data,	
	statistical evaluation of data, indeterminate errors, uses of statistics.			ses of statistics.	
Unit II	Volumetric analysis				
	General princi	iples of acid – t	base titration, prec	cipitation titration, oxidation-	
	reduction titration, iodimetry and iodometry, complexometric titrations. use of				
	EDTA for the determination of $Ca^{2+}$ and $Mg^{2+}$ and hardness of water. types of			d hardness of water, types of	
	EDTA titration	s, metal ion indica	tors.		
Unit III	Gravimetric analysis				
	Precipitation f	rom homogeneous	medium, purity of	precipitates, co-precipitation,	
post-precipitation, washing and ignition of precipitates, contamination and			tates, contamination and their		
	removal.				
Block 2					
Unit IV	Separation tec	chniques			
	Principle, tech	nique and analytica	al applications of th	e following:	
	(a) Solvent ext	raction			
	(b) Chromatog	raphy (Paper, Thin	Layer, Column an	d HPLC)	
	(c) Ion exchange	ge			
Unit V	Nano Chemist	try			

Nanomaterials – An Introduction, Size Effects, Defining Nanodimensional Materials, Potential Uses for Nanodimensional Materials, The General Methods Available for the Synthesis of Nanodimensional Materials, Precipitative Methods, Reactive Methods in High Boiling Point Solvents, Hydrothermal and Solvothermal Methods, Gas-Phase Synthesis of Semiconductor Nanoparticles, Synthesis in a Structured Medium, The Suitability of Such Methods for Scaling, Conclusions and Perspectives on the Future, Oxide Nanoparticles, Nanotubes and Nanowires. Study of different characterization tools (XRD, TEM, SEM, AFM, etc.) for Nanomaterials.

Suggested Text Book Readings:

1. Alberty, R A, Physical Chemistry, 4 th editionWiley Eastern Ltd ,2001.

- 2. Atkins, PW, the elements of physical chemistry, Oxford , 1991
- 3. Barrow, G. M, International student Edition .McGraw Hill, McGraw-Hill, 1973.
- 4. Cotton, F.A, Wilkinson, G and Gaus, P. L , Basic Inorganic Chemistry, 3rd Edition , Wiley 1995
- 5. Lee, J.D, Concise Inorganic Chemistry 4th Edition ELBS, 1977

6. Clayden, J., Greeves, N., Warren, S., *Organic Chemistry*, Second edition, Oxford University Press 2012.

7. Silverstein, R. M., Bassler, G. C., Morrill, T. C. Spectrometric Identification of Organic Compounds, John

Wiley and Sons, INC, Fifth edition.

8. Pavia, D. L. et al. Introduction to Spectroscopy, 5th Ed. Cengage Learning India Ed.

9. Willard, H.H. et al.: Instrumental Methods of Analysis, 7th Ed. Wardsworth Publishing Company, Belmont,

California, USA, 1988.

10. Christian, G.D. Analytical Chemistry, 6th Ed. John Wiley & Sons, New York, 2004.

11. Harris, D.C.: Exploring Chemical Analysis, 9th Ed. New York, W.H. Freeman, 2016.

12. Khopkar, S.M. Basic Concepts of Analytical Chemistry. New Age International Publisher, 2009.

This course can be opted as an elective by the students of following subjects:

Suggested equivalent online courses (MOOCs) for credit transfer:

 Analytical Chemistry, Prof. Debashis Ray, <u>https://onlinecourses.nptel.ac.in/noc22\_cy61/preview</u>
 Spectroscopic Techniques for Pharmaceutical and Biopharmaceutical Industries, Prof. Shashank Deep, https://onlinecourses.nptel.ac.in/noc22\_cy54/preview

Electronic media and other digital components in the curriculum:

Choose any one or more than: e-SLM/ Other electronic and digital contents

Name of electronic media: e-SLM	Year of incorporation: 202
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Course prerequisites: Chemistry in 10+2 Level			
Programme: I	3.Sc.	Year:2	Semester:3
		Subject: C	hemistry
Course Code:	UGCHE -103N	Course Title: PHYS	ICAL CHEMISTRY I (BASIC PHYSICAL CHEMISTRY)
<b>Course Objectives:</b> To get basic knowledge about computers and mathematical functions and understanding of gaseous state, critical phenomenon, liquid state, solid state, colloidal state and liquid crystals.			
<b>Course Outc</b>	omes:		
CO-1- Studen	ts would gain kno	owledge regarding t	he basic of computers and mathematical
concepts of lo	g, permutation ar	nd combination, diff	ferential and integration of some relevant
functions.			
CO-2- Studen	t would gain und	erstanding of gaseo	us state, critical phenomenon, liquid state, solid
state, colloida	l state and liquid	crystals.	
CO-3- It wou	d help students re	ecognize the import	ance of chemical kinetics and catalysis.
Credits:2			Type of Course: Core
Category of C	Course		Value-added / employability/
Max. Marks:	100	Min. Passing Mar	ks: 36
Block 1			
Unit I	Unit 1: Mathe (A) Mathematic Logarithmic re- differentiation differentiation functions; perr (B) Computer General introd and software, i computer lang different softw	ematical Concepts tical Concepts elations, curve ske of functions like $f_0$ and eciprocity nutations and comb s uction to computer nput-output device uages. Programming are in the Chemist	and Computers etching, linear graphs and calculation of slopes, $x_x$ , $e_x$ , $x_n$ , sin x, log x; maxima and minima, partial relations. Integration of some useful/relevant binations, Factorials, Probability and Regrations. rs, different components of a computer, hardware es; binary numbers and arithmetic; introduction to ing, operating systems. Use and application of ry.
Unit II	(A) Gaseous and J (A) Gaseous S Postulates of k equation of sta Critical Phen isotherms of v van der Waals Molecular Ve Qualitative di collision numb (B) Liquid Sta Intermolecular differences bet	<b>Eliquid States</b> inetic theory of gas te. <b>Iomena</b> : PV isot an der Waals equa constants, the law <b>locities :</b> Root me scussion of the I ber, mean free path <b>ite</b> forces, structure ween solids liquid	therms of real gases, continuity of states, the ation, relationship between critical constants and of corresponding states, reduced equation of state. an square, average and most probable velocities. Maxwell's distribution of molecular velocities, and collision diameter. Liquefaction of gases.

	Liquid crystals: Difference between liquid crystal, solid and liquid. Classification,
	structure of nematic, smectic and cholesteric phases and applications.
Unit III	Solid State
	Definition of space lattice and unit cell.
	Laws of crystallography:
	(i) Law of constancy of interfacial angles
	(ii) Law of rationality of indices
	(iii) Law of symmetry - Symmetry elements in crystals.
	X-ray diffraction: Derivation of Bragg's equation. Determination of crystal
	structure of NaCl, CsCl and KCl.
	A brief introduction to point defects in crystals, semiconductors, superconductors
	and nanomaterials (only qualitative idea).
Block 2	
Unit IV	Thermodynamics – I
	Definition of terms: system, surroundings, open system, isolated system, intensive
	and extensive properties, State and path functions and their differentials, reversible
	and irreversible processes, Concept of heat and work.
	First Law of Thermodynamics: Concepts of internal energy and enthalpy, heat
	capacities at constant volume and constant pressure and their relationship.
	Calculation of w, q, dU & dH for the expansion of ideal gases under isothermal and
	adiabatic conditions for a reversible process.
	Thermochemistry : standard state, standard enthalpy of formation- Hess's Law of
	constant heat summation and its applications, heat of reaction at constant pressure
	and at constant volume, Bond dissociation energy and its calculation from thermo-
	chemical data, Kirchhoff's equation.
Unit V	Electrochemistry – I and Solution
	Electrical transport - conduction in metals and in electrolyte solutions, specific
	conductance and equivalent conductance, measurement of equivalent conductance,
	variation of equivalent and specific conductance with dilution.
	Migration of ions and Kohlrausch law, Arrhenius theory of electrolyte dissociation
	and its limitations, weak and strong electrolytes, Ostwald's dilution law, its uses and
	limitations. Debye-Huckel-Onsager's equation for strong electrolytes (elementary
	treatment only). Activity and activity coefficient. Transport number, definition and
	Colution by Hillori method and moving boundary method.
	Solution Thermodynamic derivation of relation between melacular weight and elevation in
	heiling point and depression in fragging point. Experimental methods for
	determining various colligative properties. Abnormal malar mass degree of
	dissociation and association of solutes
Unit VI	Chamical Kinatics and Catalysis
	Chemical Kinetics and Catalysis
	temperature pressure solvent light and cotalust Concentration dependence of
	rates mathematical characteristics of simple chamical reactions. zone order first
	order second order pseudo order half life and mean life Determination of the
	order of reaction - differential method method of integration method of half life
Unit VI	<ul> <li>Solution</li> <li>Thermodynamic derivation of relation between molecular weight and elevation in boiling point and depression in freezing point. Experimental methods for determining various colligative properties. Abnormal molar mass, degree of dissociation and association of solutes.</li> <li>Chemical Kinetics and Catalysis</li> <li>Rate of a reaction- factors influencing the rate of a reaction such as concentration, temperature, pressure, solvent, light and catalyst. Concentration dependence of rates, mathematical characteristics of simple chemical reactions - zero order, first order, second order, pseudo order, half life and mean life. Determination of the order of reaction - differential method, method of integration, method of half life</li> </ul>

period and isolation method. Radioactive decay as a first order phenomenon.		
Experimental methods for the studies of chemical kinetics.		
<b>Theories of chemical kinetics</b> : Effect of temperature on rate of reaction, Arrhenius		
equation, concept of activation energy, Simple collision theory based on hard		
sphere model, transition state theory (equilibrium hypothesis). Expression for the		
rate constant based on equilibrium constant and thermodynamic aspects.		
Catalysis: Characteristics of catalyzed reactions, classification of catalysis,		
Industrial catalysts and enzyme kinetics.		
Suggested Text Book Readings:		
Text Books (Theory Courses):		
a. Physical Chemistry, Puri Sharma & Pathania.		
b. Pradeep Physical Chemistry, Khetrapal, Pradeep Publication.		
c. Computers and Common Sense, R. Hunt and Shelly, Prentice Hall.		
Reference Books:		
a. Physical Chemistry. G.M. Barrow. International Student Edition, McGrawHill		
b. Physical Chemistry, R.A. Alberty, Wiley Eastern Ltd.		
c. The Elements of Physical Chemistry, P.W. Atkins, Oxford.		
d. Physical Chemistry Through problems, S.K. Dogra and S. Dogra, Wiley Eastern Ltd.		
e. Basic Programming with Application, V.K. Jain, Tata McGraw Hill.		
f. Physical Chemistry, Glasstone		
Suggestive digital platforms web links		
1. https://www.coursera.org/courses?query=chemistry&languages=en		
2. https://www.mooc-list.com/tags/physical-chemistry		
3. https://www.coursera.org/learn/physical-chemistry		
4. https://ocw.mit.edu/courses/chemistry/5-61-physical-chemistry-fall-2017/		
5. http://heecontent.upsdc.gov.in/Home.aspx		
6. https://nptel.ac.in/courses/104/108/104108078/		
7. https://nptel.ac.in/courses/104/108/104108124/		
8. https://nptel.ac.in/courses/104/106/104106122/		
This course can be opted as an elective by the students of following subjects:		
Suggested equivalent online courses (MOOCs) for credit transfer:		
1. Chemical Crystallography, Prof. Angshuman Roy Choudhury, <u>https://onlinecourses.nptel.ac.in/noc22_cy48/preview</u>		
Floctronic modia and other digital components in the survisulum.		
$C_{bases only one on more then a SI M/Other electronic and digital contents$		

Choose any one or more than: e-SLM/ Other electronic and digital contentsName of electronic media: e-SLMYear of incorporation: 2021

Course prer	equisites: Chem	istry in 10+2 Lev	el	
Programme: E	3.Sc.	Year:2	-	Semester:4
0		Subject: C	hemistry	•
Course Code: <b>I</b>	Course Code: UGCHE -104N Course Title: INORGANIC CHEMISTRY II (ADVANCE INORGANIC			
	CHEMISTRY)			
Course Object	ctives:			
To giv	ve basic knowledg	ge about Chemistry	of transition and inn	ner-transition elements.
To giv	ve basic knowledg	ge about Concepts of	of coordination chem	nistry and their applications
To giv	ve basic knowledg	ge about Importance	e and different chem	ical aspects of non-aqueous
solven	nts			
Course Outco	omes:	1· , ·,·	1 ( 751 ) .	<b>1</b> . <b>1</b>
CO-I Chemist	try of transition a	nd inner-transition	elements. These insi	ghts are important as they
complexes	ional selection of	the cations of these	e elements for tanor-	made syntheses of newer
CO-2 Concept	ts of coordination	chemistry and thei	r applications	
CO-3 Importa	nce of different a	cid-base concepts v	which forms the basis	s of rational ligand designing
and coordinat	tion complex fo	rmation for speci	fic bioinorganic, n	naterials and optoelectronic
applications.	*			
CO-4 Importa	ance and different	t chemical aspects	of non-aqueous sol	vents which now-a-days are
gaining impor	tance in varied tar	geted syntheses of	drugs and materials f	for technological applications
Credits:2			Type of Course: Co	re
Category of C	Category of Course value-added / employability/			oloyability/
Max. Marks: 1	Max. Marks: 100 Min. Passing Marks: 36			
Block 1				
	Molecular Symmetry			
Unit I	Symmetry Elements, Symmetry Operations and Point groups of different			
	compounds.			
<b></b>	Character Tabl	es of $H_2O$ and $NH$	3.	
Unit II	Chemistry of Transition Elements			
	Position in per	nio rodii vorichi	onic configuration,	General Characteristics, Viz.,
	atomic and ionic radii, variable oxidation states, ability to form complexes,			Constant comparative treatment
	initiation of coloured ions and catalytic benaviour. General comparative treatment of 4d and 5d $(7r/Hf_Nh/Te_Mo/W)$ alaments with their 2d analogues with respect			heir 3d analogues with respect
	to jonic radii, oxidation states and magnetic properties			
Unit III	<b>Coordination</b>	Compounds		
	(i) Definition of	f ligand: Classifica	tion with respect to	denticity. (Examples of mono-
	to hexadentate	ligands).	Ĩ	
	(ii) IUPAC-No	menclature of Trai	nsition Metal comp	lexes.
	(iii) Werner's	postulates, Sidg	wick's effective a	atomic number concept and
	limitations, Va	lence Bond Theory	y of coordination co	ompounds, Stereochemistry of
	coordination n	umbers two, four	, five and six with	h examples of hybrid orbital
	participation in	the following :		

	$ [Ag(NH_3)_2]^+, [Ag(CN)_2]^-, [Ni(CN)_4]^{n-} (n=2 \text{ and } 4), [Cu(NH3)4]^{2+}, [Zn(NH_3)_4]^{2+}, [MnO_4]^-, [Fe(CN)_6]^{n-} (n=3 \text{ and } 4), [FeF_6]^{3-}, [Fe(H_2O)_6]^{3+}, [Fe(C_2O_4)_3]^{3-}, $				
	$[Co(NH_3)_6]^{3+}$ , $[Co(en)_3]^{3+}$ , $[Ni(NH_3)_6]^{2+}$ , $[PbCl_6]^{2-}$				
	(iv) Stability Constant of Transition Metal complexes and Chelate effect				
	(v) Various types of isomerism, viz., hydrate, ionisation, linkage, polymerization				
	and coordination position. Stereoisomerism in C.N4 and C.N6 (only ML <sub>4</sub> L' <sub>2</sub> and				
	ML <sub>3</sub> L' <sub>3</sub> complexes).				
Block 2					
Unit IV	Chemistry of Lanthanides and Actinides				
	i. Electronic Configuration,				
	ii. Atomic, Ionic radii and Lanthanide Contraction.				
	iii. Ionisation energy,				
	iv. Calculation of magnetic moments and correlation with experimental data				
	(specially for lanthanides),				
	v. Colour and spectral behaviour,				
	vi. Oxidation states and their stability,				
	vii. Ability to form complexes and examples of complexes of different coordination numbers.				
	viii. Occurrence and principle of separation of lanthanides.				
	ix. Chemistry of separation of Np, Pu and Am from U and				
	x. One synthesis each of Np to Lr.				
Unit V	Chemistry of Nobel Gases				
	Properties, Occurrence, Isolations and Applications. Chemistry of Noble Gases,				
	Compounds of Xenon & Krypton and their reactions. Clathrates.				
Unit VI	Acid - Base and Non-aqueous solvents				
	Acid - Base concept -Lewis concept, Concept and classification of hard and soft				
	acids and bases. Applications of HSAB principle.				
	Non-aqueous solvents-Classification and characteristic properties of solvents.				
	Types of chemical reactions occurring in liquid ammonia (NH <sub>3</sub> ) and liquid sulphur				
	dioxide (SO <sub>2</sub> ).				
Suggested Te	ext Book Readings:				
Text Books (	Theory Courses):				
a. Concise In	organic Chemistry, J.D. Lee, Blackwell Science Ltd.				
b. Inorganic Chemistry, Puri, Sharma, Kalia and Kaushal.					
c. Pradeep's Inorganic Chemistry, K.K. Bhasin, Pradeep Publication.					
d. Chemistry for degree students, R. L. Madan					
Reference Books:					
a. Inorganic Chemistry, J.E. Huheey, Ellen A. Keiter, Richard					
L. Keiter, Addison Wesley Longman (Singapore) Pvt. Ltd.					
b. Inorganic Chemistry, D.E. Shriver, P W. Atkins and C.H.L. Langford, Oxford.					
c. Basic Inorganic Chemistry, F.A. Cotton, G. Wilkinson and P.L. Gaus, Wiley.					
d. Concepts of Models of Inorganic Chemistry, B. Douglas, D. Mc Daniel and J Alexander,					
John Wiley.	John Wiley.				
e. Inorganic Chemistry, W.W. Porterfield, Addison - Wesley.					
t. Inorganic Chemistry, A.G. Sharpe, ELBS					
g. Inorganic Chemistry, G.L. Meissler and D.A. Tarr, Prentice-Hall.					
Suggested online links:

http://heecontent.upsdc.gov.in/Home.aspx https://www2.chemistry.msu.edu/faculty/reusch/VirtTxtJml/intro1.htm https://nptel.ac.in/courses/104/103/104103071/# https://swayam.gov.in/

This course can be opted as an elective by the students of following subjects:

Suggested equivalent online courses (MOOCs) for credit transfer: Attempt all courses

- 1. Symmetry and Group Theory, Prof. Anindya Datta NPTEL, https://onlinecourses.nptel.ac.in/noc22\_cy40/preview
- 2. Transition Metal Organometallic Chemistry: Principles To Applications , Prof. P. Ghosh , NPTEL https://onlinecourses.nptel.ac.in/noc22\_cy39/preview
- 3. Advanced Transition Metal , Prof. M S Balakrishna , NPTEL https://onlinecourses.nptel.ac.in/noc22\_cy60/preview
- 4. Chemistry of d-block elements, Quantum Chemistry and Spectroscopy, Dr. Niraj Upadhyay, Dr. Harisingh Gour Vishwavidyalaya, Sagar https://onlinecourses.swayam2.ac.in/cec22\_cy05/preview

Electronic media and other digital components in the curriculum:Choose any one or more than:(Electronic Media: Audio/Video Lectures, OnlineCounselling/Virtual Classes/E-Contents/e-SLM/OER/supplementary links forreference/Video Conferencing/Radio broadcast/Web Conferencing/ Other electronic anddigital contents)Name of electronic media: e-SLMYear of incorporation: 2021

				Course prerequisites: Chemistry in 10+2 Level				
Programme: B.Sc		Year:2		Semester:3				
Subject: Chemistry								
Course Code: SI	BSCHE-	Course Title:	ORGANIC CHEM	ISTRY II (ADVANCE				
01N			ORGANIC CHEM	AISTRY)				
Course Objective	es:							
To provid     - Alcohols	e knowledge a s Dihydric alc	about preparation a ohols: (Ethylene Gl	nd chemical reaction ycol)	s of Alcohols and Epoxides				
To provid	e basic knowl	edge about the orde	er of reactivity of diff	ferent carboxylic acid				
derivative	s and the reac	tivity of different c	arboxylic acid deriva	atives.				
To provid	e knowledge a	about mechanism o	t named reactions of	carbonyl compounds and				
Course Outcome	tion reactions	as well as their use	in food and pharmac	ceuticals.				
Course Outcome $CO_1$ The prepara	s: ation and chen	nical reactions of A	lcohols and Enoxide	es - Alcohols Dihydric				
alcohols: (Ethyler	ne Glycol)	lifeat reactions of A	iconois and Epoxide	-s - Alcohols Dinyanc				
CO-2 Understand	ing the order of	of reactivity of diffe	erent carboxvlic acid	derivatives and the				
reactivity of differ	rent carboxyli	c acid derivatives.	j					
CO-3 Able to reco	ognize structu	res of acid halides,	esters, amides, acid a	anhydrides.				
CO-4 Able to wri	te down struct	ture of phenol and p	henoxide ion and ch	nemical reactions of phenols.				
CO-5 Know the n	nechanism of	named reactions of	carbonyl compounds	s and condensation reactions				
as well as their us	e in food and	pharmaceuticals.	_					
Credits:4			Type of Course: Cor	e				
Max. Marks: 100		Min. Passing Mar	ks: 36					
Block 1								
E	lectromagnet	tic Absorption Sp	ectra					
El	lectromagneti	c Radiations, E	lectromagnetic spe	ectrum and absorption of				
Unit I ra	diations. The	e Absorption Law	s. UV-Visible spec	ctrophotometer, formation of				
A	bsorption Bai	nd. Chromatophore	e Concept, Calculat	ion of Absorption Maximum.				
In	fra Red Spec	troscopy Fundame	ntal and Application	ns.				
Unit II A	lcohols and I	Phenols						
	lassification a	ind nomenclature.	Monohydric alcohol	ls – nomenclature, methods of				
fo	rmation by	reduction of alde	hydes, Ketones, C	Carboxylic acids and Esters,				
H	ydrogen bon	ding, Acidic natu	re, Reactions of al	Icohols. Dihydric alcohols –				
no	nomenclature, methods of formation, chemical reactions of vicinal glycols,							
	oxidative cleavage [Pb(OAc) <sub>4</sub> and HIO <sub>4</sub> ] and pinacolo-pinacolone rearrangement.							
	allocation in the second secon							
	henols							
N.	omenclature	structure and bon	ding Preparation of	f phenols, physical properties				
ar	d acidic ch	aracter. Comparat	ive acidic strength	is of alcohols and phenols				
re	sonance stab	ilization of pheno	xide ion. Reaction	is of phenols – electrophilic				
ar	omatic subs	titution. acvlation	n and carboxylati	ion. Mechanisms of Fries				
re	sonance stab	ilization of pheno stitution acylation	oxide ion. Reaction	is of phenols – electrophilic				

	rearrangement, Claisen rearrangement, Gatterman synthesis, Hauben-Hoesch			
	reaction, Lederer-Manasse reaction and Reimer- Tiemann reaction.			
Unit III	Ethers and Epoxide			
	Ethers			
	Nomenclature of ethers and methods of their formation, physical properties,			
	Chemical reactions – cleavage and autoxidation, Ziesel's method. Williamson's			
	synthesis, formation and cleavage of oxonium salts, elementary idea about crown			
	etners.			
	Lpoxides Synthesis of anovides Asid and have actolyzed ring anoning of anovides			
	orientation of enovide ring opening reactions of Grignard and organolithium			
	reagents with enovides			
Block 2				
Unit IV	Aldehydes and Ketones			
Omerv	Nomenclature and structure of the carbonyl groups synthesis of aldehydes and			
	ketones with particular reference to the synthesis of aldehydes from acid chlorides.			
	synthesis of alkedydes and ketones using 1.3-dithianes, synthesis of ketones from			
	nitrites and from carboxylic acids. Physical properties. Mechnism of nucleophillic			
	additions to carbonyl group with particular emphasis on benzoin, aldol, Perkin and			
	Knoevenagel condensations, Condensation with ammonia and its derivatives.			
	Wittig reaction, Mannich reaction. Use of acetals as protecting group, Oxidation of			
	aldehydes, Baeyer-Villiger oxidation of Ketones, Cannizzaro reaction, MPV,			
	Clemmensen, Wolff-Kishner, LiAlH <sub>4</sub> and NaBH <sub>4</sub> reductions. Halogenation of			
	enolizable ketones. An introduction to $\alpha$ , $\beta$ unsaturated alkehydes and ketones.			
Unit V	Carboxylic Acids and Derivatives			
	Nomenclature, structure and bonding, physical properties, acidity of carboxylic			
	acids, effects of substituents on acid strength. Preparation of carboxylic acids,			
	Reactions of carboxylic acids, Hell-Volhard-Zelinsky reaction, Synthesis of acid			
	chlorides, esters and amides. Reduction of carboxylic acids, Mechanism of			
	decarboxylation. Methods of formation and chemical reactions of halo acids,			
	Hydroxy acids: malic, tartaric and citric acids. Methods of formation and chemical			
	formation and affect of heat and dehydrating agents			
	Corboyylic Acid Derivatives			
	Structure and nomenclature of acid chlorides esters amides(urea) and acid			
	anhydrides			
	Relative stability of acyl derivatives Physical properties interconversion of acid			
	derivatives by nucleophilic acyl substitution Preparation of carboxylic acid			
	derivatives of interespinite definition reparation of europyine definition derivatives chemical reactions. Mechanisms of esterification and hydrolysis (acidic			
	and basic).			
Unit VI	Organic Compounds of Nitrogen			
	Preparation of nitroalkanes and nitroarenes. Chemical reactions of nitroalkanes.			
	Mechanisms of nucleophilic substitution in nitroarenes and their reductions in			
	acidic, neutral and alkaline media. Picric acid. Halonitroarenes: reactivity, Structure			
	and nomenclature of amines, physical properties. Stereochemistry of amines.			
	Separation of a mixture of primary, secondary and tertiary amines. Structural			

features effecting basicity of amines. Amine salts as phase-transfer catalysts. Preparation of alkyl and aryl amines (reduction of nitro compounds, nitrites), reductive amination of aldehydic and ketonic compounds. Gabriel-phthalimide reaction, Hofmann bromamide reaction. Reactions of amines, electrophilic aromatic substitution in aryl amines reactions of amines, with nitrous acid
Synthetic transformations of aryl diazonium salts azo coupling
Synthetic transformations of aryl trazomum saits, azo coupling.
a) Organia Chamistry Morrison and Royd Prontice Hall
a) Organic Chemistry, Montson and Boyd, Frendee Hall.
c) Fundamentals of Organic Chemistry Solomons, John Wiley
d) Organic Chemistry, Vol. I. II. III. S.M. Mukherii, S.D. Singh and P.D. Kapoor
Wiley Eastern I td. (New Age International)
e) Organic Chemistry, E.A. Carey, McGraw-Hill Inc.
f) Introduction to Organic Chemistry, Streitwiesser, Hathcock and Kosover
Macmillan
g) Organic Chemistry Vol I II II. Finar
h) Spectrometric Identification of organic compounds Robert M. Silverstein
Clayton G Bassler Terence C Morril John Wiley
Suggested online links:
http://heecontent.upsdc.gov.in/Home.aspx
https://notel.ac.in/courses/104/105/104105124/
https://nptel.ac.in/courses/103/106/105106204/
https://nptel.ac.in/courses/104/105/104105034/
https://nptel.ac.in/courses/104/103/104103121/
https://nptel.ac.in/courses/104/102/104102016/
https://nptel.ac.in/courses/104/106/104106106/
https://nptel.ac.in/courses/104/105/104105120/
This course can be opted as an elective by the students of following subjects:
Suggested equivalent online courses (MOOCs) for credit transfer:
<ol> <li>Organic Chemistry-1, Dr. B. S. Balaji, Jawaharlal Nehru University, <u>https://onlinecourses.swayam2.ac.in/cec22_cy06/preview</u></li> <li>Reagents In Organic Synthesis, Prof. Subhas Chandra Pan, https://onlinecourses.nptel.ac.in/noc22_cy55/preview</li> </ol>

3. Introductory Organic Chemistry II, Prof. Neeraja DashaputreProf. Harinath Chakrapani, <u>https://onlinecourses.nptel.ac.in/noc22\_cy46/preview</u>

Electronic media and other digital components in the curriculum:				
Choose any one or more than: e-SLM/ Other electronic and digital contents				
Name of electronic media: e-SLM	Year of incorporation: 2021			

Course area	aquisites. Char	nistry in 10+21 and	vo1		
Course prer	equisites: Chen	instry in 10+2 Lev	ei		
Programme: B.Sc.		Year: 3		Semester:5	
Subject: Chemistry					
Course Code: D	Course Code: DCECHE -105N Course Title: PHYSICAL CHEMISTRY II (ADVANCE PHYSICAL				
	CHEMISTRY)				
Course Obje	ctives:	1. 1 1 ("	1 11 6	4	
thermoch	sand dasic know	enthalpy etc	and second law of	unermodynamics,	
ulernioer	iennsu'y, entropy	entilalpy etc.			
Course Outco	omes:				
CO-1- A	fter the completion	on of the semester, s	student will acquire l	knowledge of	
first law	and second law of	of thermodynamics,	thermochemistry, er	tropy enthalpy etc.	
CO-2- It	will also make th	em familiar with co	nductance, equivaler	nt conductance, Kohlrausch's	
law, Ostv	wald dilution law	, Deby-Huckel Ons	sagar equation, e.m.f	f. of cell, types of cell, liquid	
junction	potential, pH and	l pka, Henderson- H	lazel equation etc.		
Credits: 2			Type of Course: Co	re	
Category of (	Course(Please m	ention category	employability/ ski	ill development/	
of course; It i	of course: It may have more than one option)				
Max. Marks:	100	Min. Passing Mar	·ks: 36		
Block 1		·			
	Chemical Equ	uilibrium and Pha	se Equlibrium		
	Equilibrium c	onstant and free en	ergy. Thermodynai	mic derivation of law of mass	
	action. Le Cha	attelier's principle.			
	Phase Equilib	orium			
	Statement and	meaning of the te	erms - phase, comp	onent and degree of freedom,	
Unit I	derivation of C	Jibbs phase rule, p	hase equilibria of o	ne component system - water,	
	Sulpher and H	lellium. First and so	econd order phase t	ransitions. Phase equilibria of	
	two component	nt systems - solid-l	iquid equilibria, sir	nple eutectic - Pb-Ag system,	
	molting point	(Ma Zn) and an ir	nivorving compoun	point (CuSO4 H2O) Normat	
	distribution la	(Mg-ZH) and an $H$	namic derivation	point (Cu304-1120). Nemst	
Unit II	Thermodyna	mics –II			
Chit II	Second law or	f thermodynamics:	concept of entropy	v entropy as a state function	
	entropy as a fu	unction of V & T. $\epsilon$	entropy as a functio	n of P & T. entropy change in	
	physical proce	ess.	FJ		
	Gibbs and H	elmholtz functions	s; Criteria for the	modynamic equilibrium and	
	spontaneity in term of changes in entropy, Gibbs and Helmholtz functions. Concept				
	of chemical potential.				
Unit III	Electrochemi	stry – II			
	Types of reve	ersible electrodes	- gas-metal ion, me	etal-metal ion, metal-insoluble	
salt-anion and redox electrodes. Electrode reactions, Nernst equation, derivatio				Nernst equation, derivation of	
	cell E.M.F. and single electrode potential, standard hydrogen electrode-reference				

	electrodes-standard electrode potential, sign conventions, electrochemical series				
	and its significance.				
	Electrolytic and Galvanic cells - reversible and irreversible cells, conventional				
	representation of electrochemical cells.				
	EMF of a cell and its measurements. Computation of cell EMF. Calculation of				
	thermodynamic quantities of cell reactions ( $\Delta G$ , $\Delta H$ and K).				
	Concentration cell with and without transport, liquid junction potential, application				
	of concentration cells, valency of ions, solubility product and activity coefficient,				
	potentiometric titrations.				
	Definition of pH and pKa determination of pH using hydrogen, quinhydrone and				
	glass electrodes, by potentiometric methods.				
	<b>Buffers</b> - mechanism of buffer action, Henderson-Hazel equation. Hydrolysis of				
	salts.				
	Electrochemical corrosion and its prevention.				
Block 2					
Unit IV	Colloidal State and Macromolecules				
	Definition of colloids and classification of colloids. Donnan membrane theory and				
	its application. Electrokinetic Potential (Zeta potential).				
	Solids in inquids (sols): properties - kinetic, optical and electrical; stability of				
	Liquida in liquida (amulaiona): tunas of amulaiona, proportion. Emulaifiar				
	Liquids in aquids (enuisions): types of enuisions, preparation, Enuisiner.				
	Liquids in solids (gels): classification, preparation and properties, inhibition,				
	general applications of contours. Macromolecules • Determination of molecular weight of macromolecules by				
	osmotic pressure and viscosity methods. Concepts of micelles and critical micelle				
	concentrations				
	A brief introduction to conducting and light emitting polymers				
Unit V	Surface Phenomenon				
Olife V	Surface Chemistry				
	Adsorption, difference between Physical adsorption and chemisorption. Adsorption				
	isotherms - Langmuir adsorption isotherm and Freundlich adsorption isotherm,				
	Gibbs adsorption equation, BET equation, Determination of surface area.				
Unit VI	Physical Properties and Chemical Constitution				
	Molar volume, Parachor Molar refraction and Polarisation, Dipolemoment, Debey				
	equation (derivation not required) and Clausius-Mosotti equation.				
Suggested Te	ext Book Readings:				
a. Physical	Chemistry. G.M. Barrow. International Student Edition, McGraw Hill.				
b. Physical	Chemistry, R.A. Alberty, Wiley Eastern Ltd.				
c. The Elem	chemistry, Through angle S.K. Dears and S. Dears, Wiley Eastern Ltd				
d. Physical	chemistry Infolgen problems, S.K. Dogra and S. Dogra, whey Eastern Ltd.				
f Principles	s of Physical Chemistry by B R Puri L P Sharma and M S Pathania Vishal				
publication.	Jalandhar.				
Suggestive d	igital platforms web links				
1. https://www	w.coursera.org/courses?query=chemistry&languages=en				
2. https://www	w.mooc-list.com/tags/physical-chemistry				
3. https://www.coursera.org/learn/physical-chemistry					

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- 4. https://ocw.mit.edu/courses/chemistry/5-61-physical-chemistry-fall-2017/
- 5. http://heecontent.upsdc.gov.in/Home.aspx
- 6. https://nptel.ac.in/courses/104/108/104108078/
- 7. https://nptel.ac.in/courses/104/108/104108124/
- 8. https://nptel.ac.in/courses/104/106/104106122/

This course can be opted as an elective by the students of following subjects:

Suggested equivalent online courses (MOOCs) for credit transfer:

- 1. Chemistry and Physics of Surfaces and Interfaces, Prof. Thiruvancheril G. Gopakumar, NPTEL, <u>https://onlinecourses.nptel.ac.in/noc22\_cy57/preview</u>
- 2. Introduction to Chemical Thermodynamics and Kinetics, Prof. Arijit Kumar De, NPTEL, <u>https://onlinecourses.nptel.ac.in/noc22\_cy58/preview</u>

#### Electronic media and other digital components in the curriculum:

Choose any one or more than: e-SLM/Other electronic and digital contents

Name of electronic media: e-SLM	Year of incorporation: 2021
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Course prer	equisites: Chem	istry in 10+2 Level			
Programme: B.Sc.		Year:3	Semester:5		
Subject: Chemistry					
Course Code: I	Course Code: DCECHE -106 Course Title: INORGANIC CHEMISTRY III (SELECTED				
<b>TOPICS IN INORGANIC CHEMISTRY</b> )					
Course Object	ives:				
To pro	vide basic knowl	edge about chemistry of transition and	l inner-transition elements,		
Conce	pts of coordination	on chemistry and their applications			
• To pro	vide basic knowl	edge about importance of different ac	id-base concepts.		
Course Outcon	nes:				
CO-1 Chemis	try of transition a	and inner-transition elements. These	insights are important as they		
help in the rat	ional selection o	f the cations of these elements for ta	ilor-made syntheses of newer		
complexes	<b>C 1</b>				
CO-2 Concept	s of coordination	chemistry and their applications			
CO-3 Importa	nce of different a	cid-base concepts which forms the bas	sis of rational ligand designing		
and coordinat	ion complex to	rmation for specific biomorganic,	materials and optoelectronic		
CO 4 Importa	nce and differen	t chamical aspects of non aqueous so	alvents which now a days are		
gaining importa	ance in varied tar	regeted syntheses of drugs and materials	for technological applications		
Credits: 2		Type of Course: F			
Max. Marks: 1	00	Min. Passing Marks: 36			
Block 1					
	Metal-ligand I	Bonding in Transition Metal Com	olexes		
	Limitations of	valance bond theory. an elementar	ry idea of crystal field theory.		
	Crystal Field S	Stabilization Energy (CFSE), crysta	al field splitting in octahedral,		
TT • T	tetrahedral and	a square planner complexes, factor	ors affecting the crystal-field		
Unit I	parameters.		Ç .		
	Thermodynan	nic and Kinetic Aspects of Metal C	complexes		
	A brief outline	of thermodynamic stability of metal	complexes and factors affecting		
	the stability, substitution reactions of square planar complexes and trans effect.				
Unit II	Magnetic and Electronic spectra of Transition Metal Complexes				
	(a) Electronic spectra of Transition Metal Complexes				
	Types of elect	ronic transitions, selection rules for	r d-d transitions, spectroscopic		
	ground states,	spectrochemical series. Orgel-energ	gy level diagram for $d^1$ and $d^9$		
	states, discussion	on of the electronic spectrum of [Ti(	$H_2O_6]^{3+}$ complex ion.		
	(b) Magnetic I	Properties of Transition Metal Cor	nplexes		
	Types of magne	etic behavior, methods of determinin	g magnetic susceptibility, spin-		
	only formula. I	L-S coupling, correlation of $\mu$ s and $\mu$	ueff values, orbital contribution		
	to magnetic moments, application of magnetic moment data for 3d-metal				
	complexes.				
Unit III	Organometall	ic Chemistry			
	Definition, no	menclature and classification of	organometallic compounds.		
	Preparation, properties, bonding and catalytic applications of alkyls and aryls of Li,				
	AI, Hg, Sn.				
Block 2					

Unit IV	Metal Carbonyls and Nitrosyls				
	(a) Metal Carbonyls : Ligand behaviour of CO, General methods of preparation,				
	18 electron rule, nature of bonding (Synergic effect) in the mononuclear carbonyls,				
	Representation of structures of the binary carbonyls of all nuclearities of V, Cr, Mn,				
	Fe, Co and Ni.				
	(b) Metal Nitrosyls : Ligand behaviour of NO (NO <sup>+</sup> , NO <sup>-</sup> and bridging NO),				
	preparation and structures of nitrosyls of Cr, Fe and Ru; carbonyl nitrosyls and				
	cyano nitrosyls				
Unit V	Inorganic Polymers				
	Silicones and Phosphazenes				
	Silicons and phosphazenes as examples of inorganic polymers, nature of bonding				
	in triphosphazenes.				
Unit VI	Inorganic Biochemistry				
	Essential and trace elements in biological processes, metalloporphyrins with special				
	reference to oxygen carrieres hemoglobin chemistry and myoglobin. Vitamin B-12,				
	Nitogenase and Chlorophyll structure and applications. Biological role of alkali and				
	alkaline earth metal ions with special reference to $Na^+$ , $K^+$ and $Ca^{2+}$ .				
Unit VII	Environmental Chemistry and Green Chemistry				
	(a) <b>Environmental Chemistry</b> : The earth's atmosphere and its components, Lapse				
	rate, Types of pollutants and their sources (in water, Air and Soil). Green house				
	effect and global warming. Acid rains, Ozone layer (Importance and its protection).				
	(b) Green Chemistry				
	Principles and concept of green chemistry, atom economic and noneconomic				
	reactions, reducing toxicity, a few examples of environmental friendly reactions				
	and reaction media.				
Unit VIII	Nietal and Nietallurgy				
	isolation of elements. Extraction and isolation of Matals (V La Ti Zr Hf V Nh				
	To Cr. Mo W. Mn. To Do Eo Co Ni and platinum) from their minorals				
Suggested To	Ta, Ci, Mo, W, Mii, Tc, Ke, Fe, Co, Ni and platinum/ from their initierals.				
Suggested Te	ext DOOK Readings.				
a. Concise in b. Inorganic	Chemistry, Duri, Sharma, Kalia and Kaushal				
c Pradeen's I	Inorganic Chemistry K K Bhasin Prodeen Publication				
d Chemistry	for degree students R I. Madan				
Reference Bo	ooks.				
a Inorganic (	Chemistry I.E. Huheev Ellen A. Keiter Richard				
L. Keiter, Ad	ldison Wesley Longman (Singapore) Pyt. Ltd				
b. Inorganic Chemistry D.F. Shriver P.W. Atkins and C.H.L. Langford Oxford					
c. Basic Inorganic Chemistry, F.A. Cotton, G. Wilkinson and P.L. Gaus, Wiley					
d. Concepts of Models of Inorganic Chemistry, B. Douglas, D.					
Mc Daniel ar	nd J Alexander, John Wiley.				
e. Inorganic (	Chemistry, W.W. Porterfield, Addison - Wesley.				
f. Inorganic C	Chemistry, A.G. Sharpe, ELBS				
g. Inorganic	g. Inorganic Chemistry, G.L. Meissler and D.A. Tarr. Prentice-Hall.				
Suggestive dig	ital platforms web links:				
https://swayar	n.gov.in/				

https://www.coursera.org/learn/physical-chemistry https://www.mooc-list.com/tags/physical-chemistry https://www.openlearning.com/courses/introduction-to-physical-chemistry/ https://www.my-mooc.com/en/categorie/chemistry https://onlinecourses.swayam2.ac.in/nce19\_sc15/preview https://www.coursera.org/browse/physical-science-and-engineering/chemistry

This course can be opted as an elective by the students of following subjects:

Suggested equivalent online courses (MOOCs) for credit transfer:

#### **Electronic media and other digital components in the curriculum: Choose any one or more than:**e-SLM/Other electronic and digital contents

Name of electronic media: e-SLMYear of incorporation: 2022

Course prerequisites: Chemistry in 10+2 Level						
Programme: B.Sc.		Year:3		Semester:6		
Subject: Chemistry						
Course Code: D	Course Code: DCECHE -108 Course Title: ORGANIC CHEMISTRY III (SELECTED TOPICS II ORGANIC CHEMISTRY)					
Course Object	c <b>tives:</b> This cour	se will provide ba	sic qualitative and qua	antitative experimental		
knowledge of	biomolecules su	ch as carbohydrat	es, proteins, amino aci	ids, nucleic acids drug		
molecules. Up	on successful co	ompletion of this c	ourse students may ge	t job opportunities in food,		
beverage and	pharmaceutical i	industries.				
Course Outed	vines: knowledge abou	it qualitative and	quantitative experime	ntal knowledge of		
biomolecules	such as carbohy	drates proteins ar	nino acids nucleic aci	ids drug molecules		
CO2: To prov	ide knowledge a	bout Organometal	llic Compounds. Sulph	hur Containing Compounds		
and NMR Spe	ectroscopy.	C	1 / 1			
Credits:2			Type of Course: Ele	ective		
Max. Marks: 1	100	Min. Passing M	larks: 36			
Block 1		<u>a</u>				
	NMR (PMR)	Spectroscopy		1 1.11. 1		
	Proton magnetic resonance (1H NMR) spectroscopy, nuclear shielding and					
	desnielding, c	deshielding, chemical shift and molecular structure, spin-spin splitting and coupling				
Unit I	constants, are	constants, areas of signals, interpretation of IH NMR spectra of simple organic				
	ethyl acetate	molecules such as ethyl bromide, ethanol, acetaldehyde, 1,1,2-tribromoethane,				
	elucidation of	elucidation of simple organic compounds using UV IR and 1H NMR spectroscopic				
	techniques					
Unit II	Organometallic Compounds					
	Organomagnesium compounds: the Grignard reagents, formation, structure and					
	Chemical reactions. Organozinc compounds: formation and chemical reaction					
	Organolithiun	n compounds: for	mation and chemical	reactions.		
Unit III	Sulphur Containing Compounds					
	Nomenclature	, structural forma	ation, Methods of for	mation and chemical reactions		
	of thiols, thioethers, sulphonic acids, sulphonamides & Sulphaguamidine.					
Block 2						
Unit IV	Amino Acids, Peptides, Proteins and Nucleic Acids					
	Classification	, structure and ste	ereochemistry of amil	no acids. Acid-base behaviour,		
	Structure and	nomenclature of	f nentides and protoi	ins Classification of proteins		
	Pentide struc	ture determination	on end group anal	vsis selective hydrolysis of		
	pentides Clas	sical peptide syn	thesis solid-nhase n	eptide synthesis Structures of		
	peptides and r	proteins. Levels of	protein structure. Pro	tein denaturation/renaturation.		
	Nucleic acids	: Introduction. Co	onstituents of ncleic a	acids. Ribonucleosides and		
	ribonucleotides. The double helical structure of DNA.					
Unit V	Active Methylene Group					

	Preparation and synthetic applications of ethyl acetoacetate and diethyl malonate,					
	Tautomerism.					
Unit VI	Carbohydrates					
	Classification and nomenclature, Monosaccharides, mechanism of osazone					
	formation, interconversion of glucose and fructose, chain lengthening and chain					
	shortening of aldoses Configuration of monosaccharides Ervthro and three					
	diastaraomars Conversion of glucose into menness. Enguito and theoridas					
	others and estars. Determination of ring size of monoscenebrides. Cyclic structure					
	ethers and esters. Determination of fing size of monosacchandes. Cyclic structure $f(D(x))$ above. Machanizm of workerstetic Structure of eithers and descervite					
	of $D(+)$ -glucose. Mechanism of mutarotation. Structures of ribose and deoxyribose.					
	An introduction to disaccarides (maltose, sucrose and lactose) and polysaccharides					
	(starch and cellulose) without involving structure determination.					
Unit VII	Problem based on Spectroscopy (UV-Vis., IR and PMR)					
Suggested Te	ext Book Readings:					
1. Furniss, B.S	.; Hannaford, A.J.; Smith, P.W.G.; Tatchell, A.R. Practical Organic Chemistry, 5th Ed.,					
Pearson						
(2012).						
2. Mann, F.G.	& Saunders, B.C. Practical Organic Chemistry, Pearson Education.					
3. Vogel's Qua	alitative Inorganic Analysis, Revised by G. Svehla.					
4. Vogel, A.I.	A Textbook of Quantitative Analysis, ELBS. 1986					
5. Furniss, B.S	5. Furniss, B.S.; Hannaford, A.J.; Rogers, V.; Smith, P.W.G.; Tatchell, A.R. Vogel's Textbook of					
Practical Organ	Practical Organic Chemistry, ELBS.					
6. Ahluwalia, V	6. Ahluwalia, V.K. & Aggarwal, R. Comprehensive Practical Organic Chemistry, Universities Pres					
7. Cooper, T.G	a. Tool of Biochemistry. Wiley-Blackwell (1977).					
8. Wilson, K. &	& Walker, J. Practical Biochemistry. Cambridge University Press (2009).					
9. Varley, H., Gowenlock, A.H & Bell, M.: Practical Clinical Biochemistry, Heinemann,						
Suggestive dig	ital platforms web links					
1. https://www	.labster.com/chemistry-virtual-labs/					
2. https://www	.vlab.co.in/broad-area-chemical-sciences					
3. http://chemo	collective.org/vlabs					
This course c	can be opted as an elective by the students of following subjects:					
Suggested eq	uivalent online courses (MOOCs) for credit transfer: Attempt all courses					
1. Applicat	ion of Spectroscopic Methods in Molecular Structure Determination, Prof. S. Sankararaman, nlinecourses.nptel.ac.in/noc22_cy45/preview_					
2. NMR sp	ectroscopy, Prof. R. V Hosur, https://onlinecourses.nptel.ac.in/noc22_cy59/preview					
3. Organic	Chemistry In Biology, Prof. Amit Basak, https://onlinecourses.nptel.ac.in/noc22_cy62/preview					
Electronic m	edia and other digital components in the curriculum:					
Choose any o	ne or more than: e-SLM/ Other electronic and digital contents					
Name of elect	ronic media: 2022 Year of incorporation: 2022					

Course prerequisites: Chemistry in 10+2 Level					
Programme:	B.Sc.	Year:3		Semester:6	
Subject: Chemistry					
Course Code: I	Course Code: DCECHE -109 Course Title: PHYSICAL CHEMISTRY III (SELECTED TOPICS IN PHYSICAL CHEMISTRY)				
Course Object	ctives:				
•	To provide know	vledge about Quant	tum mechanics as w	well as of spectroscopy with	
	comprehensive u	nderstanding of vale	ence bond model and	l molecular orbital model.	
•	To provide know	ledge about Ultravio	olet absorption spect	roscopy, Vibrational,	
]	Rotational and El	ectronic Spectrosco	py, Infrared spectros	scopy and Bioenergetics	
Course Outco	omes:				
CO-1 Q	Quantum mechar	nics as well as of	spectroscopy. The	ey will have comprehensive	
understa	inding of valence	bond model and mo	blecular orbital mode		
CO-2 U	Itraviolet absorp	tion spectroscopy, E	seer Lambert Law, t	types of electronic transitions	
and the $CO = V^{2}$	ibrational Botati	uon and concept of	Spectroscopy of size	ixociirome.	
CO-3 V	frared spectrosco	onal and Electronic	eristic absorptions of	f various functional groups	
CO-5 Bi	ioenergetics-Gibb	by in which charact	ergies with special e	m various functional groups.	
applicati	ions		ergies with special e	inpliasis on biological	
Credits:2	10115		Type of Course: Co	ore	
Category of (	Course		value-added / em	nlovability	
Max Marks	100	Min Passing Ma	rks: 36	proyability	
(Syllabi	i should be fram	ed block wise/unit	wise: No of blocks	and units may change)	
Block 1					
	Elementary	Quantum Mechan	ics		
	Black-body ra	adiation. Planck's r	adiation law. photo	electric effect, heat capacity of	
Unit I	solids, Bohr's	model of hydroge	en atom (no derivat	tion) and its defects, Compton	
	effect. de Bro	glie's hypothesis, th	e Heisenberg's unc	ertainty principle, Hamiltonian	
	operator. Stat	ement of the Born-	Oppenheimer appro	oximation, degrees of freedom.	
Unit II	Molecular Statistics				
	The Boltzmar	n distribution. Max	well distribution la	w for distribution of molecular	
	speeds. The N	/laxwell-Boltzmanr	distribution law fo	or the distribution of molecular	
	energies. The	e partition function	ons. Thermodynam	nic quantities from partition	
	functions. Th	e Sackur-Tetrode	equation for molar	entropy of monatomic gases.	
	Rotational and	d vibrational partiti	on functions. The c	characteristic temperature. The	
	calculation of Gibbs free energy changes and equilibrium constant in terms of				
	partition functions.				
Unit III	Laws of Phot	tochemistry			
	Interaction of	radiation with matt	er, difference betwe	een thermal and photochemical	
	processes. La	ws of photochemis	try: Grothus – Drap	pper law, Stark – Einstein law,	
	Jablonski dia	gram depicting va	arious processes of	ccurring in the excited state,	
	qualitative de	scription of fluore	scence, phosphores	scence, nonradiative processes	
	(internal conversion, intersystem crossing), quantum yield, photosensitized				
	reactions – energy transfer processes (simple examples).				

Block 2			
Unit IV	Vibrational, Rotational and Electronic Spectroscopy		
	Rotational Spectrum:		
	Diatomic molecules: Energy levels of a rigid rotor (semi-classical principles),		
	selection rules, spectral intensity, distribution using population distribution		
	(Maxwell-Boltzmann distribution) determination of bond length, qualitative		
	description of non-rigid rotor, isotope effect.		
	Vibrational Spectrum:		
	Infrared spectrum: Energy levels of simple harmonic oscillator, selection rules,		
	pure vibrational spectrum, intensity, determination of force constant and qualitative		
	relation of force constant and bond energies, effect of of different functional groups.		
	Raman Spectrum: Concept of polarizability, pure rotational and pure vibrational		
	Raman spectra of diatomic molecules, selection rules.		
	Electronic Spectrum: Concept of potential energy curves for bonding and		
	antibonding molecular orbitals, qualitative description of selection rules and		
	Franck-Condon principle.		
	<b>Qualitative description of</b> $\sigma$ , $\pi$ - and n M.O., their energy levels and the respective		
	transitions.		
Unit V	Nuclear Chemistry		
	Nuclear reactions: Bethe notation, types of nuclear reactions (n, p, $\alpha$ , d and $\gamma$ ),		
	conservation of quantities (mass-energy and linear momentum) in nuclear reactions,		
	reaction cross-section, compound nucleus theory and nuclear reactions. Nuclear		
	fission: the process, fragments, mass distribution, and fission energy. Nuclear		
	reactor: the natural uranium reactor, classification of reactors, breeder reactor.		
	Nuclear fusion and stellar energy.		
	Radiation chemistry: Elementary ideas of radiation chemistry, radiolysis of water		
	and aqueous solutions, unit of radiation chemical yield (G-value), radiation		
	dosimetry (Fricke's dosimeter), units of radiation energy (Rad, Gray, Rontgen,		
	RBE, Rcm, Sievert).		
Unit VI	Bioenergetics		
	Gibbs and Helmholtz energies with special emphasis on biological applications:		
	study of energy transformations in living systems (bioenergetics): standard state in		
	biochemistry, ATP-the currency of energy, Glycolysis, limitation of applicability		
	of thermodynamics in biology.		
Suggested Te	ext Book Readings:		
1. Skoog .D	A., West. D.M and Holler .F.J., "Analytical Chemistry: An Introduction", 7th		
edition, Sau	nders college publishing, Philadelphia,(2010).		
2. Larry Hargis. G" Analytical Chemistry: Principles and Techniques" Pearson©(1988)			
Note: For the promotion of Hindi language, course books published in Hindi may be			
prescribed by the University Suggestive digital platforms web links			
1 https://www.labster.com/chemistry-virtual-labs/			
2. https://www.vlab.co.in/broad-area-chemical-sciences			
3. http://chemcollective.org/vlabs			
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This course can be opted as an elective by the students of following subjects:

Suggested equivalent online courses (MOOCs) for credit transfer:

1. Quantum Chemistry of Atoms and Molecules, Prof. Anindya Datta, <u>https://onlinecourses.nptel.ac.in/noc22\_cy41/preview</u>

Electronic media and other digital components in the curriculum:			
Choose any one or more than:e-SLM/ Other electronic and digital contents			
Name of electronic media: e-SLM	Year of incorporation: 2022		

Course prerequisites: Chemistry in 10+2 Level					
Programme: B.Sc.		Year:3		Semester: 6 <sup>th</sup>	
		Subject: Cl	nemistry	-	
Course Code:	SBSCHE-02N	Course Title: ADVA	NCED ANALYTICA	L TECHNIQUES	
Course Obje	ctives:				
To pre	ovide knowledge	about Statistical A	Analysis		
• T	'o provide basic	knowledge about	Volumetric analysi	is, Gravimetric analysis and	
S	eparation techniq	ues.			
• 1	'o provide basic k	nowledge about Na	ano Chemistry.		
Course Outco	omes:				
CO1: To gain	knowledge abou	t Statistical Analys			
CO2: To gain b	asic knowledge at	out Volumetric ana	lysis, Gravimetric and	alysis and Separation techniques	
Credite: 4	ic knowledge of N	and Chemistry.	Turna of Courses Co		
Cleans.4			Type of Course: Co		
Category of C	lourse		value-added / emp	ployability/	
Max. Marks:	100	Min. Passing Mar	ks: 36		
(Syllabi	should be frame	d block wise/unit	wise; No of blocks a	and units may change)	
Block 1					
	BLOCK-1				
	Unit 1: Statist	ical Analysis			
	Definition of	terms mean and i	median, precision,	, standard deviation, relative	
Unit I	standard devia	tion, accuracy, ab	solute error, types	of error in experimental data,	
	determinate (	systematic), indet	terminate (or rand	dom) and gross, sources of	
	errors and effects upon the analytical results, methods for reporting analytical				
<b>TT 1 TT</b>	data, statistical evaluation of data, indeterminate errors, uses of statistics.				
Unit II	Unit 2: Volumetric analysis				
	General princ	iples of acid – ba	ase titration, preci	ipitation titration, oxidation-	
	EDTA for the	dotormination of	Co21 and Mo21	plexometric utrations, use of	
	of EDTA for the	tions motal ion in	Ca2+ and Mg2+	and hardness of water, types	
	of EDTA titrations, metal ion indicators.				
Unit III	Unit III Unit 3: Gravimetric analysis				
Oline III	Precipitation	from homogene	eous medium n	purity of precipitates co-	
	precipitation	nost- precipitati	ion washing an	d ignition of precipitates	
	contamination and their removal				
Block 2					
Unit IV	Unit 4: Separa	tion techniques			
	Principle, tech	nique and analyti	cal applications of	the following:	
	(a) Solvent ex	traction	11	C	
	(b) Chromatography (Paper, Thin Layer, Column and HPLC)			and HPLC)	
	(c) Ion exchange				

Unit V	Unit 5: Nano Chemistry				
	Nanomaterials – An Introduction, Size Effects, Defining Nanodimensional				
	Materials, Potential Uses for Nanodimensional Materials, The General Methods				
	Available for theSynthesis of Na	anodimensional Materials, Precipitative			
	Methods, Reactive Methods in High	Boiling Point Solvents, Hydrothermal and			
	Solvothermal Methods, Gas-Phase S	ynthesis of Semiconductor Nanoparticles,			
	Synthesis in a Structured Medium, Th	e Suitability of Such Methods for Scaling,			
	Conclusions and Perspectives on the	Future, Oxide Nanoparticles, Nanotubes			
	and Nanowires. Study of different c	haracterization tools (XRD, TEM, SEM,			
	AFM, etc.) for Nanomaterials.				
Suggested Te	vt Book Readings:				
1. Skoog .D	A. West, D.M and Holler F.J., "Anal	vtical Chemistry: An Introduction". 7th			
edition, Sau	edition. Saunders college publishing. Philadelphia.(2010).				
2. Larry Har	2. Larry Hargis.G" Analytical Chemistry: Principles and Techniques" Pearson©(1988)				
Note: For th	Note: For the promotion of Hindi language, course books published in Hindi may be				
prescribed by the University					
1 https://www.labster.com/chemistry-virtual-labs/					
2 https://www.ylab.co.in/broad-area-chemical-sciences					
3. http://chemcollective.org/vlabs					
This course c	This course can be opted as an elective by the students of following subjects:				
Suggested equivalent online courses (MOOCs) for credit transfer: NA					
<b>Choose any one or more than</b> set $M//O$ there electronic and digital contents					
Name of elec	Name of electronic media: e-SLM Vear of incorporation 2022				

#### Year-2023-2024 Syllabus of B.Sc. Programme: [Subject Name: Computer Science] In accordance with NEP-2020

Year	Semester	Course Code	Title of Paper	Credits	Max. Marks
	-	UGCS -101N	Computer Fundamental & PC Software	2	100
	1	UGCS -101(P)N	Practical Work	2	100
1		UGCS -102N	C Programming	2	100
		UGCS -102(P)N	Practical Work	2	100
	Π	Skill Enhancement C	ourse		
		SBSCS-02N	Python Programming	4	100
		UGCS -103N	Data Structures	2	100
2	ш	UGCS -103(P)N	Practical Work	2	100
-		Skill Enhancement Course			
		SBSCS-01N	Discrete Mathematics	4	100
IV		UGCS -104N	Introduction to Database Management System	2	100
		UGCS -104(P)N	Practical Work	2	100
	Discipline Centric Elective Course				
	<b>X</b> 7	DCECS -105N	Computer Network	2	100
	v	DCECS -106N	Operating System	2	100
3		DCECS -107(P)N	Practical Work based on 106	2	100
		Discipline Centric Ele	ective Course		
	VI	DCECS -108N	C++ and Object Oriented Programming	2	100
	VI I	DCECS -109N	Software Engineering	2	100
		DCECS -110(P)N	Practical Work based on 108	2	100
			Total Credit/Max. Marks	36	1600

Course prerequisites: 10+2 with Computer science, Mathematics				
Programme: B.Sc.Year: 1Semester: I				
Subject: Computer science				
<b>Course Code:</b>	Course Code: UGCS -101N Course Title: Computer Fundamental & PC Software			
Course Object	tives:			
To demons	strate the knowledge of the basic structure, components, features and generations of			
computers.				
To describ	e the concept of computer languages, language translators and construct algorithms to solve			
problems u	ising programming concepts.			
To Compa	re and contrast features, functioning & types of operating system and computer networks.			
To demons	strate architecture, functioning & services of the Internet and basics of multimedia.			
To illustrat	e the emerging trends and technologies in the field of Information Technology.			
<b>Course Outco</b>	mes:			
CO-1 Demons	trate the knowledge of the basic structure, components, features and generations of			
computers.				
CO-2 Describe	the concept of computer languages, language translators and construct algorithms to solve			
problems using	; programming concepts.			
CO-3 Compare	and contrast features, functioning & types of operating system and computer networks.			
CO-4 Demons	the amorging trends and technologies in the field of Information Technology			
Credits: 2	Type of Course: Core			
Max Marks: 1	00 Min Passing Marks: 36			
Block 1				
DIUCK I	Computer Basics:			
	Algorithms A Simple Model of a Computer Characteristics of Computers Problem-			
	solving Using Computers			
	Data Representation: Representation of Characters in computers Representation of			
	Integers Representation of Eractions Heyadecimal Representation of Numbers Decimal			
Unit I	to Binary Conversion Error detecting codes Input & Output Devices: Description of			
	to Binary Conversion, Error-detecting codes. Input & Output Devices: Description of			
	Computer input Units, Other input Methods, Computer Output Units (Printers, Piotters)			
	Operating Systems: History and Evolution. Main functions of OS Multitasking,			
	Multiprocessing. Time Sharing, Real Time OS with Examples Database Management			
	System: Purpose and Organization of Database, Introduction to Data Models.			
Unit II	Disk operating system(DOS):			
	Introduction, history & versions of DOS, DOS basics- Physical structure of disk, drive			
	name, FAT, file & directory structure and naming rules, booting process, DOS system			
	Tiles, DOS commands- internal & external,.			
Unit III	Windows Operating System: Windows concerts Eastware Windows Structure Desister Testsher Start Many My			
	Computer Decycle Pin Windows Accessories, Colculator Netword, Start Wordned			
	Computer, Recycle Bin, Windows Accessories- Calculator, Notepau, Paint, Wordpau,			
	Unaracter Map, windows Explorer, Entertainment, Managing Hardware & Software- Installation of Hardware & Software Using Scanner, System Tools, Communication			
	Installation of Hardware & Software, Using Scanner, System 1001s, Communication,			
Block 2				
Unit IV	Word Processing, MS-Word			
	Features Creating Saving and Opening Documents in Word Interface Toolhers Dular			
	Manus Kayboard Shortout Editing Draviaging Drinting & Earmatting a Decument			
1	menus, reyooard shortcut, Editing, rieviewing, rinning,& Formatting a Document,			

	Advanced Features of MS Word, Find & Replace, Using Thesaurus, Using Auto-			
	Multiple Functions, Mail Merge, Handling Graphics, Tables & Charts, Converting a word			
	document into various formats like- Text, Rich. Text format, Word perfect, HTML etc.			
Unit V	Worksheet- MS-Excel: Worksheet basics, creating worksheet, entering into worksheet,			
	heading information, data, text, dates, al	phanumeric values, saving & quitting worksheet,		
	Opening and moving around in an exist	sting worksheet, Toolbars and Menus, Keyboard		
	shortcuts, Working with single and mu	ltiple workbook, working with formulae & cell		
	referencing, Auto sum, Coping formulae	, Absolute & relative addressing, Worksheet with		
	ranges, formatting of worksheet, Previe	ewing & Printing worksheet, Graphs and charts.		
	Database, Creating and Using macros, multiple worksheets- concepts, creating and using.			
Suggested Te	Suggested Text Book Readings:			
1. Fundamenta	1. Fundamental of Computers – By V.Rajaraman, B.P.B. Publications			
2. Fundamental of Computers – By P.K. Sinha				
3. Microsoft Office 2007 Bible – John Walkenbach, HerbTyson, Faithe Wempen, caryN. Prague, Michael				
R.groh, PeterG. Aitken, and Lisa a. Bucki -Wiley India pvt. ltd				
4. Discovering the Internet: Complete - Shelly Cashman 4th Edition - Course Technology				
Reference Books:				
1. Introduction to Information Technology - Alexis Leon, Mathews Leon, and Leena Leon, Vijay Nicole				
Imprints Pvt. Ltd., 2013.				
2. Office 2007 – By Shelly, Cengage Publication				
Suggested online links: <a href="https://www.pearsoned.co.in/prc/book/anita-goel-computer-">https://www.pearsoned.co.in/prc/book/anita-goel-computer-</a>				
fundamentals-1e-1/9788131733097				
Electronic media and other digital components in the curriculum:				
Name of electr	Name of electronic media: e-SLMYear of incorporation: 2020			
		-		

Course prerequisites: 10+2 wi	th Computer science, Mathematics		
Programme: B.Sc.	Year: 1	Semester: 1	
Subject: Computer science			
Course Code: UGCs 101P(N)	Course Title: Practical -WORK		
<b>Course Objectives:</b>			
• To understand the fundament	tal concept of computer.		
• To understand the basics cor	ncept of communication and network.		
• To explore various features of	of MS-Word and its applications.		
To develop understanding of	MS-Excel. To design the presentation	n using MS-power Point. To	
understand the fundamental	concept of database and working with	n MS-Access	
<b>Course Outcomes:</b>			
CO1. Understand the basic comp	ponent of computers, software and Ha	rdware.	
CO2. Acquire knowledge about	MS-Word and different formatting st	yles used in that.	
CO3. Acquire knowledge about	MS-Excel and different techniques us	sed in that.	
CO4. Acquire knowledge about	MS-Power Point and formatting style	s used in that.	
CO5. Acquire knowledge about	MS-Access and different techniques	such as creating form, writing	
queries used in that.	T		
Credits: 2	Type of Cours	e: Core	
Max. Marks: 100	Min. Passing Marks: 36		
Suggestive Practical List			
• Learn and Understand the basic component of computers, software and Hardware.			
Do practice on MS-Wor	• Do practice on MS-Word and different formatting styles used in that.		
Do practice onMS-Excel	l and different techniques used in that		
• Do practice on MS-Power Point and formatting styles used in that.			
• Do practice and Acquire knowledge about MS-Access and different techniques such as creating			
form, writing queries used in that.			
Suggested Text Book Readings:			
1. Fundamental of Computers – By V.Rajaraman, B.P.B. Publications			
2. Fundamental of Computers – By P.K. Sinha			
3. Microsoft Office 2007 Bible – John Walkenbach, HerbTyson, FaitheWempen, caryN.Prague,			
MichaelR.groh, PeterG.Aitken, and Lisa a.Bucki -Wiley India pvt.ltd			
4. Discovering the Internet: Complete - Shelly Cashman 4th Edition - Course Technology			
Reference Books:			
1. Introduction to Information Technology - Alexis Leon, Mathews Leon, and Leena Leon, Vijay Nicole			
Imprints Pvt. Ltd., 2013.			
2. Office 2007 – By Snelly, Cengage Publication			
Electronic media and other d	Nome of electronic media: a SLM		
Name of electronic media: e-SI	LIVI Yea	r of incorporation: 2021	

Course prerequisites: 10+2 with Computer science, Mathematics			
Programme:B.	Sc. Year:1 Semester:2		
	Subject: Computer science		
Course Code:	UGCS-102N Course Title: C Programming		
Course Object	tives:		
To describ	e & understand the problem solving techniques.		
To underst	and the concept of basic terminology used in C programming.		
To develop	p programs in C language by writing, compiling and debugging.		
To develop	p programs involving simple statements, conditional statements, iterative statements,		
array, strin	gs, functions, recursion, structure and union.		
To different	ntiate between call by value and call by reference, acquire skills of using dynamic		
memory al	locations, use of pointers and basic operations on a file		
Course Outco	mes:		
CO1. Describe	e the functional components and fundamental concepts of a digital computer system		
including num	ber systems.		
CO2. Construc	t flowchart and write algorithms for solving basic problem		
CO3. Write C	programs that incorporate use of variables, operators and expressions along with data		
CO4 Write si	mple programs using the basic elements like control statements functions, arrays and		
strings	inple programs using the basic elements like control statements, functions, arrays and		
CO5 Write ad	vanced programs using the concepts of pointers, structures, unions and enumerated data		
types	valied programs using the concepts of pointers, structures, unions and enumerated data		
Credits: 2	Type of Course: Core		
Max. Marks: 1	00 Min. Passing Marks: 36		
Block 1	Introduction to algorithms and program design		
<b>T</b> T <b>1</b> / <b>T</b>	Unit 1: Introduction to Algorithms		
Unit I	Problem solving techniques, Algorithm		
Unit II	I Pseudo-codes and Flowcharts		
	Tools of Algorithm, Pseudo codes, Flowchart		
Unit III	Program design principles		
	Introduction to computer programming, Program design principles, Programming		
	techniques, Program Errors		
Block 2	Introduction to the 'C' programming language		
Unit 1	Unit 1: Introduction		
	History of C Language, Structure of a 'C' program, Creating and Executing a 'C'		
	program		
Unit 2	Data Types in 'C'		
	Character Set of 'C' language, Tri graph characters, Tokens, Identifiers, Keywords,		
II	Constants, Data types, Variables		
Unit 5	Storage Classes		
	Scope and metime of variable, Storage classes, Automatic storage class, Register storage		
Unit 4	Input and Output Functions		
	Reading a single character. Writing a single character. Formatted Input-Output		
	Formatted Input Formatted Output		
Block 3:	Operator and Control Structures		
Unit 1:	Operators and Expressions		
	Arithmetic operators, Relational operators, Logical operators, Assignment operators,		

	Increment and decrement operators, Conditional operators, Bitwise operators, Special			
	operators, Operator Precedence and Associativity, Ivalue and rvalue, Type casting:			
	Promotion and Demotion of variable types			
Unit 2:	Decision Structures in 'C'			
	if statement, if else statement, nested if else statement, switch statement, goto statment			
Unit 3:	Loop Structures in 'C'			
	for statement, while statement, do while statement, break statement, continue statement			
Unit 4:	Arrays			
	One dimensional array, Two dimensional array, Multidimensional arrays, Strings, String			
	handling functions, Character functions			
Block 4	Advanced Features of C			
Unit 1:	Pointers			
	Pointers and Address (&) operator, Pointer declaration and Initialization, Indirection			
	operator, Pointer Arithmetic, Arrays and Pointers, Character strings and Pointers, Array			
	of Pointers, Pointer to Pointer			
Unit 2:	Functions			
	Functions, user-defined functions, categories of function, returning non-integer values,			
	function arguments, recursion, arrays as function arguments			
Unit 3:	Structures, Unions, enum and typedef			
	Structure definition, Structures within structures, Structures as function arguments,			
	Pointers to structures, Unions, Enumerated data type, Type definition			
Unit 4:	File and Memory Management in 'C'			
	Files, File Pointer Variable, Opening a file, Reading and writing to files, File Status			
	Functions, Random Access to files, Command Line Arguments, Memory management			
Unit 5:	Preprocessor Directives and Error reporting			
	Macro directives, Conditional directives, Control directives, Error reporting			
Suggested Rea	adings:			
SLM of Unive	ersity			
1. Kanetkar Y., "Let Us C", BPB Publications.				
2. E. Balagurusamy, Computer Concepts and Programming in C, McGraw Hill.				
3. Yashwant Kanetkar, "Working with C", BPB Publications.				
4. E. Balagurus	samy, "Programming in ANSI C", TMH.			
5. Reema Thareja, Computer Fundamentals and Programming in C, Oxford Publication.				
6. Jeri R. Hanly, Elliot B. Koffman, Problem Solving and Program Design in C, Pearson Education.				
7. Behrouz A.	Forouzan, Richard F. Gilberg, Thomson, Computer Science- A Structured Programming			
Approach Usin	Approach Using C, Cengage Learning.			
8. Schildt H., "C- The Complete Reference", McGraw-Hill.				

9. Goyal K. K. and Pandey H.M., Trouble Free C", University Science Press

10. Gottfried B., "Schaum's Outlines- Programming in C", McGraw-Hill Publications. Electronic media and other digital components in the curriculum:

Name of electronic media: e-SLM	Year of incorporation: 2021

Syllabus for [B.Sc.]: Subject: [Computer science]				
Course prerequisites: 10+2 with Computer science, Mathematics				
Programme: B.Sc.	Year:1	Semester:2		
Subject: Computer science				
Course Code: UGCS 102P (N)	Course Title: Pra	ctical Work Based on UGCS 102		
Course Objectives:				
1. To write, compile, debug and	execute programs in a C prog	ramming environment.		
2. To learn programs that incorpo	orate use of variables, operato	rs and expressions along with data types.		
3. To learn programs for solving	problems involving use of de	ecision control structures and loops.		
4. To learn programs that involve	e the use of arrays, structures	and user defined functions.		
5. To Write programs using file	handling operations.			
<b>Course Outcomes:</b>				
CO1. Write, compile, debug and	execute programs in a C prog	gramming environment.		
Credits:2	Type of	Course: Core		
Max. Marks: 100	Min. Passing Marks: 36			
Suggestive List of Practical				
• Program to implement condition	onal statements in C language.			
• Program to implement switch-o	case statement in C language			
• Program to implement looping	constructs inC language.			
• Program to perform basic input	t-output operations in C langu	lage.		
• Program to implement user def	ined functions in C language.			
• Program to implement recursiv	ve functions in C language.			
• Program to implement one-dir	nensional arrays in C languag	e.		
• Program to implement two-dim	• Program to implement two-dimensional arrays in C language.			
• Program to perform various operations on two-dimensional arrays in C language.				
• Program to implement multi-dimensional arrays in C language.				
• Program to implement string r	nanipulation functions in $C$ la	nguage.		
• Program to implement structur	• Program to implement structure in C language.			
• Program to implement union in C language.				
• Program to perform file handling operations in C language.				
Suggested Text Book Readings:				
SLM of University				
1. Kanetkar Y., "Let Us C", BPB Publications.				
2. E. Balagurusamy, Computer Concepts and Programming in C, McGraw Hill.				
3. Yashwant Kanetkar, "Working with C", BPB Publications.				
4. E. Balagurusamy, "Programming in ANSI C", TMH.				
5. Reema Thareja, Computer Fundamentals and Programming in C, Oxford Publication.				
6. Jeri R. Hanly, Elliot B. Koffman, Problem Solving and Program Design in C, Pearson Education.				
7. Behrouz A. Forouzan, Richard F. Gilberg, Thomson, Computer Science- A Structured Programming				
Approach Using C, Cengage Learning.				
8. Schildt H., "C- The Complete Reference", McGraw-Hill.				
9. Goyal K. K. and Pandey H.M., I rouble Free C <sup>**</sup> , University Science Press				
10. Gottfried B., "Schaum's Out	lines- Programming in C <sup>"</sup> , Mo	CGraw-Hill Publications.		
Electronic media and other digital components in the curriculum:				
ivame of electronic media: e-SLWi fear of incorporation: 2021				

Course prerequisites: 10+2 with Computer science, Mathematics				
Programme: B	Programme: B.Sc. Year: 1 Semester: 2			
U	Subject: Com	puter science		
Course Code:	SBSCS-02N Course Title: Pyth	on Programming		
<b>Course Objec</b>	tives:	× ×		
1. To acquire p	programming skills in core Python.			
2. To explore t	he use of data structures, strings, text	files, lists and dictionaries.		
3. To acquire 0	Object Oriented Skills in Python.			
4. To understa	and to solve the problems with Python	database, Python multithreading.		
5. To work wit	h Django framework, Numpy and oth	er libraries.		
Course Outco	mes:			
CO1. Understa	and and comprehend the Basics of Pyth	hon programming.		
CO2. Describe	and explain the use of the built-in dat	a structures list, sets, tuples and dictionary.		
CO3. Make us	e of functions, modules and its applica	ations.		
CO4. Demons	trate the principles of OOPs and ide	entify real-world applications using OOPs, files and		
exception hand	lling provided by Python.			
Credits: 4		Type of Course: Core		
Category of Co	ourse	Value-added / employability/		
Max. Marks: 1	00 Min. Passing Mark	(8: 36		
BIOCK I	BASICS OF PYTHON			
Unit I	Unit I UNIT – 1: Introduction: History of Python, Need of Python Programming, Applications Unit I Basics of Python Programming Using the REPL(Shell), Running Python Scripts, Python IDLE.			
Unit II	<b>Tokens and Statements</b> : Variables, Constants, Assignment, Multiple Assignment, Keywords, Punctuators, Identifiers, Input-Output, Indentation, Statements, Comments, Single Comment and Multiline Comment			
Unit III	Unit IIIData Types, Operators & Expressions: Types – Integers, Strings, Booleans; Operators- Arithmetic Operators, Comparison (Relational) Operators, Assignment Operators, Logical Operators, Bitwise Operators, Membership Operators, Identity Operators, Operators precedence, Expressions and order of evaluations Control Flow- if, if-else, if- elif.else for while break continue pass			
Block 2	DATA STRUCTURE IN PYTHO	Ň		
Unit IV	Unit IV <b>Data Structures:</b> Stack & Queue, Lists – Operations, Slicing, Methods; Tuples – Operations, Methods , Sets– Operations , Methods, Dictionaries– Operations , Methods, Sequences– Operations, Methods. Comprehensions– Operations , Methods.			
Unit V	Init VFunctions – Defining Functions, Calling Functions, Passing Arguments, Keyword Arguments, Default Arguments, Variable-length arguments, Anonymous Functions, Fruitful Functions (Function Returning Values), Scope of the Variables in a Function- Global and Local Variables			
Unit VI <b>Modules &amp; Packages</b> : Modules: Creating modules, import statement, from. Import statement, name spacing, Python packages, Introduction to PIP, Installing Packages via PIP, Using Python Packages.				
BLOCK 3	0	OPS IN PYTHON		
Unit VII	UNIT – 7: Object-Oriented Prog	ramming OOP in Python: Classes, ' self-variable'.		
	Methods, Constructor Method, Inher	ritance, Overriding Methods, Data hiding.		
Unit VIII	<b>UNIT – 8: Exception Handling :</b> E	rror, and Exceptions: Difference between an error and		

	Exception, Handling Exception, try except for block, Raising Exceptions, User Defined			
	Exceptions			
Unit IX	UNIT - 9: Python Libraries: Brief Tour of the Standard Library - Operating System			
	Interface – String Pattern Matching, Mathematic	cs, Internet Access, Dates and Times,		
	Data Compression			
Unit X	UNIT - 10: GUI Programming and Testin	ng : Multithreading, GUI Programming,		
	Turtle Graphics Testing: Why testing is required	d ?, Basic concepts of testing, Unit testing		
	in Python, Writing Test cases, Running Tests.			
BLOCK 4:	MACHINE LEARNIN	NG IN PYTHON		
Unit XI	UNIT – 11: Machine Learning Using Python : Machine Learning Basics, Features and			
	Labels, Supervised and Unsupervised Learning.			
Unit XII	UNIT - 12: Regression and Classification in Machine Learning: Simple Linear			
	Regression, Multiple Regression, Data Collection for Machine Learning, Classification –			
	Features and Types			
Suggested Tex	Suggested Text Book Readings:			
1. Kenneth A.	1. Kenneth A. Lambert, Martin, Juneja "Fundamentals of Python", Cengage Learning.			
2. Python Programming: A Modern Approach, Vamsi Kurama, Pearson.				
3. Learning Python, Mark Lutz, Orielly.				
4. Harsh Bhasi	n, "Python for Beginners", New Age Internationa	1.		
5. Ashok Name	dev Kamthane, Programming and Problem Solvir	ng with Python, TMH.		
6. Allen Downey, Learning with Python, Dreamtech.				
Suggestive digital platforms web links-				
http://docs.python.org/3/tutorial/index.html				
http://interactivepython.org/courselib/static/pythonds				
http://www.ibiblio.org/g2swap/byteofpython/read/				
Electronic media and other digital components in the curriculum:				
Name of elect	ronic media: e-SLM	Year of incorporation: 2021		

Course prerequisites: 10+2 with Computer science, Mathematics			
Programme: B.Sc. Year:2 Semester:3			
Subject: Computer science			
Course Code:	UGCS -103N	Course Title: Data	a Structures
Course Objec	tives:		
1. To develop	the understanding	of data structures, th	heir types and applications.
2. To familiar	ze with concepts	of algorithm and con	mplexity.
3. To impleme	ent, analyze variou	is data-structures as	array, linked-list, stack, queue, tree, graph, etc. in
detail and utiliz	zation of data stru	cture techniques in p	problem solving.
4. To develop	the understanding	g of various sorting a	and searching techniques.
Course Outco	mes:		
COL Explain	he concept of data	a structure, abstract	data types, algorithms and analysis of algorithms.
Linked lists on	basic data organi	list	as arrays and linked lists, implementation of
CO3 Describe	stacks and queue	-1181,	and implement various operations on them using
arrays and link	ed lists	s, then applications	and implement various operations on them using
CO4 Describe	the properties of	trees and graphs and	implement various operations such as searching
and traversal o	n them	diees and graphs and	i implement various operations such as searching
Credits: 2			Type of Course: Core
Category of Co	ourse		Value-added / employability/
Max. Marks: 1	00	Min. Passing Mark	ks: 36
Block 1			
	UNIT 1: Introd	luction to data stru	cture
Unit I	Algorithm, Basic criteria for algorithms, Data type, Data structure, Data representation,		
	linear and non li	near data structure.	
Unit II	UNIT 2: Basics of algorithm		
	Algorithm, Basi	cs of complexity of	algorithm
Unit III	UNIT 3:Array		
	Definition, Rep	presentation of arr	ay, Single and multi-dimensional array, address
	calculation (one	dimensional, two di	imensional, multidimensional), sparse matrices
Block 2	Block 2		
Unit IV	UNIT 4: Stack	entions on staals Ar	max representation and implementation of stadly infin
	prefix and po	stfix representation	and implementation of stack, mills,
	Application of s	tacks	I of expression and evaluation multiple stacks,
Unit V	<b>INIT 5: Recur</b>	sion	
Cint V	Recursive defin	ition and processes	s, some named problems of recursion, principle of
	recursion: design	ning recursive algori	ithm, how recursion works, tail recursion.
Unit VI	UNIT 6: Queue	<u> </u>	· · · · · · · · · · · · · · · · · · ·
	Definition, operation on queues, circular queue, dequeue, priority queue, Application of		
queue.			
BLOCK 3			
Unit VII	UNIT 7: Linke	d List	
	Representation a	and implementation	of single linked list, Operations in the singly linked
	list, stack and q	ueue as a linked lis	st, circularly linked list, doubly linked list, circularly
doubly linked list, Application of linked list: polynomial representation and addition,			
	garbage collection	on	
Unit VIII	UNIT 8: Tree		

	Basic terminology, binary tree, binary tree representation, complete binary tree, extended		
	binary tree, array and linked list representations, traversing binary tree, threaded binary		
	tree, binary search tree, Operations on BST, AVL tree, Operations on AVL tree, B-tree		
	Insertion and deletion in B tree.		
Unit IX	UNIT 9: Graph		
	Basic terminology Graph representation Depth first search, breadth first search,		
	topological sort, connected components, spanning tree, minimum cost spanning tree,		
	Kruskal's and prim's algorithm, Shortest path algorithms: Bellman Ford Algorithm,		
	Dijkstra's algorithm, Floyd-Warshall algorithm.		
BLOCK - 4			
Unit X	UNIT 10: Searching and sorting		
	Sequential search, binary search, comparison and analysis, Selection sort, Bubble sort,		
	Insertion sort, Heap sort, Quick Sort, Merge sort, Shell sort, radix sort.		
Unit XI	UNIT 11: Hashing		
	Hash table, hash function, collision resolution strategies, hash table implementation		
Unit XII	UNIT 12: File Structure		
	Terminology, File organization, Sequential files, Direct File organization, Indexed		
	Sequential file organization		
Suggested Te	xt Book Readings:		
1. Y. Langsam	, M. Augenstin and A. Tannenbaum, Data Structures using C and C++, Pearson Education		
Asia.			
2. Ellis Horow	vitz, S. Sahni, D. Mehta Fundamentals of Data Structures in C++, Galgotia Book Source,		
New Delhi.			
3. S. Lipschutz	z, Data Structures Mc-Graw Hill International.		
4. Data Structures: A Pseudocode Approach with C, Second Edition, Richard F. Gilberg, Behrouz A.			
Forouzan			
5. Jean-Paul Tremblay, Paul. G. Soresan, An introduction to Data Structures with Applications, Tata Mc-			
Graw Hill International Edition.			
6. A. Michael Berman, Data structures via C++, Oxford University Press.			
7. Thomas H. Cormen, Introduction to Algorithms, 3rd Edition (The MIT Press).			
8. M. Weiss, Data Structures and Algorithm Analysis in C++, Pearson Education.			
9. Tremblay & Sorenson, An Introduction to Data Structures with Applications, Mcgraw Hill.			
10. R.S. Salaria, Data Structures and Applications using C, Khanna Book Publishing.			
11. Samanta D., "Classic Data Structures", Prentice Hall India.			
12. Aho, Ullman and Hopcroft, "Design and Analysis of algorithms", Pearson Education.			
13. R. Neapoli	tan and K. Naimipour, "Foundations of Algorithms", Jones an Bartlett, Student edition.		
14. Reema Tha	reja, Data Structures using C, Oxford Univ. Press		
Suggestive dia	ital platforms woh links or onling sourse		

#### Suggestive digital platforms web links or online course-

https://www.oercommons.org/authoring/14873-data-structure/view

https://www.oercommons.org/courses/data-structure-and-algorithms

https://onlinecourses.swayam2.ac.in/cec19\_cs04/preview (online course)

### Electronic media and other digital components in the curriculum:

Choose any one or more than: e-SLM/ Other electronic and digital contents

Name of electronic media: e-SLM	Year of incorporation: 2021
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Course prerequisites: 10+2 with Computer science, Mathematics				
Programme: B.Sc. Vear: 2 Semester: 3rd				
	Subject: Computer science			
Course Code: UGCS -103(P)N	Course Title: Prac	tical Work Based	on UGCS 103	
Course Objectives:				
• To learn implementation	of various Data St	ructures.		
• To applying data structu	res in solving real l	ife problems using	C/Python.	
• To learn implementation	of various Data St	ructures.		
Course Outcomes:				
CO1. Write and execute program	is to implement diff	ferent searching alg	gorithms.	
CO2. Write and execute program	is to implement var	ious sorting algorit	hms	
CO3. Write and execute program	is to implement var	ious operations on	two-dimensional arrays	
CO4. Implement various opera	tions of Stacks an	d Queues using b	oth arrays and linked lists data	
structures.				
CO5. Implement graph algorithn	n to solve the proble	em of spanning tree	2	
Credits: 2		Type of Course	: Core	
Category of Course value-added / employability/				
Max. Marks: 100	Min. Passing Mar	ks: 36		
(	Practical Work Ba	ased on UGCS 103	3)	
• To implement addition and multiplication of two 2D arrays.				
• To transpose a 2D array.				
• To implement stack using array	/			
• To implement queue using arra	y.			
• To implement circular queue u	sing array.			
• To implement stack using link	ed list.			
• To implement queue using link	ked list.			
• To implement BFS using linke	ed list.			
• To implement DFS using linke	d list.			
• To implement Linear Search.				
• To implement Binary Search.				
• To implement Bubble Sorting.				
• To implement Selection Sorting.				
• To implement Insertion Sorting.				
• To implement Merge Sorting.				
• To implement Heap Sorting.				
• To implement Matrix Multiplication by Strassen's algorithm				
• Find Minimum Spanning Tree	Find Minimum Spanning Tree using Kruskal's Algorithm			
Electronic media and other digital components in the curriculum:				
Name of electronic media: e-SLM Year of incorporation: 2021				

Course prerequisites: 10+2 with Computer science, Mathematics			
Programme: B	.Sc. Year:2 Semester:3		
Subject: Computer science			
Course Code: SBSCS-01N Course Title: Discrete Mathematics			
Course Object	tives:		
To per	form operations on discrete structures such as sets, functions, relations.		
To app	bly mathematical arguments using logical connectives and quantifiers.		
To iden	ntify and prove properties of Algebraic Structures.		
To form	mulate and solve recurrences and recursive functions.		
<ul> <li>To app</li> </ul>	ly the concept of combinatorics to solve basic problems in discrete mathematics.		
Course Outco	mes:		
CO1. Use math	nematical and logical notation to define and formally reason about basic discrete		
structures such	as Sets, Relations and Function		
CO2. Apply m	athematical arguments using logical connectives and quantifiers to check the validity of		
an argument th	rough truth tables and propositional and predicate logic		
CO3. Identify a	and prove properties of Algebraic Structures like Groups, Rings and Fields		
CO4. Formulat	e and solve recurrences and recursive functions		
CO5. Apply th	e concept of combinatorics to solve basic problems in discrete mathematics		
Credits: 4	Type of Course: Core		
Category of Co	Skill development		
Max. Marks: 1	00 Min. Passing Marks: 36		
Block I	Language of Mathematics and its application		
Unit I	Mathematical Logic: statements, operations, truth values, tautology and quantifiers.		
Unit II	<b>Arguments:</b> Rule of Detachment, Validity of a compound statement by using 1ruth		
	Lable, Validity using Simplification Methods, Validity using Rules of Inference,		
Unit III	<b>Bealeen Algebras</b> Declean Algebras Dringinle of Duclity Jeamorphic Declean Algebras		
Unit III	Boolean Algebra: Boolean Algebra, Principle of Duality, Isomorphic Boolean Algebras,		
	Normal Form Minimization of Boolean Functions (Karnaugh Man)		
	Switching circuits and logical Circuits: Switching Circuits Simplification of circuit		
	Non-Series Parallel Circuits Relay Circuits Logic Circuits		
Block 2	Set theory and its application		
Unit I	Set theory: sets, Subsets, Operations on Sets, Complementation, Intersection and Union.		
	Laws Relating Operations. Distributive Laws and De Morgan's Laws		
Unit II	<b>Relation:</b> Relation, binary relations in a Set, Domain and Range of a Relation, Total		
	number of Distinct Relations, Relations as Sets of Ordered Pairs Types of Relations,		
	Composition of Relations Equivalence relation in a set Partition of a Set Equivalence		
	Class and Quotient set of a set		
Unit III	<b>Partitions and Distributions:</b> Equivalence Relations, Equivalence Classes, Properties of		
	Equivalence Classes, Ouotient set and Partition.		
Unit IV	<b>Function:</b> Functions, Direct and Inverse image. Inverse Functions. Operations on		
	Functions, Composite of functions, Types of Functions and Connection between		
	Equivalence relation and mapping.		
Block – 03	Counting Process		

Unit I	<b>Mathematical Induction:</b> Principle of Mathematical Induction, Second Principle of Induction and Well ordering property.
Unit II	Combinatorics: Basic counting principles, Principle of Disjunctive counting, Principle of
	Sequential counting and Ordered and Unordered Partitions.
Unit III	Permutation
Unit IV	Combination
Block – 04	Probability theory and application
Unit I	Binomial theorem: Binomial theorem, General term in a binomial expansion, Middle
	term in a binomial expansion and Binomial expansion for rational exponents
Unit II	Probability: Definition of Probability, Addition law for counting and Product law for
	counting.
Unit III	General Counting methods: General Counting method is the extension part of counting
	process. It discusses Sum and Product Rulesand The Pigeonhole Principle .
Unit IV	The Inclusion- Exclusion Principle: inclusion-exclusion principle, Alternative form of
	the inclusion-exclusion principle and Onto Functions.

Suggested Text Book Readings:

#### Suggested online links:

1. Kenneth H. Rosen, "Discrete Mathematics and Its Applications", McGraw Hill.

2. B. Kolman, R.C Busby and S.C Ross, "Discrete Mathematics Structures", Prentice Hall.

3. R.P Girimaldi, "Discrete and Combinatorial Mathematics", Addison Wesley.

4. Y.N. Singh, "Discrete Mathematical Structures", Wiley- India.

5. Swapankumar Sarkar, "A Textbook of Discrete Mathematics", S. Chand & Company PVT. LTD.V.

6. Krishnamurthy, "Combinatorics Theory & Application", East-West Press Pvt. Ltd., New Delhi.

7. Liptschutz, Seymour, "Discrete Mathematics", McGraw Hill.

8. J.P. Trembely&R.Manohar, "Discrete Mathematical Structure with application to Computer Science", McGraw Hill.

#### Electronic media and other digital components in the curriculum:

Choose any one or more than: e-SLM/ Other electronic and digital contents			
Name of electronic media: e-SLM	Year of incorporation: 2021		

Course prerequisites: 10+2 with Computer science, Mathematics			
Programme: B	B.Sc. Year: 2 Sem	ester:4th	
Subject: Computer science			
Course Code:	: UGCS -104N Course Title: Introduction to Database Manag	gement System)	
Course Objec	ectives:		
• To learn the models	the features of a database system and its application and compa	re various types of data	
• To constru	ruct an FR Model for a given problem and transform it into a relati	on database schema	
<ul> <li>To consult</li> <li>To formula</li> </ul>	ulate solution to a query problem using SQL Commands, relationa	l algebra, tuple calculus	
and domain	ain calculus.		
• To underst	stand the need of normalization and normalize a given relation to t	ne desired normal form.	
To underst	stand different approaches of transaction processing and concurren	cy control.	
Course Outco	comes:		
CO1. Describe	be the features of a database system and its application and compa	re various types of data	
CO2 Construc	uct an ER Model for a given problem and transform it into a relatio	n database schema	
CO3 Formula	late solution to a query problem using SOL Commands relation	l algebra tuple calculus	
and domain ca	calculus	i algeora, capie calculus	
CO4 Explain t	n the need of normalization and normalize a given relation to the determination $f(x)$	esired normal form	
CO5. Explain	n different approaches of transaction processing and concurrency of	ontrol.	
Credits: 2	Type of Course: Core		
Category of Co	Course employability/ skill devel	opment/	
Max. Marks: 1	100 Min. Passing Marks: 36		
Block 1			
	UNIT 1: Overview of database management		
·· · ·	Introduction, Traditional oriented approach, Three view of data	. The three level	
Unit I	Unit I architecture of DBMS, DDL, DML, data model schemas and instances. Advantage and		
	disadvantage of database management system.	ý 8	
Unit II	UNIT 2: Database Models and implementation		
	Introduction, file management system, entity relationship mod	el, the hierarchical model,	
	network model. the relational model		
Unit III	. UNIT 3: Entity relationship model		
	ER model concept, notations of ER diagram, mapping constrai	nts, keys, concept of super	
	key, candidate key, primary key, verbalization, aggregation re	duction oF ER diagram to	
	tables, extended ER model, relationship of higher degree	Ū.	
Block 2			
Unit IV	Unit IV UNIT 4: Relational Model		
	Concept, Formal definition of a relation, entirety. constraint, entity indignity, referential		
	integrity, keys constraints, domain constraint, relational algebra	relational calculus.	
Unit V	UNIT 5: SQL		
	Characteristic of SQL, Advantage of SQL, SQL data type	s, SQL commands, SQL	
operators, insertion, update and delete operations, joins, unions intersection, minus,			
	views, queries and sub queries, aggregate function, cursor in SC	)L	
Unit VI	UNIT 6: Database Design		

	Conceptual, logical and physical design, Functional dependencies, normal form, first, second third BCNE multi-valued dependencies fourth normal form join dependencies		
	fifth normal form, inclusion dependencies, lossless join decomposition, normalization		
	using FD MVD and IDs		
BLOCK - 3			
	UNIT 7: File Organization		
	Introduction, file organization, sequential file organ	ization, index-sequential file	
	organization, direct file organization, multi key file	organization.	
	UNIT 8: Transaction Processing Concept		
	Transaction system, testing of serializability, ser	ializability of schedules, conflict and	
	view serializable schedule, recoverability, recover	ry from transaction failure, Joe based	
	recovery, checkpoints, deadlock handling, concept	of concurrency.	
Suggested Tex	t Book Readings:		
1. Date, C.J., "An Introduction to Database Systems", Narosa Publishing House, New Delhi.			
2. Korth, Silbertz, Sudarshan, "Database Concepts", Tata Mcgraw-hill Education (India).			
3. Elmasri, Navathe, "Fundamentals Of Database Systems", Pearson Education New Delhi India.			
4. G.K. Gupta, "Database Management System", Tata Mcgraw-hill Education (India) Pvt. Ltd.			
5. Bipin C. De	5. Bipin C. Desai, "An introduction to Database Systems", Galgotia Publication.		
6. Majumdar& Bhattacharya, "Database Management System", Tata Mcgraw-hill Education.			
7. Ramakrishnan, Gehrke, "Database Management System", McGraw Hill (India) Pvt Ltd.			
8. Chakravarti, "Advanced Database Mnagement System" Wiley Dreamtech Publications.			
9. Ullman, J.D., "Principles of Database Systems", Galgotia Publications, New Delhi.			
10. James Mortin- Principles of Database Management Object Oriented Modeling & Design.			
Suggestive digital platforms web links			
Electronic media and other digital components in the curriculum:			
Name of electr	ronic media: e-SLM	Year of incorporation: 2021	

Course prerequisites: 10+2 with Computer science, Mathematics			
Programme: B.Sc.	Year: 2	Sei	mester:4th
	Subject: Comp	outer science	
Course Code: UGCS -104(P)N Course Title: Practical Work Based on UGCS104 (Introduction to			
	Database Managen	nent System)	
<b>Course Objectives:</b>			
• To write SQL commands to q	uery a database.		
• To develop database and write	ting queries using M	ySQL, SQL Server.	
• To write, debug and impleme	ent SQL programs in	MySQL, SQL Server.	
• To learn programming in SQI			
<b>Course Outcomes:</b>			
CO1. Describe the features of a	database system and	l its application and comp	pare various types of data
models.			
CO2. Construct an ER Model for	a given problem an	d transform it into a relati	ion database schema.
CO3. Formulate solution to a qu	ery problem using s	SQL Commands, relation	al algebra, tuple calculus
and domain calculus.			
CO4. Explain the need of normal	lization and normaliz	ze a given relation to the o	desired normal form
CO5. Explain different approach	es of transaction pro	cessing and concurrency	control.
Credits: 2		Type of Course: Core	
Category of Course		Employability/ skill dev	elopment/
Max. Marks: 100	Min. Passing Mark	is: 36	
Suggestive Practical's			
1. Installing SQL Server/MYS	QL.		
2. Creating Entity-Relationship	Diagram using case	tools.	
3. Writing basic SQL statement	ts.		
4. Restricting and sorting data.			
5. Displaying data from multipl	le tables.		
6. Aggregating data using group	p function.		
7. Manipulating data.	0		
<ol> <li>Creating and managing table</li> <li>Normalization</li> </ol>	5.		
10 Creating procedure and function	tions etc		
11 Design and implementation	of Payroll processing	rsystem	
12 Design and implementation of	of Library Information	on System	
12. Design and implementation of Student Information System.			
14. Automatic Backup of Files and Recovery of Files.			
Suggested Text Book Readings:			
1. Date. C.J., "An Introduction to Database Systems", Narosa Publishing House, New Delhi.			
2. Korth, Silbertz, Sudarshan, "Database Concepts", Tata Mcgraw-hill Education (India).			
3. Elmasri, Navathe, "Fundamentals Of Database Systems", Pearson Education New Delhi India.			
4. G.K. Gupta, "Database Management System", Tata Mcgraw-hill Education (India) Pvt. Ltd.			
5. Bipin C. Desai, "An introduction to Database Systems", Galgotia Publication.			
6. Majumdar& Bhattacharya, "D	atabase Managemen	t System", Tata Mcgraw-	hill Education.
7. James Mortin- Principles of Database Management Object Oriented Modeling & Design.			

Suggestive digital platforms web links			
Name of electronic media: e-SLM	Year of incorporation: 2021		

Course prerequisites: 10+2 with Computer science, Mathematics				
Programme: B	Sc. Year: 3 Semester:5th			
Subject: Computer science				
Course Code: DCECS -105N Course Title: Computer Network				
Course Objectives:				
• To understand the computer networks and concentrates on building a firm foundation				
• To provide the fundamental knowledge of the various aspects of computer networking				
• To understanding the OSI Reference Model and TCP/IP Model				
• To have a good knowledge of Reference Model Layers and associated protocols.				
• Analyze the requirements for a given organizational structure. Select the most appropriate				
networking	architecture and technologies and appreciate recent developments in the area.			
Course Outcomes:				
CO1. Understa	and computer network basics, network architecture, TCP/IP & OSI reference models and			
other concepts.				
CO2. Understa	and error handling concepts, data link protocols, flow-error control, multi-channel access			
protocols and Standards				
CO3. Describe network layer protocols, IP addressing, IPV4 addressing, Routing algorithms, etc.				
CO4.Describe functionality of Transport layer and related protocols; concept of QoS and techniques.				
CO5. Define various application layer protocols such as DNS, Electronic Mail, FTP, HTTP, Telnet and				
understand net	work security			
Credits: 2	Type of Course: Elective			
Category of Co	burse Employability/ skill development/			
Max. Marks: 100 Min. Passing Marks: 36				
Block 1         Computer Network Basics and Services				
Introduction to Computer Network				
Unit I	Computer networks, Network Hardware—Local Area networks, Metropolitan Area			
	networks, Wide Area networks, Wireless networks, Internetworks, Network Software:			
TT '/ TT	Protocol Hierarchies			
Unit II	OSI and TCP/IP Model			
	Design and Issue for layers, Interfaces and services, Connection oriented and Connection			
I Init III	Unit 2: The Developed Levent			
	Dhysical Layer Transmission modia twisted pair Dass hand and Drandhard accession			
	Physical Layer, Transmission media, twisted pair, Base band and Broadband coax			
Unit IV	ISDN and Switching Techniques:			
Ontry	MODEM ISDN services Switching Message Decket Circuit switching TDM and EDM			
	ATM X 25			
Block 2	Link Laver Issues and Access Protocols			
Unit V	Data Link Layer			
Data Link Layer. Error detection and Correction Protocols: Simpley Stop (				
	protocols. One bit sliding window protocol. Using Go-Back N. Flow control Slidir			
	Window Protocol. Channel Allocation Problem			
Unit VI	Multiple Access Protocol: ALOHA, CSMA protocol. Collision Free protocol. Polling.			
	FDM, TDM			

Unit VII	The Medium Access Sub Layer:			
	Framing, Static and Dynamic Channel Allocation in LANs and MANs, IEEE Standard			
	802.3, and Ethernet IEEE standard 802.4 and token Ring, IEEE Standard 802.5, Token			
	Bus			
Unit VIII	Network devices:			
	Hub, Bridges, Switch, Gateways, Routers.			
BLOCK - 3	IP Addressing and Routing Issues			
Unit IX	IP Protocol and Addressing:			
	Network layer design issue, IP Protocol, IP Addresses, subnets,			
Unit X	: Connection Management:			
	Internetworking, connectionless and connection oriented services, tunneling,			
	Fragmentation, Firewall, Internet Controls Protocols.			
Unit XI	: Routing in Network Layer:			
	Routing Algorithm, shortest path routing, F	looding, Flow-based routing, Broadcast		
	routing, Congestion Control Algorithm, Congestion control and prevention policies;			
Block 4:	Transport, Session, Presentation and Application Layer			
Unit XII:	Transport layer:			
	Transport layer connection management, flow control, error control, congestion control,			
	Establishing and releasing a connection, TCP service Model, TCP protocol			
Unit XIII:	Session and Presentation Layer:			
	Introduction to cryptography and data compression			
Unit XIV:	The Application Layer:			
	Network Security, Domain Name System, Email: Architecture and Services, Message			
	formats, Message transfer.			
Suggested Text Book Readings:				
1. Forouzen, "Data Communication and Networking", TMH A.S.				
2. Tanenbaum, Computer Networks, Pearson Education				
3. W. Stallings, Data and Computer Communication, Macmillan Press				
4. AnuranjanMisra, "Computer Networks", Acme Learning 5. G. Shanmugarathinam, "Essential of TCP/				
IP", Firewall Media				
Suggesuve digital platforms web links				
Choose any one or more then: a SI M/Other electronic and digital contents				
Nome of electronic modicy of SLM				
iname of electr	Tome media: e-SLM	rear of incorporation: 2021		
<b>Course prere</b>	quisites: 10+2 with Computer science, Mathematics			
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Programme: B	.Sc. Year: 3 Semester: 5th			
	Subject: Computer science			
Course Code: I	DCECS -106N Course Title: Operating System			
Course Objec	tives:			
To develop	p the understanding of the structure and functioning of Operating System.			
To learn at	bout Processes, Threads and Scheduling algorithms.			
• To underst	and the principles of concurrency and Deadlock.			
• To learn va	arious memory management schemes.			
• To study I/	O management and File systems.			
<b>Course Outco</b>	mes:			
CO1. Explain 1	main components, services, types and structure of Operating Systems.			
CO2. Apply th	e various algorithms and techniques to handle the various concurrency control issues.			
CO3. Compare	e and apply various CPU scheduling algorithms for process execution			
CO4. Identify	occurrence of deadlock and describe ways to handle it			
CO5.Explain a	nd apply various memory, I/O and disk management techniques.			
Credits: 2	Type of Course: Elective			
Category of Co	burse Employability/ skill development/			
Max. Marks: 1	00 Min. Passing Marks: 36			
Block 1				
	UNIT 1: Introduction			
Unit I	Basic definitions, Batch processing, Multi-programming. Time sharing, multiprocessing;			
	Structure and Functions of Operating System			
Unit II	UNIT 2: Process and thread			
	Process, Process states, State Transitions, Process Control Block, Context Switching,			
	concept of thread, comparison between process and thread, Thread model, thread usage,			
	implementing thread in kernel and user space.			
Unit III	UNIT 3: Process Scheduling			
	Scheduler, Scheduling criteria, Preemptive and non-preemptive scheduling, Process			
<b>XX 1</b> . <b>XX</b>	Scheduling, Process scheduling algorithms.			
Unit IV	UN11 4: Concurrent Process			
	Process Interaction, Shared Data and Critical Section, Mutual Exclusion,			
Dia da 2	Synchronization, Classical Problems of Synchronization, Semaphores, Monitors.			
BIOCK Z				
Unit v	UNIT 5: Deadlock condition for deadlock resource allocation graph			
	deadlock measure deadlock avoidance Denker's clearithm Deadlock detection			
	deadlock prevention, deadlock avoidance, Danker's algorithm, Deadlock detection,			
Unit VI	UNIT 6: Momory management			
	Address Binding Dynamic Loading and Linking Concepts Logical and Physical			
	Addresses Contiguous and non-contiguous memory allocation Paging Segmentation			
	Virtual Memory Demand Paging Page fault Page replacement algorithms thrashing			
Unit VII	UNIT 7: Secondary memory management:			
	Free Space management Disk Structure Disk Scheduling Formatting Swap space			
	Management			
Unit VIII	Case Study of UNIX.			
Suggested Text Book Readings:				
1. Abraham Sil	berschatz and Peter Baer Galvin, "Operating System Concepts". Addision-Wesley			

- 2. Milan Milankovic, "Operating Systems, Concepts and Design", Tata McGraw-Hill.
- 3. Harvey M Deital, "Operating Systems", Addison Wesley
- 4. Richard Peterson, "Linux: The Complete Reference", Osborne Tata McGraw-Hill.

5. SibsankarHalder and Alex A Aravind, "Operating Systems", Pearson Education

6. D M Dhamdhere, "Operating Systems : A Concept basedApproach", McGraw Hill.

7. Charles Crowley, "Operating Systems: A Design-Oriented Approach", Tata McGraw Hill Education".

8. Stuart E. Madnick& John J. Donovan.Operating Systems.McGraw Hill.

9. A. S. Tanenbaum, "Modern Operating Systems", PHI.

10. William Stallings, "Operating Systems – internals and design principles", PHI.

Suggestive digital platforms web links

Electronic media and other digital components in the curriculum:

Choose any one or more than: e-SLM/Other electronic and digital contents

Name of electronic media: e-SLM Year	of incorporation: 2021
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<b>Course prerec</b>	quisites: 10+2 wi	th Computer science, Mat	hematics	
Programme: B.	Sc.	Year: 2	Semester:4th	
		Subject: Computer so	ience	
Course Code: I	DCECS -107(P)N	Course Title: Practical W	ork Based on DCECS106N Operating System	
Course Object	ives:			
• To develop	the understanding	of the structure and functio	ning of Operating System.	
• To learn abo	out Processes, Thre	eads and Scheduling algorit	hms.	
To understand	nd the principles o	f concurrency and Deadloc	κ.	
• To learn var	ious memory man	agement schemes.		
• To study I/C	) management and	File systems.		
<b>Course Outcon</b>	mes:			
CO1. Explain n	nain components,	services, types and structure	e of Operating Systems.	
CO2. Apply the	e various algorithm	ns and techniques to handle	the various concurrency control issues.	
CO3. Compare	and apply various	CPU scheduling algorithm	s for process execution	
CO4. Identify c	occurrence of dead	lock and describe ways to h	andle it	
CO5.Explain a	nd apply various m	emory, I/O and disk manag	gement techniques.	
Credits: 2		Туре	of Course: Core	
Max. Marks: 10	00	Min. Passing Marks: 36		
Block 1				
• Installing operating systems.				
	• Run the basics Linux commands.			
	• Writing code to implement scheduling Algorithams.			
• Try to learn perform the various memory management schemes				
Suggested Tex	t Book Readings:			
1. Abraham Siberschatz and Peter Baer Galvin, "Operating System Concepts", Addision-Wesley				
2. Milan Milan	kovic, "Operating	Systems, Concepts and Des	ign", Tata McGraw-Hill.	
3. Harvey M D	eital, "Operating S	ystems", Addison Wesley		
4. Richard Pete	rson, "Linux: The	Complete Reference", Osb	orne Tata McGraw-Hill.	
5. SibsankarHalder and Alex A Aravind, "Operating Systems", Pearson Education				
6. D M Dhamdhere, "Operating Systems : A Concept basedApproach", McGraw Hill.				
7. Charles Crowley, "Operating Systems: A Design-Oriented Approach", Tata McGraw Hill Education".				
8. Stuart E. Madnick& John J. Donovan.Operating Systems.McGraw Hill.				
9. A. S. Tanendaum, Modern Operating Systems', PHI.				
10. William Sta	10. winnam Stamings, Operating Systems – internais and design principles, PHI.			
Suggestive digital platforms web miks				
Chaose any one or more than, a SI M/Other electronic and digital contents				
Name of algor	Name of electronic media: e-SI M			
	onic media. e-sL	111	1 car of meorporation. 2021	

Course prerequisites: 10+2 with Computer science, Mathematics					
Programme: B.Sc.Year: 3Semester: 6th					
	Subject: Computer science				
Course Code: I	Course Code: DCECS -108N Course Title: C++ and Object Oriented Programming				
Course Objecti	ves:				
To unc	lerstand Object Oriented Concepts using C++ Language				
• To de	velop, debug and document programs in C++ using OOP paradigms.				
• Descr	ibe the meaning of the object-oriented paradigm and implement real-world entities like				
inherit	ance, hiding, polymorphism in programming using the object-oriented design process.				
Course Outcon	nes:				
CO1 To unders	stand Object Oriented Concepts using C++ and understand/implement C++ programming				
basics as data t	ypes, variable, constants, operators, control statements, arrays, etc.				
CO2. To unde	erstand and implement concepts of inheritance, hiding, polymorphism, interfaces and				
packages etc. 1	n C++ programming.				
Credits: 2	Type of Course: Elective       00     Min Descine Marke 26				
Max. Marks: 1	00 Min. Passing Marks: 36				
BIOCK I	LINUT 1. Dringinlag of object oriented programming				
	Object oriented programming periodism. Comparison with procedural programming				
Unit I	Basic concepts of object oriented programming benefits of OOP object oriented				
	Languages advantage of $C_{++}$				
Unit II	Languages, advantage of C++.				
	Class inheritance abstraction encapsulation and information hiding polymorphism				
	overloading				
Unit III	UNIT 3: Advanced concept				
	Dynamism (Dynamic typing., dynamic binding, late binding, dynamic loading).				
	Structuring programs, reusability, organizing object oriented project,				
Block 2	Overview of C++				
Unit IV	Tokens, keywords, identifiers and constants basic data types, user-defined and derived				
	Data types, type compatibility, reference, variables type Casting, operator precedence,				
control structures, structure, function.					
Unit V	Classes and objects				
	Class specification, class objects, accessing class members, scope resolution operator,				
	data hiding, empty classes, Pointers within a class, passing objects as arguments,				
	returning objects from functions, friend Functions and friend classes, constant parameters				
	and member functions, structures and Classes, static members.				
Unit VI	Object initialization and cleanup				
	Constructors destructor, constructor overloading. order of construction and destruction,				
	Constructors with default arguments, nameless objects, dynamic initialization through				
	constructors, constructors with appartmeters and destructors, constant objects and constructor,				
DLOCK 2	static Data members with constructors and destructors, nested classes				
<b>BLUCK - 5</b>	UNIT 8: Operator overloading and type conversion				
	Defining operator overloading overloading unery operators overloading binery				
	operators overloading binary operators using friends manipulation of strings using				
	Operators, rules for overloading operators, type, conversions				
Unit VIII	UNIT 9. Inheritance: extending classes				
	Deriving derived classes, single multilevel, multiple, hierarchical, hybrid inheritance.				

	Constructors & destructors in derived classes, constructors invocation and data members			
	Initialization, virtual base classes, abstract classes, delegation.			
BLOCK-4				
	UNIT 10: Pointers, virtual functions and polymor	phism		
	Pointers to objects, this pointer. pointers to derived classes, virtual functions,			
	Implementation of run-time polymorphism, pure virtu	al functions.		
	UNIT 11: Working with files			
	Classes for file stream operations. opening and cl	osing a file, file pointers and their		
	Manipulations, sequential input and output operations, error handling during file			
	Operations, command line arguments.			
	UNIT 12: Object Oriented Modeling			
	Need of object oriented Modeling, Simulation of real life problems using OOP concept:			
	Example, Representation of problem using object and class diagrams at design level.			
Suggested Text Book Readings:				
A.R.Venugopa	A.R.Venugopal, Rajkumar, T. Ravishanker "Mastering C++", TMH.			
R.Lafore, "Obj	R.Lafore, "Object Oriented Programming using C++", Galgotia Publications.			
E. Balagurusa	E. Balagurusamy, "Object Oriented Programming with C++", TMH.			
Suggestive digital platforms web links:				
Electronic media and other digital components in the curriculum:				
Choose any one or more than:e-SLM/Other electronic and digital contents				
Name of electr	Name of electronic media: e-SLMYear of incorporation: 2022			

Course prere	Course prerequisites: 10+2 with Computer science, Mathematics				
Programme: B	Sc. Year: 3	Semester: 6th			
	Subjec	Computer science			
Course Code: I	DCECS -109N Course Tit	e: Software Engineering			
Course Objecti	ves:				
1. To unde	rstand the methodologies invo	ved in the development and maintenance of software (i.e.)			
over the en	tire life cycle.				
2. To learn	about generic models of soft	vare development process.			
3. To unde	erstand methods of capturing, s	becifying, visualizing and analyzing software			
requirement	its and analysis modeling.				
4. To know	basics of testing and understa	nding concept of software quality assurance and software			
configurati	on management process.	in and their involution			
5.10 under	rstand the different design tech	niques and their implementation.			
o. To learn	various testing and maintenar	Ouslity Assurance plan and massures			
7. 10 unde	Istand Floject management an	Quality Assurance plan and measures.			
CO1 Understa	nd and comprehend the nature	of software development and software life cycle models			
CO2 Explain	needs for software specification	ns software requirements and their gathering techniques			
and their appli	ration	is, software requirements and then gathering teeninques			
CO3 Understa	nd and comprehend software of	uality assurance techniques			
CO4. Learn an	d implement concepts of software	are design modeling and principles			
CO5. Compare	understand and learn differen	testing strategies and factic			
CO6. Understa	nd. compare and apply various	software maintenance and management techniques			
Credits: 2		Type of Course: Elective			
Max. Marks: 1	00 Min. Passin	g Marks: 36			
Block 1					
	UNIT-I Software Engine	ring Fundamentals: Definition of Software, Software			
	characteristics, Software A	plications. Software Process: Software Process Models -			
Unit I	Waterfall model, prototypi	g model, spiral model, incremental model, concurrent			
	development model. Project	management Concepts: The Management Spectrum - The			
	People. The Product The Pro	ess. The Project.			
Unit II	UNIT-II Software Process	and Project Metrics : Measures Metrics and Indicators			
	Software measurement Size	Oriented Metrics Function - Oriented Metrics Extended			
	Function point metrics Software	Project Planning · Project Planning Objectives Software			
	Project Estimation Decon	position Tachniques Problem Resed Estimation Process			
	Project Estimation, Decon	Estimation Models The COCOMO Model Bisk Analysis			
	Based Estimation ,Empirica	Esumation Models- The COCOMO Model Risk Analysis			
	and Management: Software	isks, Risk identification, Risk Projection, Risk Refinement,			
	Risk Mitigation, Monitoring and Management.				
Unit III	UNIT-III Software Quali	y Assurance: Basic concepts- Quality, Quality Control,			
	Quality Assurance, Cost o	Quality, Software Quality Assurance (SQA), Formal			
	Technical Review Softwa	Revenue Alexandre Control Charles Control Charles Control Charles Control Charles Control Charles Control Charles Char			
	Configuration Items, The SC	VI Process, Version Control, Change Control, Configuration			
	Audit, Status Reporting. An	Tysis Concepts and Principles: Requirements Elicitation for			
	Software, Analysis Princip	tion Views Specification Pomain, Modeling, Partitioning,			
	Essential and Implement	auon views, specification: specification principles,			
Dlash 2	Representation, The Softwar	Requirement specification (SKS)			
BIOCK Z					

Unit IV	UNIT-IV Design Concepts and Principles: Des	ign Principles, Design Concepts —			
	Abstraction, Refinement, Modularity, Software	Architecture, Control Hierarchy,			
	Structural Partitioning, Data Structure. Software	Procedure, Structure, Information			
	Hiding, Effective Modular Design- Cohesion, C	Coupling Software Testing: Testing			
	Objectives & principles, Unit Testing, Integration	Testing (Top Down Integration,			
	Bottom. Up Integration, Regression Testing, Smoke	e Testing), Validation Testing (Alpha			
	and Beta Testing), System Testing (Recovery Test	ing, Security Testing, Stress Testing,			
	Performance Testing).				
Unit V	UNIT-V Reengineering: Software Reengineering,	, Reverse Engineering, Restructuring,			
	Forward Engineering CASE Tools: What is CA	SE, Building Blocks of CASE, A			
	Taxonomy of CASE Tools, Integrated CASE Enviro	onments, The integration Architecture,			
	The CASE Repository.				
Suggested Text Book Readings:					
R. S. Pressma	R. S. Pressman, Software Engineering: A Practitioners Approach, McGraw Hill.				
Rajib Mall, Fundamentals of Software Engineering, PHI Publication.					
K. K. Aggar	K. K. Aggarwal and Yogesh Singh, Software Engineering, New Age International Publishers.				
Pankaj Jalote	Pankaj Jalote, Software Engineering, Wiley.				
Deepak Jain, "Software Engineering: Principles and Practices", Oxford University Press.					
Munesh C. Trivedi, Software Engineering, Khanna Publishing House.					
N.S. Gill, Software Engineering, Khanna Publishing House.					
Suggestive digital platforms web links:					
Electronic media and other digital components in the curriculum:					
Name of electronic media: e-SLMYear of incorporation: 2022					

Course prerequisites: 10+2 with Computer science, Mathematics					
Programme: B	.Sc.	Year:3	Semester:6th		
Subject: Computer science					
Course Code: 1	DCECS -110(P)N	Course Title: Practical Work based on 1	.08		
Course Objec	tives:				
1. To write	e, compile, debug a	nd execute programs in a C programming	genvironment.		
2. To learn	programs that income	orporate use of variables, operators and ex-	pressions along with data		
types.	magazana fan aalu	ing mahlama involving use of desision as	entrol structures and loops		
5. To learn	programs that inv	by the use of arrays, structures and user	defined functions		
5 To Write	e programs using fi	le handling operations	defined functions.		
Course Outco	mes:	the number operations.			
CO1. Write. co	ompile, debug and e	execute programs in a C programming en	vironment		
CO2. Write pro	ograms that incorpo	prate use of variables, operators and expre	essions along with data types.		
CO3. Write pro	ograms for solving	problems involving use of decision contr	ol structures and loops.		
CO4. Write pro	ograms that involve	e the use of arrays, structures and user def	ined functions		
CO5. Write pro	ograms using file h	andling operations.			
Credits:2		<b>Type of Course:</b> E	lective		
Max. Marks: 1	00	Min. Passing Marks: 36			
Program to implement conditional statements in C++ language.					
	• Program to implement switch-case statement in C++ language				
	• Program to implement looping constructs inC++ language.				
	Program to perf	form basic input-output operations in C+4	- language.		
	Program to imp	lement user defined functions in C++ lan	guage.		
	Program to imp	plement recursive functions in C++ langu	age.		
TTAL	Program to imp	plement one-dimensional arrays in C++ la	anguage.		
Unit I	Program to imp	lement two-dimensional arrays in C++ la	nguage.		
	Program to perf	form various operations on two-dimension	nal arrays in C++ language.		
	• Program to imp	• Program to implement multi-dimensional arrays in C++ language.			
	Program to implemented by the second se	• Program to implement string manipulation functions in C++ language.			
	• Program to imp	lement structure in C++ language.	0 0		
	Program to imp	lement union in C++ language.			
	Program to per	form file handling operations in C ++lans	guage.		
Suggested Text Book Readings:					
A.R. Venugopal, Rajkumar, T. Ravishanker "Mastering C++", TMH.					
R.Lafore, "Object Oriented Programming using C++", Galgotia Publications.					
E. Balagurusamy, "Object Oriented Programming with C++", TMH.					

#### Year-2023-2024

## Syllabus of B.Sc. Programme<u>: [Subject Name: Environmental Science]</u> In accordance with NEP-2020

Year	Semeste	Course Code	Title of course	Theory/	Credits	
	r			Practical	Marks	
	Ι	UGEVS-101N	Fundaments of Environmental	Theory	2	100
1			Sciences			
		UGEVS-101N(P)	Practical work	Practical	2	100
		Skill Enhancemen	t Course			
		SBSEVS-01(N)	Energy Resources and Green	Theory	4	100
			Technology			
		UGEVS-102(N)	Ecology and Biodiversity	Theory	2	100
			Conservation			
1	П	UGEVS-102N(P)	Practical work	Practical	2	100
•		Skill Enhancemen	t Course			
		SBSEVS-02(N)	Environmental Impact Assessment	Theory	4	100
			and Legislation			
	III	UGEVS-103(N)	Environmental Microbiology and	Theory	2	100
2			Biotechnology			
		UGEVS-103N(P)	Practical work	Practical	2	100
2	<b>TN</b> 7	UGEVS-104N	Plant Physiology and Biochemistry	Theory	2	100
2	1 V	UGEVS-104N(P)	Practical Work		2	100
Discipl	ine Centri	c Elective Course				
		DCEVS-105N	Environmental Pollution	Theory	2	100
		DCEVS-106N	Remote Sensing, GIS and Hydrology	Theory	2	100
		DCEVS-107N(P)	Practical work based on 105 &106	Practical	2	100
Discipline Centric Elective Course						
		DCEVS-108N	Statistics and Environmental Quality	Theory	2	100
			Assessment			
		DCEVS-109N	Environmental Geology	Theory	2	100
		DCEVS-110N(P)	Practical Work based on 108 &109	Practical	2	100
	Total Credit /Marks				36	1600

Programme: B.Sc		Year: 1	Semester: I	
Subject: Environmental Sciences				
Course Code: UGEVS-101N Course Title: Fundaments of Environmental Sciences				
<b>Course Objective</b>	es:			
To underst	and basics of outline o	f environment.		
<ul><li>To learn al</li></ul>	bout nature and its beha	avior for living beings.		
To underst	and modern concept of	f environment.		
To underst	and basic concept of su	ustainable development.		
<b>Course Outcome</b>	s:			
CO 1: Gain know	ledge of Bhartiya Gyar	n Parampara about nature		
CO 2: Learn abou	it the concept of enviro	nment and its components		
CO3: Able to ana	lyze ambient environm	ent and their future prospects.		
CO4: Learn about	t origin of life and theo	ry of evaluation and natural sel	ection.	
CO5: Also learn	n about environmenta	al education and their imple	ementation for sustainable	
developme	ent			
Credits: 2		Type of Course: Core		
Max. Marks: 100		Min. Passing Marks: 36		
Block 1	<b>Environmental Hist</b>	ory and Evaluation		
	Vedic Concept of En	vironment:		
Unit I	Bhartiya gyan param	para aur bhartiya vaigyanik; 1	noral and aesthetic nature of	
	environmental science	e; objectives and historic roots	of the subject.	
Unit II	Modern Concept of Environment:			
	Definition, principles and scope of environmental science, environmental studies			
	and environmental technology, concept of environmental chemistry, biotechnology			
	and microbiology nee	d for public awareness.		
Unit III	<b>Evolution:</b>			
	Origin of life and spe	ciation, Darwinism and modern	synthetic theory of	
	evolution, natural sele	ection; biochemical basis of original	gin of life; Hardy -Weinberg	
	equilibrium; genetic c	lrift.		
Block 2	<b>Environmental Edu</b>	cation		
	Segment of Environ	ment:		
Unit IV	Atmosphere, hydros	phere, lithosphere, biosphere	and anthrosphere; factors	
	affecting environmen	t, natural and artificial environr	nent, biogeochemical cycle.	
Unit V	<b>Environmental Edu</b>	cation:		
	Definition and oppor	tunity of environmental educat	tion, environmental justice,	
	Environmentalism, environmental education at primary and secondary level.			
Unit VI	<b>Environmental Issue</b>	es:		
	Integration of enviror	nmental concerns, equality and	integrity, causes and types of	
	environmental issue	, local, regional and global	environmental issues and	
	challenges, solution for environmental issues.			
Block 3	Man and Environme	ental Sustainability		
Unit VII	Man and Environme	ent:		

	Population and density, natality and mortality, biotic potential and growth form of populations, man-environment relationships; impacts of human activity on environment.			
Unit VIII	Environment and Human Health:			
	Basic understanding between environment and human health, environmental pollution diseases, allergies, respiratory, cardiovascular, and cancer, personal			
	hygiene- food - balanced diet?			
Unit IX	Environmental Sustainability:			
	Concept of sustainability and sustainable development, social, environmental and			
	economic sustainability concepts, carrying capacity, challenges for sustainable			
development.				
C				

- 1. S.C. Sandra, "Environmental Science", A new Central Book Agency, 2008
- 2. P.D. Sharma, "Ecology and Environment" Rastogi Publications, 2017
- **3.** Neerj Nachiketa, Environment and Ecology: A Dynamic Approach, G.K. Publication Ltd, 2021
- 4. V. K. Ahluwalia, "Environmental Science, Ane Books India, 2013S.
- M.C. Dash, "Concepts of Environmental Management for Sustainable Develop Concepts of Environmental Management for Sustainable Development, I K International Publishing House Pvt. Ltd

#### Suggested online links:

- 1. Origin of Environmental Science From Vedas: <u>https://youtu.be/2MJb5JrLNpA</u>
- 2. Environment: Definition, Scopes and importance: https://youtu.be/L0uF8121F-0
- 3. Environmental Education: (316) Environmental Education | World and Indian Perspective | Environmental Geography | Dr. Krishnanand - YouTube
- 4. Human Health and the Environment: <u>Untitled Document (oecd.org)</u>
- 5. Global and local environmental sustainability, development and growth: <u>FINAL POST-2015</u> global and local environmental sustainability.pdf (oecd.org)

This course can be opted as an elective by the students of following subjects: NO

# Suggested equivalent online courses (MOOCs) for credit transfer: NO **Electronic media and other digital components in the curriculum:**

Choose any one or more than: (Electronic Media: Audio/Video Lectures, Online Counseling/Virtual Classes/E-Contents/e-SLM/OER/supplementary links for reference/Video Conferencing/Radio broadcast/Web Conferencing/ Other electronic and digital contents)

Name of electronic media:Year of incorporation: 2023-24

Programme: B.S	с.	Year: 2023-24	Semester: I	
Subject: Environmental Sciences				
Course Code	: SBSEVS-01N	Course Title: Energy Resources	and Green Technology	
<b>Course Objectiv</b>	ves:			
> To un	derstand the conce	pt of energy and its sources		
> To un	derstand fossil fuel	energy		
To lease	arn about biomass e	energy production		
➤ To kn	low about green tec	hnology		
Course Out	comes:			
CO1: Able to kn	own about structure	e and composition of sun.		
CO2: Learn abou	ıt solar energy			
CO3: Gain the ki	nowledge of fuel en	ergy resource		
CO4: Learn abou	it energy production	n by water and wind		
CO5: Able to kn	ow the concept of	green technology and green buildin	ıg	
Credits: 4		Type of Course: Skill Enhance	ment Course	
Max. Marks: 100		Min. Passing Marks: <b>36</b>		
Block 1	Solar and Fossil	Fuel		
	Sun as Source of	Energy:		
Unit I	Concept of energy	y, energy use from a historical per	spective, solar energy, solar	
	radiation, Solar sh	ell, status of solar energy in India.		
Unit II	Fossil Fuel:			
	Classification and composition of fossil fuel, physico-chemical properties of fuel,			
	origin, compositio	on and types of coal, origin, compo	sition and types of liquid fuel	
	(crude oil), classification of gaseous fossil fuels, gross calorific value and r			
	calorific value of	different fuels, oil and gas reservoi	rs and reserves.	
Unit III	Renewable Energy Resources:			
	Solar energy, hyd	ro energy or water power, wind e	nergy, ocean energy (ocean	
	tidal and wave ene	ergy and ocean thermal energy conv	ersion (OTEC)), geothermal	
	energy, Indian sce	enario of renewable energy consum	ption.	
Block 2	<b>Biomass Energy</b>	and Energy Polices		
	Biomass as ener	rgy source: Biomass resources,	dedicated bioenergy crops,	
Unit IV	characteristics of	bioenergy crops, bioenergy routes f	from biomass, conversion of	
	biomass into fuelschallenges in bioenergy utilization, biomass states energy i			
	India.			
Unit V	Other Source of Energy:			
	Conventional and nonconventional energy sources, nuclear fusion for energy,			
	Ethanol and metha	anol production, pyrolysis and sourc	es gasification, composition	
	of biogas, Urban v	waste to resource recovery and recy	cling for energy.	
Unit VI	<b>Energy Policies:</b>			
	Indian emission n	orms in transportation sector, natio	nal programmes to promote	
	biomass energy p	production in India, solar photovo	oltaic programmes in India,	

	energy resources available in India, urban and	rural energy consumption, national	
	green tribunal (NGT) act, NGT activities.		
Block 3	Energy Conservation and Green Energy		
Unit VII	Energy Conservation and Green Building:		
	Definition of energy conservation, need for e	energy conservation in India, benefits	
	of energy conservation, principles of energy conservation, government initiatives		
	for energy conservation; concepts, scope and	components of green building, green	
	buildings in India, certification of green build	ling.	
Unit VIII	Green Energy:		
	Aim and scope of green technology, con	cept of green energy and green	
	technology, biomass energy production, solar	and green battery technology; Fuel	
	cell technologies and application to waste-to-	energy conversion.	
Unit IX	Green Nanotechnology:		
	Understanding green tech, sectors using green	tech, green nanotechnology necessity	
	of green technology, categories of green te	chnology; environmental profits of	
	green building, economic benefits of green	building, goals of green technology,	
	limitations of green processes and technology	·. · · · · · · · · · · · · · · · · · ·	
Suggested Text	Book Readings:		
I. S.C. Bhatia	and R. K. Gupta, Textbook of Renewable Ener	gy", WPI Publishing-2019	
2. Renu, Dhu	2. Renu, Dhupper, "Textbook on Energy Resources and Management" CBS Publishers &		
Distributors	Distributors-2015		
3. Mahmood Z	Cohoori, Advantages and Disadvantages of Green Technology; Goals, Challenges and		
Strengths, Ir	iternational Journal of Science and Engineering Applications, ISSN-2319-7560		
<b>4.</b> G.D. Kai, No.	on conventional energy sources, Khanna publication.		
5. Sameer Sarkar, Fuel Technology, New Delhi, orient longman.			
Suggested online	e IIIKS:	(nood one)	
1. Energy from	arvetion Beneveble Energy: Introduction: (a	(need.org)	
2. Ellergy Coll 3. Penewsble I	Energy and Green Growth in India: Project Par	oortTemplate (teriin org)	
J. What is Bio	mass: https://woutu.be/DueE2df52IE	<u>jorrremplate (termi.org)</u>	
Fnergy sour	rces and Conversion Process - VouTube		
This course can	be opted as an elective by the students of fol	lowing subjects: NO	
Suggested equiv	Suggested equivalent online courses (MOOCs) for credit transfor		
1 Energy Resources and conversion processes - Course (swayam? ac in)			
Electronic media and other digital components in the curriculum.			
Electronic media and other digital components in the curriculum: Choose any one or more			
than: (Electronic Media: Audio/Video Lectures, Online Counseling/Virtual Classes/E-Contents/e-			
SLM/OER/supplementary links for reference/Video Conferencing/Radio broadcast/Web			
Conferencing/ Other electronic and digital contents)			
Name of electro	Iame of electronic media: Year of incorporation: 2023-24		
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Programme: B.Sc	•	Year: 2023-24	Semester: II
Subject: Environmental Sciences			
Course Code: UGEVS-102N		Course Title: Ecology and B	iodiversity Conservation
Course Objectives:			
To underst	tand about ecolog	y and this types	
<ul><li>To learn al</li></ul>	bout ecosystem a	nd its function.	
<ul><li>To learn al</li></ul>	bout biodiversity	and its conservation.	
<b>Course Outcome</b>	s:		
CO1: Able to kno	ow the concept of	ecology and their role in understand	ling of environment.
CO 2: Gain basic	understanding of	ecosystem and its function in nature	e for natural balance.
<b>CO3:</b> Ability to u	understand the ch	aracteristic of autecology and synec	ology.
<b>CO4:</b> Learn abou	t natural diversity	v, its types and role in nature.	
<b>CO5:</b> Also learn a	assessment of bio	diversity.	
Credits: 2		Type of Course: Core	
Max. Marks: 100	T	Min. Passing Marks: <b>36</b>	
Block 1	Ecology		
	Ecology and E	nvironment:	
Unit I	Concept and	Definition of ecology and envi	ronment, types of ecology;
	Environmental	factors (Abiotic and biotic), their in	mportance and role, ecological
	Hierarchy.		
Unit II	Autecology:		
	Population characteristics- dispersion, density, natality, mortality, age, structure,		
	population growth; human population & growth; ecological niche and habitat		
Unit III	Synecology:		
	Community sti	ructure, growth forms, concept o	t keystone species, ecotone,
	ecotypes, ecoph	ene, ecological indicators; ecologica	al succession.
Block 2	Ecosystem Components of Ecosystem		
TT •/ TT 7	Components of Ecosystem:		
Unit IV	Components, st	ructure and function of ecosystem;	properties of ecosystem, major
TT */ TT	ecosystems, typ	es of ecosystem in nature, terrestrial,	, aquatic ecosystem, and biome.
Unit V	I ropnic Levels		
	Energy flow in	ecosystem, food chain and food we	b, ecological pyramid, types
	of ecological pyramid, productivity.		
Unit VI	Energy-its Flow in Ecosystem:		
	Energy-defined in ecosystem, three sources of energy in ecosystem, Y-shaped		
Dlooly 2	Piediversity		
DIOCK 5	DIOUIVEISILY Introduction to Diadiversity		
	Introduction to Biodiversity:		
	Loss of biodiversity, factors affecting biodiversity, biodiversity betanets; betanets; in		
	India		
Unit VIII	Biodivorsity C	onservation.	
	Diouiversity C	UIISUI VALIUII.	

	Conservation of natural biodiversity-approaches and conventions, In-situ conservation, Ex-situ conservation; Role of local communities and traditional		
	knowledge in conservation.		
Unit IX	Biodiversity Assessment:		
	Identification of biodiversity, measuring biodiversity, biodiversity at local, national		
	and global levels, habitat destruction, fragmentation, transformation, degradation		
	and overexploitation, causes, impacts of pesticide and pollution on biodiversity.		

- 1. S.C. Sandra, "Environmental Science", A New Central Book Agency, 2008.
- 2. P.D. Sharma, "Ecology and Environment" Rastogi Publications, 2017
- 3. Neerj Nachiketa, Environment and Ecology: A Dynamic Approach, G.K. PublicationLtd, 2021
- **4.** A text Book of Environment Studies, Asthana, D. K. and Asthana, M. 2006, S. Chand & Co.-2010.
- **5.** Dr. Y. K. Singh, "Environmental Science" New Age International Private Limited-2006 **Suggested online links:**
- 1. Textbook for Environmental Studies, Erach Bharucha <u>https://www.ugc.ac.in/oldpdf/mo</u>delcurriculum/env.pd
- 2. Environmental Science, Tom Theis and Jonathan Tomkin, OpenStax CNX, National Digital Library of India.

http://ndl.iitkgp.ac.in/document/N2tzeE1aWWpUMm04b2l1VVZEdSsvK09RckF lSkE00WI3b1Flb 2ZTNHFxST0

- 3. Environmental Science, CEC EduSat, National Digital Library of India.
- 4. Biodiversity: <u>Chapter4.p65 (ugc.ac.in)</u>

This course can be opted as an elective by the students of following subjects: Anyone

Suggested equivalent online courses (MOOCs) for credit transfer;

1. Biodiversity and Ecological Resources - Course (swayam2.ac.in)

## Electronic media and other digital components in the curriculum:

**Choose any one or more than:**(Electronic Media: Audio/Video Lectures, Online Counseling/Virtual Classes/E-Contents/e-SLM/OER/supplementary links for reference/Video Conferencing/Radio broadcast/Web Conferencing/ Other electronic and digital contents)

Name of electronic media:

Year of incorporation: 2023-24

Programme: <b>B</b> .	.Sc.	Year: 2023-24	Semester: II
Subject: Environmental Sciences			
Course Code: SBSEVS-02N		Course Title: Environmental Impact Assessment and	
		Legislation	
Course Object	tives:		
To underst	tand the basic co	oncept of EIA	
To underst	tand methodolo	gy of data collection	
To learn al	bout environme	ntal impact and social impact	
To know e	environmental le	egislation and policy	
<b>Course Outcom</b>	mes:		
CO1: Learn the	e concept of EIA	A and why it is useful in environmenta	al clearance.
CO2: Learn the	e components of	EIA and its methodology to use	
CO3: Able to k	now the role of	EIA in water, soil and air analysis.	
CO4: Learn abo	out EIA regulat	ion of in India	
CO5: Also lear	n about environ	mental law and its implementation of	f conservation of nature.
Credits:4		Type of Course: Skill Enhancemen	t Course
Max. Marks: 1	00	Min. Passing Marks: <b>36</b>	
Block 1	<b>EIA Compone</b>	ents and Data Collections	
	Basic Concept	of EIA:	
TT '/ T	Definition, prin	nciple and objectives of EIA, need	for EIA, Types of EIA,
Unit I Hierarchy in EIA, Advantages of EIA, application form of EIA, compositi			orm of EIA, composition of
	expert committee for EIA process.		
Unit II	Components of EIA:		
	EIA process, screening, scoping, baseline data, impact prediction, assessment of		
	alternatives, delineation of mitigation measure and EIA report, public hearing.		
	decision making, monitoring, environmental clearance conditions.		
Unit III	Impact Assessment and Data Collections:		
	Environmental	impact, social impact, impact ide	entification and prediction,
	baseline data collection, construction stage impacts, post project impacts.		post project impacts.
Block 2	EIA Policies and Life Cycle Assessment (LCA)		
	EIA policies:	• · · ·	
Unit IV	EIA notificatio	ns, Government of India Ministry	of Environment and Forest
	Notification (2000), list of projects requiring, environmental clearance.		
Unit V	EIA Regulations in India:		
	Status of EIA in India; current issues in EIA: case study of hydropower projects/		
	thermal projects, salient features of 2006 amendments to EIA notification		
Unit VI	Life Cycle Assessment (LCA):		
	Life cycle anal	vsis, methodology, management, flo	w of materials-cost criteria-
	case studies, introduction to ISO 14000.		
Block 3	Environmental Management, Act and Polices		
Unit VII	Environmenta	I Management:	

	Environmental appraisal, environmental im-	pact statement (EIS), environmental	
	management plan (EMP), environmental audit; sustainable development.		
Unit VIII	Environmental Act:		
	Environmental laws and protection acts, exist	sting provision of central and state	
	government on environment protection, the E	Environment (protection) act (1986),	
	the water act (1974), the air act (1981), wild l	life act (1972).	
Unit IX	<b>Guidelines and Policies:</b>		
	Guidelines and policies for control of envir	onmental pollution, Environmental	
	Policy of India, solid and hazardous w	vaste management, handling and	
	management rules.		
Suggested Te	xt Book Readings:		
<b>1.</b> S.R. Khan	ideshwar, N.S. Raman and A.R. Gajbhiye, En	vironmental Impact Assessment,	
Dreamtech	1 Press-2019.		
2. Anjaneyu	lu Yerramilli, Environmental Impact Assessme	ent Methodologies, BS Publications-	
2020.			
<b>3.</b> George Al	ex, Environmental Impact Assessment (EIA),	Blue Rose Publishers-2020.	
<b>4.</b> Teacher_n	nanual_master_EIA.pdf (iitr.ac.in)		
5. N. Mahesh	wara Swamy, Text Book on Environmental L	aw, Asia Law House-2022	
Suggested on	line links:		
1. $(187)$ Even	cything About EIA - Environmental Impact As	sessment 2006 - Draft 2020 -	
<u>YouTube</u>	Tube		
2. $(187)$ Env	ironmental Impact Assessment   EIA Process	Its Components   Benefits of EIA	
Environme	<u>ental Sci - YouTube</u>		
3. Environme	ental Science II Environmental Assessment, M	lanagement & Legislation II UGC	
<u>NET II PA</u>	<u>PER-2 - YouTube</u>		
4. $(187)$ Lect	ure 13: EIA – Law, Policy and Institutional ar	rangements for EIA system -	
YouTube			
5. Environme	5. <u>Environmental Management - ISO 14000 - 20 Nov, 6 PM - YouTube</u>		
<u>(187) Environ</u>	<u>ment Law (પંયાવરण વિધિ) - YouTube</u>		
This course c	an be opted as an elective by the students of	following subjects: Any one	
Suggested eq	uivalent online courses (MOOCs) for credit	transfer:	
1. Environmental Impact Assessment - Course (nptel.ac.in)			
Electronic m	edia and other digital components in the	curriculum:	
Choose any	one or more than:(Electronic Media	a: Audio/Video Lectures, Online	
Counseling/V	irtual Classes/E-Contents/e-SLM/OER/supple	ementary links for reference/Video	
Conferencing/	Radio broadcast/Web Conferencing/ Other ele	ectronic and digital contents)	
Name of elec	tronic media:	Year of incorporation: 2023-24	

Programme: B.Sc.	Sc. Year: 2023-24 Semester: III		Semester: III
Subject: Environmental Sciences			
Course Code: UGEVS-103N Course Title: Environmental Microbiology and Biotechno		obiology and Biotechnology	
<b>Course Objective</b>	s:		
To und	erstand the micro	bial diversity and culture.	
To lear	n about role mic	robial degradation of pollutants.	
To kno	w about nucleic	acid and gene expression.	
Course Outcomes	5:		
CO1: Able to k	now the microbi	al world and their classification	
CO2: Able to k	now the role of	microbes in environment	
CO3: Learn the	e concept of mic	robial transformation	
CO4: Knowled	lge gain about na	ture of microbial degradation	
CO5: Learn ab	out the nucleic a	cid and protein and also the DNA tec	hnology and gene expression.
Credits:2		Type of Course: Core	
Max. Marks: 100		Min. Passing Marks: 36	
Block 1	Microbial Wo	rld	
	Introduction t	o Microbes:	
Unit I	Classification	of micro-organisms, and their nor	menclature, Whittaker's five
	kingdom class	ification system and their utility	, culture media, nutritional
	requirements an	nd growth characteristics of bacteria.	
Unit II	Microbial Transformation:		
	Concept of microbial transformation, accumulation and concentration of metals,		
	metal leaching, extraction; role of microbes in copper and uranium extraction, use		
	of bioreactors for bioremediation.		
Unit III	Microbial Degradation:		
	Microbes in waste decomposition, role of microbes in soil fertility, microbes in		
	agriculture, industry, medicine and wastewater treatment use, degradation of		
	xenobiotics in environment		
Block 2	Microbial Bioremediation:		
	Microbes in gaseous production:		
Unit IV	Bioreactors fo	r bioremediation, composting, bio	oventing, biogas production;
	methane, factor effecting methane, biodegradation of hydrocarbon.		
Unit V	Ecological Restoration and Bioremediation:		
	Bioremediation and phyotoremediation, specific bioremediation technologies for		
	water and land forming, biosparging, degradative pathways of plasmids,		
	hydrocarbons, pesticides, heavy metals and heavy metal tolerance in microbes.		
Unit VI	Ecologically Sa	afe Products and Processes:	
	Plant growth Promoting Rhizobacteria (PGPR): biofertilizers, microbial		
	insecticides and pesticides, bio-control of plant pathogen, Integrated pest		
	management; development of stress tolerant plants, biofuels; mining and metal		
	biotechnology.		
Block 3	Nucleic Acid a	nd Gene Expression	

Unit VII	Structure of Nucleic acid and Proteins: Nucleoside, nucleotide and nucleic acids,
	structural forms and characteristics of DNA and RNA, physical and chemical
	properties of nucleic acid.
Unit VIII	Recombinant DNA technology:
	Recombinant DNA: origin and current status; steps of preparation; restriction
	enzymes, polymerases, R-DNA technology in environmental management.
Unit IX	Gene Expression:
	Gene vs. Allele concept, quantitative genetics and multiple factors, inheritance and
	polygenic inheritance, sex chromosome structure and sex linked inheritance, sex
	linked diseases.

- 1. R.C. Dubey and D.K. Maaheshwari, A Textbook of Microbiology, S. Chand Publication-2013.
- 2. Ian L. Pepper, Charles P. Gerba, Terry J. Gentry, A Microbiology, Academic Press-2015.
- 3. K Vijaya Ramesh, Environmental Microbiology, MJP Publication-2019.
- **4.** P.K. Mahapathra, A Textbook of Environmental Microbiology, I K International Publishing House Pvt. Ltd-213.
- **5.** Gareth M. Evans Judith C. Furlon, Environmental Biotechnolog, Theory and Application, John Wiley & Sons Ltd, 2003.
- 6. R. K. Sinha, Environmental Biotechnology, Aavishkar Publishers & Distributors-2007

#### Suggested online link:

- 1. <u>Introduction of Microorganism || B.Sc-1st Year Botany (Paper-I) Microbiology || Prahalad</u> <u>bhaiya - YouTube</u>
- 2. <u>Microorganism, Bacteria, Algae, Fungi & Protozoa Chapter 2 Microorganisms: Friend and Foe YouTube</u>
- 3. <u>BIODEGRADATION OF PESTICIDES YouTube</u>
- 4. Bioremediation | Microbiology | Environmental Microbiology YouTube
- 5. <u>Biogas (Methane) Production Process, Applications, Advantages and Disadvantages -</u> <u>YouTube</u>
- 6. (185) Recombinant DNA technology (Genetic engineering) YouTube
- 7. (185) Biomolecule | Proteins & Nucleic Acid | L4 | NEET 2022/23 | Seep Pahuja YouTube

This course can be opted as an elective by the students of following subjects: NO

Suggested equivalent online courses (MOOCs) for credit transfer: NO

**Electronic media and other digital components in the curriculum: Choose any one or more than:**(Electronic Media: Audio/Video Lectures, Online Counseling/Virtual Classes/E-Contents/e-SLM/OER/supplementary links for reference/Video Conferencing/Radio broadcast/Web Conferencing/ Other electronic and digital contents)

Name of electronic media: Year of incorporation: 2023-24

Programme: B.S.	<b>c.</b> Year: <b>2023-24</b> Semester: <b>IV</b>		
Subject: Environmental Sciences			
Course Code: UGEVS-104N Course Title: Plant Physiology and Biochemistry			
Course Objectiv	Course Objectives:		
To un	derstand the plant stress and their its control		
To kn	ow about mineral and nitrogen fixation.		
To kn	ow about mechanism of photosynthesis and respiration in plant.		
Course O	utcomes:		
<b>CO 1:</b> At	ble to understand about plant cell absorption mechanism.		
<b>CO 2:</b> Le	earn the role of plant growth regulator.		
<b>CO 3:</b> Ga	ain the concept of nitrogen fixation.		
<b>CO 4:</b> Lo	earn about plant minerals and its transportation		
<b>CO 5:</b> Al	so learn about photosynthesis and respirator regulation.		
Credits: 2	Type of Course: Core		
Max. Marks: 100	Min. Passing Marks: 36		
Block 1	Plant Cell, Stress and Growth Regulator		
	Plant Cell and Transportation:		
Unit I	Plant cell, absorption, transportation of water, properties of solution, permeabil	lity,	
	imbibitions, osmosis, ascent of sap.		
Unit II	Plants Stress:		
	Abiotic and biotic stress; salinity, water stress, chilling, heat, pathogenesis, heavy		
	metals and their impact on plant growth and metabolism.		
Unit III	Plant Growth Regulator:		
	Phytohormones and its effect on plant growth and development, regulation of p	lant	
	morphogenetic processes by light, role plant growth hormone in agriculture.		
Block 2	Minerals and Nitrogen Fixation		
	Plant Minerals:		
Unit IV	Mineral elements in plants, types of plant nutrients, classification of mine	rals	
	nutrients, availability of micro and macronutrients, essential and non-essen	itial	
nutrients, common mineral diseases in plants.			
Unit V	Nitrogen Fixation and Assimilation:		
Nitrogen cycle, Biological nitrogen fixation by free living and in sy		otic	
	association, structure and function of enzyme nitrogenase.		
Unit VI	Nitrogen Metabolism:		
	Assimilation of nitrate, enzyme of nitrate reduction and their regulation and		
assimilation of ammonia into organic compounds.			
Block 3	Photosynthesis and Respiration:		
Unit VII	Photosynthesis:		
Chloroplast structure, photosynthetic apparatus, photosynthetic app		tus,	
	photosynthetic membranes and organelles, z scheme, light dependent reactions, Hil		

	reaction, generation of NADPH and ATP, Cyclic and non cyclic
	Photophosphorylation,
Unit VIII	Carbon Assimilation:
	Calvin cycle, and photorespiration, C4 cycle and CAM cycle, carbon cycle.
Unit IX	Respiration:
	Structure of mitochondria, Types of respiration, glycolysis, regulation of plant
	glycolysis, translocation of metabolites across mitochondrial membrane, TCA
	cycle.

- 1. A Textbook of Plant Physiology, Biochemistry And Biotechnology, S K Verma, Mohit Verma, S Chand Publication, 1995.
- 2. Fundamentals of Plant Physiology, by V. K. Jain , S Chand Publication, 2017.
- 3. Introduction to Plant Physiology, Norman P. A. Hüner , William G. Hopkins, Wiley publication, 2008.
- 4. Outline Of Plant Physiology, Robert M. Devlin, Medtech Publication, 2017.
- **5.** Physiochemical and Environmental Plant Physiology, Nobel, P. S. Academic Press; 4 edition, 2009.

#### Suggested online link:

- 1. <u>Temperature stress in plants | Stress Physiology in Plants YouTube</u>
- 2. <u>Mineral Nutrition in Plants Biological Nitrogen Fixation Nitrate Assimilation YouTube</u>
- 3. <u>Photosynthesis Non cyclic Photophosphorylation YouTube</u>
- 4. Photosynthesis Pigments YouTube.
- 5. <u>Plant Respiration | Biology | NEET 2020 | Ritu Rattewal YouTube</u>

This course can be opted as an elective by the students of following subjects: Any one

Suggested equivalent online courses (MOOCs) for credit transfer: NO

#### Electronic media and other digital components in the curriculum:

**Choose any one or more than:**(Electronic Media: Audio/Video Lectures, Online Counseling/Virtual Classes/E-Contents/e-SLM/OER/supplementary links for reference/Video Conferencing/Radio broadcast/Web Conferencing/ Other electronic and digital contents)

Name of electronic media: Year of incorporation: 2023-24

Programme: B.Sc.		Year: 2023-24	Semester: V
Subject: Environmental Sciences			
Course Code: DCEVS-105N Course Title: Environmental Pollutions		onmental Pollutions	
<b>Course Objectives:</b>			
To understa	nd about envi	ronmental pollution	
To learn wa	ter, air, soil a	nd radioactive pollution	
To learn ho	w to control e	nvironmental pollution.	
<b>Course Outcomes:</b>			
CO1: Gain the know	wledge of env	ironmental pollution its source and	l sink
CO 2: Learn the ph	ysiochemical	characteristic water and its effects	on living beings.
CO3: Learn how to	detect and co	ontrol the soil and air pollution.	
CO4: Learn about t	echniques use	ed in assessment of environmental	pollution
CO5: Learn how to	ecological ba	lance is necessary to control envir	onmental pollution.
Credits: 2		Type of Course: Discipline Cent	ric Elective Course
Max. Marks: 100		Min. Passing Marks: 36	
Block 1	Water and	Air Pollution	
	Introductio	n of Pollutant:	
Unit I	Definition of pollution and pollutants, source and sink of pollutants, classification of pollutants, difference between pollutants and contaminants.		
Unit II	Water Pollution:		
	Definition and sources of water pollution, types of water pollutants, effects of		
	water pollutants on river water and potable water in India, measure of water		
	pollution.		
Unit III	Air Pollution:		
	Definition and sources of air pollution, atmospheric composition, types of air		ospheric composition, types of air
	pollutants, acid rain, particulate matter, factors effecting air pollution, control		ors effecting air pollution, control
	measure of air pollution.		
Block 2	Soil, Noise a	and Radioactive Pollution	
	Soil pollution:		
Unit IV	Definition and sources of soil pollution, nature of soil pollutants, physiochemical		
	and biological properties of soil, factor effecting soil pollution, measure of soil		
	pollution.		
Unit V	Noise Pollution:		
	Definition of noise pollution, noise exposure level, effects of noise pollution,		level, effects of noise pollution,
	measure and control of noise pollution.		
Unit VI	Radioactive Pollution:		
	Definition and sources of radioactive pollution, sources of radiations, nuclear		
	pollution, bi	ological effects of radiations, contr	ol measures, radioactive pollution.
Block 3	Effects and	Control of Pollution	

Unit VII	Thermal Pollution:		
	Definition and sources of thermal pollution, causes of thermal pollution, control		
	and measure of thermal pollution.		
Unit VIII	Effects of Environmental Pollution:		
	Effects of heavy metal pollution on natural water and soil, metal toxicity and its		
	effects on human beings, effects of environmental pollution on agriculture.		
Unit IX	Pollution Control:		
	Techniques used in water treatment, oxidation ponds, fluidized bed reactors, air		
	samplers, sequencing batch reactor, bioscrubbers, biotrickling filters,		
	Afforestation.		

- 1. S.C. Sandra, "Environmental Science", A new Central Book Agency, 2008.
- 2. A.K. De, "Environmental Chemistry" Publisher: New Age Publisher International Pvt Ltd-2016.
- **3.** Balram Pani, "Textbook of Environmental Chemistry" I K International Publishing House Pvt. Ltd-2103.
- 4. A text Book of Environment Studies, Asthana, D. K. and Asthana, M. 2006, S. Chand & Co.-2010.
- 5. Dr. Y. K. Singh, "Environmental Science" New Age International Private Limited-2006.

#### Suggested on line link:

- 1. Textbook for Environmental Studies, Erach Bharucha https://www.ugc.ac.in/oldpdf/modelcurriculum/env.pd
- 2. Environmental Science, Tom Theis and Jonathan Tomkin, OpenStax CNX, National Digital Library of India.

http://ndl.iitkgp.ac.in/document/N2tzeE1aWWpUMm04b2l1VVZEdSsvK09RckFlSkE0OWI3b1Flb 2ZTNHFxST0

- **3.** Environmental Science, CEC EduSat, National Digital Library of India.
- 4. POLLUTION <u>https://youtu.be/kOGqRMwAC6U</u>
- 5. Pollution and its Control: (316) Pollution and its Control Environmental and Ecology | Crack UPSC CSE/IAS - YouTube

This course can be opted as an elective by the students of following subjects: NO

1. <u>Environmental Pollution and Global issues - Course (swayam2.ac.in)</u> https://onlinecourses.swayam2.ac.in/cec19\_cs06/preview

#### Electronic media and other digital components in the curriculum:

**Choose any one or more than:**(Electronic Media: Audio/Video Lectures, Online Counseling/Virtual Classes/E-Contents/e-SLM/OER/supplementary links for reference/Video Conferencing/Radio broadcast/Web Conferencing/ Other electronic and digital contents)

Name of electronic media:Year of incorporation: 2023-24

Programme: B.S	с.	Year: 2023-24	Semester: V
Subject: Environmental Sciences			
Course Code: DCEVS-106N Course Title: Remote Sensing, GIS and Hydrology			Sensing, GIS and Hydrology
Course Objectiv	Course Objectives:		
To un	derstand to Remote	e sending and GIS in environn	nental management.
To un	derstand the water	hydrology and it role in envire	onmental balance.
To lease	arn how to about hy	/drological process.	
Course Outcom	es:		
CO1: Able to kn	ow the principle of	remote sensing and tools used	d in remote sensing.
CO2: Learn abou	it the role GIS in de	etermination of real time data	sampling.
CO3: Known the	e concept of hydrol	ogical maintenance of environ	iment.
CO4: Learn abou	it hydrograph, and	hydrograph Analysis for wate	r resources
CO5: Also learn	about water harves	sting and food management sy	stem.
Credits: 2		<b>Type of Course: Discipline</b>	Centric Elective Course
Max. Marks: 100		Min. Passing Marks: 36	
Block 1	Concept and Ap	plication of Remote Sensing	
	Concept of Rem	ote Sensing:	
Unit I	Electromagnetic	radiation and atmospheric win	dow, principle of remote sensing,
	types of remote sensing, data acquisition and it applications		
Unit II	Geographical In	formation System (GPS):	
	Concept of GIS, Principles, Elements and its applications, GPS principle and		
	applications.		
Unit III	Application of R	emote Sensing:	
	Application of re	mote sensing in atmospheric	and ocean studies, climate change,
	forestry, and envi	ronment.	
Block 2	Concept and Hydrological Process		
	Concept of Hydrology:		
Unit IV	Hydrologic cy	cle, water availability.	water balance, precipitation,
	evapotranspiratio	n study by remote sensing.	
Unit V	Hydrological Process:		
	Water table, aqui	ifer, evaporation and transpiration	ation, Interflow, Gravity Drainage,
	Surface Runoff, Return flow, Recharge.		
Unit VI	Hydrograph Analysis:		
	Hyetograph, Runoff, drainage basin characteristics; Hydrograph concepts,		
	assumptions and limitations, Unit of hydrograph.		
Block 3	Flood Management and Water Harvesting		
Unit VII	Reservoir:		
	Types, site selection, zones of storage, safe yield, reservoir capacity, reservoir		
	sedimentation and	d control study.	
Unit VIII	Flood Managem	ent:	

	Types of floods, Causes of flooding, Alleviation, Levees and floodwalls,	
	Floodways, Channel improvement, Flood damage analysis by remote sensing.	
Unit IX	Water Harvesting:	
	Rainwater collection, runoff collection, ponds, tanks, natural and artificial ground	
	water recharge methods, agriculture rain water harvesting.	
Suggested Text Book Readings:		
1. M. Anji R	eddy, Text Book of Remote Sensing and Geographical Information Systems,	
Publicatio	ons/BSP Books-2012.	
2 Kali Char	an Sahu, Textbook of Remote Sensing and Geographical Information Systems	

- 2. Kali Charan Sahu, Textbook of Remote Sensing and Geographical Information Systems, Atlantic Publishers and Distributors (P) Ltd-2022.
- 3. K. Subramanya, Engineering Hydrology, McGraw Hill Education, 2017.
- 4. Savindra Singh, Fundamentals of Hydrology, Pravalika,
- 5. R. N. Saxena, Elements of Hydrology and Groundwater, PHI Learning-2017.

## Suggested online link:

- 1. (185) Remote sensing in hindi | remote sensing and gis | lecture 1 YouTube
- 2. (185) GIS (geographic information systems) | introduction to gis | lecture 1 YouTube
- 3. (185) Application of remote sensing | remote sensing and gis | lecture 6 YouTube
- 4. (185) Hydrologic Processes YouTube
- 5. (185) Basics of Hydrograph Analysis and Uses Hydrograph Analysis GATE Hydrology -YouTube

(185) Hydrograph and Runoff - Hydrology - YouTube

This course can be opted as an elective by the students of following subjects: NO

Suggested equivalent online courses (MOOCs) for credit transfer:

1. <u>Basics of Remote sensing, GIS & GNSS technology and their applications - Course</u> (swayam2.ac.in)

&

2. <u>Surface Water Hydrology - Course (nptel.ac.in)</u>

## Electronic media and other digital components in the curriculum:

Choose any one or more than: (Electronic Media: Audio/Video Lectures, Online

Counseling/Virtual Classes/E-Contents/e-SLM/OER/supplementary links for reference/Video Conferencing/Radio broadcast/Web Conferencing/ Other electronic and digital contents)

Name of electronic media:Year of incorporation: 2023-24

Subject: Environmental Sciences     Course Code: DCEVS-108N   Course Title: Statistics and Environmental Quality     Assessment     Course Objectives:     > To learn graphical representation of environmental data
Course Code: DCEVS-108N   Course Title: Statistics and Environmental Quality     Assessment     Course Objectives:     To learn graphical representation of environmental data
Assessment     Course Objectives:     > To learn graphical representation of environmental data
Course Objectives:
To learn graphical representation of environmental data
To learn about data analysis techniques
To know about water and soil quality analysis
Course Outcomes:
<b>CO1:</b> Useful to known about environmental statistics and data management.
CO2: Learn about the diagram and graphs plot
<b>CO3:</b> Able to understand the probability, variable and standard deviation
<b>CO4:</b> Able to understand the analysis of variance.
<b>CO5:</b> Able to Learn about air water and soil quality assessment.
Credits: 2 Type of Course: Discipline Centric Elective Course
Max. Marks: 100 Min. Passing Marks: 36
Block 1     Graphical representation of Environmental data
Environmental Statistics:
Unit I Concept and scope of environmental statistics, role of statistics in environmental
data interpretation, environmental data accuracy and environmental quality
measurement, statistical tools
Unit II Data and Frequency:
Concept of data, types of data, grouped data, tools of data management, frequency,
frequency distribution, types of frequency distribution, frequency distribution table
and graphs.
Unit III Diagram and Graphs:
Graphical interpretation, simple diagram, multiple diagram, component bar
diagram, percentage bar diagram, Pie-diagram, histogram, frequency curve,
Dete Analysis Techniques
Block 2 Data Analysis Techniques
Unit IV Data Analysis 1001s:
Unit V Probability and Standard Distribution:
Difference of the standard distribution in the standard distribution is th
deviation and standard error testing of hypothesis
Unit VI Analysis of Variance:
Concept of statistical variance basic principles of one way and two way analysis
Block 3 Air Water and Soil Quality Assessment
Unit VII Air Quality Monitoring:
Composition of air air quality standard air sampling Particulate matter
qualitative analysis of SO <sub>2</sub> , and NO <sub>2</sub>
Unit VIII Water Quality Monitoring:

	Water quality analysis, water quality parameters study as BOD, COD, pH,
	turbidity and nitrate, salinity, test of <i>coli</i> forms, water quality standard in India.
Unit IX	Soil Quality Monitoring:
	Soil quality, organic matter determination, exchangeable calcium and magnesium,
	soil quality standard in India.

- 1. S.C. Sandra, "Environmental Science", A new Central Book Agency, 2008.
- **2.** A.K. De, "Environmental Chemistry" Publisher: New Age Publisher International Pvt Ltd-2016.
- **3.** Balram Pani, "Textbook of Environmental Chemistry" I K International Publishing House Pvt. Ltd-2103.
- **4.** Practical Statistics for Environmental & Biological Scientists, John Townened, John Wiley & Sons Inc publication.

## Suggested online link:

- 1. Environment Statistics: <u>Microsoft Word Brochure\_Environment\_2015 (un.org</u>)
- 2. Fundamentals of environment statistics: <u>Microsoft PowerPoint Session 2-1 Basic concepts</u> of environment statistics (UNSD).ppt
- 3. Mean, Median and Mode: <u>meanmedianmode (statstutor.ac.uk)</u>
- 4. Soil Quality and Methods for its Assessment: <u>Braimoh\_FM.indd (core.ac.uk)</u>
- 5. Water Quality Assessment Of Water Bodie: <u>Slide 1 (cpcb.nic.in)</u>
- 6. Air Quality Assessment: <u>Microsoft Word air\_quality\_technical\_report (nj.gov)</u>

This course can be opted as an elective by the students of following subjects: Anyone

## Suggested equivalent online courses (MOOCs) for credit transfer: NO

Electronic media and other digital components in the curriculum:

**Choose any one or more than:**(Electronic Media: Audio/Video Lectures, Online Counseling/Virtual Classes/E-Contents/e-SLM/OER/supplementary links for reference/Video Conferencing/Radio broadcast/Web Conferencing/ Other electronic and digital contents)

Name of electronic media:

Year of incorporation: 2023-24

Programme: B.S	rogramme: <b>B.Sc.</b> Year: 2023-24 Semester: VI		
	Subje	ct: Environmental Sciences	
Course Code: DCEVS-109N Course Title: Environmental Geology			Geology
Course Objectiv	/es:		
➤ To un	derstand the history of	of earth origin and their surface p	rocess.
➢ To un	derstand atmosphere	and its composition and heath ba	lance in nature.
To lease	arn how to rocks and	minerals are formed by natural particular	rocess
> To kn	ow how the mountain	and river are originate.	
Course Outcom	es:		
CO1: Useful to k	nown about earth for	mation, earth composition, atmos	sphere and hydrosphere.
CO2: Understand	ding the surface proce	ess of earth system that how the r	ocks, mineral are available
for living	being and how the ge	eographical formation are occurs	like mountain and river.
CO3: Understand	the formation of roc	ks and mineral, its composition,	and weathering process
and rock	cycling.		
CO4: Learn abou	it Plate tectonic in mo	$\frac{1}{1}$	earthquake generation.
COS: Able to une	derstand about geolog	Trans of Granner Dissipline Gran	and volcano
Credits: 2	•	Type of Course: Discipline Cer	atric Elective Course
Max. Marks: 100		Min. Passing Marks: <b>30</b>	
BIOCK I	Earth and Atmosp	nere	
I Init I	Concert of the certi	formation formation and comp	osition of some months and
Unit I	atmosphere and hyd	rosphere: chemical composition	of the earth
Unit II	Origin of Farth.	rosphere, enemiear composition	of the cartif.
	Geological time sca	le and major changes on the Far	th's surface. Holocene and the
	emergence of hum	ans. Concept of plate tectonics	and continental drift theory.
	gravitational and ma	agnetic fields of the earth.	
Unit III	Earth Atmosphere:		
	Atmosphere, evolution of earth's atmosphere, composition of atmosphere, physical		
	and optical propert	ies, earth's energy balance; ene	ergy transfers in atmosphere;
	earth's radiation budget.		
Block 2	Earth Surface Process and Rocks Formation		
	Earth Surface Processes:		
Unit IV	Circulation, interfa	v land surface processes fluxial	and glasial processes river
	and geomorphology		
Unit V	Minerals and Rock	cs Formation:	
	Composition of roc	ks and minerals rock cycle. lith	nification and metamorphism.
	Three rock laws ro	ck structure igneous sedimentar	v and metamorphic rocks
Unit VI	Weathering of Mir	erals and Rocks:	

	Physical, biogeochemical processes of weathering; physical processes of erosion, factors affecting erosion and agents of erosion; aeolian transportation and deposition of sediments by running water.		
Block 3 Mountain and River Origin and Natural Hazardous			
Unit VII	Mountain Origin:		
Plate tectonic in mountain formation, continental collision and format Himalaya; ocean floor spreading, formation of peninsular Indian mounta - western and eastern ghats. Vindhyas and Arayallis		mation of the intain systems	
Unit VIII	River Origin:		
	Perennial river systems and evolution of monsoon in Indian subcontin of Indo-Gangetic Plains, progression of agriculture in the Indian su Holocene	ent; formation ubcontinent in	
Unit IX	Natural Hazards		
	Geological hazards earthquakes and volcano characteristics of earth	auakes Types	
	of waves magnitude scales Richter scale volcanic feature type	s of volcanic	
	eruptions, active and inactive volcanoes.		
Suggested Text	t Book Readings:		
1. Savindra	a Singh, "Fundament of physical geography, Pravalika Publcations-202	2	
2. Savindra	2. Savindra Singh, Environmental Geography" Pravalika Publications-2019		
3. Thompso	3. Thompson and Turk, "Environmental Geoscience: Thomson Learning-1995		
4. Jeff Kelle	Jeff Keller, "Environmental Geology" Pearson-1999.		
5. Savindra	vindra Singh, Fundamentals of Hydrology, Pravalika - 2018		
6. Dorothy	6. Dorothy Merits, Environmental Geology: An Earth Systems Approach, W.H. Freeman-2014		
Suggested online link:			
<b>1.</b> Theories	<b>1.</b> Theories of Origin of Earth - Part 1 (Examrace - Dr. Manishika) - YouTube		
<b>2.</b> History o	2. History of the Earth: <u>History Of The Earth (Eolss.Net)</u>		
<b>3.</b> History o	of Earth: <u>History_Of_Earth - Wiki.Pdf (Content-Calpoly-Edu.S3.Amaz</u>	conaws.Com)	
<b>4.</b> Weathering	Veathering & Mass-Wasting Processes: Weathering, Erosion, and Mass-Wasting Processes		
<u>(cuny.edu</u>	<u>.edu)</u>		
5. The Evolu	Evolution of Mountain: <u>D:WPŸSGSG3ERTHHIST.99.wpd (jmu.edu)</u>		
This course can	This course can be opted as an elective by the students of following subjects: NO		
Suggested equiv	valent online courses (MOOCs) for credit transfer: NO		
Electronic med	dia and other digital components in the curriculum: Choose any	one or more	
than:(Electronic	than: (Electronic Media: Audio/Video Lectures, Online Counseling/Virtual Classes/E-Contents/e-		
SLM/OER/supplementary links for reference/Video Conferencing/Radio broadcast/Web			
Conferencing/ O	Conferencing/ Other electronic and digital contents)		
Name of electro	onic media: Year of incorporation:	2023-24	

## Year-2023-2024 Syllabus of B.Sc. Programme: [Subject Name: Mathematics]

Year	Sem.	Course Code	Paper Title	Theory/	Max.	Credits
				Practical	Marks.	
	Ι	UGMM-101(N)	Differential Calculus	Theory	100	2
		UGMM-102(N)	Analytical Geometry	Theory	100	2
1	II	UGMM-103(N)	Integral Calculus	Theory	100	2
		UGMM-104(N)	Differential Equation	Theory	100	2
		UGMM -105(N)	Mechanics-I (Statics and	Theory	100	2
			Dynamics)			
	III	UGMM -106(N)	Mechanics-II (Dynamics	Theory	100	2
2			and Hydrodynamics)			
		UGMM -107(N)	Linear Algebra	Theory	100	2
	IV	UGMM -108(N)	Calculus of function of	Theory	100	2
			several variable and Vector			
			Calculus			
		Discipline Centric E	lective Course			
		DCEMM -109(N)	Abstract Algebra	Theory	100	2
	V	DCEMM -110(N)	Number Theory	Theory	100	2
		DCEMM-111(N)(P)	Viva Voce	Practical	100	2
3		Skill Enhancement Course				
		SBSMM-03(N)	Elementary Analysis	Theory	100	4
		Discipline Centric Elective Course				
	VI	DCEMM -112(N)	Advance Analysis	Theory	100	2
		DCEMM -113(N)	Function of Complex	Theory	100	2
			Variable			
		DCEMM-114(N)(P)	Viva Voce	Practical	100	2
Total Marks/Credit 1500				1500	32	

## In accordance with NEP-2020

## Syllabus for B.Sc. Subject: MATHEMATICS

Course p	<b>Course prerequisites:</b> 10+2 with Mathematics		
Programme: B.Sc.		Year: 2023-24	Semester: I
Subject: MATHEMATICS			
Course Co	ode: UGMM-101(N)	Course Title: Different	tial Calculus
Course O	Course Objectives:		
> To und	erstand the basics concept of set	theory, function and rela	tions with their properties.
➤ To und	erstand the limit, continuity with	their applications.	
To know	w about the differentiation and the	neir application in solvin	g real life problem.
➢ To dete	ermine the derivative of some spe	ecial functions.	
Course O	utcomes:		
CO1: The	e student will be able to unders	stand about the sets, rel	ation and function with their
properties.			
CO2: The	student shall understand the imp	portance and solution pro	ocedure of problems related to
limit and c	continuity.		
CO3: The	student will get to know about	differentiation and its a	pplications in determining the
derivatives	s of higher orders.		
CO4: The	student shall understand the impo	ortance and applications of	of Rolle's theorem, Lagrange's
Mean valu	e Theorem and Cauchy Mean va	lue Theorem.	
Credits: 2	Credits: 2 Type of Course: Core		
Max. Marks: 100Min. Passing Marks: 36			
Block 1	Set, Relation, Function and its Property		
	Set and Relation: Set Theory	y, Types of sets, Operat	tions on Sets, Laws Relating
	Operations, De Morgan's Laws, Venn diagram, Cartesian product of two sets, Relation		
Unit I	Definition and Examples Domain and Range of a Relation, Types of Relations,		Relation, Types of Relations,
	Composition of Relation, Equiv	valence relation in a set,	Partition of a Set, Quotient set
	of a set, Oder Relation and Examples.		
Unit II	Functions: Functions or mapping, Direct and inverse images of subsets under maps,		
	Real valued Functions of one variable, Inverse functions, Graphs of functions,		
	Operations on functions, Composite of functions, Even and odd functions, Monotone		
	functions, Periodic functions, A	xiomatic introduction of	$\mathbb{R}$ as a complete ordered field,
	Basic properties of R, Absolute	value, Intervals on the r	eal line.
Unit III	Limits: Definition of limit of a	a function at a point of i	ts domain, Algebra of Limits,
	Infinite Limits (Limits as $x \rightarrow \frac{1}{2}$	$\pm \infty$ ), One Sided Limits.	
Unit IV	Continuity: Continuity (Defin	itions and Examples), Al	gebra of continuous functions,
	Properties of continuous functions, Local Boundedness supremum and infimum of a		
	function, Boundedness and ir	termediate value theore	em, properties of continuous
	functions over closed intervals,	Type of discontinuity, In	nage of a closed interval under
	continuous maps.		
Block 2	Differential Calculus		

Unit V	<b>Differentiability and Derivatives:</b> Differentiability of a function at a point. Definition	
	of derivative of a function and its geometrical interpretation. Derivatives of some	
	simple functions Algebra of derivatives. Chain rule Sign of derivatives and	
	monotonicity of functions Continuity versus Differentiability Derivative of	
	exponential function Logarithmic functions	
Unit VI	Derivative of Hyperbolic Functions and Some Special Functions: Definition of	
	Hyperbolic Eulerions, Derivative of Inverse Hyperbolic Eulerions, Methods of	
	Differentiation (Derivative of rD) Legenithmic Differentiation Derivatives of	
	Differentiation (Derivative of x), Logarithmic Differentiation, Derivatives of	
	functions defined in terms of a parameter, Derivatives of Implicit Functions,	
	Derivatives of Trigonometric Functions, Derivative of the Sine Function, Derivative	
	of the Cosine Function, The Derivatives of the other trigonometric functions,	
	Derivative of the Tangent Function, Derivatives of Inverse Functions, Derivatives	
	of Inverse Trigonometric Functions, Use of Transformations.	
Unit VII	Successive Differentiation: Second and third order Derivatives, n <sup>th</sup> Order	
	Derivatives. Leibnitz's Theorem. Maclaurin's Series, Taylor's Series	
Unit VIII	Mean value theorems: Rolle's theorem, Lagrange's Mean value Theorem and Cauchy	
	Mean value Theorem.	
Suggested	Text Book Readings:	
1. R.G.	Bartle & D.R. Sherbert, Introduction to Real Analysis, John Wiley & Sons, 2020.	
2. T.M.	Apostal, Calculus Vol. I, John Wiley & Sons Inc, 1991.	
3. S. Ba	achandra Rao & C. K. Shantha, Differential Calculus, New Age Publication, 2001.	
4. H. An	ton, I. Birens and S. Davis, Calculus, John Wiley and Sons, Inc., 2002.	
5. G.B. 7	Thomas and R.L. Finney, Calculus, Pearson Education, 2007.	
Suggeste	d online link:	
1. <u>https:</u>	//archive.nptel.ac.in/courses/111/104/111104092	
2. <u>https:</u>	//onlinecourses.nptel.ac.in/noc23_ma13/preview_	
3. <u>https:</u>	//archive.nptel.ac.in/courses/111/104/111104144	
4. <u>https:</u>	//archive.nptel.ac.in/courses/111/105/111105122	
This course can be opted as an elective by the students of following subjects: NA		
Suggested	l equivalent online courses (MOOCs) for credit transfer: NA	

<b>Course prerequisites:</b> 10+2 with Mathematics			
Programme: B.Sc.Year: 2023-24Semester: I			Semester: I
Subject: MATHEMATICS			
Course Code: UGMM-102(N)     Course Title: Analytical Geometry			
Course Objectives:			
To un	derstand the basic concepts of	f conic section and curve tra	acing.
To de	velop working skills with stra	ight line, sphere and cylind	er.
> To ac	quire basic knowledge about of	cones, reciprocal, envelopin	ng and right circular cone.
To lease	arn the concepts and uses of co	entral conicoids.	
Course O	utcomes:		
<b>CO1:</b> The	student will get to know abo	ut conic section and curve	tracing, and its applications in
finding the	e area, surface and volumes.		
CO2: The	student shall understand the	e concepts of straight line,	plane, sphere, intersection of
sphere and	l plane, cylinder and right circ	cular cylinder, cones and rig	ht circular cone.
<b>CO3:</b> The	subjects learn and visualize th	he fundamental ideas about	coordinate geometry and learn
to describe	some of the surface by using	analytical geometry.	1 1
CO4: In	e student will be able to und	erstand the central conicol	as with properties and their
Crediter 2		Tupo of Courses Core	
Credits: 2 Type of Course: Core   March 100 Min Daily 26			
Rlock 1	Conic Section	wini. rassing warks: 30	
DIOCK I	Conic Section: Homogeneous equation of second degree and conditions on it to		
	represent different types of c	onics Polar coordinates Po	lar equation of a line parabola
Unit I	ellipse and hyperbola when focus is taken as pole. Polar equations of the chord joining		
	two points.	focus is taken as pole. I ola	equations of the chord joining
Unit II	<b>Curve Tracing:</b> Tangent normal polar (chord of contact) pair of tangent lines		
	asymptotes. Tracing of a conic.		
Block 2	Sphere and Cylinder		
Unit III	Geometry of 3-Dimension	: Straight line and plane, d	lirection cosines and direction
	numbers, distance of a point	from a line, various form of	f the equation of a plane, plane
	passing through three given	points, angle between two	lines and two planes, distance
	of a point from a plane, equ	ation of line of intersection	of two planes, intersection of
	line and plane. Coplanar line	es shortest distance between	n two skew lines.
Unit IV	Sphere: Equation of a sphe	re, Intersection of sphere a	nd planes, Intersection of two
	sphere. Sphere passing thro	ugh a circle, Intersection o	of a straight line and a sphere.
	Tangent planes, Polar planes, Plane of contact. Power of a point. Radical planes,		
	Radical lines, Co-axel system	m of a sphere. Orthogonal s	ystem of sphere.
Unit V	<b>Cylinder:</b> Equation of a cyl	linder with given base, Cyl	inder with Axis parallel to co-
	ordinate axes. Enveloping	cylinders, Right circula	r cylinders. Rules surfaces,
	generating lines of a hyperbo	oloid of one sheet and their	simple properties.
Block 3	Cones and Central Conico	lds	

Unit VI	<b>Cones:</b> Equation of a cone with a given base, Intersection of a cone and a plane passing		
	through the vertex of cone, tangent plane, reciprocal cone, Enveloping cone, right		
	circular cone.		
Unit VII	Central Conicoids-I: Standard equation of a Central conicoid, ellipsoid,		
	hyperboloid of one sheet and two sheets, tangent planes, tangent lines, polar planes		
	and polar lines.		
Unit VIII	Central Conicoids-II: Enveloping cones and cylinders section with a given centres.		
	Diametric plane, conjugate diameters, normal, normal drawn from a given point.		
Suggested	I Text Book Readings:		
1. Rober	t J.T Bell, Elementary Treatise on Coordinate Geometry of three dimensions, Macmillan		
India	Ltd. 1923.		
2. P.R. V	2. P.R. Vittal, Analytical Geometry 2d & 3D, Pearson, 2017.		
3. S.L. L	oney, The Elements of Coordinate Geometry, McMillan & Company, London, 2016.		
4. Thoma	as, G.B. and Finney, R.L., Calculus and Analytic Geometry, 9th Edition, Pearson, Reprint,		
2002.			
This cour	se can be opted as an elective by the students of following subjects: NA		
Suggested	l equivalent online courses (MOOCs) for credit transfer: NA		

Course j	prerequisites: 10+2 with	h Mathematics	
Program	amme: B.Sc. Year: 2023-24 Semester: II		
Subject: MATHEMATICS			
Course Co	ode: UGMM-103(N)	Course Title: Integral Calculu	18
Course O	bjectives:		
≻ To un	derstand the basics conce	pts of integration and their meth	hods.
➢ To de	velop working skills with	integration and use of reduction	n formula in integration.
To lease	arn the integration of ratio	nal and irrational function and t	angent normal to the curves.
➤ To dia	scuss the applications of i	ntegration.	
Course O	utcomes:		
<b>CO 1</b> : The	e student shall understand	the Integration and their solvab	ole techniques.
CO 2: The	e student will be able to un	nderstand the principles of integra	al and learns to solve a variety
of practical	l problems in science and e	engineering.	
CO3: The	e student will get to kno	w about solutions of integration	on of rational and irrational
function.			
<b>CO4:</b> The	student will get to know	about integral calculus and its a	applications in finding areas,
surface an	d volumes.		
Credits: 2	Credits: 2 Type of Course: Core		
Max. Mai	<b>:ks:</b> 100	Min. Passing Marks: 36	
Block 1	Integration		
	Method of Integration: Standard Integrals, Algebra of Integrals, Integratio		
	Substitution, Integrals using Trigonometric formula, Trigonometric and Hyperbolic		
Unit I Substitution, Two proper		erties of Definite integrals, Int	tegration by Parts, Evaluation
	of $\int (a^2 - x^2) dx$ , $\int (a^2 - x^2) dx$	$(+ x^2) dx, \int (x^2 - a^2) dx, \int e^{-\frac{1}{2}} dx$	f(x) = f(x) + f'(x) dx.
Unit II	Reeducation Formula: Reduction formula, Integrals Involving trigonometric		
	functions, Integrals invo	olving products of trigonometric	e functions, Integrals Involving
	Hyperbolic Functions.		
Unit III	Integration of Ration	al and Irrational Function:	Integration of Rational and
	Irrational Functions Inte	gration of Rational Function, So	ome simple Rational Function,
	Partial Fraction Decon	nposition, Method of Substitu	tion, Integration of Rational
	Trigonometric Function	s, Integration of Irrational Func	tions.
Unit IV	V Tangent Normal of the Curves: Equations of tangents and normal, Angles of		gents and normal, Angles of
	intersection of two curv	es, Tangents at the origin.	
Block 2	Application of Integral Calculus		
Unit V	Tracing of curves: Cla	ssifying singular points, Asymp	ptotes (Parallel to the axes and
	oblique asymptotes. Tra	cing of curves.	
Unit VI	Area Under a Curve: A	Area of the curve in Cartesian fo	orm, Polar form, Area Bounded
	by a closed curve, Leng	th of a Plane Curve in Cartesian	Form, Parametric Form, Polar
	form.		

Unit VII	Volume of a solid of Revolution: Volume of a solid of Revolution in Parametric Form	
	Polar form, Area of Surface of Revolution in Cartesian Form, Parametric Form and	
	Polar form.	

1. P. K. Mittal, Integral Calculus, S. Chand Limited, 2005.

2. G.B. Thomas and R.L. Finney, Calculus, Pearson Education, 2007.

3. Piskunov, N., Differential and Integral Calculus, Vol 1 & 2, 2nd edition. Mir Publishers, 1974.

4. Kreyszig, E., Advanced Engineering Mathematics, 10th edition. John Wiley & Sons, 2010.

5. Malik, A.K., Mathur, P, Purohit, S.D., A text Book of Engineering Mathematics-I, Manakin Press, 2020.

Suggested online link:

1. https://archive.nptel.ac.in/courses/111/105/111105122

This course can be opted as an elective by the students of following subjects: NA Suggested equivalent online courses (MOOCs) for credit transfer: NA
# Course prerequisites: 10+2 with MathematicsProgramme: B.Sc.Year: 2023-24Semester: II

## **Subject:** MATHEMATICS

Course Code: UGMM-104(N)Course Title: Differential Equation

## **Course Objectives:**

- > To learn the first order ordinary differential equations with first degree and not of first degree.
- > To understand the applications of differential equations.
- > To acquire basic application problems described by second order linear differential equations.
- > To learn the solution of ordinary differential equations of higher orders.

# **Course Outcomes:**

**CO1:** The student shall understand the importance and solution procedure of solving the first order ordinary differential equations with first degree and not of first degree.

**CO2:** After completing this course, a student will be able to solve differential equations and applications in science and engineering.

**CO3:** The student shall understand the second order linear differential equations with constant coefficient. Furthermore, the student will be able to develop understanding towards Euler-Cauchy linear equations and Simultaneous linear differential equations.

**CO4:** To understand and able to solve the various differential equation with variable coefficients used in engineering and science.

Credits: 2	Type of Course: Core	
Max. Marks	: 100 Min. Passing Marks: 36	
Block 1	Differential Equations of First Order and First Degree	
	Differential equation: Differential equations, Types of differential equations, Order	
Unit I	and degree of differential equations, Formation of differential equation, Solution of	
Unit I	differential equation, Geometrical meaning of a differential equation, Initial value	
	problems and statement of Existence and Uniqueness Theorems.	
Unit II	Methods of solution of a differential equation of first order and first degree:	
	Methods of solution of a differential equation of first order and first degree, Method of	
	separation of variables. Solution of homogeneous equations. Equation reducible to	
	homogeneous form.	
Unit III	Linear differential equation: Linear differential equation and Bernoulli's linear	
Omt m	differential equation.	
Unit IV	<b>Exact differential equations</b> : Exact differential equations. Integrating factors to solve	
	non-exact differential equations in different cases.	
Unit V	Differential equation of the first order but not of the first degree: Differential	
	equation of the first order but not of the first degree, Equation solvable for x, y and p.	
	Clairaut's equation and singular solutions.	
Block-2	Applications of differential equation	
Unit VI	Geometrical Applications of Differential Equations: Applications of differential	
	equation, geometrical application and physical applications; Newton's law of cooling,	

	Kirchoff's law of electric circuits, motion under Gravity, rectilinear motion, simple	
	harmonic motion, rate of growth or decay, heat flow.	
Unit VII	Physical applications of differential equations of first order and first degree-I: The	
	n <sup>th</sup> order linear differential equation with constant coefficients, general solution and	
	particular integrals. Method of finding particular integrals Methods of undetermined	
	coefficient, variation of parameters.	
Unit VIII	Physical applications of differential equations of first order and first degree-II:	
	Physical applications of differential equations of first order and first degree-II, Method	
	of finding particular integrals by inverse operator methods.	
Block-3	The nth order linear differential equation with constant coefficients	
Unit IX	The nth order linear differential equation with constant coefficients: Method of	
	finding particular integrals by inverse operator methods.	
Unit X	Methods of finding particular integrals by inverse operator method: Linear	
	differential equations of second order. Transformation of the equation by changing the	
	dependent variable, independent variables and Normal forms.	
Unit XI	Linear Equations with constant coefficients: Equation reducible to Linear with	
	constant coefficients, Euler-Cauchy linear equations. Simultaneous linear differential	
	equation with constant coefficient.	
Unit XII	Linear differential equations: Linear differential equations of second order.	
	Transformation of the equation by changing the dependent variable, independent	
	variables and Normal forms.	
Suggested T	ext Book Readings:	
1. G.F. Simr	nons, Differential Equations with Application and Historical Notes, Tata –McGraw-Hill.	
2. B. Rai, D.	P. Choudhary & H. J. Freedman, A Course of Ordinary Differential Equations, Narosa	
Publicatio	n.	
3. M., D. Ra	i Singhania, Ordinary Differential Equation, S. Chand, 2020.	
4. L.E. Elsgolts, Differential Equation and Calculus of variations, University Press of the Pacific.		
5. Malik, A.K., Mathur, P & Purohit, S.D., A text Book of Engineering Mathematics-II, Manakin		
Press, 201	9.	
Suggested on	line link:	
1. <u>https://e</u>	pgp.inflibnet.ac.in/Home/ViewSubject?catid=ZLCHeZEhCZ8yCri36nSF3A==	
2. https://a	renive.nptel.ac.m/noc/courses/111 can be opted as an elective by the students of following subjects: NA	
	can be opted as an elective by the students of following subjects: NA	
Suggested e	invalent online courses (MOOCs) for credit transfer:	

Differential equations for engineers By Prof. Srinivas Rao Manam

https://onlinecourses.nptel.ac.in/noc22\_ma72/preview

Course prerequisites: 10+2 with Mathematics			
Programm	ne: B.Sc.	<b>Year:</b> 2023-24	Semester: III
Subject: MATHEMATICS			
Course Co	Course Code: UGMM-105(N)Course Title: Mechanics-I (Statics and Dynamics)		
Course O	bjectives:		
To lease	arn the stable and unstable ea	quilibrium in statics.	
➤ To de	velop competency in unders	tanding of virtual work and c	common catenary.
🕨 To un	derstand the motion in a pla	ne, tangential and normal dire	ections, rectilinear motion.
To un	derstanding the constrained	motion and motion under cer	ntral forces.
Course O	utcomes:		
CO1: The	student shall understand the	basic concepts of Statics.	
CO2: The	student will get to know abo	out virtual work, displacemen	it, tensions and string.
CO3: The	student shall understand the	e dynamics system and descri	ibe the theoretical workings
of the recti	linear motion, simple harmo	onic motion and their utility in	n real life.
<b>CO4:</b> The	student shall understand the	importance and solution proc	edure of constrained motion
and motion	n under central forces.	Γ	
Credits: 2		Type of Course: Core	
Max. Mar	· <b>ks:</b> 100	Min. Passing Marks: 36	
Block 1	Statics		
Unit I	Stable and Unstable equ	ilibrium: Stable and Unstab	ble equilibrium definition and
	examples, Stability, Condition of Stability, Heavy body on a fixed body.		
Unit II	Virtual work: Virtual works and displacement, Principal of Virtual work acting on a		
	particle, Principal of Virtual work of rigid body, tensions, string.		
Unit III	<b>Common Catenary:</b> Definition of Common Catenary, intrinsic equation, Cartesian		
	equation, definitions to Catenary, Relation for Common Catenary, Stretched wires.		
Block 2	Dynamics		
Unit IV	Motion in a Plane: Mot	tion in a plane, Velocities a	nd accelerations in Cartesian
coordinates, along radial and transverse directions, and along tangential and no			along tangential and normal
TT	directions. Determination of path under a given force.		
Unit V	<b>Rectilinear Motion:</b> Rect	illinear Motion, Simple harn	nonic motion. Elastic strings.
	Motion under inverse squa	are law and other miscellane	ous laws, Motion in resisting
	medium. Motion of particles of varying mass, Rocket motion.		
Unit VI	Constrained Motion: Con	istrained motion (Vertical cire	cle and vertical cycloid).
Unit VII	VII Motion under Central Forces: Motion under Central forces: Central orbit,		
Conservation of angular momentum, areal velocity, Kepler's laws of motion, and			
differential equations to the path of a particle.			
Suggested Text Book Readings:			
1. R.C. H	ibbeler, Engineering Mechar	nics-Statics, Prentice Hall Put	Dishers. Publishers
3. A. Nelson, Engineering Mechanics Statics and Dynamics. Tata McGraw Hill.			
4. J.L. Sy	4. J.L. Synge & B.A. Griffith, Principles of Mechanics, Tata McGraw Hill.		

Course prerequisites: 10+2 with Mathematics		
Programme: B.Sc.	<b>Year:</b> 2023-24	Semester: III
Subject: MATHEMATICS		
Course Code: UGMM-106(N)	Course Title: Mechanics-I	[ (Dynamics and Hydrodynamics)
Course Objectives:		

### > To learn about moment of inertia and D'Alembert's principle.

- $\blacktriangleright$  To learn the motion about a fixed axis, equation of motion about axis of rotation.
- > To understand the boundary surfaces, Euler's equation of motion and impulsive motion.
- > To understanding the doublets, image system of a doublet with respect to a plane, a circle, a sphere.

#### **Course Outcomes:**

**CO1:** The student shall understand the basic concepts and application of moment of inertia and D'Alembert's principle.

**CO2:** The student will get to know about motion about a fixed axis, equation of motion about axis of rotation.

**CO3:** The student shall understand the hydrodynamics system and describe the boundary surfaces, Euler's equation of motion and impulsive motion, and their utility in real life.

**CO4:** The student shall understand the importance of doublets, image system of a doublet with respect to a plane, a circle, a sphere.

Credits: 2		Type of Course: Core	
Max. Marks: 100		Min. Passing Marks: 36	
Block 1	Dynamics		
Unit I	Moment of Inertia: Mo	ment and product of inertia of some standard bodies, principle	
Unit I	axis, Momental ellipsoid	of a body.	
Unit II	D' Alembert Principle:	The general equation of motion, motion of the centre of inertia	
	and motion relative to th	e centre of inertia.	
Unit III	Motion about a fixed a	<b>xis</b> : Moment of the effective forces about the axis of rotation,	
	moment of momentum a	about the axis of rotation, kinetic energy of the body rotating	
	about a fixed axis, equation of motion about axis of rotation.		
Block 2	Hydrodynamics		
Unit IV	Boundary surfaces: Equ	ation of continuity in different coordinate system and boundary	
	surfaces, velocity potential, stream-lines.		
Unit V	Euler's equation of motion: Euler's equation of motion, steady motion, Bernoulli's		
	equation, Helmholtz equ	ation, Impulsive motion.	
Unit VI	Motion: Motion in tw	o dimensions, stream function, irrotational motion, complex	
	potential, sources and sin	nks.	
Unit VII	Doublet: Doublets, ima	ge system of a simple source with respect a plane, a circle, a	
	sphere. Image system of	a doublet with respect to a plane, a circle and a sphere, circle	
	theorem.		
Suggested	l Text Book Readings:		
1. Nag, P.K., Engineering Thermodynamics, 3rd ed., Tata McGraw-Hill, 2005.			

- 2. Robert Norton., "Kinematics and Dynamics of machinery" 1<sup>st</sup> Ed., McGraw Hill India., 2009.
- 3. R.C. Hibbeler, Engineering Mechanics-Dynamics, Prentice Hall Publishers.
- 4. A. Nelson, Engineering Mechanics Statics and Dynamics, Tata McGraw Hill.
- 5. J.L. Synge & B.A. Griffith, Principles of Mechanics, Tata McGraw Hill.

# Suggested online link:

1. <u>https://epgp.inflibnet.ac.in/Home/ViewSubject?catid=ZLCHeZEhCZ8yCri36nSF3A</u>==

This course can be opted as an elective by the students of following subjects: NA Suggested equivalent online courses (MOOCs) for credit transfer: NA

<b>Course prerequisites:</b> 10+2 with Mathematics			
Program	ne: B.Sc.	Year: 2023-24	Semester: IV
Subject: MATHEMATICS			
Course Co	Course Code: UGMM-107(N)Course Title: Linear Algebra		
Course O	bjectives:		
To un	derstand the vector space, fi	eld and their properties.	
To lease	arn about the basis and dimer	nsion of a vector space and quo	tient spaces.
To un	derstand the matrices, linear	system of equations, Eigen val	ues and Eigen vectors.
To fir	nd the characteristic polynom	ial, inner product space, bilinea	r and quadratic forms.
Course O	utcomes:		
<b>CO1:</b> The	student will be able to und	erstand the basic concepts and	d uses of vector space field and
their prope	erties.		
CO2: The	student will get to know abo	out basis and dimension of a v	ector space and quotient spaces.
<b>CO3:</b> The	student will be able to und	lerstand the basic concepts a	nd uses of matrices for solving
system the	linear system of equations,	eigen values and eigen vector	s with its importance.
<b>CO4:</b> The	student shall understand the	e characteristic polynomial, in	nner product space, bilinear and
quadratic f	forms with their applications		
Credits: 2		Type of Course: Core	
Max. Mar	: <b>ks:</b> 100	Min. Passing Marks: 36	
Block 1	Vector Space		
	Introduction to Vector	Spaces: Binary operations.	, Field, Examples $\mathbb{Q}$ , $\mathbb{R}$ , $\mathbb{C}$ , $\mathbb{Z}_p$
Unit I	(p, a prime) of fields. Definition of a vector space, Some basic properties of a vector		
space, Vector subspace, Subspace spanned by subsets with examples. Lin			sets with examples. Linearly
	dependence and independence of a subset. Finite dimensional vector spaces.		
Unit II	Basis, Dimension and Quotient Spaces: Basis of a vector space, Dimension of a vector		
	space, finite dimensional v	ector space, linear sum and di	rect sum of subspaces, Quotient
	spaces.		
Unit III	Linear transformations	: Definition examples a	nd some properties, Linear
	transformations and Isomor	rphism of vector spaces, Null s	space and range space. Rank and
	Nullity of linear transforma	ations, Fundamental theorem	of vector space homomorphism,
	Rank-Nullity theorem, Nor	1-singular, Invertible transform	nations.
Unit IV	Dual Vector Space: Hom	(V, W) as a vector space, dua	al space $\mathbf{V}^*$ of a vector space $\mathbf{V}$ ,
	Dual basis of a vector sp	ace with examples, Transpo	se of a linear transformations.
	Annihilator of a subset of a	i vector space, Rank of transp	ose of a linear transformation.
Block 2	Matrix		
Unit V	Matrices: Matrix represent	tation of linear transformation	s, equality, Algebra of matrices,
	Multiplication of matrices. Vector space of all $m \times n$ matrices over a field. Ring of all n-		
	square matrices. Invertible	e matrices (or Non-singular	matrix), Transpose of matrix,
TT '- TT	Equivalent matrices, Simila	ar Matrices, and Orthogonal n	natrices.
Unit VI	<b>Kank of a matrix:</b> Rank of	t a matrix, Row rank and colui	mn rank of a matrix, Elementary
	matrices, elementary row	and column operations of a	a matrix, elementary matrices,

	Normal form of a matrix, Echelon form of matrix, Inverse of a non-singular matrix.		
	Determinant rank of a matrix. Non-homogeneous and homogeneous linear equations.		
Unit VII	Determinants: Determinant of a square matrix, Cofactor of an element of a determinant,		
	Properties of a determinant, minor of an element of a determinant, Evaluation of a		
	determinant. Laplace expansion of a determinant. Product of two determinants (all		
	statements without proof).		
Unit	Eigen Vector and Eigen Space: Adjoint of n- square matrix. Inverse of a matrix of a		
VIII	non-singular matrix by using adjoint of the matrix. Characteristic roots or Eigen values		
	of a linear transformation and Eigen vector and Eigen space.		
Block 3	Characteristic polynomial inner product space bilinear quadratic forms		
Unit IX	Characteristic polynomial of a matrix: Characteristic polynomial of a matrix,		
	Diagonalization of a matrix, Caley - Hamilton theorem. Inverse of a matrix of a non-		
	singular matrix by Caley – Hamilton theorem. Characteristic polynomial of a linear		
	transformation, Minimal polynomial, Hermitian Matrix, characteristic roots of a complex		
	Hermitian matrix.		
Unit X	Inner product space: Definition and examples of inner product, length of a vector,		
	Cauchy Schwarz inequality, distance between two vectors, angle between two vectors,		
	Orthogonal and ortho normal sets.		
Unit XI	Bilinear, quadratic and Hermitian forms: Bilinear, quadratic and Hermitian forms,		
	bilinear form on a vector space V. Quadratic forms, matrix of quadratic form, normal		
	form or canonical form.		
Suggested	I Text Book Readings:		
1. Gilber	rt Strang, Linear Algebra and its Applications, Thomson, 2007.		
2. S. Ku	maresan, Linear Algebra- A Geometric Approach, Prentice Hall of India, 1999.		
3. Kenne	eth Hoffman, Ray Alden Kunze, Linear Algebra, 2 <sup>nd</sup> Ed., Prentice-Hall of India Pvt. Ltd.,		
1971.			
4. Malik	4. Malik, A.K., Mathur, P, Purohit, S.D., A text Book of Engineering Mathematics-II, Manakin		
Press,	2019.		
Suggested	online link:		
1. <u>https://epgp.inflibnet.ac.in/Home/ViewSubject?catid=ZLCHeZEhCZ8yCri36nSF3A</u> ==			
2. https://archive.nptel.ac.in/courses/111/104/111104125			
This course can be opted as an elective by the students of following subjects: NA			
Suggested	l equivalent online courses (MOOCs) for credit transfer: NA		
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<b>Course prerequisites:</b> 10+2 with Mathematics		
Programme: B.Sc.	<b>Year:</b> 2023-24	Semester: IV
Subject: MATHEMATICS		
Course Code: UGMM-108(N)	Course Title: Calculus of Function of Several Variables	
	and Vector Calculus	

#### **Course Objectives:**

- > To learn about the partial differentiation and jacobians with their applications.
- > To understand the maxima and minima, and Lagrange's method of undetermined multipliers with their importance.
- > To understand the basic concepts and uses of vector calculus.
- > To know about applications of vector calculus.

#### **Course Outcomes:**

**CO1:** The student shall understand the partial differentiation, Euler's Theorem on Homogeneous functions.

**CO2** The student will get to know about the maxima and minima of function s of two variables, Lagrange's method of undetermined multipliers.

**CO3:** The student will get to know about the basic concepts and uses of vector calculus.

**CO4:** The student shall understand the importance and applications of vector calculus like as curl, gradient and divergence.

Credits: 2		Type of Course: Core	
Max. Marks: 100		Min. Passing Marks: 36	
Block 1	Partial Differentiation		
	Partial Differentiation:-Partial Derivatives of first order, Partial Derivatives of Higher		
Unit I	order, Total derivative	, Homogeneous functions, Euler's Theorem on Homogeneous	
	functions, Deductions f	rom Euler's Theorem.	
Unit II	<b>Composite Functions:</b>	- Differentiation of Composite Functions, Jacobians, Properties	
	of Jacobians, Theorem	ns on Jacobians, Jacobian of Implicit Functions, Functional	
	Relationship,		
Unit III	Maxima and minima:- Maxima and minima of functions of Two variables, Conditions		
	for maxima and minima, Necessary condition for existence of maxima and minima,		
	Local and global maxima and minima of a function, Lagrange's method of		
	undetermined multipliers.		
Block 2	Vector Calculus		
Unit IV	Operations on Vectors:- Scalar triple product and its geometrical interpretation.		
	Properties of scalar triple product. Reciprocal system of vectors. Properties of the		
	reciprocal system of vectors. Scalar product of four vectors. Vector product of four		
	vectors.		
Unit V	Differentiation of a	Vector Functions and Gradient:- Scalar and vector point	
	function. Differentiation	n of a vector functions with respect to scalar. Gradient of a scalar	
	field, geometrical inte	erpretation of gradient, directional derivative, properties of	
	gradient, gradient in polar coordinates.		

Unit VI	Divergence and Curl:- Divergence of a vector point function, Physical interpretation
	of divergence, Solenoidal vector. Curl of vector point function, Physical interpretation
	of curl, Irrotational vector, Vector identities, Velocity potential, Laplace operator.
Unit VII	Integration of a Vector Function:- Line Integral, Surface Integral, Volume Integral.
Unit VIII	Applications: Green's Theorem in a plane, Gauss Divergence Theorem and Stokes's
	Theorem (Without proof) and their applications.

#### Suggested Text Book Readings:

1. Kreyszig, E., Advanced Engineering Mathematics, 10th edition. John Wiley & Sons, 2010.

2. Malik, A.K., Mathur, P, Purohit, S.D., A text Book of Engineering Mathematics-1, Manakin Press, 2020.

3. Mujumdar, N. S., Function of Several Variables and Vector Calculus, Nirali publication. **Suggested online link:** 

- 1. <u>https://epgp.inflibnet.ac.in/Home/ViewSubject?catid=ZLCHeZEhCZ8yCri36nSF3A</u>==
- 2. https://archive.nptel.ac.in/courses/111/105/111105122
- 3. https://archive.nptel.ac.in/courses/111/104/111104125

This course can be opted as an elective by the students of following subjects: NA

**Suggested equivalent online courses (MOOCs) for credit transfer:** <u>https://onlinecourses.nptel.ac.in/noc23\_ma27/preview</u> **Integral and Vector Calculus** By Prof. Hari Shankar Mahato, IIT Kharagpur.

<b>Course prerequisites:</b> 10+2 with Mathematics			
Programme: B.Sc.Year: 2023-24Semester: V			
Subject: MATHEMATICS			
Course Code: DCEMM -109 (N) Course Title: ABSTRACT ALGEBRA			
Course O	bjectives:		
To un	derstand the concepts of	group theory with their propertie	es.
To lease	arn about the homomorph	ism, subgroups and cyclic group	).
To un	derstand the normal subg	roups and symmetric groups.	
To lease	arn about the ring, field an	d ideal with their properties.	
CO1: The	e student will be able to	understand the basic concepts	and uses of group with their
properties.			
CO2: The	student shall understand	I the importance of homomorph	nism subgroups and uses of
cyclic grou	ıp.		
CO3: The	student will get to know	about the normal subgroups and	l symmetric groups.
<b>CO4:</b> The	student shall understand	the applications of ring, field and i	deals.
Credits: 2		Type of Course: Core	
Max. Mar	<b>ks:</b> 100	Min. Passing Marks: 36	
Block 1	Groups and Subgroup	S	
	<b>Elementary Group The</b>	eory: Definition of a group, abel	ian groups examples including
Unit I	$\mathbf{Z}_{\mathbf{m}}$ , $\mathbf{Z}_{\mathbf{p}}$ , $\mathbf{U}_{\mathbf{m}}$ the group	of n <sup>th</sup> roots of unity, Hamiltoni	an group, Klein's four group,
Ollit I	Permutation group, Integral power of an element of a group, order of an element of a		
	group.		
Unit II	Homomorphism, Subgroups and Cyclic Groups: Subgroups of a group and		
	examples. Homomorphism, isomorphism, Subgroup generated by a subset of a group.		
	Cyclic groups.		
	<b>Coset Decomposition of a Group:</b> Coset decomposition, left coset and right coset of		
Unit III	a subgroup of a group. Lagrange theorem. Index of a subgroup. Euler's theorem		
	Fermat's theorem.		
Block 2	Normal Subgroups and	d Symmetric Groups	
Unit IV	Normal Subgroups an	d Homomorphisms: Normal s	subgroups, Centre of a group.
	Conjugate elements, No	rmaliser of an element of a group	p. Kernel of a homomorphism.
	Direct and inverse in	nage of a subgroup and a	normal subgroup under a
	homomorphism. Quotient groups.		
Unit V	Symmetric Groups and	d Automorphisms: Fundament	al theorem of homomorphism
	of groups, Symmetric group $S_n$ , Cayley's theorem, Cycle's transposition,		
	Decomposition of a permutation, alternating groups A <sub>n</sub> . Automorphisms of groups.		
Inner automorphisms.			
Block 3	Rings and Fields:		
Unit VI	Rings and Fields: Ring	g Definition and examples, elem	entary property of a ring, zero
	divisor, Ring with or v	without zero divisor. Integral of	domain. Division ring. Field.
	Homomorphism and iso	omorphism of Rings, subrings, su	ubfield with examples. Kernel

	of a homomorphism.	
Unit VII	Homomorphisms and Embedding of rings: Direct image and inverse image of a	
	subring and a subfield under a homomorphism. Characteristic of a non-zero integral	
	domain. Imbedding of a ring into another ring the field of fractions of an integral	
	domain.	
Unit VIII	Ideals: Ideals, left ideal and right ideal, principal ideal with example. Prime ideal,	
	maximal ideal, Quotient rings. Fundamental theorem of homomorphism, rings, and	
	field.	
Suggested	l Text Book Readings:	
1. J.B. Fra	aleigh, Abstract Algebra by J.B. Fraleigh, Published by Narosa publishing House.	
2. I.N. He	erstein, Topics in Algebra, Wiley, 2006.	
3. Vijay l	K Khanna & S K Bhambri, Vijay K Khanna (Author), S K Bhambri, Vikash Publishing,	
2017.		
4. Frank	Ayres & Lloyd R. Jaisingh, Schaum's Outline of Abstract Algebra, McGraw hill, 2020.	
Suggested	online link:	
1. https://archive.nptel.ac.in/courses/111106137		
This course can be opted as an elective by the students of following subjects: NA		
Suggested	l equivalent online courses (MOOCs) for credit transfer: NA	

<b>Course prerequisites:</b> 10+2 with Mathematics			
Programme: B.Sc.Year: 2023-24Semester: V		Semester: V	
Subject: MATHEMATICS			
Course Co	Course Code: DCEMM -110 (N)Course Title: Number Theory		
Course Ob	jectives:		
➢ To under	rstand the basics concept of numb	ber theory.	
➢ To learn	about the Chinese Remainder Th	heorem and Euler's func	ction.
➢ To under	rstand the Gauss theorem, Gauss	s reciprocity theorem an	d Quadratic residues for prime-
power m	oduli.		
To know	about the arithmetic function wi	vith their applications in s	solving real life problem.
Course Ou	tcomes:		
CO1: The s	student will be able to understand	d about the basics concept	pt of number theory.
CO2: The	student shall understand the im	nportance and application	ons of the Chinese Remainder
Theorem an	d Euler's function.		
CO3: The s	student will get to know about Ga	auss theorem, Gauss rec	iprocity theorem and Quadratic
residues for	prime-power moduli.		
CO4: The s	tudent shall understand the impor	ortance and applications	of arithmetic function.
Credits: 2	T	<b>Fype of Course:</b> Core	
Max. Mark	ks: 100 N	Min. Passing Marks: 36	5
Block 1	Basic Introduction of Number	er Theory	
Unit I	Analytic Number Theory: D	Division algorithm, Eucl	lid's algorithm for the greatest
	common divisor, Prime number	ers, fundamental theorem	of arithmetic.
Unit II	Congruence's: Linear congru	uences and algorithm t	to find the solution of Linear
	congruences, Chinese Remainder Theorem, An extension of Chinese Remainder		
Theorem (with non-coprime moduli).			
Unit III	Euler's Function and applic	ication: Definition of	Euler function, examples and
	properties, Multiplicative property of Euler's function.		
Block 2	Gauss theorem & Application	ns	
Unit IV	Quadratic residues of Numb	ber Theory: The grou	p of units modulo an integer,
	primitive roots, Existence of	f primitive roots, Qua	dratic congruences, Quadratic
	residues, Legendre symbol, Eul	iler's criterion.	
Unit V	Gauss theorem: Gauss lemma	na, Gauss reciprocity th	neorem, Quadratic residues for
	prime-power moduli and arbitra	ary moduli.	
Unit VI	Arithmetic Functions and	applications: Arithm	etic Functions, multiplicative
	functions and their properties,	, Mobius function and it	ts properties, Mobius inversion
formula and its applications.			
Suggested	Text Book Readings:		
I. Niven, I.	, Zuckerman, H. S. and Montego	omery, H. L. (2003) An	Int. to the Theory of Numbers
(6th edition) John Wiley and sons, Inc., New York.			
2. Burton, D. M. (2002) Elementary Number Theory (4 <sup>th</sup> edition) Universal Book Stall, New Delhi.			
3. Balakrishnan, V. K. (1994) Schaum's Outline of Theory and Problems of Combinatorics Including			

Concepts of Graph Theory, Schaum's Outline. Suggested online link: 1. <u>https://epgp.inflibnet.ac.in/Home/ViewSubject?catid=ZLCHeZEhCZ8yCri36nSF3A</u>== This course can be opted as an elective by the students of following subjects: NA Suggested equivalent online courses (MOOCs) for credit transfer: NA

Course prerequisites: 10+2 with Mathematics			
Programme: B.Sc.Year: 2023-24Semester: V			
	Subject: MATHEMATICS		
Course Co	Course Code: SBSMM -03(N)Course Title: Elementary Analysis		
Course Ob	jectives:		
➢ To under	rstand the logical connectives, tautology and quantifiers.		
➢ To learn	about relation, equivalence relation and mapping with their properties.		
➢ To under	rstand the real number system and division in Integers, sequences and infinite series.		
To know	v about the multiple integral and their applications in areas and volume.		
Course Ou	tcomes:		
CO1: The	student will get to know about the concepts of logical connectives, tautology and		
quantifiers.			
CO2: The s	student shall understand the relations and its types, equivalence relations and mapping.		
<b>CO3:</b> The s	tudent shall understand the real number system and division in Integers, sequences and		
infinite serie	es.		
CO4: The s	student will get to know about multiple integral and its applications in finding areas,		
surface and	volumes.		
Credits: 4	Type of Course: Core		
Max. Mark	Min. Passing Marks: 36		
Block 1	Language of Mathematics, Relation and Mapping		
Unit I	Language of Mathematics: Language of Mathematics, Mathematical statements,		
	logical connectives, Tautology, quantifiers.		
Unit II	Relation: Relations (definition and examples), types of relation, composite of		
	relations, equivalence relation, equivalence class, partition of a set and order relation.		
Unit III	Mapping: Mapping (definition and examples), types of map, inverse map,		
	composition of maps, direct and Inverse images of a set.		
Block 2	Real number system and Division in Integers		
Unit IV	Real number system: Axiomatic definition of real number system as a complete		
	ordered field. Archimedean principle, relational and irrational density theorem.		
Unit V	<b>Division in Integers:</b> Division in Z, Division algorithm, greatest common divisor,		
	and least common multiple. Euclidean algorithm. Prime integers. Fundamental		
	theorem of arithmetic's.		
Block 3	Sequence and Infinite Series		
Unit VI	Sequence of Real Number: Sequences, bounded and unbounded sequences,		
	Subsequence, convergent, divergent and oscillatory sequences. Limit of a sequence.		
	Algebra of convergent sequences, Cauchy's sequences, and Cauchy's criterion for		
	convergence of a sequence.		
Unit VII	Infinite Series: Partial sums of a series. Convergence and divergence of series.		
	Series of nonnegative terms. Necessary and sufficient condition for convergence.		
	P-series theorem. Comparison tests. D'Alembert's ratio test, Raabe's ratio test,		
	Logarithmic test, Cauchy's condenses test and Root test, Alternating series.		

	Leibnitz's theorem. Absolute convergence and conditional convergence of a series.	
Block 4	Multiple Integral and Its Applications	
Unit VIII	Double and Triple integrals, Change of order of integration, surface and volume	
	integration and their applications in areas and volume.	
Suggested	Text Book Readings:	
1. Elemen	tary Analysis: The Theory of Calculus by Kenneth. A. Ross.	
2. Sharma	a and Vashishtha, Real Analysis, Krishna Publication, 2014.	
<b>3.</b> S C Ma	C Malik, Mathematical Analysis, New Age International Publishers, 2017.	
Suggested	online link:	
1. <u>https://</u>	archive.nptel.ac.in/courses/111/105/111105122	
This course can be opted as an elective by the students of following subjects: NA		
Suggested	equivalent online courses (MOOCs) for credit transfer: NA	

Course prerequisites: 10+2 with Mathematics			
Program	ne: B.Sc.	<b>Year:</b> 2023-24	Semester: VI
Subject: MATHEMATICS			
Course Co	Course Code: DCEMM -112 (N) Course Title: Advance Analysis		
Course O	bjectives:		
> To und	lerstand the basics concept of	f metric space.	
➢ To lear	n about the convergence of f	function of series and improp	er integral.
➤ To und	lerstand the convergence test	, µ-test, Abel's test, Dirichle	t's test.
To kno	w about step function, mean	value theorem for integrals a	and Change of variables.
Course O	utcomes:		
CO1: The	e student will be able to und	lerstand the basic concepts a	nd uses of metric spaces with
their prope	erties.		
CO2: The	student shall understand the	e importance of convergence	of function of series and uses
of imprope	er integral.		
CO3: The	e student will get to know ab	out the convergence, $\mu$ , Abel	's and Dirichlet's test.
CO4: The	e student shall understand th	he applications of step func	tion, mean value theorem for
integrals a	nd Change of variables.		
Credits: 2	2	Type of Course: Core	
Category	of Course	OER	
Max. Mar	r <b>ks:</b> 100	Min. Passing Marks: 36	
Block 1	Metric space, Continuity,	, Compactness and complet	eness
	Metric Space: Metric space	e (definitions and examples),	open and closed balls, interior
Unit I	points, exterior point, and l	boundary points, limit points	open and closed sets, limit of
	a sequence in a metric space. Cauchy sequence.		
Unit II	Limit and Continuity of	Functions: Limit and cont	tinuity of a function between
	metric spaces. Characteriz	ation of continuity in terms	of open sets, Closed set and
	closer of a set.		
Unit III	Compactness: Compactne	ess of metric space, Bolzan	o Weierstrass property, Total
	boundedness, sequentially compact metric spaces and countable compact metric space		
Block 2	Convergence of function	of series and Improper Inte	egral
Unit IV	<b>Complete Metric Space:</b>	Uniform continuity, Lebesg	gue number, Complete metric
	space.		
Unit V	Convergence of sequence	e and series of functions	Point wise and uniform
	convergence of a sequence	ce and series of a function	s, necessary and sufficient
	condition for a uniform cor	vergence, Weierstrass test, A	bels test and Dirichlet's test
	for uniform convergence, T	Ferm by term integration and	term by term differentiation.
Unit VI	Improper Integrals: Conv	vergence of improper integrals	s; Integral over infinite interval
	with bounded integrands a	nd intervals over finite interv	vals with unbounded integrals.
	Necessary and sufficient co	onditions for such integrals.	
Block 3	Convergence test, Rieman	nn integral	

Unit VII	<b>Convergence Test:</b> Comparison test, µ-test, absolute convergence, convergence of
	integrals of product of two functions, Abel's test, Dirichlet's test.
Unit VIII	Step Functions: Step function and their integrals, upper and lower integrals of a
	bounded function of one variable (through step functions). Integrable functions,
	Riemannian condition of integrability. Properties of Integrals of a step functions.
Unit IX	Mean Value Theorem: Mean value theorem for integrals, Fundamental theorem of
	integral calculus. Primitive of a function. Change of variables, second mean value
	theorem (statements only).
Suggested Text Book Readings:	
4 9 4	

**1.** S. Arumugam and A. Thangapandi Issac, Sequences and Series, Edition 2012, New Gamma Publishing House.

**2.** K. Chandra Sekhara Rao and K. S. Narayanan, Real Analysis Volume – I, Edition 2008, S. Viswanadhan Printers and Publishing Pvt. Ltd.

**3.** M. K. Venkatraman and Manorama Sridhar, Sequence and Series, Edition 2002, The National Publishing Company.

4. Singh, S. R. & Malik, A. K. Topology, I.K. International Publishing House Pvt. Limited, 2012.

This course can be opted as an elective by the students of following subjects: NA

Suggested equivalent online courses (MOOCs) for credit transfer: NA

<b>Course prerequisites:</b> 10+2 with Mathematics			
ogram	ne: B.Sc.	<b>Year:</b> 2023-24	Semester: VI
Subject: MATHEMATICS			
Course Code: DCEMM -113 (N) Course Title: Function of Complex Variables			
urse O	bjectives:		
To un	derstand the basic concepts of	f analytic function and Cauc	chy Riemann equations.
To de	velop working skills with con	nplex integration, series and	l singularities.
To ac	equire basic knowledge about	t calculus of residues and	Evaluation of real definite
integr	als by contour integration.		
To lea	arn the concepts and uses of co	onformal mapping and Mobi	us (bilinear) transformation.
urse O	utcomes:		
<b>1</b> : The	student will get to know abou	it analytic functions and Ca	uchy Riemann equations.
2: The	e student shall understand the	concept of complex integrat	tion, series and singularities.
<b>3:</b> Th	e student shall understand the	importance and solution pr	ocedure of problems related
calculus	s of residues and Evaluation of	f real definite integrals by c	ontour integration.
<b>4:</b> The	student will be able to unders	tand the basic concepts and	uses of conformal mapping
l Mobiu	is (bilinear) transformation wi	th their applications.	
edits: 2		Type of Course: Core	
tegory	of Course	OER	
x. Mai	<b>·ks:</b> 100	Min. Passing Marks: 36	
	Syllabi fra	med block wise/unit wise	
ock 1	Complex variables and Pov	wer series	
	Function of Complex Variable: The concept of a function of a complex variable,		
	Continuous functions, unif	orm continuity, bounded	functions, differentiable and
it I analytic functions, differentiability, analytic or regular functions Cauchy Riema		r functions Cauchy Riemann	
	equations, necessary and suf	ficient condition for a funct	ion to be analytic, construction
	of analytic function, Milne I	Thomson method.	
nit II <b>Power Series:</b> The circle of convergence of a power series, power series and analyti		ries, power series and analytic	
	functions, the exponential	functions, the trigonometr	ic functions, the logarithmic
functions.			
ock 2	Complex Integration and I	Expansion of series	
	Complex Integration: Jord	an Arcs, Rectifiable arcs, c	ontours, Complex integration,
:4 TTT	integration along a regular Arc, Cauchy theorem, the elementary form of Cauch		e elementary form of Cauchy
Juit III theorem, the general form of Cauchy theorem, extension of Cauchy's theorem		integral formula derivative of	
	an analytic function. Morero	's Theorem	integral formula, derivative of
;+ IV	Expansion in sories and	s incordin.	orios Caughy's inequalities
11 1 V	Liouville's theorem Laurent	t's series isolated singularit	ies of an analytic function the
	zeros of analytic function th	he behaviour of analytic fu	action at isolated singularities
	limiting point of zeros or po	les the behaviour of an ana	lytic function near an isolated
	essential singularity		ay the remetion near an isolated
	urse p gramm urse Co urse O To un To de To ac integr To lea urse O 1: The 2: The 2: The 3: The 3: The dis: 2 tegory x. Man ock 1 it II it II	urse prerequisites: 10+2 with Mathogramme: B.Sc.         Subj         urse Code: DCEMM -113 (N)         urse Objectives:         To understand the basic concepts of To develop working skills with control acquire basic knowledge abou integrals by contour integration.         To acquire basic knowledge abou integrals by contour integration.         To learn the concepts and uses of colurse Outcomes:         D1: The student will get to know abou 22: The student shall understand the 33: The student shall understand the 34: The student shall understand the 35: The student will be able to understand the 36: The student will be able to understand the 36: The student will be able to understand the 37: The student will be able to understand the 38: The student will be able to understand the 39: The student will be able to understand the 39: The student will be able to understand the 39: The student will be able to understand the 30: The student will be able to understand the 30: The student will be able to understand the 39: The student will be able to understand the 30: The student will be able to understand the 30: The student will be able to understand the 30: The student will be able to understand the 30: The student will be able to understand the 30: The student will be able to understand the 30: The student will be able to understand the 30: The student will be able to understand the 30: The student will be able to understand the 30: The student will be able to understand the 30: The student will be able to understand the 30: The student will be able to understand the 30: The student will be able to understand the 30: The student will be able to understand the 30: The student will be able to understand the 30:	urse prerequisites: 10+2 with Mathematics         orgramme: B.Sc.       Year: 2023-24         Subject: MATHEMATICS         urse Code: DCEMM -113 (N)       Course Title: Function of         urse Code: DCEMM -113 (N)       Course Title: Function of         urse Objectives:         To understand the basic concepts of analytic function and Cauc         To develop working skills with complex integration, series and         To acquire basic knowledge about calculus of residues and         to acquire basic knowledge about calculus of residues and         To learn the concepts and uses of conformal mapping and Mobi         urse Outcomes:         P1: The student will get to know about analytic functions and Ca         22: The student shall understand the concept of complex integrat         33: The student will be able to understand the basic concepts and         Mobius (bilinear) transformation with their applications.         edits: 2       Type of Course: Core         tegory of Course         OER         x. Marks: 100       Min. Passing Marks: 36         Syllabi framed block wise/unit wise         order Series         Funct

Block 3	The Calculus of Residues (Integration) and Evaluation of real definite integrals	
	by contour integration	
Unit V	The calculus of Residues: The Residue at a singularity, Residue at infinity, calculation	
	of reduced in some special cases, Cauchy's theorem of Residues, poles and zeros of a	
	Meromorphic function. Rouche's theorem, applications of Rouche's theorem,	
Unit VI	Evaluation of Definite Integrals by Contour Integration: Jensen's Theorem,	
	Poisson's integral formula, The evaluation of integrals of the type $\int_{0}^{2\pi} f(\cos\theta, \sin\theta)d\theta$ ,	
	The evaluation of integrals of the type $\int_{-\infty}^{\infty} f(x) dx$ .	
Block 4	Conformal Representation	
Unit VII	Conformal Representation: Mappings by analytic functions, Differentiable arc	
	through a point, conformal mapping, existence of the inverse function, Conformal	
	character and analyticity, mapping by simple functions, super facial magnification, The	
	linear transformation, The Mobius (bilinear) transformation, The transformation	
	$\omega = \frac{1}{z}$ , Geometrical inversion, fixed points of a bilinear transformation.	
Suggested Text Book Readings:		
1. Ponnusa	amy, Foundations of Complex Analysis. 2 <sup>nd</sup> Edition, Narosa Book Publication, 2008.	
2. K.P. Gupta, Functions of complex variable, Sixteen Edition, Pragati Prakashan, 2002.		
3. J. B. Conway, Functions of One Complex Variable, Narosa Publishing House, New Delhi, 2002.		
4. Dennis G. Zill, Complex Analysis, Jones and Bartlett Publishers, 3ed		
5. V. Ahlfors, Complex Analysis (Third Edition), McGraw-Hill, 1979.		
6. M. Spiegel, J. Schiller, S. Lipschutz, Schaum's Outline of Complex Variables, 2ed (Schaum's		
Outlines)		
7. James W. Brown & R. V. Churchill: Complex variables and applications, McGraw-Hill, 2006.		
Suggested	online link:	
This cour	//epgp.initionet.ac.in/Home/ viewSubject?calid=ZLCHeZEnCZ8yCh30llSF3A==	
Suggester	se can be opted as an elective by the students of following subjects. NA	
https://arcl	nive notel ac in/courses/111/103/111103070/ by Prof P A S Sree Krishna IIT Guwahati	
<u>mups.//aici</u>	by 1101.1. A. S. Sice Anshina III Ouwahan	

Year	Sem.	Course Code	Paper Title	Theory/Practical	Credits	Max. Marks
		UGPHS -101N	VECTOR, MECHANICS AND GENERAL PHYSICS	Theory	2	100
		UGPHS -101P(N)	PRACTICAL WORK	Practical	2	100
1	П	UGPHS -102N	OSCILLATION, WAVES AND ELECTRICAL CIRCUITS	Theory	2	100
		UGPHS -102P(N)	PRACTICAL WORK	Practical	2	100
	III	UGPHS-103N	ELECTROMAGNETISM	Theory	2	100
		UGPHS-103P(N)	PRACTICAL WORK	Practical	2	100
		UGPHS-104N	ANALOG AND DIGITAL ELECTRONICS	Theory	2	100
2	IV	UGPHS -104P(N)	PRACTICAL WORK	Practical	2	100
		SKILL ENHANCEN	MENT COURSE			
		SBSPHS-02	MODERN PHYSICS	Theory	4	100
		Discipline Centrie	c Elective Course			
	v	DCEPHS -105N	OPTICS	Theory	2	100
		DCEPHS-106N	THERMAL PHYSICS	Theory	2	100
		DCEPHS-107P(N)	PRACTICAL WORK	Practical	2	100
		Discipline Centrie	c Elective Course			
3	VI	DCEPHS -108N	QUANTUM MECHANICS AND SPECTROSCOPY	Theory	2	100
		DCEPHS -109N	SOLID STATE PHYSICS AND ADVANCED ELECTRONICS	Theory	2	100
		DCEPHS -110P(N)	PRACTICAL WORK	Practical	2	100
			Total Credit		32	1500

Year: 2023-24 Syllabus of B.Sc Programme: Subject: PHYSICS [UGPHS]

<b>Programme</b> :	B.Sc. Year: 1 Semester: I		
Subject: Physics			
Course Code: UGPHS -101N Course Title: Vector, Mechanics and General Physics			
Course Obje	ctives:		
• The course	provides the students of core concepts of system of particles, motion, friction,		
work, energ	gy, planetary and satellite motion.		
• Some fund	amental concept of vector calculus.		
• The fundament	nental concepts make the students to find their interest so that they can explore in		
physics, an	d they can pursue their higher degree in physics.		
<b>Course Outc</b>	omes:		
CO.1 Unders	tand and illustrate various vector calculus method		
CO-2 Unders	tand and define the laws involved in mechanics		
CO-3 Gain de	exper understanding of mechanics and its fundamental concepts.		
<b>CO-4</b> Unders	tand the fundamental ideas on conservation of laws.		
CO-5 Unders	tand objects in space as they are introduced to planetary and Satellite motions		
Credits: 2	Type of Course: Core		
Max. Marks: 1	00 Min. Passing Marks: 36		
Block 1	Vector and mechanics		
	Vector analysis		
	• Scalar and vector, polar and axial vectors, Concept of tensor.		
	• Dot and cross product of two vectors, scalar and vector triple products		
Unit 1	• Gradient of scalar, divergence of vector, curl of vector, solenoidal and		
	lamellar vector.		
	• (Line, surface, volume) integral of vectors, Gauss, Stokes		
	and Greens theorem (onlystatement).		
	Vector identities (only statement)		
Unit 2	. Dynamics of a particle		
	• Force, momentum, impulse.		
	• Work, Power, energy.		
	• Conservative and non-conservative forces.		
	• Work-energy theorem, mechanical energy.		
	• Conservation of momentum and conservation of mechanical energy.		
	Elastic and inelastic head on collision.		
Unit 3	Angular and rotational motion		
	• Equation of motion and fundamental definitions.		
	• Angular momentum, torque, rotational K.E. angular impulse.		
	• Conservation of angular momentum and its applications.		
	• Moment of inertia, radius of gyration, theorem of parallel and		
	perpendicular axes.		
	• Expression for moment of inertia for ring and disc, hollow and solid spheres, hollow and solid splitter, this rod and plates.		
	(derivation is not required)		
	<ul> <li>Rolling without sliding and sliding without rolling motion, total kinetic</li> </ul>		

	energy.		
	<ul> <li>Motion of body along inclined plane in both cases.</li> </ul>		
	Dynamics of many particles		
	• Centre of mass and centre of gravity of a system		
	• Centre of mass and laboratory frame of reference.		
Unit 4	• Motion of centre of mass of a system.		
	• Linear momentum, angular momentum, torque, kinetic energy,		
	potential energy, mechanicalenergy for a system of particles.		
	• Difference between conservation laws (linear momentum, angular		
TT :	momentum, mechanicalenergy) for a particle and system of particles		
Unit 5	Dynamics of rigid body		
	• Concept of rigid body and its characteristics.		
	• Equations of rotational motion when the directions of angular		
	momentum coincide and donot coincide with axis of rotation.		
	• Relation between angular momentum, moment of inertia and angular		
	velocity in tensor form.		
	• Moment and product of inertia, inertia tensor.		
	Processional motion.		
Block-II	General Physics		
Unit 6	Gravitation		
	• Gravity and gravitation, inertial and gravitational mass.		
	• Variation of gravity with shape and rotation of earth, height and depth		
	from surface of earth.		
	<ul> <li>Gravitational field and potential due to spherical shell and solid sphere.</li> <li>Gravitational calf energy.</li> </ul>		
	• Gravitational self-energy.		
	• Orbital motion of satellite.		
	• Escape velocity of body.		
II.'. 7	Communication satellite and weightlessness condition.		
Unit /	Notion under central force		
	<ul> <li>Central force and its characteristics.</li> <li>Beduced mass. Reduction of two body central force problem to one</li> </ul>		
	<ul> <li>Reduced mass. Reduction of two body central force problem to one hody problem</li> </ul>		
	<ul> <li>Expression for transverse and radial acceleration of a body moving</li> </ul>		
	• Expression for transverse and radial acceleration of a body moving		
	<ul> <li>Acceleration of planet moving around sun</li> </ul>		
	<ul> <li>Kenler's laws of planetary motion (statement, derivation and</li> </ul>		
	applications)		
	<ul> <li>Expression for total energy of earth and condition to different paths</li> </ul>		
	<ul> <li>Newton's law of gravitation from Kepler's law.</li> </ul>		
Unit 08	Elasticity		
	• Kinetic model for solids (F-r and U-r graphs).		
	• Behavior of loaded wire (graphs and definitions).		
	• Poisson ratio, elastic constants and inter-relationship among them.		

	• Angle of twist and shear. Torsion of cylinder. Torsional rigidity.
	• Bending of beam, bending moment, geometrical inertia and flexural
	rigidity.
	• Centiliver (negligible weight and finite weight), expression for depression.
	• Elastic potential energy of stressed and twisted wire
Unit 9	Fluid mechanics and viscosity
	• Ideal fluid, critical velocity, stream line and turbulent motion.
	• Compressible and incompressible fluid, lamellar and
	nonlamellar motion, steady andvariable motion.
	• Equation of continuity and its significance.
	• Euler's equation and its application to deduce
	Bernoulli's equation, Application of Bernoulli's
	theorem (velocity of efflux, spinning of ball).
	• Newton's formula for viscous force. Kinematical and dynamical
	viscosity (CGS, MKS and SI units).
	• Poiseuille's law (statement, derivation, limitations), Series
	and parallel combinations of capillaries.
11.4.10	• Stokes's law for viscous force, terminal velocity.
Unit 10	Surface tension
	• Adhesive and conesive force. Shape of meniscus. Angle of contact.
	• Surface tension, surface energy, unison of small drops and bubbles.
	• Effect of temperature and impurity on surface tension and angle of
	contact.
	• Excess pressure inside air bubble and soap bubble.
	• Rise and fall of liquid inside capillary.
Suggested T	• Importance and application of capillarity.
Suggested 1	ext Book Readings:
An Int     Macha	roduction to mechanics, D. Kieppner, R.J. Kolenkow, 1975, McGraw-Hill.
Niecita     Dhugia	uncs, berkeley Physics, vol.1, C.Kittel, W.Kilight, et.al. 2007, Tata McGraw-Hill.
• Fliysic	is, Resilick, Hamuay and Walker o/e. 2006, whey.
• Anary	an Lactures, Vol. L. P. P. Fourman, P. P. Laighton, M. Sanda, 2008, Deerson
• Feynn Educa	tion
<ul> <li>Introdu</li> </ul>	uction to Special Relativity, R. Resnick, 2005, John Wiley and Sons
Univer	rsity Physics Ronald Lane Reese, 2003 Thomson Brooks/Cole
This course c	an be opted as an elective by the students of following subjects: NA
Suggested eq	uivalent online courses (MOOCs) for credit transfer: NA
Electronic m	edia and other digital components in the curriculum:
Choose any	one or more than: e-SLM/ Other electronic and digital contents
Name of elec	tronic media: e-SLM Year of incorporation: 2020

Programme: B	.Sc. Year:1 Semester:2		
	Subject: Physics		
Course Code:	Course Code: UGPHS-102N Course Title: Oscillation, Waves and Electrical Circuits		
Course Objec	tives:		
• The co circuits	urse provides the students of core concepts of oscillations, waves and electrical s.		
• The co	oncepts make the students to find their personal interest, so that they can explore		
their k	ey personal area in physics and pursue their masters in future.		
Course Outco	mes:		
CO-1 Underst	and in detail the concepts of oscillations.		
CO-2 Underst	and in detail the concepts of waves.		
CO-3 Recogn	ize basic terms in electrical circuits.		
CO-4 Apply th	neorems to construct and solve electrical circuits.		
CO-5 Ability	to design and conduct experiments as well as to analyze and interpret data		
CO-6 Buildup	strong problem-solving skills by effectively formulate a circuit problem into a		
mathematical	problem using circuit laws and theorems.		
Credits:2	Type of Course: Core		
Max. Marks: 1	00 Min. Passing Marks: 36		
Block 1	Mechanical oscillations		
Unit I Unit II	<ul> <li>Undamped oscillator</li> <li>Periodic motion and its classification.</li> <li>Electrical analogous of mechanical quantities.</li> <li>Undamped oscillations and its characteristics, kinematical and dynamical definition of SHM.</li> <li>Derivation of differential equation using energy consideration and its definition.</li> <li>Examples of SHM (mass-spring system, general pendulum compound pendulum, floatingcylinder, liquid column in U-tube), effective mass of spring.</li> <li>Damped oscillator</li> <li>Damped oscillation and its characteristics, comparison with undamped oscillation.</li> <li>Derivation of differential equation using energy consideration and its solution for heavy, critical and weak damping.</li> <li>Condition for oscillation, frequency of damped oscillation.</li> </ul>		
Unit III	<ul> <li>Relaxation time, energy dissipation, logarithmic decrement, quality factor.</li> <li>Forced oscillator</li> <li>Forced oscillations and its examples.</li> <li>Differential equation and steady state solution.</li> <li>Amplitude resonance and velocity resonance, mechanical impedance.</li> <li>Amplitude and velocity resonance frequency, phase difference among position velocity andforce.</li> <li>Power absorption and power dissipation.</li> <li>Quality factor, band width, sharpness of resonance.</li> </ul>		

	Coupled oscillator
	• Nature and condition of Lissajous figures (for 1:1 & 1:2 frequencies).
TT '4 TT 7	• Normal co-ordinate, degree of freedom, normal modes of vibrations.
Unit IV	Oscillations of two coupled masses.
	Oscillation of two coupled pendulums.
	• Energy of two coupled systems.
Block 2	Waves
Unit V	
	Wave motion
	Basic definitions, types of propagation, concept of phase.
	• Expression and properties of plane progressive wave.
	• Differential equation of wave motions, wave front.
	• Plane progressive wave in fluid and stretched string. Displacement wave and
	pressure wave.
	Plane progressive wave in stretched string.
	• Intensity and energy transportation in wave.
Unit VI	Waves at boundaries of two media
	• Free and bounded medium.
	Specific acoustic impedance, characteristic impedance.
	• Reflection and transmission coefficient of amplitude at joints of two
	media/strings.
	• Reflection and transmission coefficient of energy at joints of two media/strings.
	Discussion of results for various conditions of impedance of both media
Unit VII	Superposition of waves
	• Principles of superposition (statement, limitations, phenomenon observed).
	• Reflection of sound waves at free surface and rigid surface.
	• Stationary waves (formation and characteristics), SWR.
	• Mode of natural oscillations of stretched string and air column.
	• Fundamental frequency, harmonics and overtones.
	• Difference between interference and beats in sound.
Block III	Electrical circuits
Unit VIII	Transient phenomenon and galvanometer
	• Transient state and steady state, Time constant.
	• Transient response LR, CR, LC and LCR circuits.
	• Theory of moving coil galvanometer (dead beat and ballistic),
	critical resistance anddamping.
	• Sensitivity (current, charge and voltage) of moving coil galvanometer.
	• Applications to measurement of high resistance by leakage method.
Unit IX	Alternating current
	• J-Operator and phasor notations, reactance, impedance, susceptance,
	admittance.
	• Instantons, Peak, RMS and Average value of alternating voltage and current,
	Form factor.
	• Angle of lag and lead, wattful and wattless current, average power

	consumed (active reactive and apparent)	nowar factor	
	consumed (active, reactive and apparent), power factor.		
	• Phasor and vector diagram of CR, LR, LCR series, LCR parallel, LR		
	in series with C inparallel circuits.		
	• Parallel and series resonance, sharpnes	ss of resonance, Quality factor,	
	Bandwidth Resonancefrequency.		
Unit X Network analysis (For both AC and DC)			
Circuit elements and various networks circuits.		rcuits.	
	• T and $\pi$ networks and their equivalence.		
	• Kirchhoff's current and voltage laws. N	Mesh and nodal analysis of electrical	
	circuits.(Matrices and determinant metho	ods).	
	• Concept of constant current and constant	voltage source. Thevenin and Norton's	
	theorem.		
	• Maximum power transfer theorem, superposition theorem, reciprocity theorem		
Suggested Text Book Readings:			
1. Wave	1. Waves: Berkeley Physics Course, vol. 3, Francis Crawford, 2007, Tata McGraw-Hill.		
2. Funda	2. Fundamentals of Optics, F.A. Jenkins and H.E. White, 1981, McGraw-Hill		
3. Princip	3. Principles of Optics, Max Born and Emil Wolf, 7th Edn., 1999, Pergamon Press.		
4. Optics	4. Optics, Aiov Ghatak, 2008, Tata McGraw Hill		
5. The Pl	5. The Physics of Vibrations and Wayes, H. J. Pain, 2013, John Wiley and Sons.		
6. The Pl	6. The Physics of Waves and Oscillations, N.K. Bajaj, 1998, Tata McGraw Hill		
This course can be opted as an elective by the students of following subjects: NA			
Suggested equivalent online courses (MOOCs) for credit transfer: NA			
Electronic m	edia and other digital components in the	curriculum:	
<b>Choose any one or more than:</b> e-SLM/ Other electronic and digital contents		c and digital contents	
Name of electronic media: e-SLM Year of incorporation: 2020		Year of incorporation: 2020	
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Programme: B.Sc.		Year: 2 <sup>nd</sup>	Semester: 3 <sup>rd</sup>
		Subject: Physics	
Course Code:	UGPHS-103N	Course Title: Electromagnet	ism
Course Objec	tives:		
• To make students understand the electrostatic fields, potential, and capacitance, by			, potential, and capacitance, by
applyi	ng Coulomb's la	w and Gauss's law.	
• To ma	ke students unde	erstand the magnetostatic field	Is and inductance by applying Biot
Savart	's law and Ampe	re's law to find.	
To imp	part knowledge o	n the concepts of magnetostatic	cs, magnetic flux density, scalar and
		applications	··· ··· ··· ··· · · ··· · · · · · · ·
• 10 imp	bart knowledge o	n the concepts of Faraday's lav	w, induced emi and Maxwell's
equation	Dn		
Course Outco	mes:	athematical concepts related to	o alastromagnatia vastar fields
COI-Onderst	and the basic ma	all estructures to the solutions	of much lama malating to algorithm field
CO2- Apply t	ne principies of	electrostatics to the solutions (	of problems relating to electric field
and electric po		y conditions and electric ener	gy density.
CO3- Apply t	ne principles of	magneto statics to the solution	is of problems relating to magnetic
field and mag	netic potential, t	oundary conditions and magn	letic energy density.
CO4- Underst	and the concept	s related to Faraday's law, inc	duced emf and Maxwell's equations
CO5- Apply N	Maxwell 's equat	tions to solutions of problems	relating to transmission lines and
uniform plane	wave propagati	on.	
Credits: 2 Type of Course: Core			
Max. Marks: 1	00	Min. Passing Marks: 36	
Block I	Electrostatics		
	Electric charge	e, force and fields	
	Concept of	charge, Coulomb's law, elect	ric field, electric flux.
	• Gauss law	(statement and derivation, inte	egral and differential form).
Unit I	<ul> <li>Application</li> <li>Coulomb's</li> </ul>	n of Gauss law for charge distr	ribution (linear, cylindrical, spherical).
	<ul> <li>Coulonio s</li> <li>Electric field</li> </ul>	Id due to charged ring, charged	d infinite rod and charged disc
	from Could	omb'slaw	i minite fou and charged dise
	<ul> <li>Laws of ele</li> </ul>	ectrostatics.	
Unit II	. Electric poter	itial and dipole	
	• Electric po	tential and electrostatic potent	ial energy.
	• Electric fie	lds, potential gradient and the	ir relationship.
	• Electrostati	c self-energy (conducting and	l dielectric sphere).
	• Electric po	tential due to spherical charge	ge distribution (hollow
	and solid),	graphicalrepresentation.	
	• Electric dip	ole and its behavior in uniform	m and non-uniform electric field.
	• Electric fie	d and potential due to electric	dipole at a point in Cartesian and polar
	coordinates	<b>.</b>	

• Force between two electric dipoles.

Unit III	Dielectrics	
	• Capacitor and its capacity, principle of capacitor, energy stored in field of capacitor	
	• Capacity of partially filled parallel plate capacitor, expression for induced	
	charge.	
	• Effect of dielectrics slab introduced inside plates of charged capacitor when its remainsconnected with battery and when it is disconnected	
	from battery.	
	<ul> <li>Change in electrical properties when N small charged drops coalesce to form a</li> </ul>	
	large drop.	
	<ul> <li>Three electric vectors (<b>D</b>, <b>E</b>, <b>P</b>), dielectric constant, dielectric</li> </ul>	
	strength, electricalsusceptibility.	
	• Polarization, surface and volume charge density, Gauss law in dielectrics.	
	Macroscopic and microscopic properties of dielectrics. Clausius – Mossotte     formula	
Block 2	Magnetostatics	
Unit IV	Electric current and magnetic fields	
	<ul> <li>Electric current and current density. Ohm's law and Joule's law, drift velocity.</li> <li>Magnetic field around stationary charge, moving charge and current carrying</li> </ul>	
	conductor.	
	• Biot-Savart law and its application to straight conductor, circular loop,	
	<ul> <li>Magnetic field due to moving charge. Lorentz force</li> </ul>	
	France between two moving charge, Electric force	
	<ul> <li>Force between two current carrying conductor and two moving charges.</li> <li>Cyclotron (principle, construction, working, limitations and modification), Distance</li> </ul>	
Unit V	Belalfon.	
Unit v	• Lines of forces Gauss law in magnetostatics	
	<ul> <li>Ampere circuital law (statement and derivation), its applications to</li> </ul>	
	current carrying rod(hollow and solid).	
	• Inconsistency of Ampere circuital law with equation of continuity.	
	• Modification of Ampere circuital law by Maxwell with introducing concepts of	
	displacement currents and its importance. Comparison of displacement current	
	and conduction current.	
	<ul> <li>Vector potential and its expression due to straight conductor and circular loop.</li> <li>Derivation of magnetic flux density using vector notantial for singular loop.</li> </ul>	
Unit VI	Derivation of magnetic flux density using vector potential for circular loop.     Magnetic materials	
	• Magnetic properties (magnetic flux density B magnetizing	
	field H, intensity of magnetization Im, susceptibility, relative	
	and absolute permeability).	
	• Magnetization, cycle of magnetization, hysteresis loop, retentivity, residual	
	magnetism.	
	• Three magnetic vectors (B, H, Im), three magnetic currents (free, bound and	

	total).	
	Curl of intensity of magnetization.	
Block III	Electromagnetic Phenomenon	
Unit VII	Electromagnetic induction	
	<ul> <li>Faraday's law of electromagnetic induction (statement, integral form, differential form) andanalogy with Newton's laws of motion in mechanics.</li> <li>Condition for existence and depending factors of induced charge, induced voltage, inducedcurrent and induced power.</li> <li>Dynamic induced EMF and derivation of its expression,</li> <li>Self and mutual induction and inductance, static induced EMF (self and mutual).</li> <li>Reciprocity theorem and Neuman's relation.</li> <li>Relation between self and mutual inductance of two coupled coils, energy of coupledcircuits.</li> <li>Transformer and its equivalent circuit, condition for ideal transformer (expression forefficiency and voltage gain) transformer losses</li> </ul>	
BLOCK IV	Electromagnetic Theory	
Unit VIII	Fundamental equations	
	<ul> <li>Four Maxwell's equations (statement and physical significance).</li> <li>Maxwell's equations and features of their general plane wave solution in source free space.</li> <li>Maxwell's equations and features of their general plane wave solution in simple dielectrics.</li> </ul>	
	<ul> <li>Differential equation and velocity for electromagnetic waves in source free space anddielectric medium.</li> <li>Characteristics of electromagnetic waves, impedance, refractive index.</li> <li>Skin depth and its importance.</li> </ul>	
Unit IX	Fnergy and momentum of an electromagnetic wave	
	<ul> <li>Differential equation of plane electromagnetic wave</li> <li>Differential equation of plane electromagnetic waves in conducting media and its solution.</li> <li>Behavior and property of electromagnetic waves for good dielectric and good conductors.</li> <li>Poynting theorem (statement and derivation).</li> <li>Expression for electromagnetic energy density.</li> <li>Momentum density vector and its importance</li> <li>Maxwell's stress tensor (statement and derivation).</li> </ul>	
Unit X	Fresnel's equation	
	<ul> <li>Boundary conditions at discontinuity for D, E, B and H.</li> <li>Reflection and refraction at normal and oblique incidence of electric vectors perpendicular toboundary.</li> <li>Reflection and refraction at normal and oblique incidence of electric</li> </ul>	

vectors parallel toboundary.		
• Total internal reflection, Brewster's law, degree of polarization.		
• Plane wave propagation in plasma and its properties (qualitative), metalli		
reflection.		
Elementary theory of dispersion.		
Suggested Text Book Readings:		
1. Electricity and Magnetism, Edward M. Purcell, 1986, McGraw-Hill Education		
2. Electricity & Magnetism, J.H. Fewkes & J.Yarwood. Vol. I, 1991, Oxford Univ. Press		
3. Electricity and Magnetism, D C Tayal, 1988, Himalaya Publishing House.		
4. University Physics, Ronald Lane Reese, 2003, Thomson Brooks/Cole.		
5. D.J.Griffiths, Introduction to Electrodynamics, 3rd Edn, 1998, Benjamin Cummings.		
This course can be opted as an elective by the students of following subjects: NA		
Suggested equivalent online courses (MOOCs) for credit transfer: NA		
Electronic media and other digital components in the curriculum:		
Choose any one or more than: e-SLM/ Other electronic and digital contents		
Name of electronic media: e-SLMYear of incorporation: 2021		

Programme: B.Sc.	Year:2 <sup>nd</sup>		Semester:4 <sup>th</sup>
	Subject	t: PHYSICS	
Course Code: UGPHS -104N	Course Title: Analog and Digital Electronics		
Course Objectives:			
• This course helps the stud	dents to gain basic id	eas of the construct	tion and working of electronic
devices and circuits.			
• The aim of this course is	to make students acq	uire knowledge ab	out Boolean algebra, logic Circuits.
Course Outcomes:			
CO-1- Be familiar with the	basic concepts of con	nstruction and work	ting of electronic devices and
Bipolar junction transistor.			
CO-2- Apply the knowledge to understand the working of amplifiers and oscillators			
CO-3- Apply the knowledge to understand the working of special types of Diodes			
CO-4- Apply the principles of feedback in amplifiers and oscillators			
CO-5- Understand the concepts and techniques in digital electronics.			
CO-6- Understand various number systems and their importance in digital designing.			
CO-7- Analyze and construct various digital circuits.			
CO-8- Design combination and sequential circuits.			
Credits:2		Type of Course: Co	re
Max Marks: 100	Min Passing Marks	· 36	
(Syllabi framed block wise/unit wise)			

Block 1	Electron devices
	Semi-conductor physics
	Band theory of solids and classification of solids on its basis. Intrinsic and attrinsic
	(n-type p-type) semiconductors
	<ul> <li>Conductivity mobility drifts motion and diffusion motion of free electron and holes</li> </ul>
	<ul> <li>conductivity, mobility, drifts motion and diffusion motion of free election and notes.</li> <li>p-n junction (formation of depletion region and potential barrier)</li> </ul>
Unit I	<ul> <li>Forward and reverse biasing of p-n junction forward and reverse current</li> </ul>
	<ul> <li>Diode equation and characteristics, static and dynamics, resistance, knee voltage</li> </ul>
	<ul> <li>Breakdown mechanism (Zener and Avalanche) transition and storage capacitance</li> </ul>
	<ul> <li>Dreakdown mechanism (Zener and Available), transition and storage capacitance.</li> <li>Zener diode (statement, characteristics and comparison with p-n junction)</li> </ul>
	<ul> <li>Zener diode as voltage regulation (circuit and analysis)</li> </ul>
Unit II	Zener under as vonage regulation (circuit and analysis).
Omt n	<ul> <li>Non regulated and regulated power supply</li> </ul>
	<ul> <li>Circuit and mathematical analysis of rectifier (HWR EWR and BR)</li> </ul>
	<ul> <li>Ripple factor, rectification constant, voltage regulation and efficiency of rectifier.</li> </ul>
	<ul> <li>Filtering by RL RC and LC circuit (only qualitative)</li> </ul>
	<ul> <li>Photonics device I ED (principles and applications)</li> </ul>
	<ul> <li>Photodiode and photo transistor, photoconductivity</li> </ul>
	<ul> <li>Solar cell (principle, construction, working and characteristics)</li> </ul>
Unit III	Binolar junction transistor
Omem	NPN and PNP transistor and their action
	<ul> <li>Types of configurations region of operation, thermal runaway</li> </ul>
	<ul> <li>Characteristics and parameters of CE_CB and CC configuration of transistor</li> </ul>
	<ul> <li>Emitter efficiency base transport ratio and current gain in CB configuration</li> </ul>
	<ul> <li>Leakage current hybrid parameters</li> </ul>
	<ul> <li>Transistor biasing and their merits &amp; demerits thermal stability</li> </ul>
	<ul> <li>DC and AC current gains in all three configurations of transistors and their inter-</li> </ul>
	relationship of transistors
Unit IV	Uninolar transistors
Child I V	• FET (construction, classification, symbol, principle of operation).
	• Internal and external biasing of FET. Fundamental definition related with FET.
	• Characteristics and parameter of FET.
	• Comparison of FET with BJT.
	• MOSFET (enhancement and depletion mode), construction, symbol and working.
	• NMOS and PMOS. CMOS as switch. BJT as switch.
	• Storage and transition time (definition and importance).
	• Schottky diode and Schottky transistors (statement, symbol and importance).
Block 2	Electronic circuits
Unit V	Amplification
	• Introduction to different gains (impedance, current, voltage and power).
	• Small signal hybrid equivalent circuits of transistor in three configurations.
	• Inter-relationship among hybrid parameters in CF CR and CC configuration of
	transistors.

	Importance of voltage divider biasing of transistor, operating points.		
	DC and AC load lines (statement and analysis).		
	• Classification of amplifiers on the basis of coupling, range of		
	operations, uses and frequency.		
Unit VI	Voltage and power amplifier		
	• RC coupled amplifier (single and multistage), components and their functions.		
	• Analysis of frequency (low, medium and high) response curve with the help of		
	equivalent circuits.		
	Push-pull amplifier (characteristics, advantages and disadvantages).		
	• Single and double tuned amplifier (characteristics and importance).		
	• Audio and radio frequency amplifier (characteristics and importance)		
Unit VII	Oscillator		
	• Feedback amplifier (positive and negative), open and closed loop gains.		
	• Merits and demerits of negative feedback amplifier over positive amplifier.		
	• Barkhausen criterion for sustained oscillation (statement and proof).		
	Component of oscillator and their functions		
	• Tuned collector, Hartely and Colpit oscillator (circuit and working).		
	• RC oscillator, phase shift oscillator and Wiens bridge oscillator.		
	Crystal oscillator.		
Block III	Digital Electronics		
Unit VIII	Number system and codes		
	• Number system (decimal binary octal and heya-decimal) radiy		
	<ul> <li>Rules for interconversion of one number, system into other number systems</li> </ul>		
	<ul> <li>Rules for interconversion of one number system into other number systems.</li> <li>1s and 2s compliments of binary numbers. Binary arithmetic's</li> </ul>		
	<ul> <li>Is and 2s compliments of onlary numbers. Dinary antimictic s.</li> <li>Different types of codes (PCD code, Excess 2 codes, Cray code, ASCII code)</li> </ul>		
	• Different types of codes (BCD code, Excess 5 codes, Grey code, ASCII code, ERICDIC code and error code)		
Unit IX	Boolean algebra and logic gates		
	Boolean algebra and its features		
	<ul> <li>Logic gates (Switching circuit Truth table Venn diagram Boolean function)</li> </ul>		
	<ul> <li>Design gates (Switching circuit, Truth table, Venn diagram, Boolean function).</li> <li>Primary gates (AND, OR, NOT) and their representation using p-n diodes &amp;</li> </ul>		
	• Transistors		
	• Universal gates (NAND, NOP) and realization of other gates using these		
	<ul> <li>Universal gates (IVAND, NOR) and realization of other gates using these.</li> <li>Evolutive gates (VOP, VNOP) acquivelent and non acquivelent gates, characteristics.</li> </ul>		
	• Exclusive gates (AOK, ANOK), equivalent and non-equivalent gates, characteristics		
	• AND OP OP AND NAND NOP NOP realization of Boolean expression		
Unit Y	AND-OK, OK-AND, NAND-NAND, NOK-NOK realization of Boolean expression.      Boolean Theorems and combinational logics		
Unit A	DOULTAIL FILEDICIUS AUD COMDINALIONALIOPICS		
	<ul> <li>De Morgan's laws, commutative laws, associative laws, distributive laws</li> </ul>		
	<ul> <li>De-Morgan's laws, commutative laws, associative laws, distributive laws, absorptive laws of Boolean algebra</li> </ul>		
	<ul> <li>De-Morgan's laws, commutative laws, associative laws, distributive laws, absorptive laws ofBoolean algebra.</li> <li>Dual and compliment of Boolean function</li> </ul>		
	<ul> <li>De-Morgan's laws, commutative laws, associative laws, distributive laws, absorptive laws ofBoolean algebra.</li> <li>Dual and compliment of Boolean function.</li> <li>Minterms, maxterms, SOP form, POS form of Boolean functions.</li> </ul>		
	<ul> <li>De-Morgan's laws, commutative laws, associative laws, distributive laws, absorptive laws ofBoolean algebra.</li> <li>Dual and compliment of Boolean function.</li> <li>Minterms, maxterms, SOP form, POS form of Boolean functions.</li> <li>Minterms and maxterms are conjugate to each other.</li> </ul>		
	<ul> <li>De-Morgan's laws, commutative laws, associative laws, distributive laws, absorptive laws ofBoolean algebra.</li> <li>Dual and compliment of Boolean function.</li> <li>Minterms, maxterms, SOP form, POS form of Boolean functions.</li> <li>Minterms and maxterms are conjugate to each other.</li> <li>Kompush's monning and its memoratorian (for 2, 2 and 4 variables) in terms of a second s</li></ul>		

	<ul> <li>Minterms and maxterms.</li> <li>Simplification and Boolean expression by Boolean laws and K-mapping.</li> <li>Half and full adder, half and full subtractor.</li> </ul>
Suggested Text Book Readings:         1. Integrated Electronics, J. Millman and C.C. Halkias, 1991, Tata Mc-Graw Hill.	

2. Electronics: Fundamentals and Applications, J.D. Ryder, 2004, Prentice Hall.

- 3. Solid State Electronic Devices, B.G.Streetman & S.K.Banerjee, 6th Edn., 2009, PHI Learning
- 4. Electronic Devices & circuits, S.Salivahanan & N.S.Kumar, 3rd Ed., 2012, Tata Mc-Graw Hill
- 5. OP-Amps and Linear Integrated Circuit, R. A. Gayakwad, 4th edition, 2000, Prentice Hall
- 6. Electronic circuits: Handbook of design & applications, U.Tietze, C.Schenk, 2008, Springer
- 7. Semiconductor Devices: Physics and Technology, S.M. Sze, 2nd Ed., 2002, Wiley India
- 8. Electronic Devices, 7/e Thomas L. Floyd, 2008, Pearson India

This course can be opted as an elective by the students of following subjects:NA

Suggested equivalent online courses (MOOCs) for credit transfer: NA

Electronic media and other digital components in the curriculum:		
<b>Choose any one or more than:</b> e-SLM/ Other electronic and digital contents		
Name of electronic media: e-SLM	Year of incorporation: 2021	

Programme: H	3.Sc. Year:2 <sup>nd</sup> Semester:4 <sup>th</sup>			
Subject: PHYSICS				
Course Code: S	SBSPHS-02 Course Title: MODERN PHYSICS			
Course Objectives:				
The aim of th	is course is to make the students learn and discuss about the Modern Physics.			
Course Outc	comes:			
CO-1 Unders	tand the concept of Reference point.			
CO-2 Unders	tands the concept of Special and General theories of relativity.			
CO-3 To esta	blish a relationship between mass and energy.			
CO-4 Descrit	be the series of spectra and know Frank-Hertz experiment			
Credits: 4	Type of Course: Skill Enhancement Course			
Max. Marks:	100 Min. Passing Marks: 36			
	(Syllabi framed block wise/unit wise)			
Block I	Special Theory of Relativity			
	Emergence of special relativity:			
	• Frame of reference (inertial and non-inertial), Events (simultaneous,			
	colocal andcoincidence)			
Unit I	Centripetal force, centrifugal force, and Coriolis force.			
	Classical relativity, Galilean variant and Galilean in-variant.			
	• Compatibility of electromagnetism with principle of relativity and mechanics.			
	Michel Jon-Morley experiment-significance of negative result.			
<b></b>	Postulates of special theory of relativity.			
Unit II	Relativistic kinematics:			
	• Lorentz transformations (statements and derivation).			
	• Relativity of simultaneity and length contraction.			
	• Relativity of co locality and time dilation.			
	• Experimental verification of length contraction and time dilation.			
Relativistic transformation of velocity, resultant of two successi				
transformations.				
• Relativistic velocity addition theorem (statement, derivations and applications).				
	• Aberration of stars (statement, derivation and comparison with classical result).			
	• Relativistic Doppler effect (statement, derivation and discussion of result).			
Unit III	Relativistic dynamics:			
	• Non-relativistic and relativistic particles.			
	• Einstein's mass and energy equivalence relation, relativistic kinetics energy.			
	• Variation of mass with velocity (only qualitative) and its importance.			
	• Fundamental equations of relativistic motion, longitudinal and transverse mass.			
	Momentum and energy transformation equations.			
	Minkowski time space diagram and its applications.			

Block II	Atomic Physics		
Unit IV	Atomic models:		
	• Bohr's theory of hydrogen likes atoms, Bohr radius, Sommerfeld fine structure		
	constant, Rydberg & Rydberg Constant, Binding energy.		
	• Spectral series of hydrogen atom, $H_{\alpha}$ , $H_{\beta}$ , $H_{\gamma}$ , H Balmer lines		
	<ul><li>Reduced mass, effect of nuclear motion, isotopic shift,</li><li>Ground, Excited and ionized state, emission and absorption spectra</li></ul>		
	• Excitation, resonance and ionization potential, ionization energy of atom		
	Bohr's correspondence principle (statement, proof and importance)		
	Qualitative discussion of sommerfeld atom model.		
Unit V	X-Ray spectra		
	Production of X-rays (qualitative discussion of Roentgen tube and Coolidge tube)		
	Properties and application of X-rays		
	Continuous and characteristic X-rays, Bremstrahlung radiation		
	• Continuous and line X-ray spectra (K-series, L-series, M-series)		
	• Duane-Hunt's law, cutoff frequency and cutoff wavelength,		
	• Moseley's law (statement, derivation and applications), absorption edge		
	Comparison of optical and X-rays spectra		
Unit VI	Atomic structure:		
	• Vector atom model (need, statement and importance)		
	• Space quantization, concept of electron spin and quantum numbers		
	• Stern-Gerlach experiment (principal theory and importance of results)		
	• Magnetic moment of atom, Bohr magneton, Gyro magnetic ratio,Larmor precession and frequency		
	• Intensity rules, selection rules, spectral terms, sodium $D_1$ and $D_2$ lines, Fine		
	structure of Halines,		
	• Coupling scheme (L-S and j-j), spectra of alkali and alkaline earth elements.		
Unit VII	Dualism nature:		
	• Planck's quantum theory and Einstein modifications, Photon and its characteristics		
	• Photoelectric effect (statements laws and mathematical explanation), quantum efficiency		
	• Compton effect (statement and explanation, expression for Compton shift and recoil		
	energy),		
	• Dualism in nature, de-Broglie hypothesis, matter waves and its importance.		
	Comparison of matter waves with electromagnetic waves and mechanical waves		
	• Davisson-Germen experiment (principle, working and importance of result)		
	• Wave packet, phase velocity and group velocity, wave and particle velocity,		
	relation among them		
	Uncertainty principle (statement, significance and application)		
Block III	Nuclear physics		
Unit VIII	Radioactivity:		
	• Natural and artificial radioactivity, emission of alpha particle, electron, positron		
	and gammaparticles		

	• Size of nucleus, classification of nuclei (isotopes, isobars, isotones,			
	Isomers andisodiapheres).			
	<ul> <li>Radioactive series, successive radioactive decay, radioactive equilibrium</li> <li>Forth define and each on define, artificial nuclear transmutation</li> </ul>			
	<ul> <li>Earth dating and carbon dating, artificial nuclear transmutation</li> <li>Discovery of neutrons and radioisotons in everyday life</li> </ul>			
	<ul> <li>Discovery of neutrons and radioisotope in everyday life</li> <li>Nuclear force and its Yukawa (Massa) theory</li> </ul>			
LL 4 IV	Nuclear force and its Yukawa (Meson) theory.			
Unit IX	Nuclear energy:			
	<ul> <li>Mass delect, packing fraction, binding energy, specific binding energy,</li> <li>Diading energy energy englanction of medical fraction medical fraction on the termination of medical fraction.</li> </ul>			
	• Binding energy curve, explanation of nuclear fission, nuclear fusion and release of nuclearenergy,			
	• Kinematics of nuclear reaction, Q-value of reactions			
• Bohr's-Wheeler model, activation and excitation energy, normal and enriched				
	Uranium,			
	• Liquid drop model, semi-empirical mass formula,			
	• Shell model, magic number, collective model			
Unit X	X Elementary particles			
	• Classification of elementary particles on the basis of mass, spin and interaction,			
	• Particles and anti-particles.			
	• Process of annihilation and process of production of matters,			
	• Quantum number (Lepton number, Baryon number, iso-spin number, hyper			
	charge number, strange number)			
	Conservation laws and concept of Quarks.			
Suggested Text Book Readings:				
1. Introd	actory nuclear Physics by Kenneth S. Krane (Wiley India Pvt. L	.td., 2008).		
2. Conce	pts of nuclear physics by Bernard L. Cohen. (Tata Mcgraw Hill,	, 1998).		
3. Introduction to the physics of nuclei & particles, R.A. Dunlap. (Thomson Asia, 2004).				
4. Introduction to High Energy Physics, D.H. Perkins, Cambridge Univ. Press				
5. Introduction to Elementary Particles, D. Griffith, John Wiley & Sons				
6. Quarks and Leptons, F. Halzen and A.D. Martin, Wiley India, New Delhi				
7. Basic ideas and concepts in Nuclear Physics - An Introductory Approach by				
K. Heyde (IOP- Institute of Physics Publishing, 2004).				
8. Radiation detection and measurement, G.F. Knoll (John Wiley & Sons, 2000).				
This course can be opted as an elective by the students of following subjects:NA				
Suggested equivalent online courses (MOOCs) for credit transfer: NA				
Name of electronic media: e-SLMYear of incorporation: 2021				
Programme: B	.Sc.	Year:3 <sup>rd</sup>	Semester:5 <sup>th</sup>	
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Subject: PHYSICS				
Course Code: DCEPHS- 105N				
Course Objectives:				
• The aim of this course is to make the students learn and discuss about the Ontics and				
their re	their related experiments.			
• Studen	ts also correlate	them with the correspon	nding theory, through the standard set of	
experin	ments.	L		
Course Outco	mes:			
CO-1 Student	will learn basics	of Optics.		
CO-2. To impa	art basics knowle	edge of laser and hologr	aphy	
CO-3. Student	will familiar wit	th nature of light.		
Credits:2	00	IV	pe of Course: ELECTIVE	
Max. Marks: 1	00 should be frame	Min. Passing Marks:	20 • No of blocks and units may change)	
Block 1	Geometrical a	and Quantum ontics	, ito of blocks and units may change)	
DIOCK I	Co-axial system	m of lenses		
	Cardinal pe	oints (focal points, prin	ncipal points and nodal points).	
	Analytical methods for analysis of cardinal points.			
Unit I	Matrix met	thods for analysis of ca	ardinal points.	
	• Equivalent	Equivalent lens, problems on combination of thin lenses.		
	• Eye pieces (Ramsdon and Huygens), Ray diagram and characteristics, n			
	and demerits.			
	Aplanatic points and its importance			
Unit II	Unit II Laser and holography:			
	• Coherence	(Temporal and Spatial	1)	
	• Stimulated	and spontaneous em	ission, Einstein co-efficient and their inter-	
	relationshi	р		
	• Basic idea	about laser and its con	ponents, pumping and population inversion	
	Compariso	n of laser light and ord	linary light	
	• Ruby laser	, Helium-Neon laser, s	emiconductor laser and their applications	
	Holograph	y and hologram, comp	arison of hologram with photography.	
	Recording	and reconstruction of	hologram and its applications	
Unit III	. Fiber optics			
	Construction	ons and materials used	in optical fibers,	
	Principle o	t fiber optics and prop	agation of light in optical fiber,	
	Advantage	s and disadvantages of	optical fiber communication	
	• Numerical	aperture, acceptance a	ngle, V-parameters, meridional and skew	
	rays' analy	SIS		
	• Types of fi	bers (SIF, GIF, Single	mode and multimode), fiber profile.	
	• Phase inde	x and group index in o	ptical fiber, slowest and fastest	
	• Attenuation	opagation moptical fit	1018, ical fibers parameter (attenuation loss	
	<ul> <li>Types of fi</li> <li>Phase inde mode of pr</li> <li>Attenuation</li> </ul>	bers (SIF, GIF, Single x and group index in o ropagation inoptical fit	mode and multimode), fiber profile. ptical fiber, slowest and fastest pers, ical fibers parameter (attenuation loss	

	dispersion)		
	• Qualitative discussion of coupler, splices and connecter.		
Block 2	CONCEPT OF LIGHT		
Unit IV	Nature of light:		
	<ul> <li>Statement, merits and demerits of Newton's corpuscular theory, Huygens longitudinalwave's theory and Fresnel transverse wave theory.</li> <li>Huygens principle and its explanation, laws of reflection and refraction.</li> <li>Electromagnetic wave theory (statement, consequences and limitations)</li> <li>Fermat's principles (statement and applications).</li> <li>Perception of light (human vision, color vision and color receptor)</li> <li>Scattering of light and its importance</li> </ul>		
Unit V	Concept of polarization:		
	<ul> <li>Cause and concept of polarization, plane of vibration and plane of polarization</li> <li>Un-polarized light and types of polarized light (linear, circular and elliptical).</li> <li>Plane polarized light by reflection and refraction, Brewster's law, piles of plates</li> <li>Plane polarized light by selective absorption (dichroism) and double reflection (E &amp; Orays), birefringence.</li> </ul>		
	• Polarizer and analyzer, Nicol prism, law of Malus		
	• Huygens theory of double reflection by uniaxial crystal, negative and positive		
	<ul> <li>crystal, optic axis</li> <li>Superposition of two plane polarized lights along mutually perpendicular directions</li> </ul>		
Unit VI	Detection of polarized light:		
	<ul> <li>Retardation plates (quarter and half wave plates), features and applications.</li> <li>Production and detection of elliptically and circularly polarized light</li> <li>Analysis of different polarized light.</li> <li>Babinet compensator (principle, theory, application).</li> <li>Optical rotation and specific rotation, optical activity</li> <li>Fresnal theory of optical rotation.</li> <li>Polarimeters (Half Shade and Biquartz), their merits and demerits.</li> </ul>		
Block III	Interference and diffraction		
Unit VII	<ul> <li>Concept of interference</li> <li>Statement and essential conditions for observation of interference</li> <li>Constructive and destructive interference, shape of fringes, visibility of fringes</li> <li>Youngs double slit experiment, shift in fringes pattern due to thin sheet of transparent material.</li> <li>Fresnel biprism, Fresnel bimirror, non-localized fringes.</li> <li>Lloyds single mirrors, achromatic fringes.</li> <li>Formation of coherent sources due to division of wave front in above devices and comparison of their fringe pattern.</li> <li>Fringes pattern with white light.</li> </ul>		

Unit VIII	Interference by division of amplitudes		
	Stokes analysis of phase change on reflection		
	• Color in thin films (parallel and Wedge shaped) due to white light.		
	Newton's rings (principle, theory and applications)		
	Haidenger fringes, localized fringes.		
	• Michelson interferometer (principle and working), conditions for different shape of fringes comparison with Newton's ring fringes		
	<ul> <li>Febry-Perot interferometer intensity distribution coefficient of finesse</li> </ul>		
	visibility of fringes, sharpness of fringes, superiority over Michelson		
	interferometer fringes,		
	• L-G plates		
Unit IX	Fresnel diffraction		
	• Difference between interference and diffraction.		
	Classification of diffraction and their conditions.		
	• Fresnel construction of half period's zones, rectilinear propagation of light.		
	• Diffraction at straight edge and circular aperture.		
	• Zone plate and its comparison with convex lens		
	• Cornu's spiral (theory, applications and merit)		
Unit X	Fraunhoffer diffraction:		
	• Single slit fraunhoffer diffraction (theory and graph for maxima and minima)		
	• Double slit fraunhoffer diffraction (theory and importance), missing spectra.		
	• Plane diffraction grating (principle and analysis), condition for absent spectra		
	Grating spectra and prism spectra.		
	• Concave diffraction grating (theory and types), superiority over plane grating.		
	• Rayleigh criterion of resolution, limits of resolution of eye.		
	• Resolving power of Grating, Prism, Telescope, and Microscope.		
Suggested Te	ext Book Readings:		
1. Funda	mental of optics, F. A. Jenkins & H. E. White, 1981, Tata McGraw hill.71		
2. LASERS: Fundamentals & applications, K.Thyagrajan & A.K.Ghatak, 2010,			
3. Tata McGraw Hill			
4 Fibre ontics through experiments M R Shenov S K Khijwania et al 2009 Viva Books			
5 Nonlinear Optics, Robert W. Boyd, (Chapter-I) 2008, Elsevier			
6 Ontice Karl Dieter Moller Learning by computing with model examples 2007 Springer			
7. Optical Systems and Processes. Joseph Shamir. 2009. PHI Learning Pvt Ltd			
8. Optoelectronic Devices and Systems S.C. Gupta 2005 PHI Learning Pvt 1 td			
9 Ontical Physics A Linson S G Linson H Linson 4th Edn 1996 Cambridge Univ Press			
Suggested equivalent online courses (MOOCs) for credit transfor: Ontics antel courses			
https://youtu be/U20hpa27mm4			
<u>maps.//youtu.</u>			
Electronic m	edia and other digital components in the curriculum:		
Choose any	one or more than: e-SLM/ Other electronic and digital contents		
Name of elec	tronic media: e-SLM Year of incorporation: 2022		

Programme: E	3.Sc. Year: 3 <sup>rd</sup> Semester:5 <sup>th</sup>			
Subject: PHYS	SICS			
Course Code: DCEPHS-106N Course Title: THERMAL PHYSICS				
Course Objectives: The aim of this course is to make the students learn and discuss about the				
Thermodynamics.				
Course Outcomes:				
CO-1 Student	will learn basics of Thermodynamics.			
CO-2. To imp	art basics knowledge of laws of Thermodynamics.			
CO-3. Student	will be familiar with term Entropy.			
Crodita: 2				
Cleans. 2	Type of Course. Elective			
Max. Marks: 1	100 Min. Passing Marks: 36			
(Syllabi	should be framed block wise/unit wise; No of blocks and units may change)			
Block 1	Thermodynamics			
	Fundamental of thermodynamics			
	• Thermodynamic systems, thermodynamic variables.			
	• Thermodynamic equilibrium (thermal, mechanical and chemical).			
Unit I	• Equation of state, equation of constraints.			
Unit I	• Zeroth law of thermodynamics, concept of temperature.			
	• Macroscopic and microscopic variables, extensive and intensive variables.			
	• Quasi-static and non-quasi-static processes. reversible and irreversible			
	processes.			
	Conditions for reversibility.			
Unit II	Unit II First law of thermodynamics			
	• Expression for thermal work done. Types of work done.			
	Point function and path function.			
	• Internal energy and enthalpy both as point function.			
	• First law of thermodynamics (statement, derivation, limitations and			
	applications).			
	• Degrees of freedom, atomic heat ratio, Mayer's relation.			
	• Processes (cyclic, isobaric, isochoric, isothermal and adiabatic).			
	P-V indicator diagram and its importance.			
Unit III	nit III Second law of thermodynamics			
	• Need of second law.			
	• Carnot cycle (P-V and T-S diagram).			
	Carnot heat engine and Refrigerator.			
	Thermal efficiency and coefficient of performance.			
	• Carnot theorems (statement, proof and significance).			
	Clausius & Kelvin statement of Second law and their equivalence.			
	• Absolute scale of temperature and its comparison with perfect gas scale of			
	temperature.			
Unit IV	Entropy			
	• Entropy (need, statement and characteristics).			
	• Clausius theorem and Clausius inequality of entropy.			

	• Change of entropy (in mixing, for change of state and for perfect gas).		
	• Available and non-available energy, ordered and disordered state.		
	• The principle of increase of entropy and degradation of energy.		
	• Consistency of Clausius & Kelvin statement with definition of		
	second law in terms of entropy.		
Block 2	Thermodynamic relations		
Unit V	Maxwell's relations		
	• Reciprocal theorem and reciprocity theorem in thermodynamics		
	• Maxwell's relations (statement significance and derivation from laws of		
	thermodynamics)		
	<ul> <li>Thermodynamic potentials (statement significance and applications)</li> </ul>		
	<ul> <li>TdS equations (statement, proof and applications).</li> </ul>		
	<ul> <li>Host capacity equations in different forms</li> </ul>		
	• Heat capacity equations in different forms.		
11	• Atomic neat ratio.		
Unit VI	Phase Transition		
	• First order phase transitions and its characteristics. Clausius Clapeyron		
	equations.		
	• Second order phase transition and its characteristics. Ehrenfest equations.		
	• First and second latent heat equations from Maxwell's relation, effect of		
	pressure on meltingpoint of solids and boiling point of liquids.		
	• Joule's expansion, Joule's coefficient, Energy equation.		
	• Free expansion and conservation of internal energy in it.		
	Change of entropy in isothermal irreversible process.		
Unit VII	Third law of thermodynamics		
	• Joule's Thomson expansion, conservation of enthalpy.		
	• Enthalpy equation, Joule-Kelvin coefficient, inversion curve.		
	• Different methods of coolings, liquefaction of gas, adiabatic demagnetization.		
	• Comparison of Joule-Thomson expansion with Joule expansion and adiabatic		
	expansion.		
	• Throttling process, change of entropy in an irreversible adiabatic process.		
	• Third law of thermodynamics and its consequences.		
	• Statistical Mechanics: Maxwell-Boltzmann law - distribution of velocity -		
	Ouantum statistics - Phase space - Fermi-Dirac distribution law - electron gas -		
	Bose-Einstein distribution law - photon gas - comparison of three statistics.		
Block III	Heat		
Unit VIII	Kinetic Theory of Gases		
	• Perfect gas equation and conditions for its validity to real gas.		
	• Comparison of ideal gas and real gas. Vander Waal equation of state,		
	Vander Waal constants.		
	• Andrews experiment on CO <sub>2</sub> , critical state, gas and vapors.		
	• Critical constants (statement and inter-relationship).		
	• Mean free path (qualitative) and its applications.		
	• Transport phenomenon in gases (viscosity, thermal conductivity and self		

	diffusion, derivation of expressions for each and their inter-relationship).			
	Brownian's motions and its features.			
Unit IX	Conduction and convection			
	• Modes of transfer of heat.			
	• Steady and variable states. Thermal conductivity, Thermal diffusivity.			
	• Temperature gradient, heat flow through combination of slabs, thermal			
	resistance.			
	• Formation of ice layer and its consequences.			
	• Fourier equation of heat. Discussion of result	ts for exposed and covered		
	rods, Ingen-Hauseexperiment.	-		
	• Periodic flow of heat (qualitative) and its ap	plications.		
	• Natural and forced convection (qualitative).	-		
Unit X	Radiation			
	• Radiant energy, black body radiation, white	adiation. Reflectivity, absorptivity		
	andtransmitivity.			
	• Kirchoff's law for radiation, (statemer	t, derivation, significance and		
	applications).	-		
	• Stefen-Boltzman law (statement and deriva	tion), Stefan's constant, Newton's		
	law of cooling.			
	• Average energy of quantum oscillator and	d classical oscillator. Number of		
	modes per unitvolume in frequency range.			
	• Planck's law for radiations (need, statement and derivation), ultraviolet			
	catastrophe.			
	• Derivation of classical laws (Stefan's, W	ien's displacement. Wien's fifth		
	power, Rayleigh-Jean) from Planck's law.	, ··		
	• Spectrum of black body radiations at different temperatures.			
Suggested T	Fext Book Readings:	<b>I</b>		
1. Heat a	and Thermodynamics, M.W. Zemansky, Richard I	Dittman, 1981, McGraw-Hill.		
2. A Tre	eatise on Heat, Meghnad Saha, and B.N.Srivastava	, 1958, Indian Press		
3. Therm	3. Thermal Physics, S. Garg, R. Bansal and Ghosh, 2nd Edition, 1993, Tata McGraw-Hill			
4. Modern Thermodynamics with Statistical Mechanics, Carl S. Helrich, 2009, Springer.				
5. Thermodynamics, Kinetic Theory & Statistical Thermodynamics, Sears & Salinger.				
1988,	, Narosa.			
6. Conce	epts in Thermal Physics, S.J. Blundell and K.M. B	lundell, 2nd Ed., 2012, Oxford		
Unive	ersity Press			
This course c	can be opted as an elective by the students of fo	llowing subjects:NA		
<u> </u>				
Suggested eq	Suggested equivalent online courses (MOOUs) for credit transfer: Thermal Physics: Nptel			
<u>mups://youtu.</u>	I.De/ISJKPeyOA I U			
Electronic m	media and other digital components in the cu	rriculum:		
Choose any	one or more than: e-SLM/Other electronic and	d digital contents		
Name of elec	ectronic media: e-SLM Year of	incorporation: 2022		

Programme: E	S.Sc. Year:3 <sup>rd</sup> Semester:6 <sup>th</sup>			
Subject: PHYSICS				
Course Code: DCEPHS-108N   Course Title: QUANTUM MECHANICS AND SPECTROSCOPY				
Course Objectives:				
<ul> <li>To provide basics knowledge of Quantum mechanics.</li> </ul>				
To ma	ke the students familiar with the Schrodinger's equation.			
<b>Course Outc</b>	omes:			
CO-1 Student	will learn basics concepts of quantum mechanics.			
CO-2 student	s will be able to understand the various operators used to represents dynamic			
variables				
CO-3 The eig	en values and eigen functions of linear harmonic oscillator			
CO-4 Hydrog	en atom will help students to understand the behaviors of microscopic systems.			
Credits: 2	Type of Course: Elective			
Max. Marks:	00 Min. Passing Marks: 36			
(Syllabl Block 1	Should be framed block wise/unit wise; No of blocks and units may change)			
BIOCK I	Wave mechanics			
	Need of quantum theory and quantum mechanics			
	• Fundamental equation of wave mechanics and its representation in various			
Unit I	forms			
	• Wave function and its interpretation by Max-Born and Schrodinger			
	Separation of variables, stationary states			
	Probability density and probability current density			
	Equation of continuity			
Unit II	Condition of wave function:			
	• Normalizable and unnormalizable wave function			
	<ul> <li>Condition for normalized orthogonal ortho normal and complete wave</li> </ul>			
	function			
	<ul> <li>Expectation values of thermo dynamical variables</li> </ul>			
	<ul> <li>Expectation values of merino dynamical valuables</li> <li>Dirac-Delta function and Kronecker delta function</li> </ul>			
	<ul> <li>Dirac-Dena function and Kronecker dena function</li> <li>Non degenerate states and degenerate state</li> </ul>			
	<ul> <li>Floorfact theorem (statement, derivation and significance)</li> </ul>			
Unit III	Onerator algebra			
	• Concept of operator null operator inverse operator operator algebra			
	<ul> <li>Operator in quantum mechanics (position momentum energy velocity)</li> </ul>			
	kinetic energy and angular momentum)			
	<ul> <li>Physical operators (Linear, Hermitian, parity and their properties)</li> </ul>			
	• Commutator and non-commutator operators, simultaneous wave			
	function and uncertaintyrelation			
	• Eigen function, Eigen values and Eigen values equations			
	• Commutator rules among components of orbital angular momentum			
	momentum, position. $L^2$ .			
	• Ladder operates (L+ and L <sup>-</sup> ), commutator rules amongst L+,L-, Lz and L <sup>2</sup> .			

Block 2	Applications of Schrodinger's equation		
Unit IV			
	One- and three-dimensional problems		
	• Free particles, particle in box.		
	<ul> <li>Potential steps, potential barrier(tunneling).</li> </ul>		
	<ul> <li>Potential well of infinite denth and finite denth</li> </ul>		
	<ul> <li>Harmonic oscillator (classical and quantum) series solution</li> </ul>		
	<ul> <li>Figen values and Figen function of harmonic oscillator Hermite polynomial</li> </ul>		
	<ul> <li>Zero-point energy and parity of oscillator</li> </ul>		
Unit V	Snherically symmetric systems:		
Oline V	• Spherically symmetric potential components of angular momentum in polar		
	coordinate		
	<ul> <li>Spherical harmonics and their orthogonality Legendre Polynomial</li> </ul>		
	<ul> <li>Schrödinger equations for rotator with free axis and its series solutions</li> </ul>		
	• Schrödinger quatim för hydrogen atom, solution of r-equation, $\theta$ -equation,		
	b equation, Eigenfunction and Eigen values of hydrogen atom, Bhoris		
	Tadius		
	<ul> <li>Degeneracy and quantum number of hydrogen atoms</li> <li>Comparison of Sohrödinger atomic model with Dohr's atomic models</li> </ul>		
Block III	Comparison of Schrödinger atomic model with Bonr's atomic models.  Identical particles and parturbation		
Linit VI	Identical particles		
	<ul> <li>Distinguishable and indistinguishable particles</li> </ul>		
	<ul> <li>Symmetric and anti-symmetric wave functions</li> </ul>		
	<ul> <li>Concept of spin and spin angular momentum</li> </ul>		
	<ul> <li>Pauli spin matrices (definition commutation anti-commutation) spin wave</li> </ul>		
	function		
	• Exchange operator exchange degeneracy		
	<ul> <li>Equation of motion, condition for constant motion.</li> </ul>		
	<ul> <li>Pauli exclusion principle (statement and importance)</li> </ul>		
Unit VII	Approximation methods:		
	<ul> <li>Born-approximation (statement and application)</li> </ul>		
	• Variation methods (statement and application)		
	• Perturbation method (time independent and time dependent)		
	• Time independent perturbation (non-degenerate and degenerate, first order		
	and second ordercorrection)		
	• Application of perturbation theory to helium atom, ortho and para helium.		
	• Application of perturbation theory to anharmonic oscillator, comparison		
	of harmonic and anhormonic oscillator		
Unit VIII	Atomic spectra:		
	• Lande-g factor (statement and derivation)		
	Shift in energy of atom when placed in magnetic field		
	Zeeman effect (statement and classification)		
	Classical and quantum theory of normal and anomalous Zeeman effect, Paschen		
	Back effect		

	• Zeeman pattern for sodium lines and other various transitions Stark effect and its above staristics		
	Its characteristics		
BIOCK IV	Molecular spectroscopy Unit		
Unit IX	Types of spectroscopies:		
	• Electronic, Rotational, Vibrational and Rotational-vibrational spectroscopy		
	• Selection rules, energy and frequency	of vibrational spectra	
	• Selection rules, energy and frequency	of Rotational spectra	
	• Raman effect, stokes and anti-stokes	lines (quantum and classical explanation)	
	• Infrared spectroscopy		
	• Fluorescence and phosphorescence s	pectroscopy	
Unit X	Techniques of spectroscopy:		
	• Electronic spectra,		
	Electronic transitions		
	Frank-Condon principle		
	Singlet and triplet states		
	• Fine structure and hyper fine structur	e	
	• NMR (principle and importance)		
	ESR (principle and importance)		
Suggested Te	ext Book Readings:		
1. A Tex	ext book of Quantum Mechanics, P.M.Mathews and K.Venkatesan, 2nd Ed.,		
2010,	), McGraw Hill		
2. Quant	tum Mechanics, Robert Eisberg and Robert Resnick, 2nd Edn., 2002, Wiley.		
3. Quant	tum Mechanics, Leonard I. Schiff, 3rd Edn. 2010, Tata McGraw Hill.		
4. Quant	ntum Mechanics, G. Aruldhas, 2nd Edn. 2002, PHI Learning of India.		
5. Quant	antum Mechanics, Bruce Cameron Reed, 2008, Jones and Bartlett Learning.		
6. Quant	6. Quantum Mechanics: Foundations & Applications, Arno Bohm, 3rd Edn., 1993, Springer		
7. Quant	/. Quantum Mechanics for Scientists & Engineers, D.A.B. Miller, 2008, Cambridge		
Suggested eq	Suggested equivalent online courses (MOOCs) for credit transfer: NA		
Buggesieu eq	urvarent onnine courses (14100005) for th		
Electronic m	Electronic media and other digital components in the curriculum:		
Choose any	one or more than:e-SLM/Other electron	nic and digital contents	
Name of elec	tronic media: e-SLM	Year of incorporation: 2022	

Drogramma: I	<b>S S a</b>	Voor 2rd		Samastar: 6 <sup>th</sup>
Flogramme. B.Sc.		Subject D	HVSICS	Semester.0
Course Code: DCEPHS-109N		Course Title: SOL	ID STATE PHYSIC	CS AND ADVANCED
Course Objectives: The sim of this course is to make the students learn and discuss about the				
basia's concer	<b>Course Objectives:</b> The aim of this course is to make the students learn and discuss about the			and discuss about the
basic's concepts of sonds and Advanced electronics.				
<b>Course Outco</b>	omes:			
CO-1: Unders	tand the knowled	ge of Crystal and it	s structure.	
CO-2: Students will be able to understand Advanced analog electronics.			nics.	
CO-3: Studer	nts will be able to	understand Adva	nce digital electroni	cs.
Credits: 2			Type of Course: Ele	ctive
Max. Marks:	100	Min. Passing Mar	·ks: 36	
(Syllabi	should be frame	d block wise/unit	wise; No of blocks a	ind units may change)
Block 1	Basic concepts	s of solids		
	Crystal and its	structure		
	Crystalline	and amorphous st	tate of solids, liquid	crystal and its characteristics
	(qualitative	a).		
	• Simple cry	stal structure (SC,	FCC, BCC).	
I Init I	• Unit cell ar	d Bravais lattice.		
Omt I	• Classification of lattices and types of crystals on the basis of Bravais lattice.			
	• Direct and reciprocal lattice, Miller indices and planes.			
	• X-ray diffraction, Bragg's law.			
	• Generalized Hooke's law for Anisotropic body, elastic constants of cubic			
	crystals			
Unit II	Band theory of solids			
	• Need of free electron quantum theory			
	Sommerfel	d Fermi model bar	nd theory.	
One dimen		sional motion of e	lectron in periodic p	ootential (Bloch theorem).
	Kronning-H	enny model (featu	res and its importat	nce).
Fermi surf		ice. effective mass	of charge carriers (	electron and holes).
	Concentration		tors.	,
	Hall effect	(qualitative).		
Unit III	Unit III Lattice vibrations			
	Interatomic	force and classifi	cation of solids.	
	Lattice ene	rgy of ionic crysta	ls.	
	Vibration o	f monoatomic and	diatomic linear chai	in, acoustic and optical modes.
	phonon			,
	Thermal ca	pacity of solids of	assical theory of spe	cific heats (Dulong and Petit's
	law)		assical moory of spe	ente neuts (Durong und Fett 5
	• Experiment	tal results and need	d of quantum theory	of specific heat of solids
	Finctoin's	theory of sno	cific heats (noor	statement
	assumption	derivations and	limitations)	i, statement,
	Debye theo	by of specific hea	ts (need statement	assumptions derivations and
	limitations	i specific field	is (need, statement,	assumptions, convations and

	Concept of Einstein's temperature and Debye temperature.		
Unit IV	Magnetism and superconductivity		
	• Comparison of features of diamagnetic and paramagnetic materials with		
	examples. Curie lawand Curie Wiess law.		
	• Classical and quantum theory of diamagnetism and paramagnetism.		
	• Qualitative discussion of ferromagnetism, anti-ferromagnetism and		
	ferrimagnetism.		
	• Superconductivity and its characteristics, magnetic behavior of superconductor.		
	• Meisner's effect, BCS theory (qualitative).		
	• Types of superconductors (examples, properties and applications).		
	Josephson effect, quantum Hall effect.		
Block 2	Advanced analog electronics		
Unit V	Different modes of operations.		
	• Eber's moll model for PNP and NPN transistors.		
	• Expressions for various currents and voltage.		
	• Saturation parameters and its importance.		
	• Conditions for cut off mode, saturation mode, inverse mode and active mode.		
	Comparison among all modes of operations.		
Unit VI	Transmission and reception		
	Basic elements of radio communication systems.		
	• Requirements of transmitter, medium and receiver.		
	• Modulation (need, types and statements).		
	• Analysis of AM, FM and PM, modulation index.		
	• Frequency spectrum and power in modulations.		
	Circuit of modulator.		
	• Demodulation (need and statements).		
	Circuit for demodulator.		
Unit VII	Operational amplifier		
	• OP-amplifier (symbol, number code, power supply and characteristics).		
	• Input-output relationship, input-offset and output offset voltage.		
	• Differential input and output resistance.		
	• Common mode rejection ratio, output current, power consumption,		
	• Characteristics of OD emplifier comparators and detector		
	<ul> <li>Unaracteristics of OF- amplifier, comparators and detector.</li> <li>Inverting and non-inverting amplifier.</li> </ul>		
	<ul> <li>Differentiator and basic integrator</li> </ul>		
Block III	Advance digital electronics		
Unit VIII			
eint vin			
	Introduction and classification of logic families.		
	• Input and output characteristics.		
	• Fan-in and fan-out.		
	Noise margin and noise immunity.		
	• Rise and fall time.		

	• RTL (circuit, analysis and applications).		
	• DTL (circuit, analysis and applications).		
	• TTL (circuit, analysis and applications), totem-pol.		
	Comparison of RTL, DTL and TTL.		
Unit IX	Sequential circuits		
	Difference from combinational circuit.		
	• Flip-flops (RS, D, JK) master slave.		
	• Register (function and types).		
	• Counter (function and types).		
	Memory (function and types).		
	• Convertors (A/D and D/A).		
Unit X	Integrated circuits and devices		
	• Introduction of integrated circuit and its comparison with discrete circuits.		
	Classification of IC on the basis of construction and operation.		
	Monolithic IC (basics structure and fabrication).		
	• Cathode ray oscilloscope (principle, construction, block diagram, working and		
	application).		
	• Multimeter (principle, types, construction and function).		
	• Ultrasonics (production, detection, velocity measurements and applications),		
	Hypersonic and ultrasonics.		
Suggested T	ext Book Readings:		
1. Introdu	iction to Solid State Physics, Charles Kittel, 8th Edition, 2004, Wiley India		
Pvt. Ltd.			
2 Elements of Solid-State Physics I.P. Srivastava 2nd Edition 2006 Prentice-Hall of			
2. Elements of Sond-State I hysics, J.I. Sitvastava, 2nd Edition, 2000, I tenuce-fiall of India			
mara			
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